

Substitutional p-Type Doping in NbS₂–MoS₂ Lateral Heterostructures Grown by MOCVD

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Monolayer MoS₂ has attracted significant attention owing to its excellent performance as an n-type semiconductor from the transition metal dichalcogenide (TMDC) family. It is however strongly desired to develop controllable synthesis methods for 2D p-type MoS₂, which is crucial for complementary logic applications but remains difficult. In this work, high-quality NbS₂-MoS₂ lateral heterostructures are synthesized by one-step metal-organic chemical vapor deposition (MOCVD) together with monolayer MoS₂ substitutionally doped by Nb, resulting in a p-type doped behavior. The heterojunction shows a p-type transfer characteristic with a high on/off current ratio of $\approx 10^4$, exceeding previously reported values. The band structure through the NbS₂-MoS₂ heterojunction is investigated by density functional theory (DFT) and quantum transport simulations. This work provides a scalable approach to synthesize substitutionally doped TMDC materials and provides an insight into the interface between 2D metals and semiconductors in lateral heterostructures, which is imperative for the development of next-generation nanoelectronics and highly integrated devices.

1. Introduction

Semiconducting 2D transition metal dichalcogenides (TMDCs) have been extensively investigated for next-generation nanoelectronics and optoelectronics because of their atomic scale

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The ORCID identification number(s) for the author(s) of this article can be found under https://doi.org/10.1002/adma.202209371.

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DOI: 10.1002/adma.202209371

thickness,^[1] layer-dependent band structure,^[2] and favorable electrical properties.^[3] Among them, monolayer molybdenum disulfide (MoS₂) is most widely studied due to its strong photoluminescence^[4] (PL) and high stability.^[3] Monolayer MoS₂ with a large intrinsic bandgap of $1.8 \text{ eV}^{[5]}$ and a high on/off current ratio of 108 in n-type FET devices^[1] behaves as an n-type semiconductor, which is attributed to the electron doping by sulfur vacancies,^[6,7] Moreover, MoS₂-based integrated nanoelectronics are emerging rapidly in recent years, such as in-memory devices^[8] and artificial neural networks (ANNs),^[9] which highlights their potential to extend Moore's law in advanced technologies.^[10,11] Complementary logic applications based on MoS₂ are desirable for the realization of a platform for low energy-consuming and high effective computing, which requires

the reliable synthesis of p-type MoS₂.^[12] Several methods have been reported to achieve p-type doped MoS₂, such as plasma treatment,^[13–15] charge transfer,^[16,17] and contact engineering.^[12,17] However, these non-intrinsic p-type doped MoS₂ were limited to multilayer thicknesses and their further applications are hindered by the complicated process

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and low controllability,^[18] showing that the synthesis of highquality and controllable p-type doped monolayer MoS_2 remains a challenge. Recently, some attempts have been reported to introduce substitutional metal doping with Nb atoms inside MoS_2 lattice structure by chemical vapor deposition (CVD),^[18– 20] realizing charge transfer by electron occupation of metal d-orbitals.^[21,22] Nevertheless, these p-type MoS_2 FET devices show a low on/off ratio and carrier mobility because of their large contact resistance related to strong Fermi level pinning on metal contacts.

2D metal-semiconductor (M-S) junctions based on TMDC heterostructures are expected to result in a reduced contact resistance due to weakened Fermi level pinning and modulation of the Schottky barrier.^[23] Furthermore, the interface of these 2D M-S heterostructures is one of the crucial factors determining their performance, requiring comprehensive characterization, such as identifying geometrical distortions, interfacial coupling, and electronic states.^[24] Although TMDC heterostructures can be obtained using the transfer technique thanks to van der Waals interactions,^[25] CVD growth is scalable toward large area growth, layer number control, high reproducibility, as well as low residue^[26] concentration. Several examples of lateral and vertical TMDC heterostructures grown by CVD have been reported, including semiconductor-semiconductor heterostructures (e.g., MoS₂/WS₂,^[27] MoS₂/MoSe₂^[28]) and metal-semiconductor heterostructures (e.g., NbS₂/WS₂^[29]).

Here, we demonstrate the one-step simultaneous MOCVD synthesis of NbS₂-MoS₂ lateral heterostructures with p-type Nb substitutional doping of monolayer MoS₂. Atomic force microscopy (AFM), Raman spectroscopy, photoluminescence (PL), aberration-corrected scanning transmission electron microscopy (STEM) combined with energy-dispersive X-ray spectroscopy (EDX) were used to confirm the high-quality lateral heterostructure growth and substitutional Nb doping. The FET device based on our heterojunction exhibits a p-type transfer characteristic with the highest on/off current ratio of 10⁴ ever reported in p-type MoS₂ realized via substitutional metal doping. Density-functional theory (DFT) simulations were performed to calculate the band structures of Nb-doped MoS₂ and reveal the Fermi level alignment at the interface with the critical role of doping. Temperature-dependent I-V characterization and selfconsistent quantum transport simulations were carried out to extract the Schottky barrier height (SBH) at the interface of the p-type heterojunction.

2. Results and Discussion

2.1. 2D NbS2-MoS2 Lateral Heterostructures Grown by MOCVD

Our NbS₂–MoS₂ lateral heterostructures were synthesized by sequentially growing each component on a *c*-plane sapphire wafer in a home-built MOCVD system, as schematically shown on **Figure 1a**. Since NbS₂ requires a higher growth temperature than MoS₂ according to our previous results,^[30,31] the quartz tube was first heated to 850 °C with temperature maintained for 10 min for monolayer MoS₂ growth. During this time, molybdenum hexacarbonyl (Mo(CO)₆) and hydrogen sulfide (H₂S) precursors were simultaneously introduced into the

growth tube. Subsequently, the precursor supply was turned off and the tube was continuously heated until 950 °C. Monolayer MoS₂ on sapphire substrate was in turn employed for the preparation of heterostructures. H2S flow was resumed, reacting with the pre-coated NbCl₅ and resulting in NbS₂ growth. Initial nucleation of NbS2 preferably takes place at the edges of as-grown MoS2 due to its high chemical reactivity. NbS2 growth continues outward onto the surface of sapphire owing to its lower Gibbs free energy compared to the surface of MoS_{2} ,^[32] resulting in lateral NbS_2 -MoS₂ heterostructures. Typical as-grown NbS2-MoS2 lateral heterostructures are shown in Figure 1b, characterized by MoS₂ triangles with a lateral size of $\approx 30 \ \mu m$ and surrounded by NbS₂ with a width of $\approx 5 \,\mu\text{m}$. Notably, the successive growth of both components without air exposure prevents the oxidation of as-grown MoS₂ edges, which is essential for the formation of high-quality lateral heterostructures. Additional details on the growth method are available in the Experimental Section.

Figure 1c shows the AFM image of an as-grown NbS₂-MoS₂ lateral heterostructure on the sapphire substrate, indicating a smooth surface with a thickness of 3 nm for NbS₂ and a height of 0.7 nm for monolayer MoS2. By controlling the different growth parameters, we were able to tune the thickness of NbS2 within the heterostructures between 3 and 10 nm. Figure 1d presents the Raman spectra obtained from different regions of the heterostructure acquired at room temperature. The Raman spectrum from the central region (orange curve) contains two typical peaks at 383 cm⁻¹ (E_{2g}^1) and 402 cm⁻¹ (A_{1g}), confirming the monolayer nature of MoS_2 . The peripheral region shows peaks in the Raman spectrum (red curve) located at 340 cm⁻¹ (E_{2g}^1) and 379 cm⁻¹ (A_{1g}), in agreement with the vibration modes of NbS2. Raman intensity maps corresponding to the characteristic modes of NbS₂ and MoS₂ are shown in Figure 1e. Uniform signals indicate uniform chemical distribution and the clear boundaries can be seen between the NbS2 and MoS2 regions, confirming the formation of in-plane heterostructures without alloying.

To investigate the atomic structure and interface of the heterostructure, aberration-corrected annular dark-field (ADF)-STEM imaging was performed with images shown in Figure 1f,g. Due to varying thickness, few-layer NbS2 shows stronger contrast in comparison to monolayer MoS2. The magnified and filtered image shows a clear and sharp interface between the two materials. Moreover, the fact that NbS₂ is vertically stacked in a hexagonal lattice arrangement (2H-phase) with no visible defects in the individual regions suggests a high crystalline quality of our heterostructure samples. A relative orientation angle of $\approx 30^{\circ}$ is estimated from their atomic lattice structure of the heterostructure, which can be further confirmed from the corresponding fast Fourier transformation (FFT) image shown in the inset. Energy-dispersive X-ray spectrometry (EDX) spectrum and mappings shown in Figure S1 (Supporting Information) present the elemental spatial distribution, also confirming the chemical compositions of NbS2 and MoS2. Geometric phase analysis (GPA) of the corresponding ADF-STEM image shows compressive in-plane strain along the interface with MoS₂ (see Figure 1h). Distribution of strain is clearly observed along the interface, which is attributed to the lattice mismatch and layer variations of the two compounds.

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а b Gas flow sapphire substrate Quartz tube Atmospheric pressure T = 850 °C ~ 950 °C 850 °C, 10 min 950 °C, 30 min H_sS + Mo(CO) H_S (Nb-doped) MoS (Nb-doped) MoS NhS Sapphire substrate Pre-coated NaCl and NbCl, Pre-coated NaCl and NbCl, d С е NbS₂ nm MoS₂ 379cm E1. 340cm ntensity (a.u.) Sapphire 402cm Sapphire 300 350 400 450 6um Raman shift (cm⁻¹) f h g Monolayer MoS, NbS. 5 nm 2 nm

Figure 1. NbS₂–MoS₂ lateral heterostructures grown by MOCVD. a) Schematic illustration of the MOCVD process for NbS₂–MoS₂ lateral heterostructure growth. b) Optical microscopy image of as-grown NbS₂–MoS₂ lateral heterostructures grown on sapphire. Inset: Zoomed-in image presenting a MoS₂ triangle surrounded by NbS₂. Scale bar: 10 μ m. c) AFM image and the corresponding height profile of as-grown NbS₂–MoS₂ lateral heterostructure. The thicknesses of NbS₂ and MoS₂ are \approx 3 nm and 0.7 nm, respectively. d) Raman spectra at room temperature for NbS₂–MoS₂ lateral heterostructures on sapphire. The red curve was measured on peripheral NbS₂, while the orange curve was taken on central MoS₂. e) Raman mapping images based on the intensity of the typical modes on NbS₂ region and MoS₂ region from the heterostructure sample. Scale bar: 10 μ m. f,g) STEM images taken on the interface of NbS₂–MoS₂ lateral heterostructure. The darker side shows monolayer MoS₂, while the brighter sides correspond to few-layer NbS₂. A mis-orientation angle of 30° can be observed. Inset of (g) FFT of the ADF-STEM image shown in (g). The separated diffraction spots indicate the in-plane growth of the heterostructure. h) 2D strain mapping of NbS₂–MoS₂ heterostructure for in-plane (ε_{xx}) direction, which is analysed from the corresponding image shown in (g).

2.2. p-Type Nb-Doped Monolayer MoS₂ in the Heterostructures

Our monolayer MoS_2 in the heterostructures was substitutionally doped with Nb atoms during the MOCVD process. Nb dopants were introduced from pre-coated NbCl₅ into the MoS_2 lattice during the initial, MoS_2 growth phase, at 850 °C (Figure 1a, left panel). Notably, the substitutional doping (together with MoS_2 growth) and the formation of the heterostructure interface with NbS_2 and subsequent NbS_2 take place at different stages and temperatures of growth process, independently of each other.

A typical ADF-STEM image acquired in the individual MoS_2 region in Figure S2a (Supporting Information) shows the large and contamination-free region. Further magnified ADF-STEM image in **Figure 2**a shows the perfect atomic lattice structure of monolayer MoS_2 . Several sulfur vacancies, a

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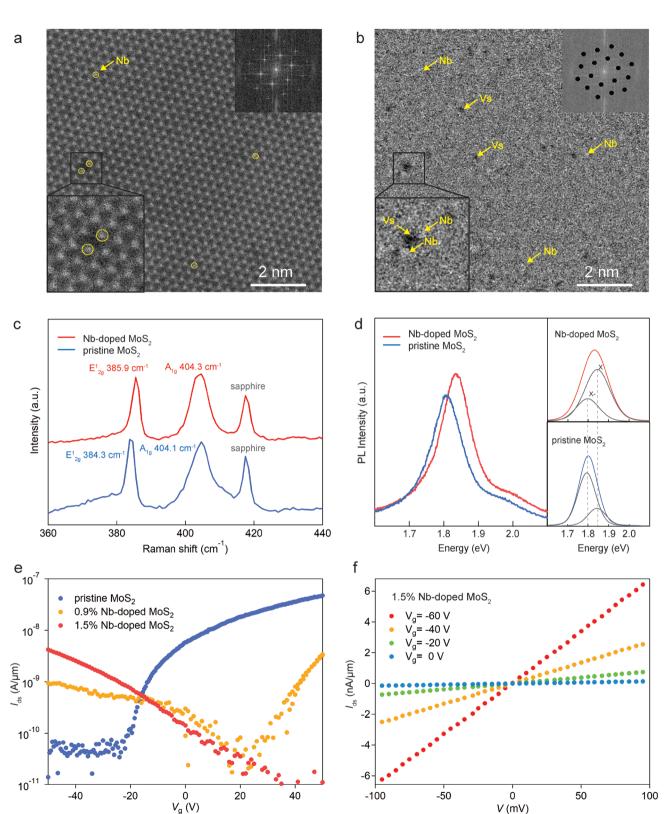


Figure 2. Nb-doped MoS_2 in the lateral heterostructures. a,b) ADF-STEM image of the Nb-doped MoS_2 film and inverse fast Fourier transform (IFFT) filtered image, respectively. Top-right insets show the corresponding FFT images. c,d) Raman and PL spectrum of Nb-doped MoS_2 compared with pristine MoS_2 on sapphire substrates measured at room temperature. e) Transport properties of Nb-doped MoS_2 with various doping concentrations. A Ti/Au stack is utilized as an electrode contact. Pristine MoS_2 shows an n-type behavior, 0.9% Nb-doped MoS_2 shows both p-type and n-type behavior, and 1.5% Nb-doped MoS_2 shows a p-type behavior. Source–drain voltage was fixed at 200 mV during the measurement. f) *I–V* curves of 1.5% Nb-doped MoS_2 under various back-gate voltages at room temperature.

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common defect in TMDCs, can be observed. The fast Fourier transform (inset) pattern from the corresponding image further confirms the high-quality growth. In principle, different atoms can be identified directly from their atomic contrast because of the relation between ADF intensity and the Znumber ($\approx Z^2$). However, since Nb and Mo differ from each other by only one Z number, it is not straightforward to distinguish them in ADF-STEM images. Therefore, the inverse fast Fourier transition (IFFT) method was used to identify Nb dopants in MoS₂,^[33] as shown in Figure 2b. Nb atoms are highlighted with yellow circles in Figure 2a and can be seen as bright dots at the corresponding location in Figure 2b. Sulfur vacancies (V_s) are also visible as black dots in Figure 2b. Quantitative ADF-STEM simulations were also performed to distinguish Nb as substituent atoms (see in Figure S2c, Supporting Information), using the DFT calculated atomic models with similar imaging parameters as in the experiments. The line profiles from both experimental and simulated images show a lower intensity along Nb atoms compared with Mo atoms. In addition, to extract the doping concentration precisely, EDX spectroscopy was further employed in the MoS₂ region as shown in Figure S3 (Supporting Information). Elemental mapping and spectrum from the corresponding region show the spatial distribution of S, Mo, and Nb atoms. Although the K and L edge energies for Nb and Mo are very close, peak deconvolution can be used to separate them, highlighting the uniform distribution of the three elements within the measured region showing an atomic concentration of S > Mo > Nb. Average Nb dopant concentration extracted from the EDX is ≈1.5%.

Figure 2c compares Raman spectra acquired from the Nbdoped MoS₂ and pristine MoS₂ on sapphire substrate. Both of the characteristic peaks (E_{2g}^1 and A_{1g}) show a separation of ~20 cm⁻¹, indicating the monolayer nature of the pristine MoS₂ and Nb-doped MoS₂. The Raman signals of Nb-doped MoS₂ show a blueshift compared with the pristine MoS₂, consistent with previous literature.^[18] The same location of sapphire signals confirms that the shift is not due to noise. The shift of Raman spectra depends on the coexistence of strain and charge doping.^[34,35] In our case, the blueshift of Raman signals can be explained by growth-induced tensile strain and reduced electron-phonon scattering, which results from the distorted lattice and hole injection due to the substitutional Nb-doping.^[18] respectively.

Figure 2d presents PL spectra of Nb-doped monolayer MoS₂ and pristine monolayer MoS₂, as well as their corresponding fitting curves. A strong PL peak located at ~1.80 eV was observed in pristine monolayer MoS₂, in-line with its directbandgap nature.^[4] Compared with pristine MoS₂, the Nb-doped monolayer MoS₂ shows a blueshifted PL peak with enhanced intensity, which is consistent with previous results on p-type doping of MoS₂,^[18,36] To explain the spectral changes, the PL peaks can be considered as a combination of the trion (X⁻) peak at ~1.79 eV and the exciton (X) peak at ~1.85 eV, as shown by the gray curves. In pristine MoS₂, the PL spectra are dominated by the trion (X⁻) peak because of the strong electron doping.^[37] With Nb doping, the exciton (X) peak becomes dominant due to the reduced electron density and a suppressed electron concentration. www.advmat.de

To examine electrical transport properties of our Nb-doped MoS₂, transfer characteristics were studied on multiple samples with different doping concentrations with results shown in Figure 2e. Notably, a Ti/Au stack was used as a conventional 3D contact to our MoS₂ channel. The variation of Nb dopant concentration is due to different amounts of spincoated NbCl₅ precursor during growth, which shows a range of 0.9% to 4.6% according to our EDX analysis (see in Figure S4, Supporting Information). A clear transition from n-type to p-type is visible from the transfer curves, as the Nb dopant concentration increases. In 1.5% Nb-doped MoS₂ sample, a typical p-type behavior with an on/off ratio of 1×10^3 and fieldeffect mobility of \approx 1.43 cm² V⁻¹ s⁻¹ was observed. The mobility is lower than that of pristine MoS2, which can be attributed to the increased structural distortion because of substitutional Nb doping in MoS₂. Nevertheless, in a sense of material development, p-type MoS₂ achieved by substitutional doping is already an advancement compared to pristine MoS₂. Additionally, a linear I-V curve displays ohmic contacts between Nbdoped MoS₂ and Ti electrodes within our measured range, as shown in Figure 2f.

2.3. Atomistic and Electronic Structures of $\mathsf{NbS}_2\mathsf{-}\mathsf{MoS}_2$ Heterojunction

To shed theoretical insight into the heterojunction between metallic NbS_2 and p-type Nb-doped monolayer MoS_2 , first-principle calculations were performed based on DFT. Density of states (DOS) and the band structure for the heterojunction are shown in **Figure 3**. The detailed calculation parameters are available in the Supporting Information.

To model the NbS2-MoS2 lateral heterostructure at the quantum-mechanical (QM) level, we exploit the virtual crystal approximation (VCA),^[38] i.e., we define "mixed" chemical elements composed of a given percentage of Nb and Mo elements. This allows us to rapidly analyze the structural and electronic properties of phases with Mo/Nb mixed composition. From this analysis, we create our QM structural model: a reasonably large unit cell (depicted in Figure 3a) made of: i) pure NbS₂, ii) a Nb-doped MoS₂ buffer, and iii) a slightly Nb-doped core/pristine MoS₂ (the Nb-doping content is ≈1.5% at the experimental level, set to 1.66% for convenience in the simulations). The buffer phase (ii) is introduced to mediate the 5% lattice mismatch between MoS₂ (3.19 Å at the DFT level) and NbS₂ (3.35 Å). Its doping level is set to 8.3%, as this doping is energetically the most stable, i.e., it has the minimum gradient of total energy and work function with respect to the lattice parameter (see the Supporting Information for more details).

The VCA analysis of the pure phases detailed in the Supplementary Information confirms the dependence of band structure upon lattice parameters.^[40,41] Moreover, a fragment analysis^[39] of the system, illustrated in Figure 3e, leads us to two further conclusions. First, the top of the valence band (TVB) of semiconducting MoS₂ is pinned to the Fermi level of the system. Second, there are two localized barriers in the electrostatic potential at the NbS₂/[8.3%Nb-doped-MoS₂]/[1.7%Nb-doped-MoS₂] junctions of $\Delta_{\rm h} \approx 72$ meV and

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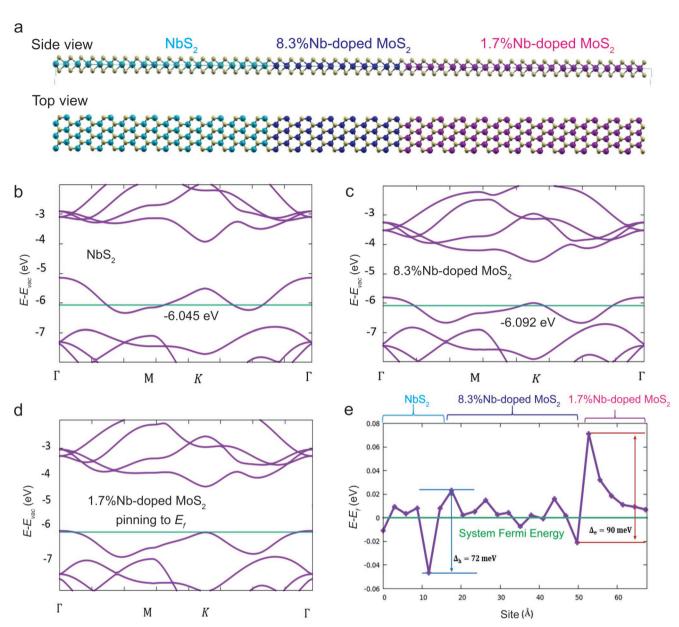


Figure 3. QM modeling of the NbS₂/MoS₂ lateral heterostructure in terms of a composite three-phase NbS₂/8.3% Nb-doped MoS₂/1.7% Nb-doped MoS₂ system. a) Atomistic depictions in side and top views. b–d) Band structure of the pure fragments, aligned as derived from the analysis of the electrostatic potential in the scattering system. e) Profile of local Fermi energy or local top of the valence band profile along the transmission direction derived from a fragment analysis of the electrostatic potential on metal atoms,^[39] with a highlighted potential jump at the buffer-phase/MoS₂ junction of Δ = 90 meV.

 $\Delta_{\rm e} \approx 90$ meV, which suggests a suppression of transmission at the heterojunctions, despite the uncertainty in the values of these barriers due to the deformation of the band structure following the formation of charge dipoles at the interface (Figure S11, Supporting Information). In synthesis, QM modeling predicts that the lattice and electronic mismatch at the NbS₂–MoS₂ lateral heterostructure is overcome by Nb doping of MoS₂ (as produced by growing NbS₂ in excess of Nb onto pre-prepared MoS₂ flakes and evidenced by EDX and ADF-STEM measurements) mediating structurally and electronically the NbS₂→MoS₂ transformation, and giving rise to a p-type NbS_2 -MoS₂ heterojunction, also possessing good electrical transport properties.

2.4. p-Type NbS₂-MoS₂ Heterojunctions

To characterize the transport properties of our MOCVD-grown NbS_2/p -type MoS_2 lateral heterostructures, FET devices based on the heterojunctions were fabricated on Si/SiO_2 substrate (shown in Figure S5, Supporting Information), where NbS_2 is used as a 2D contact compared to the devices with conventional

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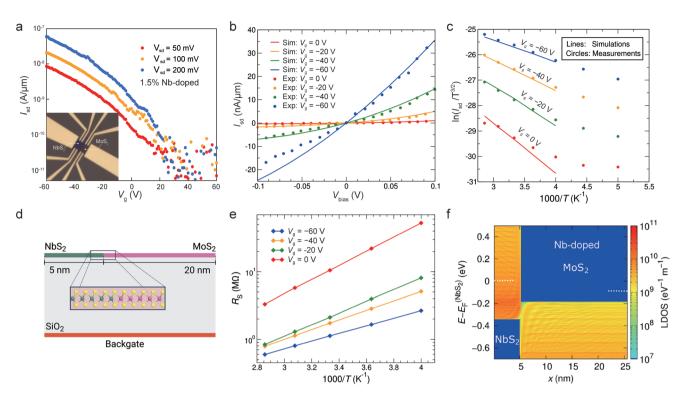


Figure 4. a) Room-temperature transfer characteristic of the FET device based on the NbS₂–MoS₂ lateral heterojunction with 1.5% of Nb doping concentration, shown in the inset. NbS₂ is utilized as a 2D contact. Scale bar: 10 μ m. Back-gate voltage V_g is applied to the substrate with a source-to-drain voltage ranging from 50 to 200 mV. P-type behavior is demonstrated from the transfer curves. b) Comparison of measured and simulated I_{ds} –V characteristics of an NbS₂–MoS₂ lateral heterojunction at room temperature for different values of back-gate voltage. c) Comparison of the Arrhenius plots obtained from measurements (dots) and simulations (lines). The probe voltage was fixed to 100 mV during the measurement. d) Sketch of the simulated device (not to scale). e) Sheet resistance R_s extracted from simulations as a function of temperature. f) Colormap of the density of states along the device close to flat-band conditions. The dotted lines denote the Fermi level at the NbS₂ and MoS₂ contacts.

3D contacts shown in Figure 2e. A linear I-V curve confirms metallic behavior without gate modulation of our NbS2 at room temperature with a total resistance (R_{tot}) of 770 Ω (see in Figure S6, Supporting Information). Figure 4a demonstrates the room-temperature transfer characteristic curves collected from the FET devices shown in the bottom-left image with a 1.5% Nb doping concentration. A typical p-type behavior with an on/off ratio of 1×10^4 was observed and the four-probe fieldeffect mobility is calculated to be 1.46 cm² V⁻¹ s⁻¹ (defined as $\mu = \frac{dI_{ds}}{dV_g} \times \frac{L}{WC_iV}$, where *L* and *W* are the channel length and width between two voltage probes, respectively, while C_i is the capacitance between Nb-doped MoS₂ channel and the back gate per unit area). The four-probe *I–V* curves in Figure 4b present the output characteristics of the NbS2-MoS2 lateral heterojunction at room temperature. A non-symmetric current rectification behavior is visible from the I-V curves, indicating the presence of a Schottky contact at the heterointerface. To study the Schottky barrier height (SBH) at the NbS₂/p-type MoS₂ junction, temperature-dependent transport measurements were carried out under various gate voltages, as shown in Figure 4c. In order to obtain an accurate estimate of the barrier height, we fit the measurement data with the results of quantum transport simulations of the heterojunction. We have used the following transport model to perform our simulation: NbS2 and MoS2 monolayers were described by a nearest-neighbor, two-orbital tight-binding Hamiltonian. The Hamiltonian parameters were calibrated to obtain the best agreement with the DFT band structures of the uncoupled materials in the energy range of interest for transport phenomena. This model leaves as free parameters the hopping amplitude between the monolayers at the heterointerface and the height of the Schottky barrier in flat-band conditions. The value of both of these quantities was iteratively adjusted in order to obtain the best fit of the I_{ds} -V characteristics. The value $t_{\rm IF}$ of the hopping amplitude considered in simulations also accounts for the cumulative effect of the defects and disorder localized at interface, which is otherwise assumed to be ideal in our model. We estimated $|t_{\rm IF}| \approx 2 \times$ 10^{-6} eV, a very small value, which suggests a strongly defective interface in addition to the localized barriers already observed in Figure 3e.

The cross section of the simulated device is sketched in Figure 4d. The simulation domain in the transport direction is restricted to a neighborhood of the heterointerface. For a given I_{ds} current, the voltage drop over this region is smaller than the experimentally measured *V*, due to the presence of a residual parasitic sheet resistance R_s associated to the $\approx 5 \,\mu$ m-long MoS₂ region between the heterojunction and the voltage contact. In order to map the source-to-drain bias over the simulation region into *V*, and thus, obtain results directly



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comparable with the experimental data, we estimated $R_{\rm s}$ from the $I_{\rm ds}$ –V measurements at room temperature by means of the Cheung method.^[42] The dependence of $R_{\rm s}$ on the temperature was modeled by assuming $R_{\rm s} \propto \exp\left(\frac{\Delta E_{\rm v}}{k_{\rm B}T}\right)$, where the distance

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 $\Delta E_{\rm v}$ between the top of the valence band and the Fermi level in the MoS₂ was estimated from the simulation of the metal/SiO₂/MoS₂ stack in equilibrium conditions. It is shown in Figure 4e for different values of $V_{\rm g}$.

To take into account the substitutional doping due to Nb atoms, MoS₂ was assumed p-doped at a concentration of 6 \times 10¹² cm⁻². This value is in reasonable agreement with the doping levels extracted from the EDX data. Furthermore, a uniform density of states $D_{\rm it} = 3.4 \times 10^{13} {\rm ~cm^{-2}}$ was included in the MoS₂ bandgap to model the presence of traps at the interface with SiO₂. The simulation results for the I_{ds} -V characteristics and for the Arrhenius plot are compared with the corresponding experimental data in Figure 4b,c, respectively. The plots show that our numerical model closely reproduces the dependence of the current on both the back-gate voltage and the bias voltage, for temperatures sufficiently close to 300 K and higher. The deviations of the experimental Arrhenius curves from the exponential behavior for temperatures ≤250 K are likely due to disorder-induced fluctuations of the Schottky barrier height along the NbS2-MoS2 interface.[43] This effect is not taken into account in our simulations. Figure 4f shows the local density of states along the device close to flat band conditions. According to the map, the corresponding Schottky barrier height turns out to be $\Phi_{\rm B}^{\rm FB} \approx 230$ meV.

3. Conclusion

One-step MOCVD method was utilized to synthesize NbS₂-MoS₂ lateral heterostructures with Nb dopants present in monolayer MoS₂. A visible interface and high quality of the as-grown heterostructures were proven by detailed characterization. Due to the increased hole concentration from Nb doping, monolayer MoS₂ shows a p-type transfer characteristic with a high on/off current ratio of $\approx 10^4$. A detailed compositional model of the heterojunction interface, describing the transition between the pure lattice-mismatched phases of NbS2 and MoS2 through a buffer phase of 8.3% Nb-doped MoS₂ was derived through DFT simulations. Furthermore, a value of 230 meV for the Schottky barrier height at the NbS2-MoS2 interface was estimated by fitting the electrical measurements with the results of selfconsistent quantum transport simulations. Our work makes an instructive combination of substitutional doped TMDC materials and 2D metal-semiconductor heterostructures, and paves the way to designing innovative nanoscale devices and CMOSlike 2D circuits.

4. Experimental Section

Material Synthesis: NbS₂–MoS₂ lateral heterostructures were produced by metal–organic chemical vapor deposition (MOCVD) method in the home-built system shown in Figure 1a. Niobium chloride (NbCl₅) was chosen as the Nb precursor because of its low melting point and relatively high solubility compared to other precursors (e.g., Nb₂O₅). The Mo precursor, $Mo(CO)_6$, was stored inside of a bubbler whose temperature was maintained at 20 °C. Sodium chloride (NaCl) was used to reduce growth temperature by suppressing the reaction energy barrier.^[44] A c-plane sapphire substrate was annealed in air for 6 h and spin-coated with 0.3 mol L⁻¹ NaCl solution in deionized water as well as 0.00125 mol L⁻¹ NbCl₅ solution in isopropanol (IPA) before being loaded into the chamber. During the growth process, MoS2 was synthesized first at 850 $^\circ\text{C}$ with 5 sccm of H_2S and 10 sccm of Ar, which carries \approx 1.8 \times 10⁻⁴ sccm of Mo(CO)₆ flowing into the tube. A flow of 4 sccm of H₂ was introduced into the growth chamber to efficiently remove carbon contamination generated from Mo(CO)₆. MoS₂ growth lasted for 10 min and the precursor supply was abruptly cut-off. Subsequently, the tube was heated up to 950 °C, and then 35 sccm of H₂S was introduced into the tube for 30 min to react with pre-coated NbCl₅ and form NbS₂-MoS₂ heterostructures. Last, the furnace cooled down to room temperature naturally with a mixture gas flow of 5 sccm of H₂S and 200 sccm of Ar in order to provide a sulfur-rich environment. The growth environment was maintained at atmospheric pressure during the whole process.

Raman and Photoluminescence (PL) Measurements: Raman and PL measurements were conducted by using a Renishaw inVia Confocal Raman microscope at room temperature. A laser beam with wavelength of 532 nm and a power of 1 mW was utilized to excite the samples. To obtain suitable spectral ranges and resolution, diffraction gratings of 3000 g mm⁻¹ and 300 g mm⁻¹ were employed in Raman and PL measurements, respectively.

Sample Transfer: To transfer as-grown heterostructures from the sapphire substrate, the sample was spin-coated with poly(methyl methacrylate) (PMMA) 4% in anisole (A4) at a speed of 2000 rpm for 1 min and baked on a hot plate at 70 °C for 15 min for drying. The edges of the PMMA film were scratched by a sharp tweezer and then the sample was immersed in water for 5 min. Subsequently, the PMMA film attached with heterostructure samples was detached from sapphire by water tension and picked up by the target chip. In order to promote the adhesion, the sample was immersed in acetone overnight and annealed in high vacuum at 250 °C for 6 h to remove the PMMA residue.

Scanning Transmission Electron Microscopy Measurements: Scanning transmission electron microscopy experiments were conducted using a double aberration-corrected FEI Titan Themis, equipped with a X-FEG, Super-X EDX detector, and a Wein-type monochromator, operated at 80 kV acceleration voltage. All STEM images were acquired using the following conditions: the probe convergence angle was set to 20 mrad, camera length was chosen to be 185 mm which corresponds to the high-angular annular dark field (HAADF) detector 49.5-198 mrad collection angle, the estimated probe current of ≈16–20 pA, and images were recorded with a resolution of 512 \times 512 pixels and 8 μs dwell times. EDX measurements were performed in both Titan Themis and FEI Talos TEM. The intensities of STEM-EDS elemental maps and spectrum shown in Figures S1 and S3 (Supporting Information) were recorded using Nb L α , Mo K α , and S K α X-ray emissions, respectively. Velox software, ThermoFisher Scientific was used to process the EDX spectrum and elemental mapping data. Band pass and Inverse fast Fourier transform filtering was used to process the images in ImageJ. Multislice algorithm-based ADF-STEM simulations were carried out with the QSTEM software package using the DFT calculated Nb-MoS₂ atomic model.

Device Fabrication: The heterostructure samples were transferred on Si/SiO₂ substrate before fabrication. The sample was first etched into a suitable geometry by using e-beam lithography and XeF₂ etching. A second step of e-beam lithography was used to make electrode patterns and a layer composed of 2 nm/80 nm-thick Ti/Au was thermally evaporated as electrodes. Finally, a lift-off process was performed in acetone to remove PMMA layer.

Transport Measurements: Electrical transport measurements were carried out in a Janis closed-cycle cryogen-free cryostat. Prior to measurements, the sample was annealed in high vacuum at 150 °C for 6 h. DC electrical measurements were done by applying DC source-drain voltage and back-gate voltage with Agilent E5270B sourcemeter

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while measuring four-probe DC voltage and current with Keithley 6514 electrometer.

DFT Simulations: Density-functional theory (DFT) simulations were performed using the Quantum Espresso (QE) package,^[45] using 50 Ry as energy cutoff for the wave function and 500 Ry as density cutoff, and $1 \times 22 \times 1$ Monkhorst–Pack k-meshes to sample the Brillouin zone. A plane-wave basis set, a gradient-corrected exchange-correlation (XC) functional (Perdew–Burke–Ernzerhof (PBE)),^[46] and scalar-relativistic ultrasoft pseudopotentials (US-PPs) were utilized for all atoms, and also to produce "mixed" pseudopotentials within the virtual crystal approximation (VCA).^[38]

Transport Simulation: Transport simulations were performed within the ballistic non-equilibrium Green's function framework, by using the self-consistent Schrödinger–Poisson solver NanoTCAD ViDES.^[47] Periodic boundary conditions were applied in the direction transverse to transport, and a set of 30 transverse wave vectors was used to sample the first irreducible Brillouin zone.

The effect of the parasitic sheet resistance R_s was taken into account by replacing in simulations the experimental value of the source-to-drain bias V with $V' = V - R_s I_{ds}$. The latter represents a further equation to be self-consistently coupled to the transport and Poisson equations.

Supporting Information

Supporting Information is available from the Wiley Online Library or from the author.

Acknowledgements

The authors acknowledge the help of Z. Benes (CMI) with electron-beam lithography. Device fabrication was carried out at the EPFL Center for micro and nanotechnology. Electron microscopy imaging was performed at the EPFL Interdisciplinary Centre for Electron Microscopy. This work was financially supported by the European Union's Horizon 2020 research and innovation program under grant agreements No 829035 (QUEFORMAL) and No 785219 and 881603 (Graphene Flagship Core 2 and Core 3), European Research Council (grants no. 682332 and 899775), the Swiss National Science Foundation (grants no. 175822 and 157739) and the CCMX Materials Challenge grant "Large area growth of 2D materials for device integration". Computational support from Cineca Supercomputing Center (Italy) is gratefully acknowledged.

Open access funding provided by Ecole Polytechnique Federale de Lausanne.

Conflict of Interest

The authors declare no conflict of interest.

Author contributions

A.K. initiated and supervised the project. Z.W. performed MOCVD growth with supervision of A.R. M.T. performed STEM measurements and EDX spectroscopy. Z.W. performed Raman and PL measurements with help of Y.Z. Z.W. fabricated the devices and performed electrical measurements with assistance of G.M.M. G.I. designed and supervised the numerical simulation of transport. G.L, F.M., and D.L. developed the quantum transport model, performed the device numerical simulations, and extracted the Schottky barrier height and the MoS₂ series resistance. A.F. performed conception and design of QM modelling. Z.G. and P.K. conducted QM simulations and modelling under supervision of L.S. A.K. and Z.W. analysed the experimental data with input from H.G.J. and M.T. All the authors contributed to the writing of the manuscript.

The data that support the findings of this study are openly available in Zenodo at https://doi.org/10.5281/zenodo.7382708, reference number 7382708.

Keywords

metal–organic chemical vapor deposition (MOCVD), p-type MoS_2 , substitutional doping, TMDC heterostructures

Received: October 11, 2022 Revised: December 30, 2022 Published online:

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