

1 **Hydrodistillation of *Trachelospermum jasminoides* Lindl. flowers. An analysis of**
2 **essential oil, hydrolate and polyphenols content of the process wastes.**

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34 **ABSTRACT**

35 *Trachelospermum jasminoides* (Lindl.) is a woody evergreen plant of the Apocynaceae family
36 mainly used as ornamental/hedge plant. Its stems and leaves are used in traditional Chinese
37 medicine with the name of “*Luoshiteng*”. Although some studies have been devoted to the
38 phytochemistry of *luoshiteng*, little is known about the chemical composition of its flowers. In
39 this study, the volatile composition of the flowers of *T. jasminoides* has been studied. The Gas
40 chromatographic - Mass spectrometry analysis (GC/FID-MS) of the essential oil and hydrolate
41 allowed identifying 43 compounds with linalool as main compound for both of them.
42 Phytochemical investigation was completed by the High Performance Liquid Chromatography
43 – Mass spectrometry (HPLC/DAD-ESI-MS) analysis of hydrodistillation wastes (liquid and
44 solid) which revealed the presence of biologically active compounds such as chlorogenic acids
45 and other polyphenols. This study contributes to the definition of the phytochemistry of *T.*
46 *jasminoides* and provides the first data on the composition of the hydrodistillation wastes
47 preparatory to their valorisation in the future.

48 **KEYWORDS**

49 *Trachelospermum jasminoides*, hydrodistillation, essential oil, hydrolate, polyphenols, wastes.

50 **1. Introduction**

51 *Trachelospermum jasminoides* (Lindl.) (*T. jasminoides*) is a lianas woody plant of the family
52 Apocynaceae and mainly spread throughout in the Southern Chinese province of the Yangtze River
53 (1). It is an evergreen plant, robust and with prolonged abundant flowering from spring to early
54 summer. These features have made it very common as an ornamental/hedge plant all over the world.
55 Its stems and leaves are a traditional Chinese medicine known as “Luoshiteng” due to their biological
56 activities (2). Several studies have been conducted to elucidate the phytochemistry of this medicinal
57 plant (3-5) and to consolidate its traditional pharmaceutical activities. Some compounds of ethanolic
58 extract of *T. jasminoides* are able to suppress the lipopolysaccharide (LPS) stimulated inflammatory
59 responses in macrophages (6). Likewise, its water extract is an effective anti-inflammatory agent in
60 carr-induced inflammation exerting its anti-inflammatory effects by suppressing TNF- α and nitric
61 oxide serum levels (7). Furthermore, antitumor and antiviral activities of the water/ethanolic extracts
62 or their single compounds have been reported (8). Up to now, only one study was published on the
63 composition of *T. jasminoides* volatile fraction (9) and none on the chemical composition of the non-
64 volatile fraction of its flowers. *T. jasminoides* flowers contain metabolites with biological activity as
65 per as the wastes (liquid and solid) of their hydrodistillation. The aim of this study is to contribute to
66 the knowledge on volatile and non-volatile chemical composition of the flowers of *T. jasminoides*
67 collected in the southern part of Italy (Eastern Sicily) by means the study of their essential oil (EO)
68 and hydrolate (Hyd), together with the polyphenols content of the solid residue extracted with ethyl
69 acetate (AcSR), ethanol (EtSR) and lyophilized water waste (lyoWW) obtained from the
70 hydrodistillation process.

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72 **2. Experimental section**

73 **2.1 Chemicals**

74 All solvents used were high-purity American Chemical Society (ACS) solvents from VWR (Milan,
75 Italy); acetonitrile and water (VWR, Milan, Italy) were of HPLC grade. Pure reference standards:
76 Caffeic acid, *p*-Coumaric acid, Quercetin-3-*O*-rutinoside and Luteolin were obtained from Fluka
77 (Milan, Italy); volatile standards were purchased from Aldrich Chemical Co., Extrasynthese (Lyon,
78 France) and Fluka (Milan, Italy).

79 **2.2 Plant material**

80 Samples of the flowers of *T. jasminoides* were collected in a private garden located in Catania (Sicily),
81 GPS 37, 577450N 15,101926E. The sampling has been carried out in the early mornings during the
82 month of May 2019. Samples after rapid transportation to the lab were immediately hydrodistilled.
83 The sample was identified and botanically classified by the Department of Biological, Geological and
84 Environmental Sciences of the University of Catania. A voucher specimen was photographed, dried
85 and stored in the herbarium of the Institute of Biomolecular Chemistry of Catania, with the number
86 0001/2019.

87 **2.3 Hydrodistillation and GC-FID, GC-MS analysis of essential oil and hydrolate**

88 Fresh flowers (248 g) were subjected to hydrodistillation in a Clavenger-type apparatus with 2 L for
89 3 hours. The oil (0.3 ml) was recovered with 1 ml of hexane and stored under N₂ in a sealed vial until
90 required. The hydrolate was recovered from the apparatus and immediately the volatile part was
91 extracted shakering 5 ml of water solution with 1 ml of hexane. After separation, the organic phase
92 was immediately injected to the gas chromatographer. Gas chromatographic (GC) analyses were run
93 on a Shimadzu gas chromatograph, Model 17-A equipped with a flame ionization detector (FID). The
94 analytical conditions and oven temperature were the same as those reported previously (10).
95 Percentages of compounds were determined from their peak areas in the GC-FID profiles. Gas-
96 chromatography-mass spectrometry (GC/MS) was carried out in the fast mode on a Shimadzu GC/MS
97 mod. GCMS-QP5050A, with the same column and the same operative conditions used for analytical
98 GC-FID. The identity of components was confirmed on the basis of their GC retention index (relative
99 to C₉-C₂₀ *n*-alkanes on SPB-5 column), computer matching of spectral MS data as already reported
100 (11-12) and whenever possible, co-injections with authentic samples.

101 **2.4 Filtration of hydrodistillation water waste**

102 The hydrodistillation water wastes (2L) were cooled at room temperature and then filtered on filter
103 paper (Whatman, cat. n° 1004-930, grade 4). After filtration was frozen up to -18°C and then freeze-
104 dried. 16.53 g of lyophilized powder (lyoWW) was obtained.

105 **2.5 Extraction of hydrodistilled solid residue**

106 The solid residue obtained from hydrodistillation was air dried at room temperature up to no
107 difference of weight was observed. 27.54 g of solid residue was obtained. 1.58 g of this residue were
108 finely ground and then defatted with 30 ml of hexane (24 hours under stirring). After filtration, the
109 defatted residue was extracted with 30 ml of Ethyl acetate (24 hours, under stirring) and then filtered

110 on filter paper. 1 ml of this solution was filtered on on PTFE 0.45 mm filters (PALL Corporation),
111 and put into 2 mL amber vials for HPLC analysis (AcSR). The solid residue was further extracted
112 with 30 ml of ethanol (24 hours, under stirring) and then filtered on filter paper. 1 ml of this solution
113 was filtered on on PTFE 0.45 mm filters (PALL Corporation), and put into 2 mL amber vials for
114 HPLC analysis (EtSR).

115 ***2.6 HPLC-DAD and HPLC-ESI-MS analysis of water waste and extracts***

116 Lyophilized water wastes and the solid residue extracts were analysed for their polyphenol content
117 on an Ultimate3000 equipped with a binary high-pressure pump and a photodiode array detector
118 Thermo Scientific, Italy). The data were processed through the Chromeleon Chromatography
119 Information Management System v. 6.80. All chromatographic runs were performed using a reverse-
120 phase column (Gemini C₁₈, 250 x 4.6 mm, 5 µm particle size, Phenomenex, Italy). Samples were
121 eluted with a gradient of 5%-90% buffer B (2.5% formic acid in acetonitrile) over 50 min after which
122 the system was maintained for 7 min at 100% Buffer B (13). The solvent flow rate was 1 mL/min.
123 Quantifications were carried out as follows: at 330 nm for cinnamic acids using Caffeic acid and *p*-
124 Coumaric acid as standards ($R^2 = 0.9997$ and $R^2 = 0.9996$ respectively) and 350 nm for Quercetin
125 derivatives using Quercetin-3-*O*-rutinoside ($R^2 = 0.9962$) and Luteolin ($R^2 = 0.9998$) as standards.
126 LC-MS analysis were performed using the same HPLC apparatus described above, whilst the ESI
127 mass spectra were acquired using a Exactive Plus Orbitrap MS (Thermo Fisher Scientific, Inc., Milan,
128 Italy), and a heated electrospray ionization interface. Mass spectra were recorded while operating in
129 the negative ion mode, in the *m/z* range 120-1500 at a resolving power of 25000 (full-width-at-half-
130 maximum at *m/z* 200). This was performed under the conditions already reported in a previous study
131 (24). Data acquisition and analyses were performed using the Excalibur software.

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133 **3. Results and discussion**

134 ***3.1 Essential oil and hydrolate chemical composition***

135 The yield in EO is very low (0.12 % v/w) compared with the only data reported in literature (9), but
136 nevertheless the GC-MS analysis allows to identify 43 components covering 73.11% of the total
137 composition of the essential oil (see Table 1, and Figure 1). Also the chemical composition seems to
138 be very different. The essential oil composition of the sample collected in Italy shows a predominance
139 of oxygenated monoterpenes (almost 42%), followed by sesquiterpenes (almost 17%). Both
140 monoterpene hydrocarbons and other compounds are below 10%. In the Italian sample the main
141 compound is linalool (26.2%) followed by geranyl propanate (8.7%) and *Z*-jasmone (7.3%). *cis*-
142 Farnesol (5.1%), *trans*-nerolidol (3.3%), limonene (3.0%) and α -terpineol (2.6%) complete the main

143 compounds list. It is not easy to draw conclusions about the compositional differences of essential
144 oils when the available comparative data are so lacking. What is certain is that the chemical
145 composition of an oil is strongly conditioned by the pedoclimatic conditions in which the plant grows.
146 The chemical composition of the essential oil sampled in the south of Italy lets one incline towards
147 the presence of a different chemotype in this case, linalool-chemotype. Moreover, even in Pansanit
148 and Pripdeevech (9) work significant compositional differences emerge in two samples from different
149 regions of Thailand. Unfortunately, the lack of further phytochemical data does not allow to confirm
150 this hypothesis as well as to enlarge the comparative considerations.

151 Under a biological point of view, the high percentage of linalool present in the EO of *T.*
152 *jasminoides* flowers makes it interesting for the potential application in the medical field. Linalool is
153 a monoterpene alcohol with anti-inflammatory and anti-oxidant properties. A recent study indicates
154 the linalool may be therapeutically useful in treating neurodegenerative conditions due to its
155 beneficial impact on mitochondrial integrity as well as ability to limit oxidative stress (15). Linalool
156 effectively exerts a protective role in ovalbumin-induced airway inflammation and its protective
157 effects are closely related to the downregulation of inflammatory mediators and MAPKs/NF- κ B
158 signalling (16). Interesting is the recent study on effective protection of intranasal linalool in ischemic
159 rats (17).

160 Hydrolates consist of the water saturated with the more hydro soluble fraction of the EO and
161 contain biologically active volatile compounds. In the case of plants with a low content of essential
162 oils, as in our case, could be difficult isolate the essential oil and hydrolates could be considered an
163 interesting secondary product or by-products. Hydrolates have a much softer scent and lower
164 biological activity than the corresponding essential oils but certain of them also show antimicrobial
165 properties (18). Chemical composition of hydrolate obtained from Italian sample is dominated by
166 oxygenated monoterpenes. This class cover more than 78% of the total hydrolate composition.
167 Monoterpene hydrocarbons and sesquiterpenes are absent and the other compounds are below 6%.
168 For the same reasons of the EO, the high presence of linalool (45.7%, see Table 1 and Figure1) in the
169 hydrolate of Italian *T. jasminoides* makes this secondary product potentially of great interest in the
170 pharmacological and microbiological field.

171 **3.2 Polyphenol composition of water and solid wastes of the hidrodistillation process**

172 The hydrodistillation process produce two kind of wastes. The water in contact with the plant material
173 and the plant solid residue. Hydrodistillation, as well as steam distillation, is a long and high-energy
174 consumption process. For this reason, in recent years the interest in the possibility to exploit the waste
175 of these industrial processes has intensified. The large volume of by-products generating during
176 distillation is nowadays a concern that some industries try to overcome by producing energy or

177 compost from the biomass (19). In reality, these by-products are rich of fine chemicals and
178 pharmaceutical building blocks, which can be valued in another way if identified and quantified. They
179 can contain both biologically active thermostable substances and fragments of molecules generated
180 mainly by thermal degradation that still maintain a high biological activity. In this study, the
181 polyphenolic profile of *T. jasminoides* flowers hydrodistillation wastes has been analysed and results
182 are summarized in Table 2. The lyophilized water waste (lyoWW) profile (Figure 2) is dominated by
183 three compounds (1, 3 and 4, table 2); UV and mass spectra allow to identify these components as
184 caffeoylquinic acids derivatives. They represent the 6% w/w of the lyophilized powder obtained from
185 hydrodistillation. Traces of other biologically active compounds in lyoWW such as quercetin-3-*O*-
186 rutinose (8) has been also recorded. Identification of caffeic acid (5) and quercetin-3-*O*-rutinose
187 (8) was based on the comparison of their UV, mass spectra and retention times with those of reference
188 substances. Luteolin-7-*O*-glucoside (13), quercetin (15) and luteolin (16) have been identified in
189 AcSR and EtSR (Table 2) at quantitative level lower than 1 mg in 1 g of dried solid residue. For the
190 quantity and quality of polyphenols content, the lyoWW seems to be the more interesting waste
191 fraction due the presence mainly of chlorogenic acids which have wide biological activities, including
192 anti-oxidant, anti-inflammatory and neuroprotective (20-21). These compounds have been detected
193 in other flowers extracts and infusions also used in traditional medicine as a folk remedy (22-24).

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195 **4. Conclusions**

196 This work is the first complete study on the volatile and non-volatile chemical profiles of *T.*
197 *jasminoides* flowers growing in the South of Italy. The volatile composition of the flowers has been
198 studied by means of the gas chromatographic analysis of the essential oil and the hydrolate (an
199 interesting secondary product of the distillation of this plant) both rich in linalool. In addition, the
200 problem of the exploitation of the distillation wastes was addressed in this study. Through the analysis
201 of the polyphenols content of the lyophilized water wastes and of the solid plant residue, this study
202 shows that part of the biologically active chemical profile, mainly chlorogenic acids, remains after
203 the thermal treatment of hydrodistillation maintaining its interest for an exploitation and potential use
204 in the pharmaceutical and cosmetic fields.

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Table 1. Chemical composition of Sicilian *T. jasminoides* essential oil and hydrolate

# ^a	RI ^b	RI ^c	Components	% Essential oil	% Hydrolate
1	830	859	3Z-Hexenal	-	1.78 (±0.01)
2	842	862	2E-Hexenol	-	1.33 (±0.01)
3	971	979	octen-3-ol	0.34 (±0.02)	1.48 (±0.01)
4	979	985	6-methyl-hepten-2-one	0.23 (±0.01)	0.78 (±0.01)
5	982	990	myrcene	1.20 (±0.00)	-
6	1011	1024	<i>p</i> -cymene	0.26 (±0.00)	-
7	1019	1026	<i>o</i> -cymene	0.11 (±0.01)	-
8	1024	1029	limonene	2.95 (±0.01)	-
9	1033	1037	<i>cis</i> -ocimene	0.15 (±0.00)	-
10	1055	1050	<i>trans</i> -ocimene	0.28 (±0.00)	-
11	1072	1072	<i>cis</i> -linalool oxide	-	1.60 (±0.03)
12	1084	1088	terpinolene	0.59 (±0.01)	-
13	1087	1090	dehydro-linalool	-	4.37 (±0.03)
14	1097	1096	linalool	26.24 (±0.04)	45.67 (±0.11)
15	1111	1111	6-camphenol	0.20 (±0.00)	-
16	1118	1122	<i>trans-p</i> -mentha-2,8-dien-1-ol	-	5.82 (±0.10)
17	1133	1137	<i>cis-p</i> -mentha-2,8-dien-1-ol	0.58 (±0.09)	3.87 (±0.12)
18	1151	1159	citronellal	0.08 (±0.00)	-
19	1165	1170	<i>p</i> -mentha-1,5-dien-8-ol	-	0.51 (±0.00)
20	1173	1177	terpinen-4-ol	0.65 (±0.01)	1.42 (±0.01)
21	1186	1188	α -terpineol	2.57 (±0.01)	11.42 (±0.05)
22	1194	1196	<i>cis</i> -piperitol	0.27 (±0.01)	-
23	1204	1205	verbenone	0.18 (±0.01)	2.51 (±0.07)
24	1212	1223	linalool formate	0.18 (±0.00)	-
25	1222	1229	nerol	0.77 (±0.01)	-
26	1247	1252	geraniol	1.00 (±0.13)	-
27	1248	1243	carvone	-	0.84 (±0.11)
28	1265	1265	geranial	0.36 (±0.01)	-
29	1298	1299	geranyl formate	0.21 (±0.01)	-
30	1301	1299	carvacrol	-	0.84 (±0.03)
31	1302	1324	<i>E</i> -jasmonol	0.12 (±0.01)	-
32	1325	1335	δ -elemene	0.32 (±0.00)	-
33	1330	1341	5-indanol	0.63 (±0.00)	-
34	1358	1364	eugenol	0.09 (±0.01)	-
35	1376	1390	α -copaene	0.40 (±0.00)	-
36	1382	1391	<i>E</i> -jasmone	0.30 (±0.00)	-
37	1403	1419	<i>Z</i> -jasmone	7.30 (±0.21)	-
38	1430	1439	α -guaiane	0.60 (±0.01)	-
39	1436	1444	6,9-guaidiene	0.67 (±0.01)	-
40	1439	1457	alloaromadendrene	0.94 (±0.01)	-
41	1462	1495	γ -amorphene	1.19 (±0.00)	-
42	1483	1505	geranyl propanoate	8.66 (±0.03)	-
43	1498	1506	valencene	0.18 (±0.01)	-
44	1508	1519	δ -cadinene	0.38 (±0.01)	-
45	1535	1532	<i>trans</i> -nerolidol	3.28 (±0.21)	-
46	1557	1583	caryophyllene oxide	0.18 (±0.00)	-
47	1566	1595	β -calacorene	2.18 (±0.06)	-
48	1684	1698	<i>cis</i> -farnesol	5.11 (±0.27)	-
49	1790	1822	farnesyl acetate	1.38 (±0.25)	-
			Monoterpenes hydrocarbons	5.28	-
			Oxygenated monoterpenes	41.95	78.87
			Sesquiterpenes	16.81	-
			Others	9.07	5.37
			Total	73.11	84.24

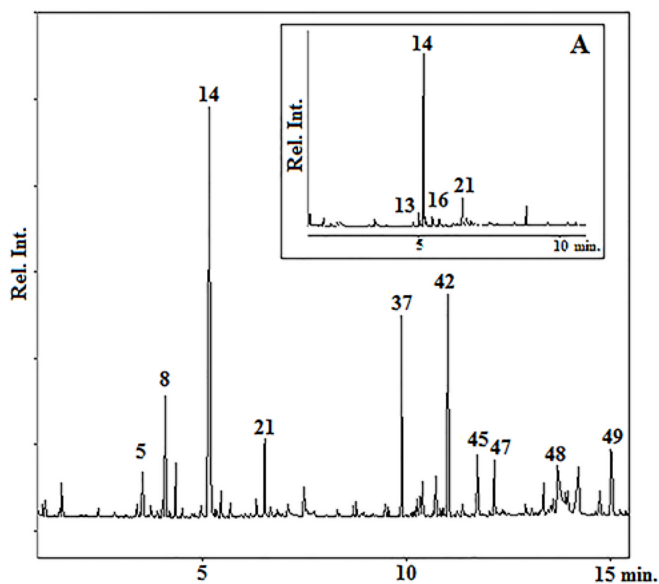
^a The numbering refers to elution order, and values (relative peak area percent) represent averages of 3 determinations ^b Retention index relative to standard mixture of *n*-alkanes on SPB-5 column; ^c Retention index from literature.

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Table 2. Main polyphenols identified in *T. jasminoides* hydrodistillation wastes

# peak	RT min.	Compound	HPLC-DAD λ -max (nm)	ESI(-)-MS	Mass Accuracy Values (ppm)	ESI(-)-MS fragments	mg/g of lyoWW	mg/g of AcSR	mg/g of EtsR
1	7.13	Caffeoylquinic acid derivative	243, sh295, 324	353.083	11.3	191.053, 179.034	20.3	-	-
2	9.21	<i>p</i> -Coumaroylquinic acid	310	337.089	8.3	-	0.62	-	-
3	10.24	Caffeoylquinic acid derivative	243, sh300, 325	353.083	11.3	173.044	18.13	-	-
4	10.89	Caffeoylquinic acid derivative	243, sh300, 325	353.083	11.3	191.053	20.15	-	-
5	11.82	Caffeic acid	244, sh300, 322	179.032	10.6	135.044	0.72	-	-
6	12.80	<i>p</i> -Coumaroylquinic acid	311	337.089	8.3	191.054	0.43	-	-
7	14.07	<i>p</i> -Coumaroylquinic acid	313	337.089	8.3	191.054	0.89	-	-
8	19.80	Quercetin-3- <i>O</i> -rutinoside	255, 353	609.138	11.5	301.918	0.58	-	-
9	20.68	Quercetin-3- <i>O</i> -glucoside	255, 353	463.082	11.0	301.918	0.88	-	-
10	21.21	Quercetin-3- <i>O</i> -galactoside	255, 351	463.082	11.0	301.918	1.71	-	-
11	21.71	di-caffeoylquinic acid	242, sh300, 332	515.113	10.5	190.927, 178.975	2.32	-	-
12	22.87	di-caffeoylquinic acid	242, sh300, 326	515.113	10.5	190.927, 178.975	1.06	-	-
13	24.06	Luteolin-7- <i>O</i> -glucoside	255, 348	447.088	9.4	285.907	-	0.08	0.08
14	24.84	Quercetin rhamnoside	255, 350	447.088	9.4	301.915	1.15	-	-
15	33.29	Quercetin	254, 370	301.032	7.6	-	-	0.98	0.21
16	34.99	Luteolin	253, 265, 346	285.037	8.4	-	-	0.37	0.17

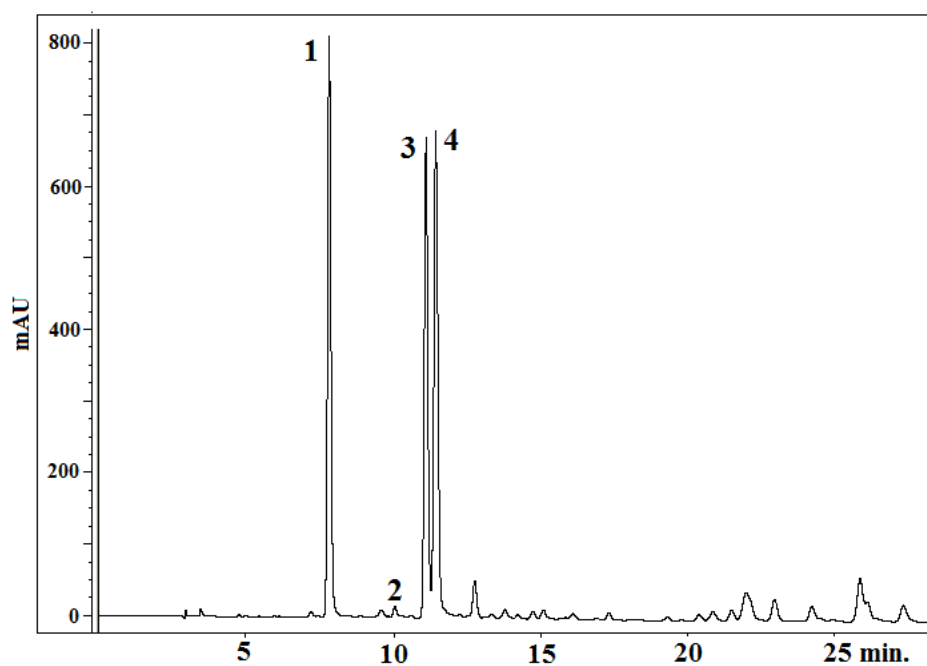
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315 **Figure 1.** GC-MS chromatograms of *T. jasminoides* essential oil and hydrolate in the box A.

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318 **Figure 2.** HPLC-DAD chromatogram of the lyophilized hydrodistillation waste of *T. jasminoides*.

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