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Thermoplastic Blends Based on Poly(Butylene Succinate-co-Adipate) and Different Collagen Hydrolysates from Tanning Industry—II: Aerobic Biodegradation in Composting Medium

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Abstract

Two different raw hydrolyzed collagens (HCs), by-products of the Tannery industry, were investigated in blends with a bioplastic, as poly(butylene succinate-co-adipate) (PBSA), for the production of thermoplastic items for possible applications in agriculture. Chemical characterization of selected PBSA/HC blends and phytotoxicity assays on garden cress seeds (*Lepidium sativum* L.), used as spy species, were carried out; in addition, biodegradation and disintegration of specimens were assessed under controlled composting conditions at different temperature (58 and 25 °C). Although one of the HC investigated released sodium chloride in the aqueous extract, all PBSA/HC blends, up to 20 wt.% HC, resulted no-phytotoxic and showed considerable amounts of macro- and micro- nutrients for plants (mainly nitrogen). Regardless the amount added, HCs enhanced the biodegradation rate of PBSA/HC blends in compost at 58 °C compared to pure PBSA; lowering the temperature at 25 °C, as expected, biodegradation rate slightly lowered using the same compost. Most disintegration tests, performed on dog bone samples, corroborated the results of the biodegradation tests, thus suggesting that plastic mixtures could reasonably end their life cycle in a composting facility without decreasing the quality and the safety of the resulting compost. The outcomes achieved encourage the use of raw collagen hydrolysates from tanning industry in the production of PBSA-based thermoplastic blends to produce compostable items (mulching films and/or plant pots) for more sustainable uses in agriculture and/or plant nurseries. In addition, the use of these low-cost by-products can lower the cost of final product and give it fertilizing properties for plants given the presence of organic nitrogen in the hydrolysates.

Keywords Poly(butylene succinate-*co*-adipate) \cdot PBSA \cdot Hydrolyzed collagen \cdot Aerobic biodegradation \cdot Disintegration tests

Introduction

Plastic is a crucial and ubiquitous material in our daily lives. Over the past 50 years, the role and economic importance of plastics has consistently grown. Global production of plastics has increased more than 20-fold since the 1960s,

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² Department of Civil and Industrial Engineering, University of Pisa, Largo Lucio Lazzarino 2, 56126 Pisa, Italy reaching 359 million tons (Mt) in 2019 and it is expected to double again over the next 20 years (Source: European Bioplastics, nova-institute, 2019).

However, too often the way plastics are produced, used and discarded fails to capture the benefits of a typical "circular economy" approach and harms the environment: globally, 5–13 Mt of plastic waste end up in the oceans every year, causing growing public concern [1].

To face that, biodegradable plastics, specifically designed for short duration at their end-of-life, are seen as solution, although the global production capacity of biodegradable polymers still remains very low, about 1.16 Mt, corresponding to only 0.32% of the global plastic production (Source: European Bioplastics, nova-institute, 2019).

Besides, the introduction in the market of new plastics with enhanced biodegradable properties (expected to increase by 13.6% in the next five years), could even exacerbate the risk of release of micro-plastics into the environment, especially in the absence of clear labelling and marking for consumers proving their authentic biodegradability.

Biodegradation refers to decomposition of substances occurring with the prevalence action of microorganisms: it proceeds in several steps, including biodeterioration, biofragmentation and assimilation that finally lead to the recycle of carbon, the mineralization of organic compounds and the generation of new biomass.

Different standard methods have been defined to evaluate the biodegradability of plastics under different environment conditions. However, the unique proof that a plastic polymer, essentially consisting of an organic carbon skeleton, is consumed by microorganisms is its conversion into carbon dioxide; therefore, regardless of the inoculum used (compost, soil or water) the majority of methods involve measuring the carbon dioxide emitted during biodegradation in controlled environmental conditions [2, 3].

Among the biodegradable plastics suitable to replace conventional polyolefins, poly(butylene succinate-*co*-adipate) (PBSA), recently available on the market even of bio-based origin, is an interesting candidate because it shows good technological properties (i.e. melt processability) and, in addition, due to lower crystallinity and higher flexibility of polymer chains compared to poly-butylene succinate (PBS), enhanced biodegradability performance in different environments. In fact, several studies have demonstrated that films and molded items of PBSA significantly biodegraded within few months in soil, sea water and water enriched with activated sludge [4–8].

Unfortunately, the actual high cost of PBSA slows down its large use for commodity services: for this reason, in the last years, PBSA has been successfully blended with low cost natural polymers, such as corn starch [6, 9, 10], to lower the cost of the final product, while ensuring a high biodegradability.

Similarly, and for the same purpose, waste protein hydrolysates, such as hydrolyzed collagen (HC), a typical by-product of the tannery industry, could be used in blend with bioplastics [11], given the well-known good adhesive properties of proteins and their good processability in the melt.

In the present study, PBSA/HC blends containing up to 20 wt.% of HCs, obtained by alkaline and enzymatic hydrolysis of tanned leather shavings [12], were investigated in terms of their environmental sustainability. After a preliminary elementary characterization, PBSA/HC blends were assessed for their ultimate aerobic biodegradability under controlled composting conditions, simulating both "industrial" (58 °C) and "home" composting (25 °C). Moreover, disintegration tests of dog-bone specimens were carried out under composting, using a specific equipment [13] able to simulate on lab-scale a typical "industrial" process. Disintegration tests

were also performed under "home" composting, at room temperature while the curing phase had been running.

Materials and Methods

Materials

Pellets of PBSA (tradename BioPBSTM FD92PM) were purchased from MCCP Germany GmbH (Mitsubishi Chemical Co., Tokyo, Japan). FD92PM is produced by copolymerization of bio-based succinic acid and adipic acid with 1,4-butanediol [14, 15], having a content of butylene adipate of about 20 wt.% [16]. This PBSA grade is certified industrial/home compostable and biodegradable in soil by TUV Austria [17]. Two types of HC were used for making PBSA/HC blends: one derived from alkaline hydrolysis of shavings, (HCa), supplied by Consorzio SGS (Santa Croce sull'Arno, Italy) and another derived from enzymatic hydrolysis of shavings (HCe), supplied by ILSA S.p.A. (Verona, Italy). Some chemical features of the two HC are reported in Table 1.

As reported in the concomitant study carried out by Seggiani et al. [12], PBSA/HC blends were prepared by melting extrusion and provided in different shapes (pellets, dog-bone specimens and films) for running investigations following described. Blends selected for testings were PBSA/HCe (80/20, w/w), PBSA/HCa (95/5) and PBSA/HCa (80/20).

Chemical Characterization of PBSA/HCs Blends

The ash content of samples, previously dried at 105 °C and reduced to size < 1 mm, was determined as weight loss at 650 °C for 24 h in a muffle furnace. Total carbon,

 Table 1
 Properties of the HCs used in preparation of PBSA/HC

 blends (data provided by the suppliers)

Property	НСа	НСе
Dry matter (at 103 °C, wt.%)	97.0	99.9
Ash (at 550 °C, wt.% db)	7.3	<1
Bulk density (kg/L)	0.50	0.55
Total nitrogen (wt.% db)	14.1	15.9
Organic nitrogen (wt.% db)	13.2	15.9
Organic carbon (wt.% db)	43.7	49.5
Organic matter (wt.% db)	87.4	99.0
NaCl (wt.% db)	7.8	<1
Total amino acids (wt.% db)	88.9	99.0
Free amino acids (wt.% db)	39.7	0.6
pH in aqueous solution	5.3	6.1
Water solubility (1:4, w/v)	Total	Almost total

db dry basis

hydrogen, nitrogen and sulphur were determined on aliquots (50–100 mg) of samples, using a CHNS analyzer (Vario Macro Cube, Langenselbold, Germany). Electrical conductivity (EC) and pH were determined on deionized water extracts obtained after 24 h extraction (1:10 w/v).

The elemental analysis was carried out on dry samples after acid digestion performed in a closed vessel microwave system (Ethos D, Milestone, Leutkirch im Allgäu, Germany), in accordance with the method US-EPA 3051A [18]; digested samples were then analyzed using Inductively Coupled Plasma-Optical Emission Spectrometry (ICP-OES) analyzer ICAPTM 7200 (ThermoFisher Scientific, Waltham, MA, USA).

Biological Assays

Germination tests on *Lepidium sativum* L. seeds were performed on composts used as inoculum in the ultimate aerobic biodegradation tests and on composts sampled after disintegration tests: these experiments were performed as described by Zucconi, et al. 1981 [19], adding deionized water on compost samples up to achieve 85% of water content (wet weight).

Furthermore, the same test was also used to check preliminarily if the release of salts in water, expected from the PBSA/HC blends (mainly in HCa), could have a detrimental effect on the germination of L. sativum seeds. Test was conducted on water extracts (1:50 w/v) of neat PBSA and selected PBSA/HCs blends, and on HCe and HCa water solutions (1:50 w/v). Crushed pellets of neat PBSA and the selected PBSA/HCs blends were extracted with deionized water (1:50 w/v) for 24 h using a horizontal shaker, and the water extracts, obtained by centrifugation and filtration through a 0.45 µm membrane filter, were used as germination medium. A Whatman filter paper (no. 42), placed into 9 cm Petri dish, was wetted with 1 mL of extract and 9 seeds of L. sativum were placed on the paper. Deionized water was used as the control germination medium and five replicates were carried out for each sample. Petri dishes were wrapped with Parafilm® to minimize water loss and allow adequate aeration, and kept in the dark for 42 h at 24 °C. At the end of incubation, the number of germinated seeds and primary root lengths were measured and expressed as percentage of the Germination Index (GI) in the control. GI was calculated using the following equation:

$$GI(\%) = \frac{G_s \cdot L_s}{G_c \cdot L_c} \times 100$$

where G_s and G_c are the number of seeds germinated for the sample and for the control, respectively; L_s and L_c are the mean root lengths for the sample and for the control, respectively.

Ultimate Aerobic Biodegradation Tests in Composting Conditions

The biodegradability of neat PBSA and selected PBSA/ HC blends was evaluated according to the standard method UNI EN ISO 13,432 and 14,855–1 [20, 21], using a specific Lab equipment, set up at ISAFOM-CNR and previously described [22].

About 20 g of 200 μ - thick film sample, previously cut in small pieces (< 1 cm²) by scissors, were kept at 58 °C (thermostatic chamber, Gallenkamp, UK) in 2L airtight vessels (bioreactors) and exposed to about 240 g of wet inoculum (50% water content), providing a constant CO₂-free air flow rate of about 20 L/h all along the test. Two InfraRed Gas Analyzer (IRGA) sensors were used for measuring the CO₂ concentration in the exhausted air flowing over each vessel: Leybold-Heraeus Binos® (Germany), for levels below 3000 ppm, and Gascheck by Edinburgh Sensors (UK), for higher concentrations up to 100,000 ppm. An alternate measuring/cleaning sequence was ensured to properly monitor the CO₂ concentration, recording up to six series of data per day.

The biodegradation test was conducted in triplicates, using an olive mill waste compost, obtained using the procedure reported in [23], as inoculum (blank) and cellulose powder (Sigmacell Type 20, 20 μ m, Sigma-Aldrich) as certified biodegradable material (positive reference). The automation, monitoring, recording and elaboration of data were performed by a customized software developed in Lab-VIEW® environment.

Bioreactors were frequently shaken to stimulate and homogenize the biodegradation process, while the moisture content was estimated by weighting the vessels after shaking, and kept at about 50% all over the test in order to ensure an ideal environment for microbes driving the biodegradation.

The biodegradation extent of each sample was calculated as percentage, corrected for the inoculum (blank) vessel emissions, of the overall theoretical amount of CO_2 (Th_{CO2}) that could be released, in case of complete mineralization, and calculated on the basis of the initial organic carbon content of the sample:

Biodegradation (%) =
$$\frac{\sum CO_{2sample} - \sum CO_{2inoculum}}{Th_{CO2}} \times 100$$

where $\sum CO_2$ is the cumulative amount of CO₂ evolved from the sample or inoculum all over the test.

In order to study "home" composting, the ultimate biodegradation test was repeated at 25 °C, using the same equipment and procedure above described, according to the Australian Standard AS 5810–2010 [24]. This standard requires that the test sample shall degrade at least 90% in total or equal to maximum degradation of a suitable reference substance (cellulose in this case).

Disintegration Tests in Compost

Dog-bone shaped specimens (length 90 mm, max width 10 mm; min. width 4.8 mm; thickness 1.35 mm, weight 1.2 g), produced by injection molding using PBSA neat and selected PBSA/HCs blends [12], were subjected to the disintegration tests under simulated "industrial" composting conditions. Experiments were carried out using an automatic lab equipment, Composter (Figs. 1 and 2), recently designed and set up at CNR-ISAFOM [14], and mostly taking EN 14,045 as reference [25]. Composter includes two aerated and basically adiabatic 35 L bioreactors, able to reproduce and control an "industrial" composting process, characterized by typical sequence of mesophilic-thermophilic-mesophilic phases. It records in real time the temperature and weight of the biomass, and quantifies O₂ consumption and CO₂ emission resulted in the exhausted air from the aerobic biodegradation process driven by the microbial populations (mainly bacteria and fungi) colonizing the compost.

The organic mixture intended for running disintegration tests included organic waste (OW) such as: olive pomace (51%), dry olive leaves and twigs (4%), woodchips (14,5%), wheat straw (5.5%) and pig slurry (22%). Moreover, about



Fig. 2 Composter, lab-scale equipment arranged for assessing the disintegration degree of plastic specimens under "industrial composting" conditions



Fig. 1 Schematic diagram of the Composter

3% Etixamin, supplied by ILSA S.p.A. (Verona, Italy), was added for adjusting the C/N ratio of the biomass. The starting OW mixture prepared for the test showed a C/N = 26.3, moisture = 50.3%, volatile solids content = 93.5% (w/w), pH = 5.5. OWs were selected considering hygroscopic features, physical structure and C/N ratio of each ingredient, measured out in order to get an ideal batch to compost. It is well-known that piles with a C/N ratio ranging around 25-30, a moisture content 40-60 wt.% and a balanced microand macro-porosity (bulk density around 0.5 kg L^{-1}), compost in a right way. It usually results from an appropriate mixing of OW rich in water with others dry; in our case, wet olive pomace and pig slurry were mixed with hygroscopic bulking agents, such as wheat straw, woodchips and dry leaves. At the beginning each bioreactor was loaded with 11.7 kg of the batch described, and used for running disintegration tests.

The air flow rate of bioreactors is managed by two separate mass flow controller (Model F-201CV-10 K-AGD-22-V, 500 ln h⁻¹, Bronkhorst® High Tech, NL) able to keep the oxygen concentration in the exhausted air above 10 vol.%, thus ensuring a full aerobic biodegradation. Temperature of the biomass, pressure and temperature of the inlet and outlet airflows are monitored continuously during the trial, as well as weight loss of the bioreactors placed on scales. Concentrations of oxygen and carbon dioxide are recorded from the exhausted air by means of specific online gas detectors: zirconium dioxide oxygen sensor (model XYA5M, equipped with oxygen sensor circuit board ZBXYAF, FirstSensor, DE) and Gascheck CO₂ Sensor (0–10 vol.%CO₂) by Edinburgh Sensors Ltd (UK). Acquisition of signals, control of actuators, remote software supervision, data elaboration and monitoring are performed by means of fit-for-purpose integration of hardware and software created in NI LabVIEW environment. Parameters are stored, plotted and promptly displayed in tables, graphs and spreadsheets, showing on video in real time trends of biodegradation occurring in both bioreactors. Without stopping data collection, bioreactors can be easily opened for turning the mixture, thus warranting a homogeneous biodegradation process. At the end of the thermophilic phase, which lasts 5-7 weeks, bioreactors are emptied and compost arranged in a pile kept at room temperature to allow the mesophilic phase to take place (compost curing), while ensuring an adequate moisture content (40-60 wt.%) in the biomass up to the end of disintegration tests.

From the beginning, dog bone specimens were buried into the OW mixture loaded in the bioreactors in order to promote their biodegradation and disintegration, together with the rest of the biomass. Each type of specimen was placed into separate polyethylene (PE) net sacks (1 mm holes) for a successive easier recovery by sieving of the sample residues, likely reduced in small pieces. At the end of test lasted 84 days, visual assessments on dog-bone specimens were carried out for estimating the percentage of disintegration: a positive result occurs when the residual material with size higher than 2 mm is less than 10% of the initial mass weight. However, it was impossible to perform reliable weight loss assessments on dog-bone residues, due to an enhanced biodegradation occurred. Therefore, the degree of disintegration, as suggested by other authors [26], was estimated by measuring the area of plastic residues > 2 mm, using ImageJ, a public domain Java image-processing program (https:// imagej.nih.gov/ij/). In that case, the thickness of fragments was not taken into account, although it appeared thinned if compared to the initial one; therefore, outcomes shall be considered underestimated, at least partially.

Disintegration in compost was also performed at room temperature, simulating "home" composting. Tests were carried out in buckets left indoor (Fig. 3) and different ingredients were utilized for making composts used as degradation media (Table 2). Six dog-bone specimens, for each PBSA/ HC blend, were buried in different compost and left to biodisintegrate for 259 days.

At the end of both disintegration tests, the potential phytotoxicity of composts, sampled in the surrounding area where dog-bone specimens disintegrated, were evaluated by germination test on cress, as described above in the Biological assays section.

Results and Discussion

Chemical Characterization and Biological Assays

Both solutions of HCs (1:50, w/v) showed high salinity, with higher values for HCa, while PBSA revealed, as expected, low EC (Table 3). These data agreed with the intermediate values for salinity found on the PBSA/HC blend extracts, with higher values related to higher HC content.



Fig. 3 Dog-bone specimens buried into compost kept indoor at room temperature, undergoing the mesophilic *curing phase* for the home composting disintegration test

Table 2Compositionof composts used asbiodegradation media inthe "Home composting"disintegration test

Sample	PBSA	PBSA/HCe 80/20	PBSA/HCa 80/20	PBSA/HCa 95/5
	mix 1	mix 2	mix 3	mix 4
Compost components				
Broadleaf prunings, %	43.7	22.1	43.5	15.0
Conifer prunings, %	0.0	22.1	0.0	5.0
Pig slurry, %	48.4	47.6	56.5	0.0
Rabbit manure, %	4.7	4.8	0.0	0.0
Chicken manure, %	3.1	3.4	0.0	0.0
Olive husk, %	0.0	0.0	0.0	80.0
Starting compost composition				
Carbon, % dm	45.5	45.4	46.9	43.6
Nitrogen, % dm	1.3	1.5	1.2	1.0
C/N	34.0	31.2	39.0	43.6

dm dry matter

Table 3Chemical-physical andbiological characterization ofPBSA, HCs and selected PBSA/HC blends

					PBSA/HC		
	Unit	PBSA	HCe	НСа	HCe 20 wt.%	HCa 5 wt.%	HCa 20 wt.%
pН		7.45	5.10	5.19	5.60	4.81	4.71
EC	$dS m^{-1}$	0.051	1.728	3.160	0.999	0.590	2.350
	wt. db%						
ash		0.05	4.27	6.13	0.72	0.19	1.07
С		56.67	45.73	42.08	54.47	56.02	53.81
N		0.53	16.54	14.41	3.50	0.68	2.89
S		0.01	0.35	0.65	0.08	0.02	0.08
Н		6.99	7.06	7.59	7.20	7.05	7.01
	${\rm g~Mg^{-1}}$						
Na		6.71	4840.1	17,670.6	887.9	796.7	3712.4
Κ		26.9	608.9	724.2	137.29	103.31	241.3
Р		185.5	8.33	3978.8	60.4	187.4	672.7
Ca		12.1	9832.7	2128.7	2007.2	399.3	476.6
Fe		2.69	1.64	72.1	3.47	6.88	29.5
Mg		30.0	37.2	158.4	31.9	41.9	63.9
Cu		0.28	0.80	0.80	bdl	0.66	4.4
Zn		0.32	2.32	59.30	0.68	6.58	18.4
Al		0.94	7.37	10.72	2.08	3.24	6.4
Pb		bdl	bdl	bdl	0.25	0.10	0.12
Cr		bdl	10.43	37.39	1.98	1.67	8.58
Mn		0.03	0.72	4,74	0.22	0.27	0.96
Ni		bdl	0.85	1.15	1.24	bdl	bdl
Мо		0.09	0.70	0.29	0.15	0.24	0.38
As		0.45	0.24	0.19	0.52	bdl	bdl
Cd		bdl	0.03	0.01	bdl	0.04	0.05
GI	%	114.0	75.4	39.9	82.9	92.1	67.3

db dry basis, bdl below detection limit; pH, EC and GI determined on 1:50 (w/v) solutions for HCe and HCa

Ash analyses corroborated these findings: in fact, HCa and related blends showed higher values, while PBSA neat showed a negligible ash content. As regard pH, PBSA solution resulted nearly neutral, while both HC solutions showed a sub-acidic reaction, also confirmed in the PBSA/HC blends.

Given the HCs origin [12], CNHS analyses proved a relevant content of nitrogen in the pure HCs (about 15%), confirming data provided by the suppliers (Table 1). Consequently, higher nitrogen content resulted in the blends containing higher content of HCs, up to 3.5 wt.% N for PBSA/HCe 80/20 (wt%/wt%) (Table 3). Since nitrogen, as a key nutrient supply, usually boosts either microbial or plant growth, it can be assumed that the addition of HCs to the bioplastic can positively influence its biodegradation rate as well as the release of nitrogen, acting as fertilizer, provided that these mixtures end up in the growth media/ soil for plants.

ICP-OES analyses showed a relevant content of Na in both HCs, with values for HCa more than 3 times higher than those for HCe. Most of Na derives from the salted skins used in the tanning process. As for N, Na content was found in the PBSA/HCs blends as a function of the amount of HCs added in the blend. Moreover, the high water solubility of Na salts is responsible for the high EC (salinity) showed in both HCs solutions and related PBSA/HCs blends. This feature could be of detrimental effect on plant growth when Na is released directly into the growth media. However, in case of excess, Na could be easily removed by leaching irrigations because it is weakly bonded on soil/growth media colloids [27].

In addition to N, ICP-OES analyses revealed interesting amounts of other nutrients for plants, such as Ca (mainly in HCe), P (mainly in HCa), Mg and K in both HCs and related PBSA/HCs blends, while some micronutrients, such as Fe, Cu, Zn, Mn, Mo, even at low content, were found still within detectable values. Besides, it is worth noting that all checked heavy metals (Ni, Cd, Pb, Cr, As) resulted largely below limits required by the UNI EN ISO 13,432 standard [20]. Only Cr (total) in the pure HCa, appeared rather high but, even when loaded in the blends at the highest rate (PBSA/HCa 20%), Cr resulted 8.58 mg kg⁻¹, more than five times lower than the maximum admitted limit (50 mg kg⁻¹).

The GI values confirmed the potential safety for plants of the PBSA/HCs blends, even regardless the amount HCs added: in fact, all tested blends, as well as PBSA neat and HCe, showed GI % higher than 60%, threshold limit considered safe for water extracts obtained from compost [28]. Potential harmful GI was only found for the HCa solution (about 40%), probably due to the high salinity that may inhibit the germination of seeds and/or the growth of the primary roots in the spy species tested (*L. sativum* L.).

Summarizing from an agronomic point of view, in case PBSA/HCs blends ended up in soil/growth media for plants, the resulting positive outcomes arising from the biodegradation could reasonably overcome potential drawbacks.

Fable 4 Characterization of compost used as <i>inoculum</i> in the altimate aerobic biodegradation ests		Unit	Value
	pН	_	8.06
	EC	$dS m^{-1}$	2.960
		wt.%	
	ash		26.08
	GI		69.00
	С		38.17
	Ν		3.97
	C/N	-	9.61



Fig.4 Ultimate aerobic biodegradability of PBSA and PBSA/HCe 20% blends conducted under controlled industrial composting conditions (UNI EN ISO 14,855–1, at 58 $^{\circ}$ C)

Ultimate Aerobic Biodegradation in Composting Medium

The main chemical and biological features of the compost used as inoculum in biodegradation tests, are reported in Table 4 compost shows pH and EC values typical of mature products, with a GI% higher than 60% which certifies its safety. In addition, the nitrogen abundance (C/N < 20) makes it a perfect substrate for stimulating microbial growth, thus allowing ideal biodegradation tests.

The first biodegradability test, conducted at 58 °C, showed after 20 days cumulated emissions of 47.9 ± 5.6 and 54.0 ± 2.5 gCO₂ per vessel, in PBSA and PBSA/HCe 20%, respectively, corresponding to 69.7 ± 13.6 and $87.5 \pm 6.4\%$ of biodegradation (Fig. 4).

In the second test at 58 °C, after 20 days, cumulated emissions of 38.6 ± 10.5 and 40.2 ± 1.9 gCO₂ per vessel were recorded in PBSA/HCa 5% and PBSA/HCa 20%, respectively, corresponding to 74.3 ± 28.5 and $81.8 \pm 5.5\%$ of biodegradation (Fig. 5).

Analyzing outcomes from the positive reference (cellulose), both tests conducted at 58 °C showed little differences between vessels (less than 20% in terms of biodegradation), satisfying the conditions required by the UNI EN 14,855–1



Fig.5 Ultimate aerobic biodegradability of PBSA/HCa 5% and PBSA/HCa 20% blends conducted under controlled industrial composting conditions (UNI EN ISO 14,855–1, at 58 $^{\circ}$ C)

standard. In fact, cumulated emissions of 39.0 ± 2.4 and 31.0 ± 2.5 gCO₂ per vessel, corresponding to 61.4 ± 7.5 and $67.6 \pm 8.7\%$ biodegradation, were recorded for cellulose in the first and second test, respectively, corresponding to 12.2% and 12.9% as CV (coefficient of variability). Moreover, also the inoculum complied with the 14,855–1 standard which requires the release of 50–150 mg of CO₂ per gram of VS (volatile solids) after ten days of incubation; indeed, inoculum alone (blank vessels) emitted 153.8 mg ± 6.1 and 104.1 mg ± 7.3 of CO₂ per gram of VS, in the first and second biodegradation test, respectively.

Figures 4 and 5 show fast biodegradation rates from the beginning for all specimens tested, especially for blends with higher HC content. The presence of amino acids in both HCs enhanced the nitrogen availability for the microbes colonizing the inoculum so that they were stimulated to growth rapidly and use PBSA as carbon source, thus resulting in a quick biodegradation. It must be emphasized the fact that the high EC found in both HCs and consequent release of Na salts in the media-inoculum, did not affect negatively the biodegradation process. On the other hand, also pure PBSA, tested at 58 °C, reached 70% of biodegradation in very short time (about 20 days), thus fully accomplishing the 14,588-1 standard, and showing, like PBSA/HCs blends, even higher biodegradation rates than cellulose (positive reference). It can be seen that after 20 days of incubation at 58 °C, the biodegradation percentage found for PBSA and cellulose have been significantly higher than those reported by other authors in similar tests [29].

The results of the biodegradation tests carried out at 25 °C, are reported in Fig. 6 e 7. In the first test, after 55 days, cumulated emissions of 36.6 ± 1.5 and 26.5 ± 4.2 gCO₂ per vessel were recorded in PBSA and PBSA/ HCe 20%, respectively, corresponding to 71.7 ± 3.7 and $48.4 \pm 10.1\%$ of biodegradation (Fig. 6). In the second test, after 55 days, cumulated emissions of 29.3 ± 0.6 and



Fig. 6 Ultimate aerobic biodegradadation of PBSA and PBSA/HCe 20% blends conducted under controlled composting conditions (Australian standard 5810/2010, at 25 °C)

 23.9 ± 2.8 gCO₂ per vessel were recorded in PBSA/HCa 5% and PBSA/HCa 20%, respectively, corresponding to 70.9 ± 1.7 and $56.6 \pm 8.6\%$ of biodegradation (Fig. 7).

Looking at the biodegradation of cellulose triplicates conducted at 25 °C, as it happened at 58 °C, the difference between vessels was always less than 20%, as CV %. In fact, cumulated emissions of 28.4 ± 3.5 and 20.8 ± 1.9 gCO₂ per vessel, corresponding to 65.5 ± 11.1 (CV % = 16.9) and 56.9 ± 7.1 (CV % = 12.5), as % biodegradation, were recorded in the first and second test, respectively.

Regarding CO₂ emissions from the blank vessels loaded with the inoculum alone, the amount of CO₂ recorded was 20.4 ± 2.7 and 33.0 ± 4.0 mg of CO₂ per gram of VS, in the first and second test at 25 °C, respectively.

Tests carried out at 25 °C corroborated findings achieved at 58 °C showing that most of the bio-based plastic biodegrades quite rapidly even at lower temperature. In fact, they all showed a biodegradability comparable to that of the reference material (cellulose) in the same period of time



Fig. 7 Ultimate aerobic biodegradation of PBSA/HCa 5% and PBSA/HCa 20% blends conducted under controlled composting conditions (Australian standard 5810/2010, at 25 °C)

(55 days); only the PBSA/HCe 20% blend showed a biodegradability lower than 90% relative to the cellulose reference. The lower biodegradation rate showed by the PBSA/ HCe 20%, compared to neat PBSA and PBSA/HCa blends, can be reasonable attributed to the still present secondary structure of the HCe, as reported by Seggiani et al. [12], as well as a negligible content of free amino acids (Table 1), which can make the protein material of the blend less attackable by mesophilic microorganisms operating at such low temperatures.

As regards inoculum, comparing trends of biodegradation at different temperatures, it results that the temperature strongly affects the biodegradation rate. In fact, inoculum alone showed, at 58 °C, a respiration rate, expressed as mg CO₂ evolved per gram of volatile solids, 4.8 and 3.8 times higher than that registered at 25 °C, after 10 and 20 days, respectively (Table 5). On the other hand, analyzing the biodegradation trends for vessels added with bioplastics or reference material (cellulose), it results that higher temperature mainly boosts the process in the first period of biodegradation, especially for the bioplastic specimens. Indeed, cellulose showed at 58 °C, as respect to 25 °C, only a little increase in biodegradation rate (10-20%) while, after 10 days at 58 °C, PBSA, PBSA/HCe 20%, PBSA/ HCa 5% and PBSA/HCa 20% showed a biodegradation rate 3.4, 7.1, 1.8 and 1.8 times higher than those performed at 25 °C, respectively. In addition, these differences tend to reduce over time as shown in Table 6. The reasons of such behavior could be explained with the fact that microbial populations colonizing the inoculum may need some time to familiarize themselves with a new carbon source, as that contained in the bio-based plastics. On the other hand, cellulose represents a well-recognized food for all microbes and does not require any period of adaptation; therefore, it is rapidly metabolized from the beginning, regardless of the temperature.

Disintegration Tests in Compost

Figure 8 shows the weight loss and temperature of compost during the first phase of disintegration tests carried out

Table 5 Cumulated CO_2 emission released by the *inoculum* after 10 and 20 days of incubation at 25 °C and 58 °C (mean values from ultimate aerobic biodegradation tests)

Temperature	25 °C	58 °C
mg g ⁻¹	VS	
Mean CO ₂ ,—10 days	26.7	129.0
Mean CO ₂ ,—20 days	46.4	176.8

VS volatile solids

Table 6 Biodegradation (%) recorded after 10 and 20 days of incubation at 25 $^{\circ}$ C and 58 $^{\circ}$ C (ultimate aerobic biodegradation tests)

Sample/Temperature	25 °C	58 °C	58 °C / 25 °C (ratio)
Cellulose–10 days	41.4	44.0	1.1
Cellulose–20 days	53.6	64.5	1.2
PBSA-10 days	8.4	28.3	3.4
PBSA-20 days	46.9	69.7	1.5
PBSA/HCe 20%–10 days	7.7	54.1	7.1
PBSA/HCe 20%–20 days	19.7	87.6	4.4
PBSA/HCa 5%–10 days	29.1	51.3	1.8
PBSA/HCa 5%–20 days	59.1	74.3	1.3
PBSA/HCa 20%-10 days	28.8	50.6	1.8
PBSA/HCa 20%–20 days	50.3	81.8	1.6

under "industrial composting" conditions simulated by the Composter.

Having used an ideal OW mixture for composting, mass loaded in the biorectors started to self-heat very soon, showing an intense microbial activity from the beginning: temperature reached high values in less than 1 day, thus soon entering the thermophilic phase that proceeded for about 30 days. In the meantime, at test day 20th, in order to prevent parts of the biomass from drying out or showing an excess of moisture, without disturbing data logging, bioreactors were opened and the content thoroughly mixed. This action resulted in peaks of oxygen consumption and related carbon dioxide emissions, as reported in Fig. 9, as well as further rapid increase in temperature of the pile (Fig. 8) due to an intense revitalized microbial activity that releases heat in the biomass.

The thermophilic phase of composting lasted about 30 days giving the way to the mesophilic phase, partially



Fig.8 Temperature and weight of compost during the first phase (45 days) of industrial composting disintegration test conducted in Composter



Fig. 9 O_2 consumption and related CO_2 emission released from each bioreactor during the first phase (45 days) of industrial composting disintegration test conducted in Composter

carried out into the bioreactors (up to 45th day); later on, both bioreactors were emptied and the OW mixture left to compost in a tank (Fig. 10) at room temperature $(23.8 \pm 3.3 \text{ °C})$ up to the end of disintegration test (84th day). At this stage, residues of plastic specimen, if any, have been pulled out from compost for visual inspections and for assessing the extent of disintegration occurred.

An intermediate visual assessment, done after 20 days, evidenced clear signs of disintegration in all specimens under investigation, with particular reference to the neat PBSA ones, which showed clear breaks in their structure (Fig. 11).

Figure 12 depicts residues of plastic specimens pulled out from compost after 84 days of disintegration test, while Table 7 reports the extent of disintegration estimated.

According to the reference taken into account [25], only PBSA/HCe 20% dog-bone specimens showed positive findings, with % residues larger than 2 mm below the threshold limit admitted (10%). All other specimens did not pass the test, even though residues showed evident signs of degradation and resulted extremely fragile to the touch.

Phytotoxic test, conducted on compost sampled in the area (net bags, Fig. 10) where specimens disintegrated, confirmed the absence of potential detrimental effects caused by the biodegradation of bioplastics; in fact, they all showed GI values (Table 7) largely above the threshold limit (60%) considered safe for compost [28].

On the other hand, it is noteworthy that the Composter environment performed, in terms of biodegradation activity, similarly to an industrial composting plant; indeed, in a short time (less than 3 months), it demonstrated to yield a mature and safe compost, for plants. Compost showed a sub-acid reaction (pH 5.90) and low EC (1.23 dS m⁻¹); moreover, it reached a high biological stability; in fact, after 45 days, it showed a very low dynamic respiration index (DRI = 139 mgO₂ h⁻¹ kg⁻¹VS), with values largely below the threshold limit (800 mgO₂ h⁻¹ kg⁻¹VS) set by the new European legislation [30] for high quality compost. The Composter is able to monitor in real time the DRI, thus allowing an objective evaluation on the compost quality achieved during the tests.

The quality of compost was also confirmed by the high GI index (102.8%), determined on a significant sample collected random from the whole final biomass.

As regards disintegration test conducted under "home composting", (Fig. 13), a first visual inspection, to monitor the process, was done after 90 days only on PBSA neat specimens: although conducted in a less aggressive environment, PBSA neat specimens showed clear signs of disintegration.

Disintegration was then proven at the end of the trial (259th days) for the majority of specimens tested. In fact, PBSA, PBSA/HCe 20% and PBSA/HCa 20% specimens resulted completely disintegrated, leaving no visible residues in the biomass, sought after through an accurate manual sieving, while PBSA/HCa 5% specimens, surprisingly, did not show any disintegration at all (Fig. 14). Such a different behavior could be explained by the different composition



Fig. 10 Different specimens in separated PE net bags and buried into compost at room temperature (second curing phase, from 45 to 84th days)



Fig. 11 Intermediate visual inspection on neat PBSA specimens after 20 days of industrial composting disintegration test occurred in Composter: on the right side the starting dog-bone specimens

of the compost to which the 5% PBSA / HCa samples were exposed, as reported in Table 2.

These findings support the hypothesis that compost of different origin would have different performances, as a means of disintegration; therefore, particular attention must be paid to the quality of the compost used to perform the disintegration tests.

Conclusions

Blends based on PBSA and two different collagen hydrolysates, derived by alkaline (HCa) and enzymatic hydrolysis (HCe) of tannery byproducts, were investigated in terms of aerobic biodegradation and compostability. Most of the items produced by melt extrusion using PBSA/HC blends up to 20 wt.% HC, in different form (pellets, films and molded dog-bone) resulted compostable and with good biodegradation rates even in home composting conditions.

All investigated PBSA/HC blends did not show negative effects on the germination of *L. sativum* seeds while showing appreciable amounts of important macro- and micro- nutrients for plants (mainly nitrogen).

In comparison to neat PBSA, the presence of HCs in the PBSA-based blends enhanced the biodegradation rate in the composting tests carried at 58 °C. While, at 25 °C, the different secondary structure and free amino acid content of the two hydrolysates leads to a different behavior of the blends, showing for the blends containing 20 wt.% HCe a lower biodegradation rate compared to neat PBSA and PBSA/ HCa blends.

In conclusion, the developed PBSA/HC blends appear a sustainable opportunity for valorizing by-products of the tanning industry in the production of biodegradable/compostable items with fertilizing properties, given the presence of HC, suitable for applications in plant nursery as mulch films or molded products (plant pots, small containers).







PBSA neat

PBSA/HCe 20%





PBSA/HCa 5%

PBSA/HCa 20%

Table 7 Quantification of plastic residues in compost and evaluation of some quality parameters of the compost sampled in the area where specimens disintegrated during the "Industrial composting" disintegration test

Residues, %	GI, %	pН	EC, dS m ⁻¹
86.4	92.6	6.53	1.096
6.0	99.1	6.65	1.186
54.6	102.8	6.49	1.089
11.4	87.5	6.65	0.965
	Residues, % 86.4 6.0 54.6 11.4	Residues, % GI, % 86.4 92.6 6.0 99.1 54.6 102.8 11.4 87.5	Residues, % GI, % pH 86.4 92.6 6.53 6.0 99.1 6.65 54.6 102.8 6.49 11.4 87.5 6.65



Fig. 13 First visual inspection of neat PBSA specimens after 90 days of home composting disintegration test



Fig. 14 Specimens of PBSA/HCa 5% after 259 days of home composting disintegration test

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