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Supporting Information

A Spectroscopic Study on the Amyloid- β Interaction with Clicked Peptide-Porphyrin Conjugates: a Vision Toward the Detection of A β Peptides in Aqueous Solution

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SUPPORTING INFORMATION

A spectroscopic study on the Amyloid- β interaction with clicked peptide-porphyrin conjugates: a vision toward the detection of A β peptides in aqueous solution.

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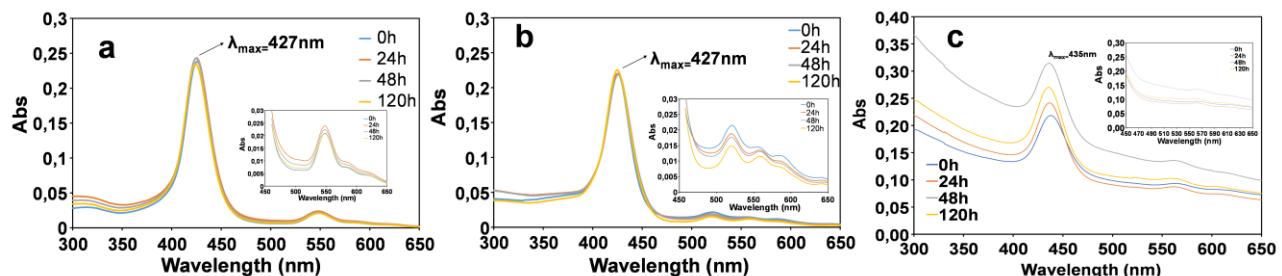


Figure 1S. UV-Vis spectra acquired at different time intervals, in the region between 300-650 nm of a) Cu-Ph-Pep, b) Ph-Pep, c) Zn-Ph-Pep.

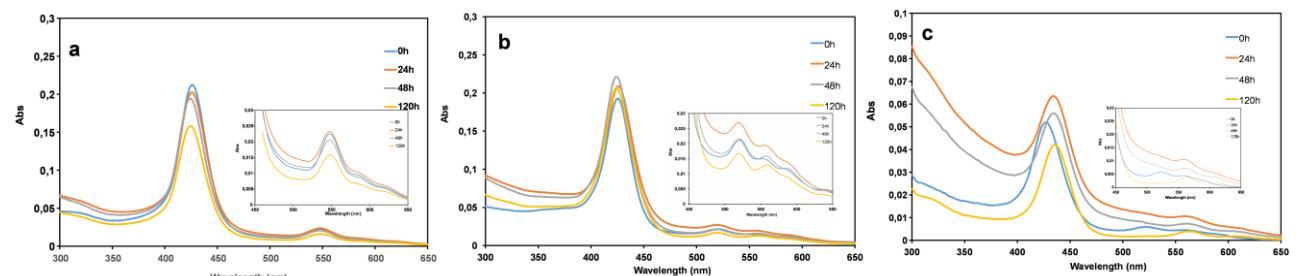


Figure 2S. UV-Vis spectra of porphyrin-peptide conjugates recorded in presence of A β ₄₂ monomers, a) A β ₄₂/Cu-Ph-Pep; b) A β ₄₂/Ph-Pep; c) A β ₄₂/Zn-Ph-Pep.

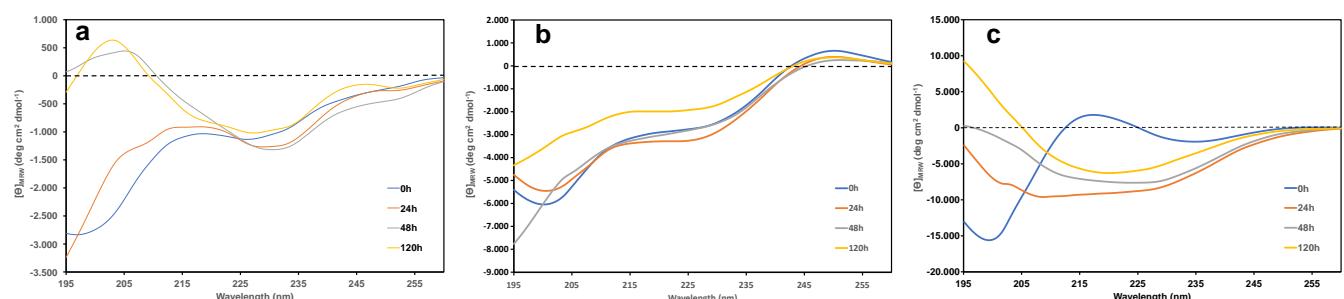


Figure 3S. CD spectra of conjugates at 5 μ M concentration recorded at different time intervals a) Cu-Ph-Pep; b) Ph-Pep; and c) Zn-Ph-Pep.

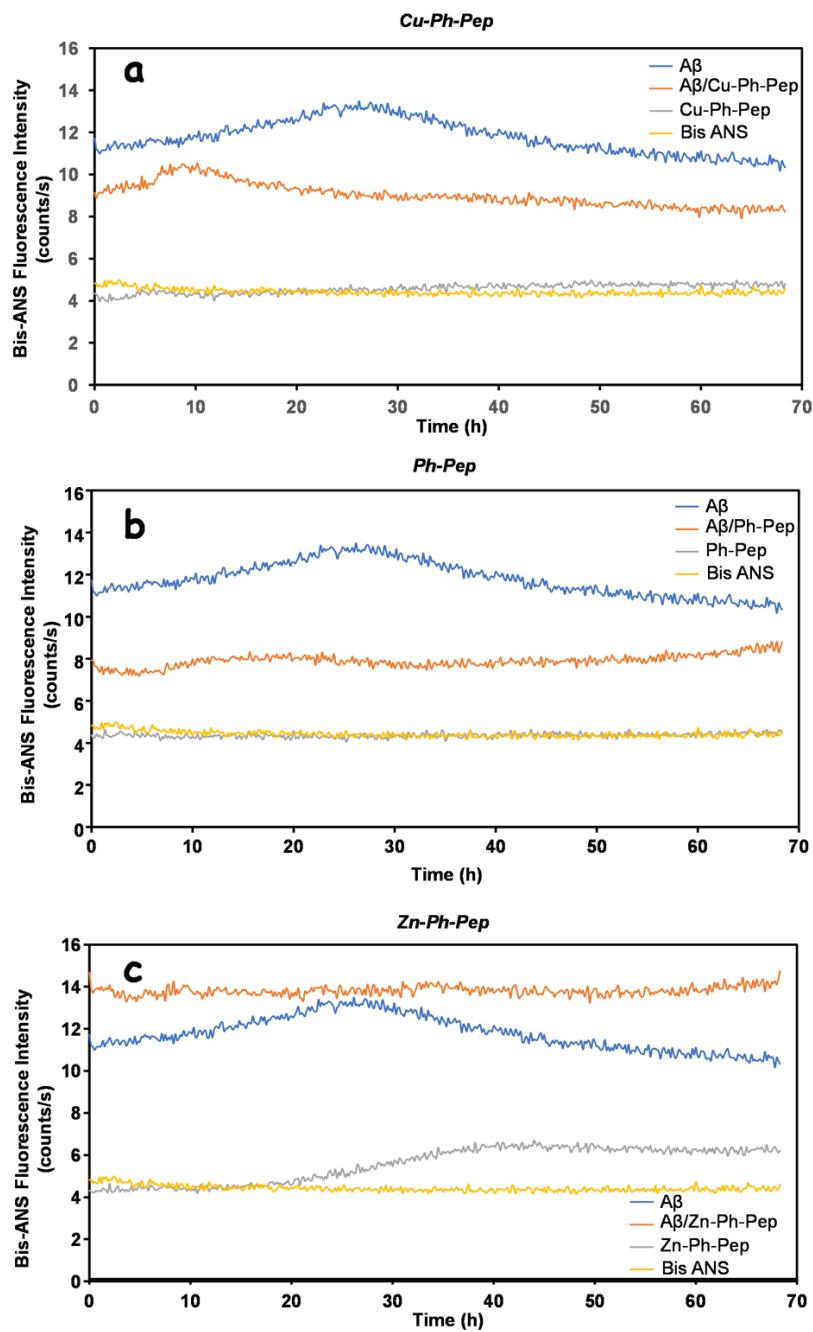


Figure 4S. bis-Ans fluorescence kinetics of A β 42 (20 μ M), A β 42/porphyrin-peptide conjugates at a 1:1 molar ratio and peptide conjugates alone. a) **Cu-Ph-Pep**; b) **Ph-Pep**; c) **Zn-Ph-Pep**.

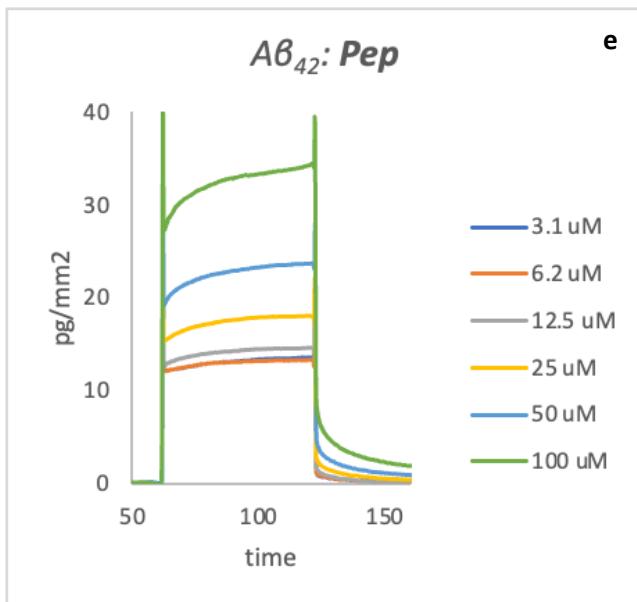
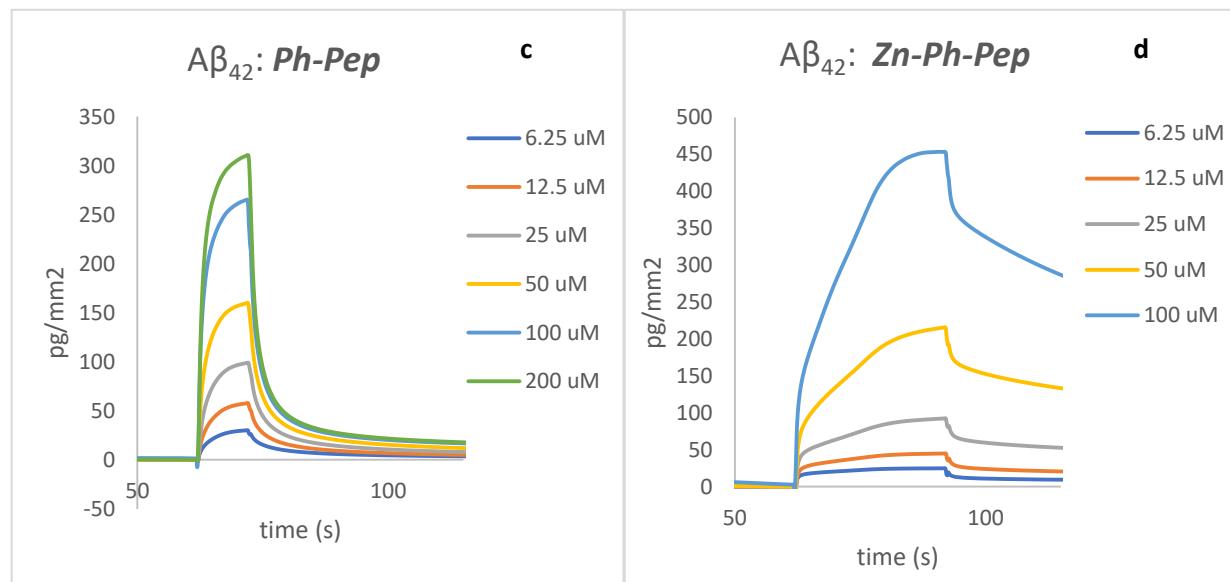
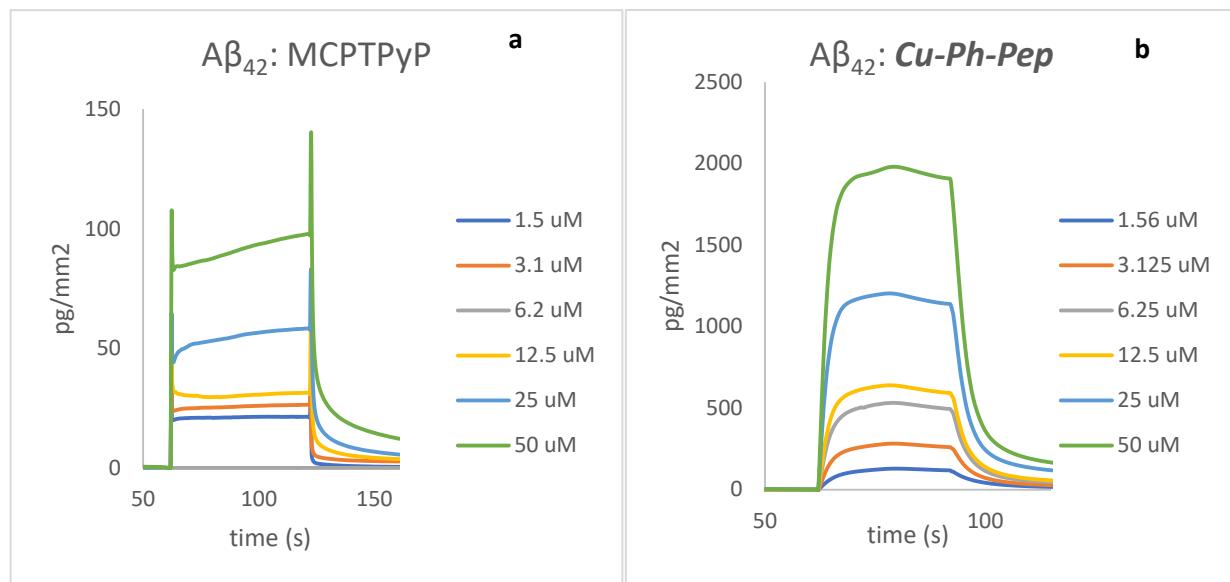


Figure 5S. Quantitative binding kinetics of $\text{A}\beta_{42}$ vs a) MCPTPyP; b) **Cu-Ph-Pep**; c) **Ph-Pep**; d) **Zn-Ph-Pep**; and e) **Pep** measured by GCI. All curves were blank subtracted. Experiments were performed in triplicate. All the quality assessments (i.e., χ^2 shown in Table 1 of the main text, parameter errors, and residual plots were acceptable; the sensorgrams had sufficient curvatures and the kinetic constant k_{off} were within the measurable range) were fulfilled. Descriptive Statistical Analysis was employed to calculate the mean of the obtained values of k_{on} , k_{off} and K_D as a measure of central tendency and standard deviations of the former values were computed as a measure of dispersion.

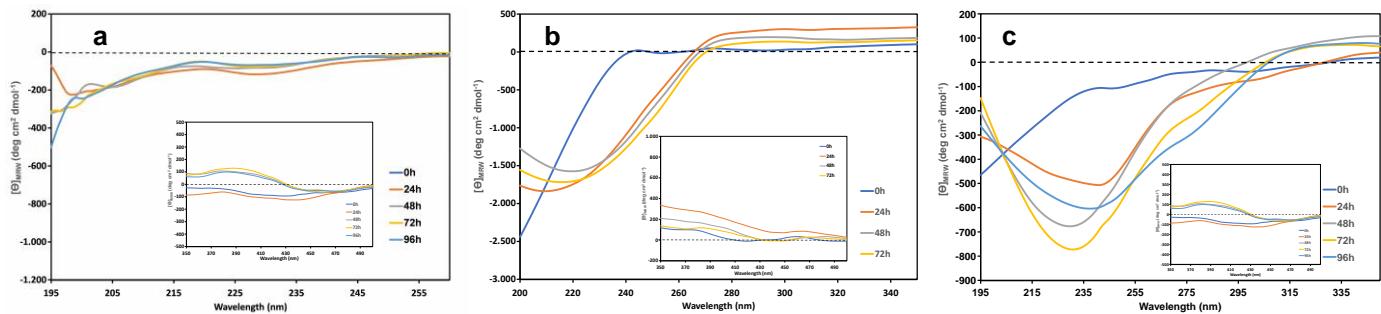


Figure 6S. CD Spectra the porphyrin-peptide conjugates at 20 μM in the Far-UV region (195-350 nm). a) **Cu-Ph-Pep**; b) **Ph-Pep**; c) **Zn-Ph-Pep**. inset: CD spectra in the near UV region (350-500 nm).

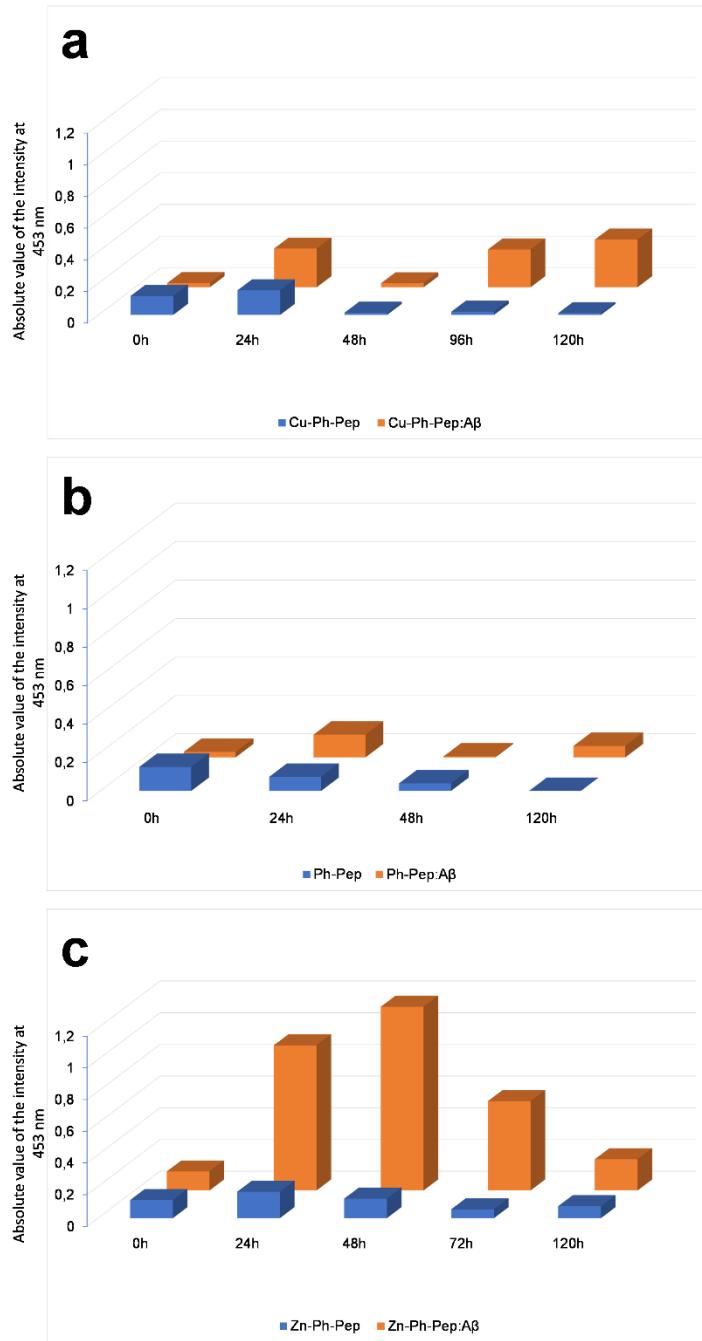


Figure 7S. Comparative graphic showing the absolute value of CD intensities at maximum absorption wavelength for each peptide derivative either in the presence or absence of A β ₄₂ at the same interval of time.

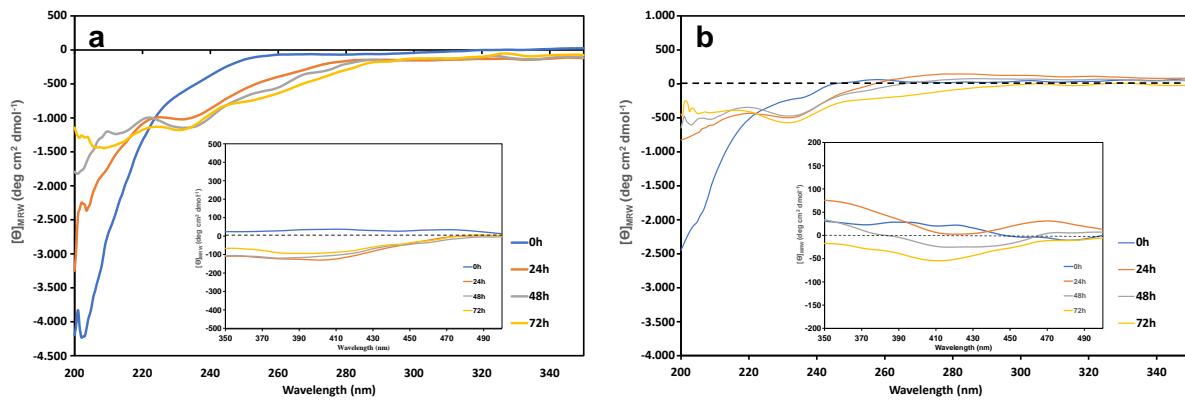


Figure 8S. Panel **a**: CD Spectra of $\text{A}\beta_{8-20}$ (20 μM) in the Far-UV (195-350 nm) and visible (350-500 nm, inset) regions. Panel **b**: CD Spectra of $\text{A}\beta_{8-20}/\text{Zn-Ph-Pep}$ equimolar mixture (20 μM) in the Far-UV (195-350) and near-UV (350-500 nm) regions.

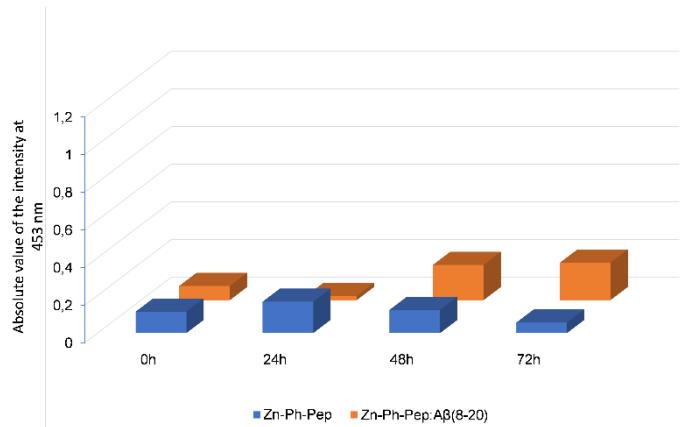


Figure 9S. Comparative graphic showing the absolute value of CD intensities at maximum absorption wavelength for conjugate **Zn-Ph-Pep** either in the presence or absence of $\text{A}\beta_{8-20}$ at the same time interval.

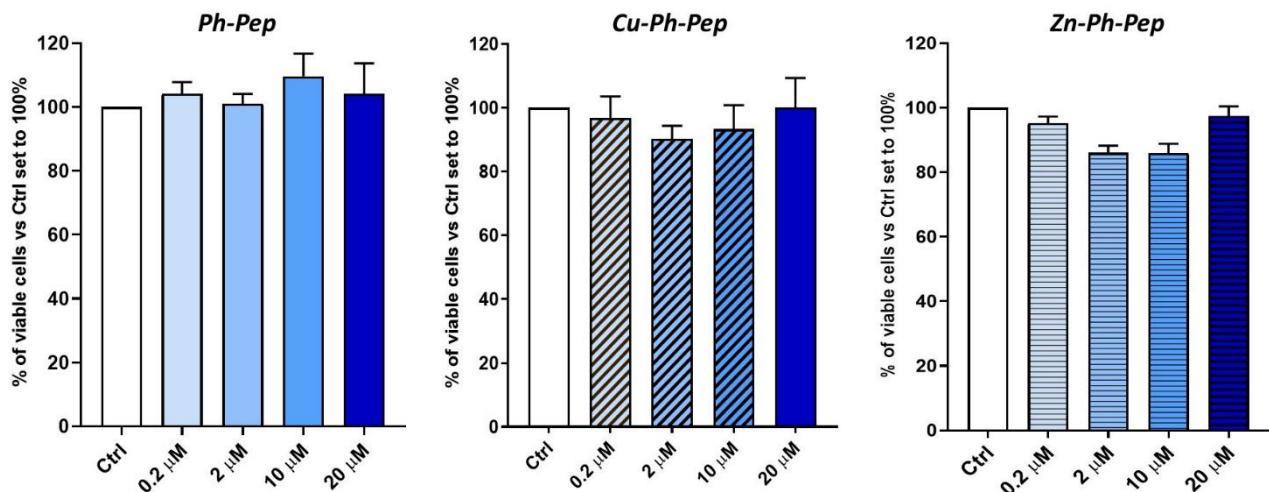


Figure 10 S. MTT Assay of fully differentiated SH-SY5Y treated for 48 hours with increasing concentrations of **Ph-Pep** and its derivatives (0.2, 2, 10, 20 μ M). Bars represent means \pm SEM of three independent experiments with n=3 each.

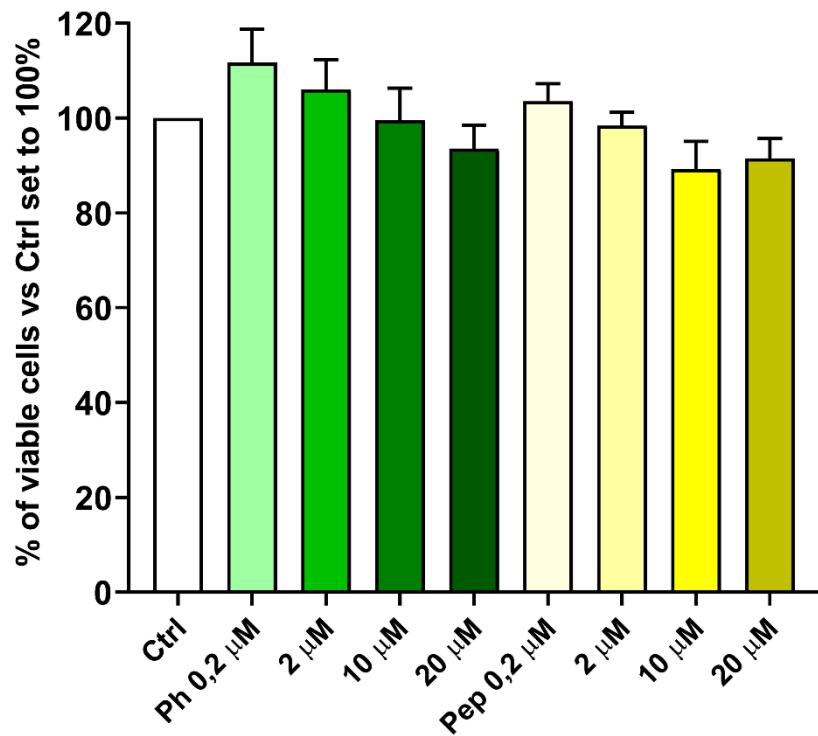


Figure 11S. MTT Assay of fully differentiated SH-SY5Y treated for 48 hours with increasing concentrations of pristine MCPTPyP (**Ph**) and Ac-GK(N₃)PGKLVFF-NH₂ Peptide (Pep) (0.2, 2, 10, 20 μ M), separately added to evaluate the potential toxicity of the single components of the compounds. Bars represent means \pm SEM of three independent experiments with n=3 each.

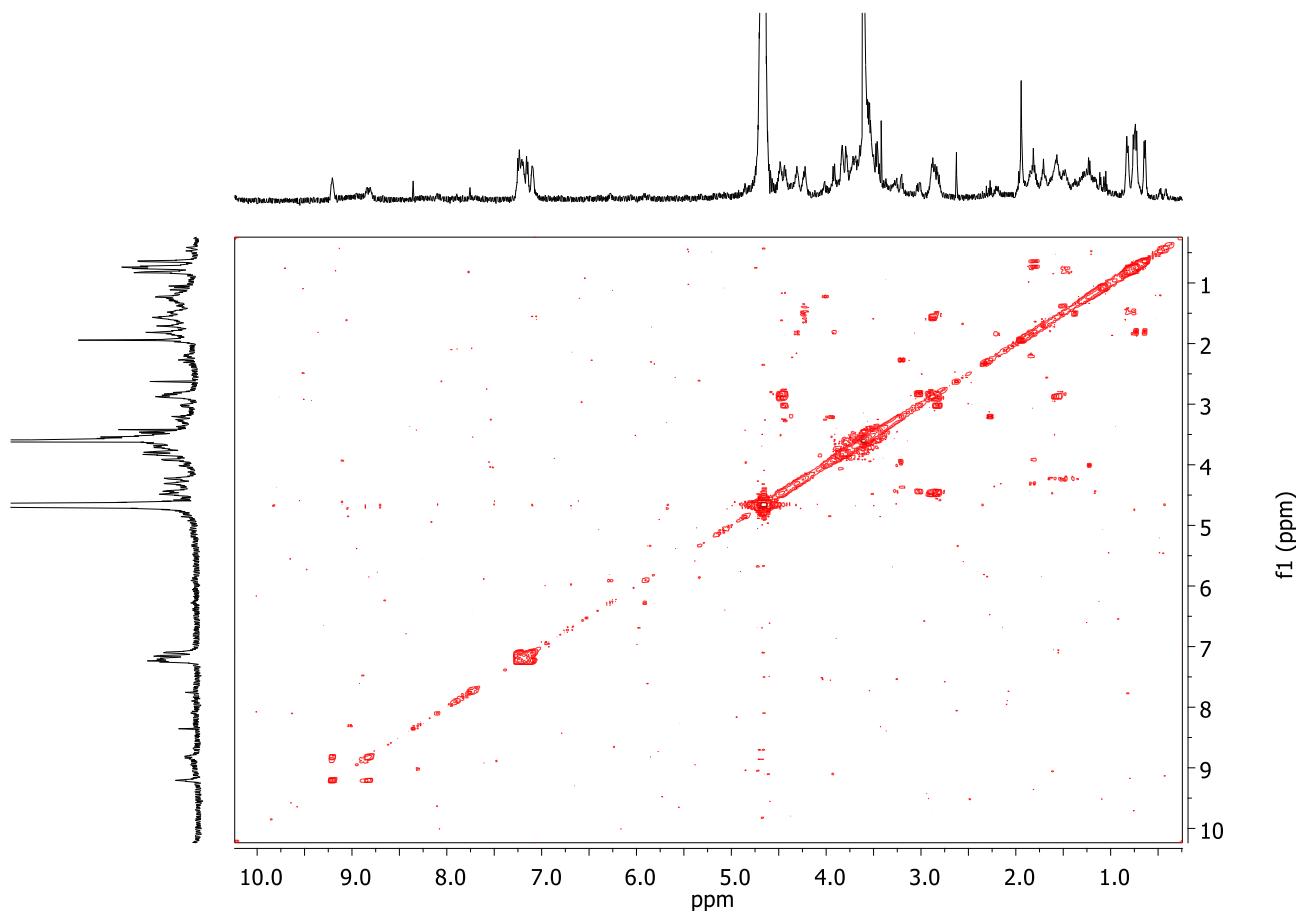


Figure 12S 2D COSY *Ph-Pep*

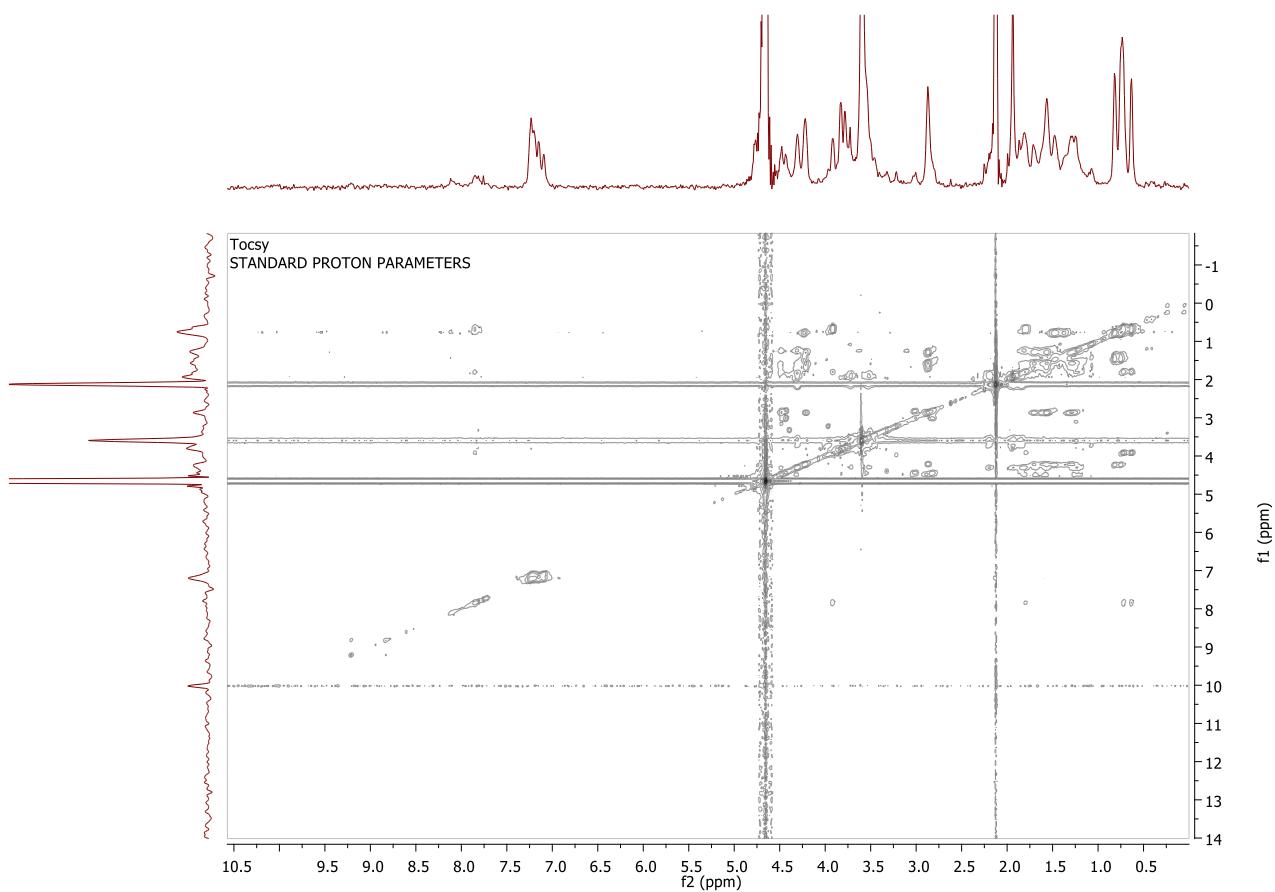


Figure 13S 2D TOCSY **Ph-Pep**

Table 1S. Proton Chemical shifts (ppm) for the peptide-conjugate **Ph-Pep**.

Position	H α	H β	H γ	Others
(1,2,3-triazol-4-yl)AlaOMe	3.89;	3.79; 3.75		O-CH ₃ 3.72
Lys(N ₃)	4.23	1.61; 1.58	1.31	δ CH ₂ 1.68; ε CH ₂ 2.90
Gly(Ac)*	3.83;3.78			CH ₃ 1.95
Pro	4.31	2.21; 1.83	1.94	δ CH ₂ 3.54;3.71
Gly*	3.34; 3.71			
Lys	4.48	1.69;1.58	1.35	δ CH ₂ 1.68; ε CH ₂ 3.05
Leu	4.23	1.50	1.49	CH ₃ 0.81; CH ₃ 0.75
Val	3.93	1.80	CH ₃ 0.72; CH ₃ 0 .64	
Phe	4.43	3.03; 2.84		ar7.09; ar7,22
Phe	4.48	2.82; 2.91		ar7.09; ar7.22
MCPTPyP moiety				
Piridyl _{2,6} 9.18	Piridyl 3,5 8.18	Pirrolyl 3,4 8.93; 8.72	Benzyl 2,6; 7.17 Benzyl 3,5 6.51	N-CH ₃ 3,61