Resolving a structural issue in cerium-nickel-based oxide: single compound or a two-phase system?

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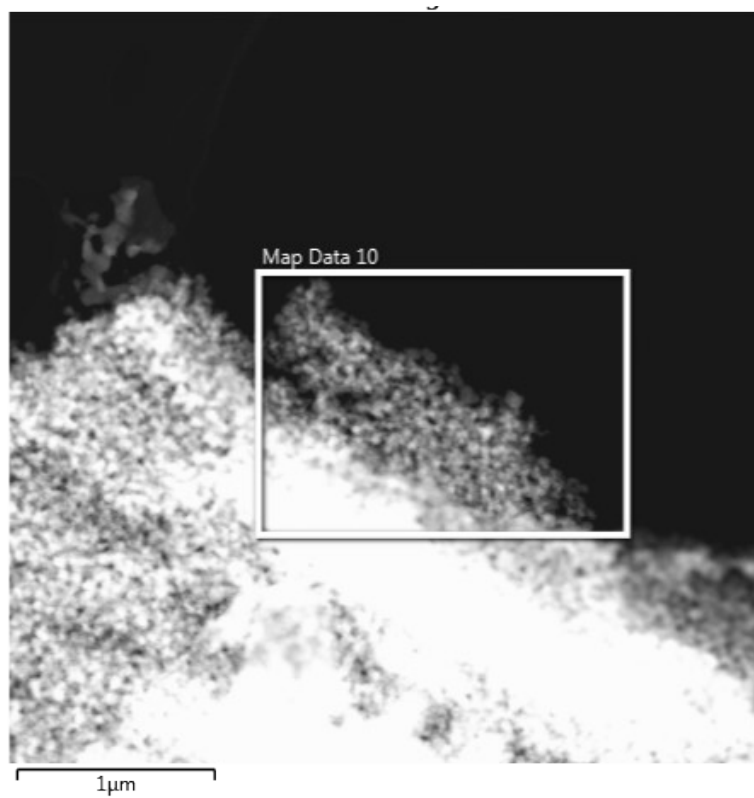* igor.djerdj@kemija.unios.hr

Table S1. Amounts of precursors for the synthesis of compounds.

Compound	Element	Amount (mmol)
RB1 ($CeNiO_3$)	Ce	1
	Ni	
CeO_2	Ce	1
NiO	Ni	1
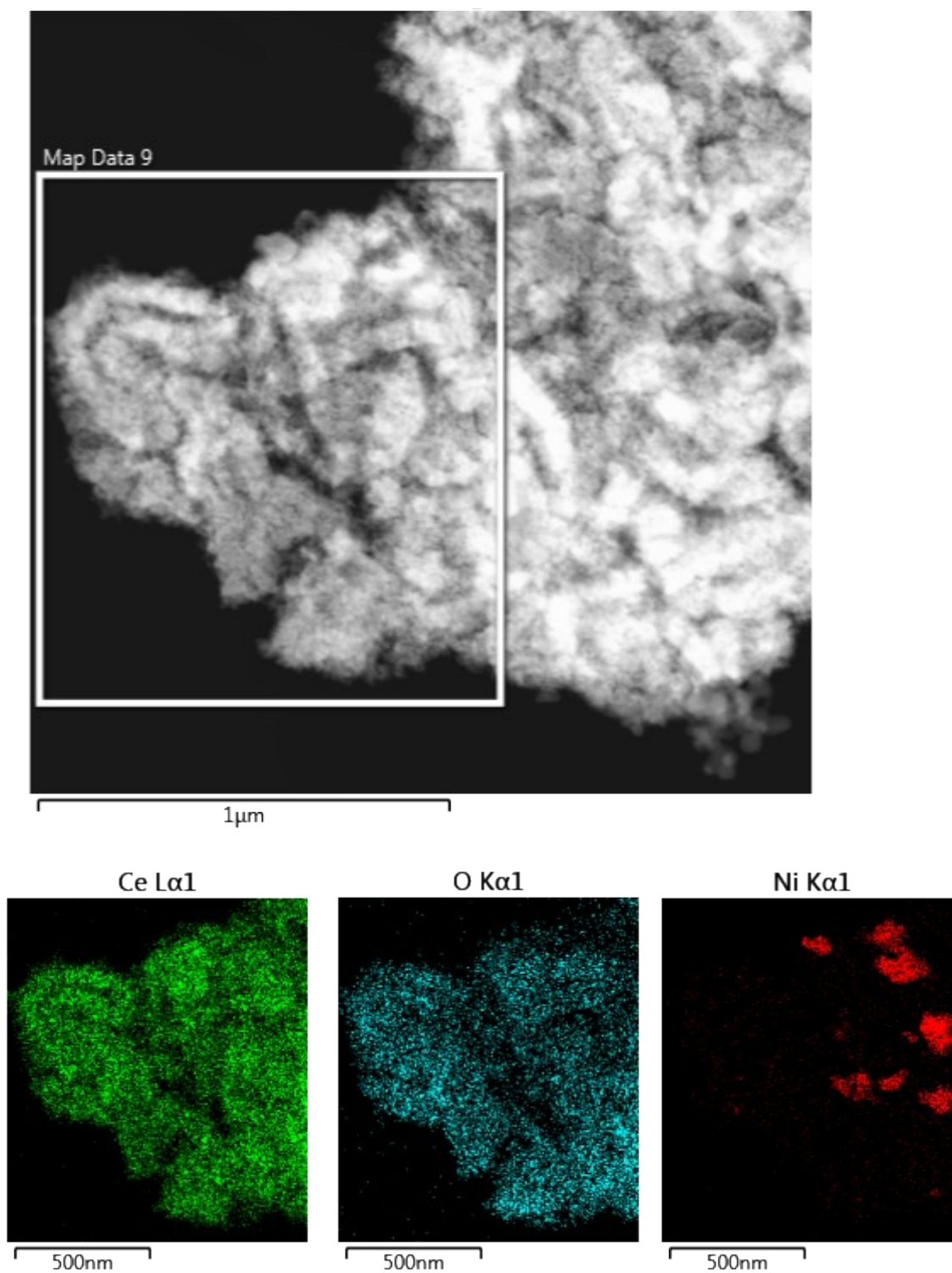
Figure S1. STEM-EDS image of a mixture CeO_2 + NiO ($n : n$ = 1:1) at one spot showing a non-uniform distribution of Ce, Ni and O. The measured spot contains more cerium than nickel.
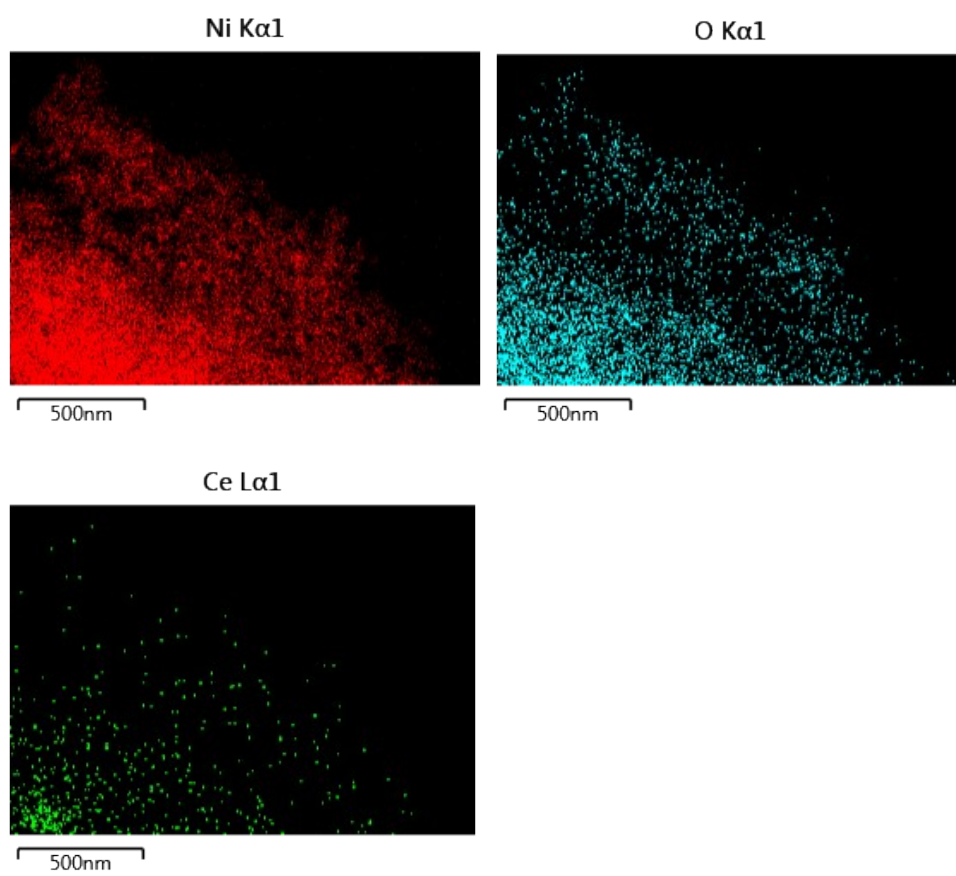
**Figure S2.** STEM-EDS image of a mixture CeO_2 + NiO ($n : n$ = 1:1) at one spot showing a non-uniform distribution of Ce, Ni and O.The measured spot contains more nickel than cerium.

EXAFS analysis of CeO$_2$ and RB1 at Ce L$_3$-edge

Structural parameters of the average local Ce neighbourhood (the type and the average number of neighbours; the radii and Debye–Waller factor of neighbour shells) in CeO$_2$ are quantitatively resolved from the EXAFS spectra by comparing the measured EXAFS signal with the model signal. The FEFF model for the crystalline CeO$_2$ nanoparticles is based on the cubic crystal structure of CeO$_2$ with the space group *Fm–3m* with the lattice constant $a = 5.411$ Å (PDF4+ database card #00-004-0593), where Ce is coordinated with 8 oxygen atoms at a distance of 2.34 Å, 12 Ce atoms at 3.83 Å and 24 oxygen atoms at 4.47 Å. The FEFF model comprised three single scattering and ten significant multiple scattering paths up to 4.6 Å, with 7 variable parameters: coordination shell distance (R) for the first (O) and second (Ce) neighbour shell, Debye–Waller factor (σ^2) of the single scattering path for (Ce-O) whereas Debye–Waller factors of other paths are evaluated using Debye model via Debye temperature (θ_d). The amplitude reduction factor (S_0^2) and shift of the energy origin of the photoelectron (ΔE_o), common to all scattering paths, are introduced. The shell coordination numbers were fixed to the crystallographic values in the fit, and the structural parameters of multiple scattering paths are constrained to those of the corresponding single scattering paths. A very good EXAFS fit (**Fig. S3**) is obtained in the k range of 2–9.9 Å and the R-range of 1.3–4.6 Å. The best-fit structural parameters are given in **Table S2**. For the RB1 sample, the EXAFS spectra were fitted with the FEFF model as it was done for CeO$_2$ (cubic *Fm–3m*) and the FEFF model for CeO$_2$ with the space group *P4$_2$/nmc* with the lattice constant $a = 3.818$ Å, $b = 3.818$ Å and $c = 5.426$ Å (PDF4+ database card #04-025-2756). For fitting the RB1, the amplitude reduction factor (S_0^2) was fixed to the value obtained for CeO$_2$ and shell coordination numbers were used as variable parameters. The best-fit structural parameters are given in **Table S2**. The results show that EXAFS cannot resolve between cubic and tetragonal structures for RB1 sample. The obtained structural parameters for RB1 are the same for both FEFF models and in agreement with those obtained for CeO$_2$.
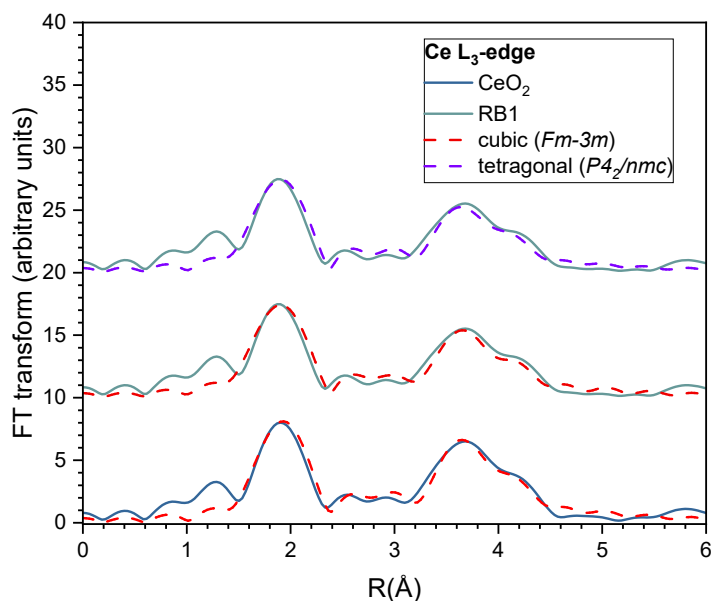
Figure S3. Fourier transform magnitude of the k^3-weighted Ce L_3-edge EXAFS spectra of the CeO_2 and RB1 samples calculated in the k range of 2–9.9 Å. Experiment – (solid line); the best fit EXAFS model calculated in the R-range of 1.3 to 4.6Å – (red dashed line for cubic system, violet dashed line for tetragonal system). Graph curves shifted vertically for clarity.

Table S2. Parameters of the nearest coordination shells around Ce cations in the CeO_2 and RB1 sample: coordination number (N), distance (R), and Debye-Waller factor (σ^2). Uncertainty of the last digit is given in parentheses. The best fit is obtained with the amplitude reduction factor $S_0^2 = 0.66$ and the shift of the energy origin $\Delta E_o = 8(1)$ eV.
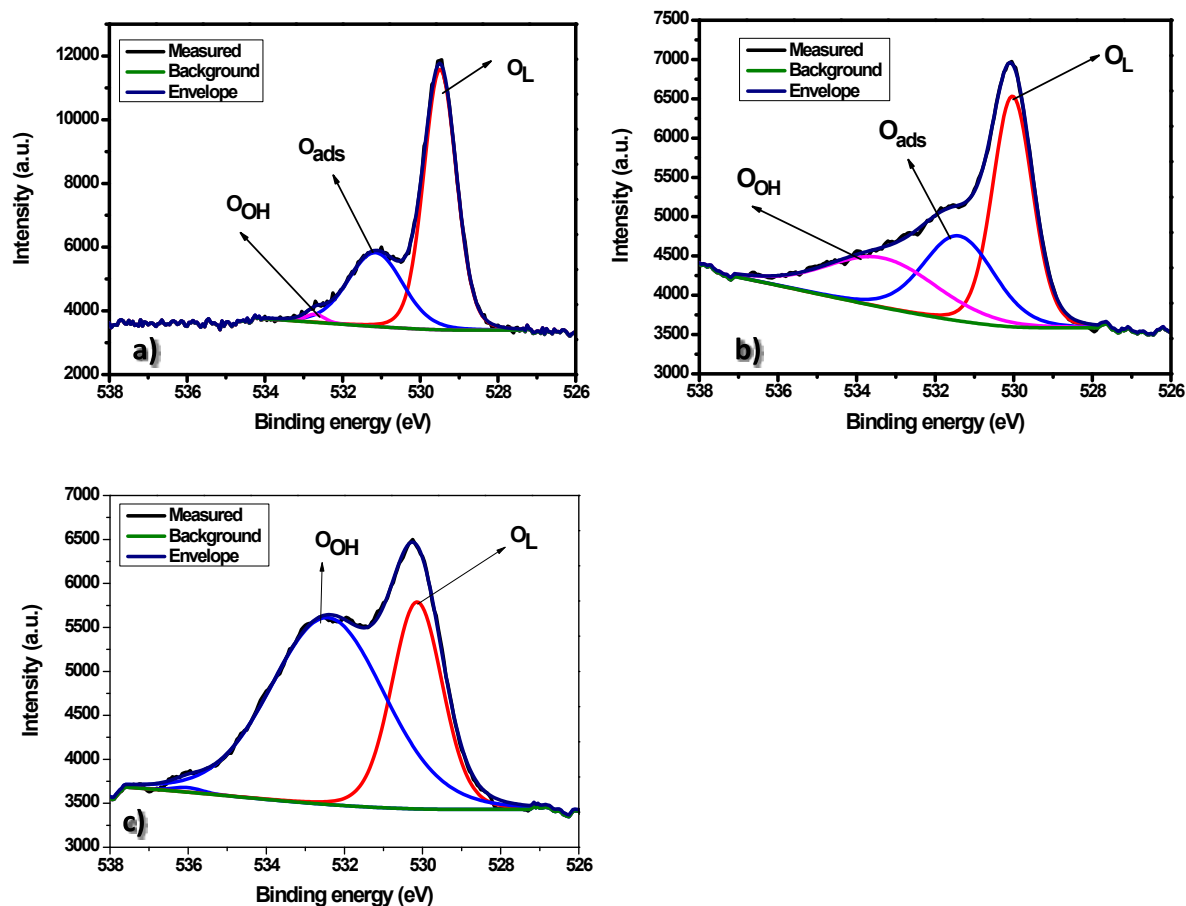
		CeO_2		
Neighbour	N (fixed)	R (Å)	σ^2 (Å^2)	R-factor
O	8	2.34(1)	0.004(3)	
Ce	12	3.86(1)	0.004(3)	0.028
O	24	4.41(1)	0.02(1)	
RB1 - fit with cubic *Fm–3m* structure model				
Neighbour	N	R (Å)	σ^2 (Å^2)	R-factor
O	8(2)	2.33(1)	0.006(3)	
Ce	10(3)	3.85(1)	0.004(3)	0.041
O	21(7)	4.41(2)	0.02(1)	
RB1 - fit with tetragonal *P4$_2$/nmc* structure model				
Neighbour	N	R (Å)	σ^2 (Å^2)	R-factor
O	8(2)	2.35(1)	0.005(3)	
Ce	8(3)	3.86(1)	0.004(3)	0.042
O	20(8)	4.3(1)	0.03(1)	

Table S3. High-resolution Ce 3d spectra peak position analysis for CeO$_2$ and RB1.

CeO$_2$	RB1	Assignment	Orbital splitting
Binding energy (eV)	Binding energy (eV)		
883.1	883.3	Core level	3d$_{5/2}$
889.8	890.1	Satellite 1	
899.3	899.3	Satellite 2	
902.3	901.9	Core level	3d$_{3/2}$
908.8	908.7	Satellite 3	
917.8	918.4	Satellite 4	

**Table S4.** High-resolution Ni 2p spectra peak position analysis for NiO and RB1.

NiO	RB1	Assignment	Orbital splitting
Binding energy (eV)	Binding energy (eV)		
853.9	854.4	Core level	2p$_{3/2}$
855.7	855.8	Core level	
861.0	857.3	Satellite 1	
864.4	861.4	Satellite 2	
866.6	864.1	Core level	2p$_{1/2}$
871.4	Not possible to read	Core level	
873.2	874.0	Core level	
879.3	Not possible to read	Satellite 3	

Figure S4. Deconvoluted high-resolution O 1s spectrum of a) NiO b) CeO_2 and c) RB1.

Table S5. Deconvoluted O 1s spectra report for CeO_2, NiO and RB1.

NiO		CeO_2		RB1		
Binding energy (eV)	Area	Binding energy (eV)	Area	Binding energy (eV)	Area	Assignment
529.5	8909.8	530.03	3902.3	530.1	3977.2	Lattice oxygen (O_L)
531.2	4252.8	531.4	2554.4	-	-	Adsorbed oxygen species (O_{ads})
532.7	161.7	533.3	2347.0	532.4	8113.3	Hydroxides (O_{OH})
Total area	**13324.2**		**8803.7**		**12142.1**	
Oxygen species concentration	**31.9 %**		**29.0 %**		**-**	
Hydroxide concentration	**1.2 %**		**26.7 %**		**66.8 %**	

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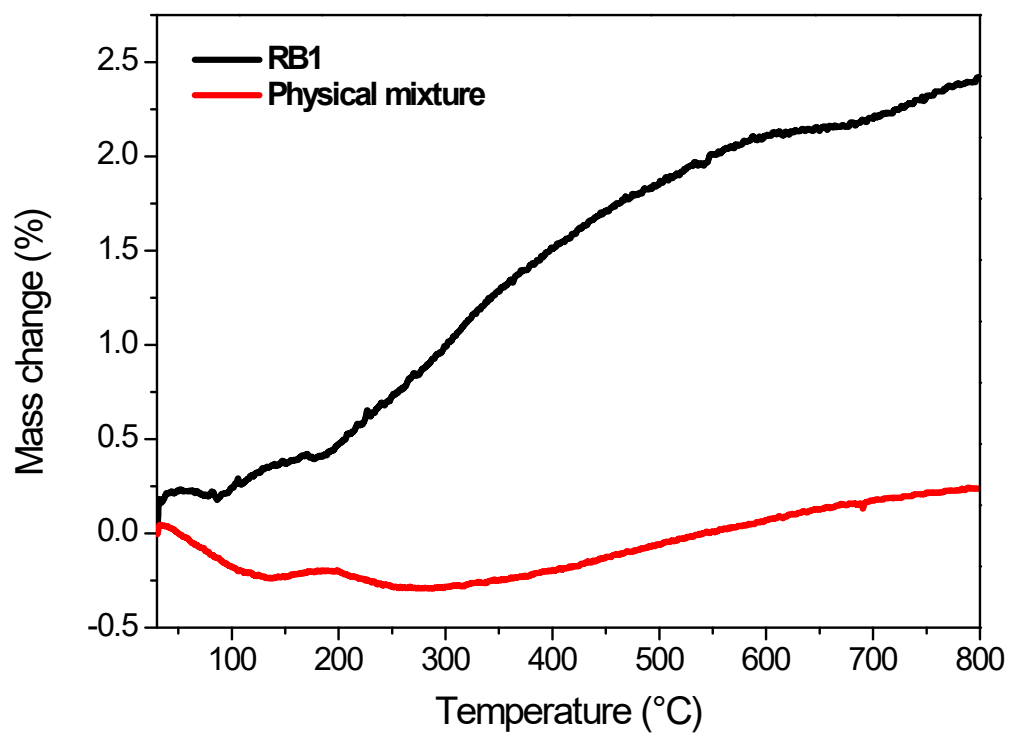
Figure S5. Thermogravimetric curves of RB1 (black) and a physical mixture of CeO_2 + NiO (red) during heating in an oxidative atmosphere (10 °C/min) from 30 °C to 800 °C.

Table S6. Crystallographic data obtained from the Rietveld refinement of 3% Ni doped CeO_2.

Compound	CeO_2 (3% doped Ni)
Chemical formula	$Ce_{0.97}Ni_{0.03}O_2$
Crystal system	Tetragonal
Space group	$P\,4_2\,/\,n\,m\,c$
Z	2
Calculated density (g/cm³)	7.076
Unit cell parameters (Å)	a= 3.8319(2) c= 5.4183(1)
Unit cell volume (Å³)	79.5600(1)
Phase content (wt. %)	100
Average crystallite size (nm)	8.5
Average apparent microstrain ($\times 10^{-4}$)	0.95
R_B	5.58
R_p, R_{wp}, R_e	13.2, 10.7, 8.23
χ^2	1.68