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The vacuum ultraviolet absorption spectrum of norbornadiene: Vibrational analysis of the singlet and triplet valence states of norbornadiene by configuration interaction and density functional calculations

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The vacuum ultraviolet absorption spectrum of norbornadiene: Vibrational analysis of the singlet and triplet valence states of norbornadiene by configuration interaction and density functional calculations



A synchrotron-based vacuum ultraviolet (VUV) absorption spectrum of norbornadiene (NBD) is reported, and the extensive vibrational structure obtained has been analyzed. The previously known 5b₁3s-Rydberg state has been reinterpreted by comparison with our recent high-resolution photoelectron spectral analysis of the X^2B_1 ionic state. Additional vibrational details in the region of this Rydberg state are observed in its VUV spectrum when compared with the photoelectron 2B_1 ionic state; this is attributed to the underlying valence state structure in the VUV. Valence and Rydberg state energies have been obtained by configuration interaction and time-dependent density functional theoretical methods. Several low-lying singlet valence states, especially those that arise from $\pi\pi^*$ excitations, conventionally termed NV₁ to NV₄, have been examined in detail. Their Franck–Condon (FC) and Herzberg–Teller (HT) profiles have been investigated and fitted to the VUV spectrum. Estimates of the experimental 0⁰ band positions have been made from these fits. The anomaly of the observed UV absorption by the 1A_2 state of NBD is attributed to HT effects. Generally, the HT components are less than 10% of the FC terms. The calculated 5b₁3s lowest Rydberg state also shows a low level of HT components. The observed electron impact spectra of NBD have been analyzed in detail in terms of triplet states.

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I. INTRODUCTION

Recently, we reported synchrotron-based, high-resolution photoelectron spectra (PES) for both norbornadiene (NBD) and

quadricyclane (QC).¹ These isomeric compounds, shown in Fig. 1, readily interconvert as discussed below, and this has led to numerous studies of the system having the potential for conversion of sunlight to electricity. We now present the previously unknown vacuum

ultraviolet (VUV) spectrum for NBD. The large amount of details exposed necessitates our corresponding study of QC to be delayed to a following paper. The PES for both compounds were previously analyzed in considerable detail by a combination of Møller–Plesset fourth order perturbation theory, which included single, double, and quadruple [MP4(SDQ)] substitutions, configuration interaction (CI), multi-configuration self-consistent field (MCSCF), and density functional theoretical (DFT) methods. Both are isoelectronic with cycloheptatriene (CHT), which we also studied recently, but lack conjugated double bonds.^{2,3} Our theoretical approaches will be similar to those that were successful with CHT.

Interconversions between NBD and QC and their derivatives are very important since a promising approach to energy storage from the sun lies in the use of strained organic molecules.^{4–7} This justifies a brief summary. These so-called molecular solar thermal (MOST) binary systems absorb solar energy via the alkene form (here, NBD) leading to the strained form (here, QC); the stored energy can then be released from the latter catalytically, regenerating NBD.

A related application for the NBD and QC systems is as a switch for an optical memory system.^{8,9} The "OFF" form of a photo-switch occurs when NBD or one of its derivatives is converted to the corresponding QC upon irradiation. Conversion back to NBD gives the "ON" form of the switch. The stored energy is released as electricity when the photo-switch interacts with a semiconducting electrode surface.

The NBD \Leftrightarrow QC interconversion occurs via an endothermic photoinduced [2 + 2] reversible cycloaddition.¹⁰ The reverse reaction where QC regenerates NBD with the release of heat can occur via thermal or catalytic interaction, or even fluorescence.⁷ When triplet sensitized using acetophenone, the NBD process appears to involve the two triplets, ³NBD and ³QC, followed by relaxation to QC,¹⁰ but not all sensitizers behave similarly.¹¹ Since the nature of the triplet states involved in these processes is obscure, this has led us to include the study of the electron impact (EI) spectra of NBD. During the interconversion process, up to 100 kJ mol⁻¹ of chemical energy is stored, a value comparable to contemporary batteries.¹²



FIG. 1. The compounds norbornadiene (NBD) and quadricyclane (QC). The classical bond switching between the two systems occurs during UV excitation. For simplicity, H atoms are not marked for the cyclic moieties.

Since the UV onset of NBD is 267 nm, a combination of donor and acceptor groups in the NBD + QC system is necessary to give an improved solar spectrum match since the main wavelengths of sunlight lie between 300 and 700 nm. Examples have been reported with an onset of absorption of up to 529 nm.^{10,13,14} The UV–visible absorption spectra of the substituted NBD and QC show a strong blue shift when in the QC form.

NBD, isolated in argon, xenon, or nitrogen matrices at 20 K, gives quadricyclane when irradiated with UV light.¹⁵ The selectivity of the photochemical reaction of NBD to QC has been rationalized in terms of matrix rigidity; this may exert a constraint on the transition state, where preference is given to the reaction where the product shape best fits the matrix site occupied by the reacting molecule.¹⁶ The NBD \leftrightarrows QC reaction is not exclusive in the gas phase since cyclopentadiene + acetylene or even toluene can be formed.^{15,17} All these alternative products are thought to be formed by breaking of different C–C bonds and rearrangement of the resulting diradicals.^{15,17}

While technological aspects of the NBD \leftrightarrows QC equilibrium lie outside the scope of the current paper, the electronically excited states for the parent molecules of NBD and QC are crucial to understanding these more complex interactions. Our theoretical work is directed to performing this at a more rigorous level than is currently available. However, the large amount of spectral study for NBD requiring our detailed analysis makes it essential to deal with NBD first and defer our parallel study of QC to a later paper. We do not discuss the details of the dynamics of the NBD \leftrightarrows QC process but refer interested readers to Refs. 4–9 for recent discussions.

We analyze the VUV spectrum of NBD following the same procedures as with CHT and cyclooctatetraene^{18,19} previously in a detailed manner. We have determined both adiabatic excitation energy (AEE) and vertical excitation energy (VEE) for singlet and triplet states of NBD. Where possible, we have retained C_{2V} symmetry, but for some states, structural minimization led to saddle points rather than true minima; this led to relaxation of symmetry to C₁. The AEE methods used are mainly based on time-dependent density functional theory (TDDFT), a version of single excitation configuration interaction (CIS). VEEs were determined by multi-reference multi-root singles and doubles configuration interaction (MRD-CI) calculations; since these are performed at the X¹A₁ ground state structure, the results are VEE. Rydberg states are important, and these were treated theoretically through the use of very diffuse Gaussian-type basis functions.

The close proximity of the two (non-conjugated) double bonds in NBD leads to direct spatial overlap. This splits the degenerate isolated ethylenic moiety energy into symmetric (S) and antisymmetric (A) combinations for both π and π^* orbitals,^{20–22} as discussed below. Much earlier literature on NBD used Mulliken's terminology of N(normal), V(valence upper state), T(triplet), and R(Rydberg);²³ typical terms are NV for valence states. We combine this with more recent classifications involving the molecular orbitals (MOs) involved.²⁴

Since NBD is an archetypal example of such through space interactions, there are a number of previous studies of its ultraviolet absorption (UV) spectrum,^{25–30} as well as electron impact (EI)^{31,32} and resonant multiphoton ionization (REMPI) spectra.^{29,30} These have been critically reviewed by Robin.¹⁶ Previous theoretical studies

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report analyses with varying levels of sophistication, which are integrated with our work below.

We report the UV + VUV absorption spectra and analyze these by high-level computational methods. Our interpretations are applied to each of the NBD absorption, EI, and REMPI spectra. Much of our work has been devoted to adiabatic theoretical studies, enabling vibrational analyses by both Franck–Condon (FC) and Herzberg–Teller (HT) methods. Correlation of the theoretical envelopes with the experimental spectra allows for the identification of the 0^0 band origins, and hence AEE for the electronic states.

II. METHODS

The NBD sample, CAS registry number 121-46-0 and systematic name bicyclo[2.2.1]hepta-2,5-diene, was a commercial sample (ABCR, assay 99%) and used with no further purification.

A. The VUV absorption spectrum of NBD

This was obtained at room temperature on the AU-UV beamline of the ASTRID2 synchrotron in Aarhus, Denmark, using methods described previously.^{1,33} The overall photoabsorption spectrum is measured in small sections, in order that an appropriate pressure of sample gas can be chosen, depending on the local cross section. For each region, an I₀ scan is first measured with the cell evacuated. The signal recorded from the photomultiplier tube (PMT) is a measure of the light intensity passing through the cell with a measurement time of ~2 s per wavelength step. After filling the cell with the gaseous sample, two It scans of the attenuated light are measured. The cell is evacuated again, and second I₀ is measured; the process is repeated for each region of the spectrum measured. Using the number density (n) obtained through an accurate measurement of the pressure and the path length of the gas cell (l), the absolute photoabsorption cross sections (σ) are then determined using the following Beer-Lambert relationship:

$$I_t = I_0 \exp(-n\sigma l).$$

The full UV + VUV spectrum for NBD was covered by 3633 data points in the range 330 (3.857)-116 nm (10.751 eV). The data points are separated by 0.02 nm for most of the range, 0.1 nm up to 229.8 nm, and 1 nm from 280 nm to higher wavelengths.

The NBD spectrum, shown in Fig. 2, exhibits a fine structure in the 5.5–9 eV region superimposed on a series of broad peaks. One of these broad peaks is best seen by the logarithmic plot in the supplementary material (SM1 as Fig. SM1a).

B. Theoretical methods

As in our recent studies of NBD,¹ cycloheptatriene (CHT),^{2,3} and cyclooctatetraene (COT),^{18,19} we use several computational chemistry suites since no single suite can offer us a complete analysis. These include two Gaussian versions (G-09 and G-16),³⁴ as further discussed in the supplementary material under SM2. Vertical excitation energies (VEEs) were determined by use of the MRD-CI method³⁵ in GAMESS-UK.³⁶ Vibrational features of the



FIG. 2. Electron impact (EI) including electron-energy-loss (EEL) spectra of Frueholz and Doering *et al.* vs the present VUV spectrum for NBD. Several of the peaks above 6.5 eV are also present in the VUV spectrum and, hence, are singlet states overlaying the current triplet ones. Conversely, the fine structure close to 8 eV in the VUV cannot be a Rydberg state.

excited states, both FC and HT, were processed by the Pisa Group software,³⁷⁻³⁹ as implemented in Gaussian.

The equilibrium structures of both valence and Rydberg states were determined by time-dependent density functional theory (TDDFT).^{40–42} The functionals used included the Becke three-parameter Lee–Yang–Parr (B3LYP) hybrid functionals⁴³ and a long-range-corrected version of B3LYP, the Coulomb-attenuating method, CAM-B3LYP.⁴⁴

The TDDFT suite in both G-09 and G-16³⁴ was used to determine the adiabatic excitation energies and their equilibrium structures for several valence states of each symmetry. It is necessary to correct these AEEs to the energy difference at equilibrium between the X^1A_1 and excited state geometries since both suites give the AEE as the energy difference at the excited state structure. This correction is performed automatically in both the FC and HT modules.

C. The principal basis sets

Modern bases deliberately have a wide range of exponents, which can be used for both the valence and Rydberg state determination. We have a clear distinction of purpose between these two types. Various older basis sets contain Gaussian-type orbitals (GTOs), which are strictly valence in type; we used triple-zeta valence with single polarization (TZVP and $6-311G^{**}$).⁴⁵⁻⁴⁸

We calculate Rydberg states by adding very diffuse exponents to TZVP. These have exponents 0.021, 0.008, and 0.0026, positioned on C_7 as the unique atom, and have the same Gaussian exponents for s-, p-, d-, and f-states.

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III. RESULTS AND DISCUSSION

Previous NBD experimental studies include UV absorption,²⁵⁻³⁰ EI,^{31,32} REMPI,^{29,30} and circular dichroism (CD).⁴⁹ Several previous NBD theoretical studies have historical significance^{20–22,30,50–52} but do not contribute to the current level of understanding. Zgierski and Zerbetto⁵³ identified several theoretical criteria for the interpretation of the NBD spectra but were only able to pursue these aims using semi-empirical methods. McDiarmid and co-workers^{29,30} contributed extensive spectroscopic studies of NBD, which were augmented by vertical excitation energy studies of some excitation processes for Rydberg states.

Our VUV absorption spectrum has a much larger energy range when compared with the REMPI study of the lowest Rydberg state by Xing *et al.*²⁹ Indeed, the spectral details exposed in the present VUV are almost identical to those in the REMPI study, as shown in the supplementary material under SM3. Our recent PES study¹ of NBD showed much more vibrational detail than in earlier studies. Overlay of this new PES with its theoretical assignment, as in the supplementary material under SM4, onto our UV + VUV absorption spectrum enables us to pinpoint the origins of several Rydberg states more precisely; this is performed below.

All orbitals and electrons are included in our computations, but we use valence shell numbering for occupied and virtual molecular orbitals (MOs and VMOs) to facilitate comparison with previous work. With the exception of a multiconfigurational second-order perturbation theory (CASPT2) study,³⁰ most previous discussion of the valence MOs has been in a solely π -electron context, although mixing with the σ -MOs was understood.^{20–22} In the current C_{2V} coordinate system, the NBD valence shell is 7a₁ + 5b₁ + 4b₂ + 2a₂ with core MOs 3a₁ + 1b₁ + 2b₂ + 1a₂. In studies where alternative NBD C_{2V} coordinate axes are used, interchanges of both b₁/b₂ MOs and B₁/B₂ states occur.

The degenerate π -MOs and their antibonding conjugates form symmetric (S) and anti-symmetric (A) combinations; these were conventionally termed normal valence (NV₁ to NV₄) in Mulliken's notation²³ or SS (a₁), AS (b₁), A^{*}S^{*} (b₂), and A^{*}A^{*}(a₂) in Hoffmann's terms.²² In practice, the mixing of π - and σ -atomic orbitals is very apparent, as shown in the supplementary material as SM5.

The ground state NBD molecular structure has been determined by both electron diffraction⁵⁴ and microwave Fourier transform spectroscopy.⁵⁵ Our X¹A₁ structures for all bases used are very similar, but since the structural results for several singlet and triplet valence states of NBD are not central to this theoretical analysis, they are shown in the supplementary material as SM6.

A. The NBD theoretical singlet and triplet state manifolds

The principal triplet and singlet state equilibrium structure results are shown in Tables I and II. Excitation energies are corrected to be the differences between the ground and excited state energies, where both are at their respective equilibrium structures; this follows standard spectroscopic practice.

The highest occupied molecular orbital (HOMO) and lowest virtual unoccupied molecular orbital (LUMO) sequence numbers,

TABLE I. Triplet state vertical excitation energies (VEEs) using the MRD-CI and TDDFT methods. The leading configurations have valence shell numbering, with eigenvectors showing the proportions where there are two leading terms. The TDDFT determinations were at the X¹A₁ equilibrium structure. The correlation between the two determinations is VEE_{TDDFT} = 1.032(117) * VEE_{MRDCI} + 0.690(782), where the standard deviations for slope and intercept are in parentheses; the correlation coefficient, adjacent R-square, is 0.818.

	TDDFT		
VEE (eV)	Symmetry	Leading configurations	VEE (eV)
3.056	${}^{3}A_{2}$	$0.891~(5b_15b_2{}^*)-0.210~(7a_13a_2{}^*)$	3.264
3.386	${}^{3}B_{2}$	$0.701~(7a_15b_2{}^*)-0.517~(5b_13a_2{}^*)$	3.711
5.122	${}^{3}A_{1}$	$0.903\;(5{b_17}{b_1}^*)-0.193\;(7{a_18}{a_1}^*)$	7.573
5.909	${}^{3}B_{1}$	$0.920 (5b_18a_1^*)$	7.149
6.081	${}^{3}A_{1}$	$0.915 (5b_1 6b_1^*)$	7.691
6.556	${}^{3}B_{2}$	$0.591~(7a_15b_2{}^*) + 0.671~(5b_13a_2{}^*)$	6.100
6.559	$^{3}A_{2}$	$0.895~(4{b_1}5{b_2}^*)-0.163~(7{a_1}3{a_2}^*)$	6.814
6.585	${}^{3}A_{1}$	$0.806~(5{b_17}{b_1}^*) + 0.441~(7{a_18}{a_1}^*)$	8.222
6.756	${}^{3}B_{1}$	$0.896 (5b_1 11a_1^*)$	7.213
6.841	${}^{3}A_{2}$	$0.541~(7{a_1}3{a_2}^*)-0.215~(5{b_1}6{b_2}^*)$	8.010
6.957	${}^{3}B_{2}$	$0.861~(5b_14{a_2}^*) + 0.262~(7a_16{b_2}^*)$	8.282
7.122	${}^{3}A_{2}$	$0.202~(7a_13a_2^{\ *}) + 0.625~(5b_16b_2^{\ *})$	8.166
7.244	${}^{3}A_{1}$	$0.818\ (7{a_1}8{a_1}^*) + 0.426\ (5{b_1}7{b_1}^*)$	8.234
7.625	${}^{3}B_{1}$	$0.896 (5b_1 10a_1^*)$	7.934
7.698	${}^{3}B_{1}$	$0.896 (5b_19a_1^*)$	8.523
7.784	${}^{3}B_{2}$	$0.854 (7a_16b_2^*) + 0.337(5b_14a_2^*)$	8.941
7.966	${}^{3}B_{1}$	$0.913 (7a_17b_1^*)$	8.904
8.058	${}^{3}A_{2}$	$0.817 (5b_17b_2^*)$	8.729

26 and 27 in the Aufbau energy sequence, apply irrespective of symmetry changes. We note that two pairs of MOs, $4b_2$ and $2a_2$ and $6b_2^*$ and $6b_1^*$, are nearly degenerate, but, in general, we express our discussion in terms of states and their energies; the constituent MOs only occur when denoting the composition of configurations.

Using the TDDFT method with the $6-311G^{**}$ basis set, we find 13 singlet state vertical excitation energies (VEEs) below 8.0 eV. Our state binding energies in C_{2V} are shown in Fig. 3. Where linear combinations of configurations are shown, both eigenvectors are greater than ± 0.1 , with the largest value first. The adiabatic excitation energies all lie lower in energy, a result of structural relaxation, but by variable amounts, as discussed below.

Strictly, the $\pi\pi^*$ excitations, as originally defined,^{22,23} involve promotions from doubly occupied MOs 24 (7a₁) and 25(5b₁) to 26 (5b₂*) and 27(3a₂*), leading to NVn with n = 1–4. However, for NBD, mixing of the local π or π^* with σ or σ^* AOs occurs, leading to additional complexity, as demonstrated in the supplementary material as Fig. SM5. Simple combinations of A*S* and S*S* do not occur; this was first appreciated by Zgierski and Zerbetto⁵³ in their (semi-empirical) spectroscopic parameterized complete neglect of the differential overlap method (CNDO/S) study. They proposed major structural changes occurred between the ground state (S₀) and the lowest two singlet states (S₁ and S₂), but we do not concur with that view, as discussed in the supplementary material as SM7. **TABLE II.** Adiabatic excitation energies (AEEs) and oscillator strengths for singlet states using the CAM-B3LYP functional at the TDDFT level. Comparison with the MRD-CI VEE method, which includes single and double substitutions in the reference configurations. AEE corrections to the TDDFT AEE are included, as described in the supplementary material, SM10. The leading term is shown first, and the orbitals use valence shell numbering. The linear correlation between the two determinations is AEE_{TDPFT} = 1.070(192) * VEE_{MRDCI} - 1.516 (1510), where the standard deviations for slope and intercept are in parentheses. The correlation coefficient, adjacent R-square, is low at 0.750. The VEE correlation between TDDFT and MRD-CI over the tabulated values is poor; the adjacent R-square is 0.684, and the slope is 0.68(13). An extended list of the TDDFT singlet states is shown in the supplementary material.

		TDI	OFT			MRD-CI	EI ³²	Roos et	al. ³⁰
State	Root	Leading configurations	Corrected AEE (eV)	Oscillator strength f(r)	VEE (eV)	VEE (eV)	VEE (eV)	CASSCF (eV)	PT2F (eV)
1^1A_2	NV_1	$5b_15b_2^*$	3.996	0.0	3.545	5.818	5.23, 5.25	8.18	5.28
$1^{1}B_{2}$	NV_2	$7a_15b_2^* + 5b_13a_2^*$	5.762	0.0246	5.756	6.854	5.92, 5.95	9.42	6.20
2^1A_2	NV_3	$7a_13a_2^* + 5b_15b_2^*$	6.216	0.0	7.095	7.795	6.65	9.78	6.48
$1^{1}B_{1}$		$5b_18a_1^* - 5b_19a_1^*$	6.856	0.0098	6.036	7.541			
1^1A_1		$4b_25b_2^*$	6.899	0.0122	6.954	7.244			
2^1B_2	NV_4	$5b_13a_2^* + 7a_15b_2^*$	6.901	0.2005	6.960	7.756	7.50	10.40	7.36
2^1B_1		$2a_25b_2^* - 5b_18a_1^*$	7.272	0.0243	7.133	8.456			
2^1A_1		$5b_16b_1^* + 7a_19a_1^*$	7.582	0.0021	7.841	7.588		7.97	7.49
3^1B_1		$5b_18a_1^* + 2a_25b_2^*$	7.782	0.0011	7.291	8.950			
3^1B_2		$6a_15b_2^*$	7.853	0.0025	7.942	9.068			
3 ¹ A ₂		$5b_15b_2^* + 7a_13a_2^*$	7.929	0.0	7.973	8.499			

B. The electron impact spectral onset for NBD

These have been obtained for NBD by Doering and McDiarmid,^{28,29} Frueholz *et al.*,³¹ and Allan³² and occur in the 2.8–5.5 eV range. The profiles of EI spectra vary considerably with a change in



FIG. 3. The lowest set of singlet state vertical excitation energies using the TDDFT method with a triple-zeta basis set. Where more than one leading term is shown, both eigenvectors are greater than ± 0.1 , with the configuration with the largest value first. Where two states are nearly degenerate, as in ${}^{1}A_{1} + {}^{1}B_{2}$ and ${}^{1}A_{2} + {}^{1}B_{1}$, all close to 7 eV, the energy level is thickened. In some cases, such as the two ${}^{1}A_{1}$ states at 6.96 and 7.84 eV, the nature of the two states is very different; these are 5b_16b_1* and 4b_25b_2*, respectively.

the impact electron energy and scattering angle; these are further discussed in the supplementary material as SM8.^{28,31,32} A comparison of two of these EI spectra with the VUV spectrum is shown in Fig. 2. Considerable differences in the relative peak intensities are exhibited in the two EI methods and the VUV spectrum.

The presence of a strong peak in all the spectra shown in Fig. 2 is indicative of a valence state. The one close to 8.0 eV must be the highest of the $\pi\pi^*$ -states, NV₄. Variations in the EI intensity with scattering angle for the 5.23 and 7.5 eV peaks suggest that these are forbidden relative to the 5.95 and 6.65 eV peaks. Doering and McDiarmid²⁸ summarized the observed electron impact $\pi\pi^*$ VEE as follows: ¹A₂ (5.25, b₁b₂*), ¹B₂ (5.95, a₁b₂*), ¹B₁ (6.65, b₁a₂*), and ¹A₂ (7.5eV, a₁a₂*); those at 5.95 and 6.65 eV overlap with 3s- and 3p-Rydberg states, respectively. These proposals, based on variations in intensity with high and low incident energies, have widely been accepted.

Our vertical excitation triplet state energy results were determined using the MRD-CI method³⁵ in GAMESS-UK;³⁶ all single and double substitutions from the ground state wave-function (at equilibrium) are included. The calculated triplet state VEEs in Table I are relatively close to Allan's spectral maxima³² and are displayed in Fig. 4. These were obtained using a strictly valence TZVP basis set, so no Rydberg state participation can occur through these calculations. Table I also shows the single excitation TDDFT results for the VEE determined at the X¹A₁ structure. Although the two lowest triplet state energies are close between the two methods, there is considerable scatter for the higher values when comparing similar states of the same symmetry. However, the overall correlation has a slope of 1.032 [where the MRD-CI values (x) are lower], and the line apparently goes through the origin since the intercept is smaller than its standard deviation. Since the MRD-CI values use a higher level of correlation than TDDFT, we regard those values as more realistic.



FIG. 4. Assignment of Allan's triplet state manifold³² using vertical excitation energies calculated by the MRD-CI method.

The lowest electronically excited triplet state (T1) is the HOMO \rightarrow LUMO process, generating the ³A₂ state (largely 5b₁5b₂*). This has calculated AEE (using the TDDFT method) and VEE (MRD-CI) of 2.189 and 3.056 eV. These are significantly lower values than the apparent EI onset of 3.47 eV and maximum 3.88 eV, respectively.³² Experimentally, T_1 only appears as a shoulder on the leading edge (lower energy) side of the second triplet state (T_2) . The combined T_1 + T_2 unit, shown in Fig. 4, covers the energy range 2.9-4.5 eV.³¹ Allan suggested that the ${}^{3}B_{2}$ state (T₂) has the onset at 3.76 eV, but the present AEE and VEE are again significantly lower at 3.029 and 3.386 eV, respectively. T₂ shows the vibrational structure³² with apparent frequency 1210 cm^{-1} , which is close to an a_1 mode, which we calculate at 1170 cm⁻¹ for the ³B₂ state. The present symmetry sequence of $T_1 \mbox{ and } T_2$ is in agreement with the previous VEE study.³⁰ The a_1 frequencies for the ${}^{3}A_2$ and ${}^{3}B_2$ states are shown in the supplementary material, Sec. SM9.

The energy region near 5.1 eV in Fig. 4 corresponds to the ${}^{3}A_{1}$ state (T₃) calculated energy at 5.122 eV, but this region also contains the lowest singlet state, the ${}^{1}A_{2}$ (NV₁) state, as discussed below. The two states calculated at 5.909 (${}^{3}B_{1}, 5b_{1}8a_{1}^{*}$) and 6.081 eV (${}^{3}A_{1}, 5b_{1}6b_{1}^{*}$) are assigned to the EI peak at 6.0 eV. The EI region above 6.5 eV, shown in Figs. 2 and 4, is crowded by calculated triplet states. The peak at 7.873 eV both in the VUV and the Doering²⁵ and Frueholz³¹ EI studies is not present in the Allan study³² and hence must be a valence state; this is further discussed below. It is important to recognize that the theoretical basis set used here is purely valence in character and, hence, cannot generate Rydberg states.

C. The onset of optical absorption for NBD

The EI spectra display two well-defined broad peaks at 5.25 and 5.95 eV, which are best seen in the spectrum of Frueholz *et al.*³¹ These have been assigned as ${}^{1}A_{2}$ ($b_{1}b_{2}^{*}$) and ${}^{1}B_{2}$, ($a_{1}b_{2}^{*}$), respectively. The onset of the optical spectrum shows only a shoulder at 5.4 eV.^{25–30} This shoulder has been attributed to the lowest $\pi\pi^{*}$

singlet excitation (NV₁) of norbornadiene. The work of Robin and Kuebler studying NBD at various temperatures established the 0–0 band as 5.608 eV (45 230 cm⁻¹).²⁶ Lightner *et al.*,⁴⁹ using circular dichroism, also observed this state in ethanol solution, as a shoulder at 5.391 eV (230 nm). Its presence in solution indicates a singlet valence state.^{24,50} The adiabatic ionization energy (AIE₁) for NBD is 8.279 eV.¹ The present band with origin 5.608 eV is separated from this by 2.671 eV, which is not suitable for a 3s-Rydberg state.

The NV₁ state symmetry, first established by Zgierski and Zerbetto⁵³ as the ¹A₂ state, was calculated at 5.982 eV (208 nm); this was an important, but fortuitous advance in 1993 since such CI energies were very dependent on the empirical parameters used. Their analysis,⁵³ part of a study on the ground, first, and second excited states (S₀, S₁, and S₂) of NBD, concluded that major structural changes occurred for both excited states. This included lengthening of the C–H bonds and closing of the C₂C₁C₆ angle between the two local planes containing the double bonds ("wings"). This angle, and also one dihedral angle, is incorrectly labeled⁵³ C₄C₃C₅ but is unambiguous from their citation with structural data.⁵⁴

We have computed the equilibrium structures for all three states using the CAM-B3LYP method under TDDFT conditions. We agree that the $C_2C_1C_6$ angle is considerably different in the X^1A_1 state (107.1°) from that in the 1A_2 state (88.2°); however, the C–H bond lengths and CCH angles differ in trivial amounts between the two states. The structural bond length changes involving the H atoms, claimed⁵³ to be 0.37 Å, must be an artifact of their calculations. The details of the structures are given in the supplementary material under SM6 and SM7.

Similarly, in order to determine the vibrational contributions to these electronic states, we sought the AEE using the TDDFT method with the CAM-B3LYP functional and the TZVP basis set. Since the details of these calculations are not central to the current theme, they are shown in Table II but discussed in the supplementary material as SM10.

Although the lowest singlet state of NBD is generally agreed to be the 1^1A_2 state, this is optically forbidden under Franck–Condon rules.^{20–22,30,50–52} This dilemma is resolved by the presence of non-symmetric vibrations, which are allowed under Herzberg–Teller (HT) rules as shown below.

D. The Herzberg-Teller (HT) profile of the 1¹A₂ state

Since the theoretical behavior of this 1^1A_2 state is typical of HT states, it is treated in more detail than some succeeding states. The standard convention for vibrational mode sequence for NBD in C_{2V} symmetry is $a_1: 1-12$, $a_2: 13-20$, $b_1: 21-29$, $b_2: 30-39$. While we adhere to this convention, we note that the most prominent modes are those of lowest frequency and, hence, highest sequence numbers. Hence, we retain both systems, noting that when "mode" is used we refer to standard usage; in contrast, "sequence number" refers to the ascending frequency sequence. All such results in G-09 and G-16 use this convention. Full sets of harmonic frequencies and the G-09/G-16 labeling system for the lower excited states are in the supplementary material at SM11. In this section and all following sections, FC and/or HT intensities are given as the molar absorption coefficient (dm³ mol⁻¹ cm⁻¹). The "position" is relative energy from the 0–0 (0⁰) calculated band.

The HT analysis for NV₁ shows that several sequences of binary and tertiary combination bands occur; these have maximum intensity distant from the generally low intensity of the 0-0 position. One series containing combinations of modes 12 $(12a_1, 368 \text{ cm}^{-1})$ and 39 (10b₂, 306 cm⁻¹) has its maximum with the tenth member at 4465 cm⁻¹ from the 0-0 band. The most intense bands of the HT profile, shown in Table III, consist of several non-symmetric vibrations combined with a $12a_1^n$ carrier, where *n* is the number of quanta; generally, n is greater than 10 for significant intensity to occur, while the maximum intensity occurs for n = 18. Other vibrations, having minor intensity in Table III, also involve a1 modes as carriers. Since both the HT (and FC) vibrational states generate a large number of vibrational states, we have limited the tables to combinations of two simultaneously excited modes, while all states are included in Fig. 5. HT bands are generally relatively weak in comparison with FC bands.

The vibrational contributions giving the most intense binary combination series are 12^939^1 through to $12^{30}39^1$ with maximum intensity at $12^{19}39^1$. The most intense HT band overall combines these with mode 5 (a₁, 1431 cm⁻¹). This corresponding triple combination series has maximum intensity for $5^112^{19}39^1$. In addition to modes 12, 39, and 5, shown in Table III, other binary modes participating in the HT calculations for the ${}^{1}A_2$ state, in conjunction with 12^n , are 10^1 , 10^3 , 10^5 , 10^7 (all b₂), 19^1 , 18^1 , 17^1 (all a₂), 38^1 , 38^3 , 37^1 , 34^1 , 33^1 (all b₂), and 28^1 , 26^1 , 25^1 , 24^1 , 23^1 (all b₁), where the quanta are given in superscripts.

The UV absorption, shown in Fig. 2, shows a clearly sigmoidal rise of signal strength toward the onset of the 5.95 eV Rydberg state sequence (as discussed below), and there is an overlap of the vibrations for the HT band with the following Rydberg state. The present HT calculations suggest that absorption starts significantly below 4.5 eV. The calculated origin for the 0–0 transition is 3.873 eV (31241 cm^{-1}). As expected for a formally forbidden transition, the 1^1A_2 (S₁) state shows no Franck–Condon vibrational bands. The

TABLE III. The most intense $1^{1}A_{2}$ state Herzberg–Teller modes determined using the CAM-B3LYP functional in the TDDFT method. The complete envelope contains 880 vibrational states for binary and tertiary combinations; an extended list is shown in the supplementary material as SM8. The terminology 12^{16} ; 39^{11} refers to a binary combination mode, where 16 quanta of sequence mode 12 are combined with 1 quantum of mode 39; in conventional terms, this is mode $12^{16} + 39^{1}$.

Mode	Position (cm^{-1})	Intensity
12^16;39^1	6191	86
12^17;39^1	6558	101
12^18;39^1	6926	112
12^19;39^1	7294	115
12^20;39^1	7662	110
12^21;39^1	8030	99
12^22;39^1	8397	84
5^1;12^16;39^1	7621	95
5^1;12^17;39^1	7989	111
5^1;12^18;39^1	8357	121
5^1;12^19;39^1	8725	123
5^1;12^20;39^1	9092	117
5^1;12^21;39^1	9460	104
5^1;12^2;39^1	9828	87

J. Chem. Phys. **155**, 034308 (2021); doi: 10.1063/5.0053962 © Author(s) 2021 extensive HT vibrational profile obtained is exemplified in Fig. 5. In order to correlate the observed sigmoidal rise of the experimental spectrum with the calculated HT profile, we have increased the calculated HT energies by 0.335 eV to fit the electron impact maximum at 5.23 eV.

Not all the bands involve combinations of a_1+b_2 vibrations. Various combinations of HT bands occur, mainly from the low- and mid-frequency fundamentals. The high frequency C–H stretching modes are absent. Since all calculations of the S₁ state (¹A₂) are carried out independently from those for the S₂ (¹B₂) and S₄ (¹B₁) states below, there is no possibility of b₂ modes, inducing intensity in S₁ by stealing from either the S₀ to S₂ or S₀ to S₄ transitions. Thus, it is not essential to postulate the stealing proposed by Zgierski and Zerbetto when considered in the HT as opposed to FC context.⁵³ Furthermore, all the modes showing HT activity above for the ¹A₂ state have significantly lower frequencies than 1600 cm⁻¹, C=C (b₂ stretching vibration) previously thought⁵³ to be prominent in the intensity of S₁.

E. The NV₂ valence state: The 4.5-6.5 eV range

The sigmoid rise in absorption here carries a set of sharp bands on top of rising intensity. This indicates the presence of at least two states in this region. Zgierski and Zerbetto⁵³ denied the presence of a Rydberg state in their CNDO/S study of the 5.827 eV (47 000 cm⁻¹) region and claimed that it contained only a ¹B₂ valence state; however, the nature of their calculations would not have detected a Rydberg state. This becomes important below since their ¹B₂ valence state frequencies were later used in the REMPI study of a Rydberg state by Xing *et al.*²⁹ Doering and McDiarmid²⁸ suggested that the NV₂ (¹B₂) valence state is the EI peak



FIG. 5. The onset of absorption for NBD following Frueholz *et al.* in the 4.0–6.5 eV region; the electron-energy-loss spectrum is at a scattering angle of 20° and 50 eV incident energy. The Herzberg–Teller (HT) profile is superimposed with vibrational bands having half-width at half-maximum of 10 (red), 70 (blue), and 400 cm⁻¹(magenta). All bandwidths are expressed as HWHM, but with varying widths. The intensity of the HT peak is scaled by a factor of two relative to Table III.

at 5.95 eV. This valence state was observed by Robin²⁷ in the presence of a high pressure of helium gas, which selectively reduces the cross section of the Rydberg state.²⁷ In summary, a number of UV + VUV absorption, REMPI, and theoretical calculations have all agreed that this NV₂ valence state has a Rydberg state superimposed.^{20–22,29,30,50–52}

The high pressure VUV absorption spectrum²⁷ of Robin *et al.* is combined with both the EI spectrum of Frueholz *et al.* and the present Franck–Condon profile in Fig. 6. Although the intensities of the FC profile shown differ from the experiment considerably, the principal local maxima from the experiment are close to the red set of calculated peaks, which have Half-Width at Half-Maximum (HWHM) of 10 cm⁻¹; this represents a close correlation that enables assignment of the 0–0 band from the calculated value. There is generally good agreement between the FC profile and the high pressure VUV spectrum, and this appears to establish the NV₂ state origin as 5.730 eV for the ¹B₂ state.

Figure 6 shows that there is considerable intensity in the 0–0 band. Several of the prominent modes excited under Franck–Condon conditions and given in Table IV show that all a_1 modes except mode 1 contribute to the profile.

The mode 12^n sequence where n is the number of quanta has an unexpected dip in the intensity of the n = 3 member, compared with those with n = 2 and 4. We note that mode 11 is almost exactly double the frequency of mode 12. As a consequence, 12^3 is almost degenerate with the binary combination $11^{1}12^{1}$; both have much lower intensities than other members of their sequences. Other accidental near-degeneracies at the harmonic level are mode 9 with modes 11 + 12 and mode 7 with modes 10 + 11. Some of these near-degenerate levels can be expected to show significant interactions in anharmonic calculations, but these were not further pursued in this work. Among the more intense binary



FIG. 6. The NV₂ band (blue) exposed from the VUV spectrum by 136 atm of helium gas and from the electron impact spectrum (black) at a scattering angle of 20° and 50 eV incident energy. The calculated Franck–Condon profile with the energy scale increased by 0.1288 eV is superimposed in red. The 0–0 band is at 5.731 eV (46 206 cm⁻¹).

TABLE IV. The most intense 1 ¹ B ₂ state Franck–Condon modes determined using the
CAM-B3LYP method. The complete envelope contains 880 vibrational states from
binary through to quintet combinations. The most intense band 29^1;1^1 refers to
a binary combination, where 1 quantum of sequence mode 29 is combined with 1
quantum of mode 1, from Table V; in conventional terms, this is mode $29^1 + 1^1$.
The position is relative energy from the 0-0 calculated band; intensity is the molar
absorption coefficient ($dm^3 mol^{-1} cm^{-1}$) throughout the text and tables of this study.

Mode	Position (cm^{-1})	Intensity	
1^1	260	6871	
1^2	520	2745	
21^1	1115	4385	
21^1;1^1	1375	4718	
29^1	1496	3609	
28^1;1^1	1699	2451	
29^1;1^1	1756	7287	
21^2;1^1	2490	2626	
29^1;21^1	2611	4239	
29^1;21^1;1^1	2871	4755	
29^1;1^1	3131	2234	
29^1;28^1;1^1	3194	2402	
29^2;1^1	3252	3598	
29^1;21^2;1^1	3985	2430	

combinations are $12^n 20^2$; an a_2^2 series; and a_1 combinations $11^1 12^n$, $10^1 12^n$, and $7^1 12^n$.

F. The ${}^{1}B_{1}$ 3s-Rydberg state and its vibrational properties: The 4.5–6.5 eV range

The most detailed previous study of the Rydberg state is the resonant-enhanced multiphoton ionization [(2 + 1) REMPI] of Xing *et al.*²⁹ with additional features shown by Roos *et al.*³⁰

Both UV absorption and two REMPI studies led to the assignment of the series of bands between 5.4 and 6.2 eV as the 5b₁3s-Rydberg state; these conclusions preceded our high-resolution PES study and were based on the largely unresolved broad PES band known at the time, a successful prediction. A comparison of the VUV absorption spectral results of this state with the X^2B_1 ionic state profile is shown in Fig. 7.

The experimental PES X^2B_1 ionic state (in red) can be superimposed upon 5.5–6.5 eV of the VUV spectral region (in blue), as shown in Fig. 7, by a shift of 2.837 eV to lower energy. Application of the Rydberg energy equation leads to a value of $n - \delta = 2.199$; this implies the principal quantum number n = 3 with quantum defect $\delta = 0.801$. The previously reported¹ individual calculated vibrational FC peaks, with a narrow (10 cm⁻¹) linewidth, generally offer a good interpretation of the apparent underlying broad region below the Rydberg state, as shown in Fig. 8. Some very weak peaks lying below the Rydberg state, which do not occur in either the ionic state itself or the theoretical profile, which describes it, can be attributed to the underlying valence state structure.

Variable temperature studies²⁶ suggested that the 0–0 band for the Rydberg state was 5.608 eV (45 230 cm⁻¹). There is a direct correspondence between the most intense set of vibrations in the theoretical envelope, with frequency 360 cm⁻¹, and our calculated lowest $a_1 \mod 12$ (381 cm⁻¹); the calculated values (MP2 was used here) are 5.8% high, which were scaled to perform a complete fit. Weak



FIG. 7. Superposition of the experimental photoelectron spectrum (in red), shifted by 2.8374 eV to lower energy, and the VUV state (in blue) in the 5.5 and 6.5 eV regions. It is clear that the VUV region contains a number of weak peaks, which are not present in the ionic state; these indicate the presence of an underlying valence state.



FIG. 8. Comparison of the background subtracted VUV spectrum of the 3s-Rydberg state ($^{1}B_{1}$) of NBD (in black) with the hot band structure for the $X^{2}B_{1}$ state (in red). The half-width at half-maximum (HWHM) of the theoretical study is 10 cm⁻¹.

bands attributed to the underlying ${}^{1}B_{2}$ state have no ${}^{2}B_{1}$ state counterpart, and this supports the ${}^{1}B_{2}$ assigned lines described above. The REMPI study of this Rydberg state by Xing *et al.*²⁹ assigns a considerable number of line positions for both hot and cold bands. The principal cold band series identified by Xing *et al.*²⁹ was the main PES sequence 12_{0}^{n} where the quanta are n = 1-8; clearly, this fits with both our PES analysis and the current Rydberg state study.

The principal focus of the paper of Roos *et al.*³⁰ was Rydberg states based on limiting ionic states ${}^{2}B_{1}$ and ${}^{2}A_{1}$; this included

the observed ${}^{1}B_{1}$ 5b₁3s-state and numerous higher Rydberg states. Their principal method was CASSCF, which was limited to the four π -orbitals discussed above, one from each irreducible representation, but was augmented for the Rydberg state studies. Their work had a second level of approach based on the multi-configurational second-order perturbation theory method (CASPT2, the PT2F variant). The latter gives very different VEE from the CASSCF values and is much closer to experimental energy values. Our TDDFT corrected results are similar to those of Roos *et al.*,³⁰ but NV₂ is low by about 0.1288 eV when compared with the EI results.

Rydberg states present very similar vibrational envelopes to those of the limiting ionic state for the series; our previous analysis¹ of the ²B₁ ionic state shows that a_1 modes, especially 4, 5, 7, 9, and 10, occur in binary combinations with mode 11. Some of these occur in quaternary combinations, and some occur in combination with even quanta of non-symmetric modes (e.g., $20a_2^2$). Herzberg–Teller calculations were performed, similar to those for the ¹A₂ state above; the harmonic frequencies and labeling system used by G-09/G-16 are shown in the supplementary material at

TABLE V. The most intense $1^{1}B_{2}$ state Herzberg–Teller modes determined using the CAM-B3LYP method. The most intense band 29^{1} ; 1^{1} refers to a binary combination, where 1 quantum of sequence mode 29 is combined with 1 quantum of mode 1, from Table V; in conventional terms, this is mode $29^{1} + 1^{1}$. The position is relative energy from the 0–0 calculated band; intensity is the molar absorption coefficient (dm³ mol⁻¹ cm⁻¹) throughout the text and tables of this study. The complete envelope contains 1232 vibrational states for binary through to quintet combinations.

Mode	Position (cm^{-1})	Intensity	Symmetry
2^1	325	18	A ₂
3^1	465	2	B_2
7^{1}	546	4	A_2
2^1;1^1	586	29	B_1
9 [^] 1	679	19	B_2
10^{1}	705	1	A_2
9^1;1^1	941	34	B_2
2^3	975	2	A_2
19^1	1049	36	A_2
17^1;1^1	1258	23	B_2
19^1;1^1	1310	41	A_2
24^1;1^1	1505	24	A_2
19^1;1^2	1571	17	A_2
21^1;2^1;1^1	1701	22	A_2
21^1;9^1	1794	23	B_2
29^1;2^1	1821	18	A_2
21^1;9^1;1^1	2055	23	B_2
29^1;2^1;1^1	2082	31	A_2
21^1;19^1	2163	29	A_2
29^1;9^1	2175	19	B_2
21^1;19^1;1^1	2425	44	B_1
29^1;9^1;1^1	2436	36	B_2
29^1;19^1	2544	37	A_2
24^1;21^1;1^1	2620	19	A_2
29^1;17^1;1^1	2754	25	B_2
29^1;19^1;1^1	2805	44	B_1
29^1;24^1;1^1	3001	26	A ₂



FIG. 9. The 7.1–8.3 eV range with the assignment of the ¹B₁ state. Although the intensities of the local peaks vary erratically, their separation is more systematic, thereby enabling the ¹B₁ state, which is dominated by groups of frequencies, to be assigned. The linewidth for the theoretical spectrum, in red, is 10 cm⁻¹.

SM11. The most intense HT vibrations for the $1^{1}B_{2}$ state, depicted in Table V, show that modes of all symmetries occur in the (total 1200) combination bands. However, the HT intensities are only ~1% when compared with the FC set.

C. NBD absorption in the 7.1–8.1 eV range: The NV_3 1B_1 state

The VUV envelope shows considerable changes in cross section here, but with no simple pattern being evident. We assign this to the two highest valence states, NV₃ and NV₄, with the underlying additional valence and/or Rydberg structure. Two calculated valence states for this region are the lowest $1^{1}B_{1}$ with the $2^{1}B_{2}$ state lying slightly higher in energy. NV₃ has two leading configurations in the antisymmetric combination, but very unequal proportions, as follows: $0.690 \ 5b_{1}8a_{1}^{*} (25 \rightarrow 27) - 0.105 \ 5b_{1}9a_{1}^{*} (25 \rightarrow 30)$.

We assign the lower energy region to the 1^1B_1 state to a calculated origin at 6.856 eV (55 295 cm⁻¹) with f(r) 0.0098, as shown in Fig. 9; it shows an almost Gaussian set of bands that become more complex with the number of vibrational satellites as the energy increases. We do not wish to imply that this calculated state is the sole participant in the 7.1–8.1 eV region. We believe that (possibly several) other states lie under the envelope here. However, the *positions* of the calculated multiplet and the observed main peaks do give a relatively close correlation, and this is evidence for the assignment.

A summary of the principal vibrations is shown in Table VI. The ${}^{1}B_{1}$ state is dominated by the FC vibrations, while the fundamentals are dominated by the lowest a_{1} frequency (mode 12, 363 cm⁻¹), but with low intensity. Not all the a_{1} fundamentals are active. Mode (12 a_{1}) forms binary combinations with several other fundamentals, including modes 11 (779), 10 (825), 8 (945), 7 (1085), 4(1586), 3 (2706) 2 (2925), and 1 (2993 cm⁻¹). The binary combinations are generally 100 times more intense than the fundamentals.

Position (cm^{-1}) Position (cm^{-1}) Mode FC intensity HT intensity Mode FC intensity HT intensity 2^1;1^1 0 0-0 46 5 1105 44 2^1;1^3 1^{1} 363 275 16 1468 80 1^{2} 2^1;1^4 725 723 47 1831 93 1^3 2^1;1^5 1088 1785 76 2194 141 1^{4} 2^1;1^6 1451 2411 84 2556 112 1^5 2^1;1^8 1814 2258 112 3282 57 1^{6} 7^1;1^7 17 2177 2564 89 3318 1^{7} 9^1;1^2 53 2539 1742 1550 24 $1^{\circ}8$ 9^1;1^4 891 42 2902 2276 109 2^{1} 2 9^1;1^5 380 0 3463 71 7^{1} 779 12 0 3001 9^1;1^6 154 9^{1} 9^1;2^3;1^3 825 42 1 2293 70 11^{1} 9^1;2^1;1^4 0 3 80 860 2655 15^{1} 9^1;2^1;1^5 0 945 3018 127 4 7^{2} 16^{1} 0 970 1 1557 1 0 20^{1} 7^1;1^1 1078 0 3 1141 66 21^{1} 0 7^1;1^2 2 249 1085 1504 27^{1} 7^1;1^3 1288 1 1867 472 29^1 2^2;1^4 2210 1418 1 14 $31^{\wedge}1$ 22 2573 2^2;1^5 13 1586 1 $31^{\circ}2$ 3171 4 1 2986 3^2;1^6 41 $33^{\wedge}1$ 2 2592 7^1;1^5 818 2706 $37^{\wedge}1$ 7^1;1^6 2925 1 2955 688 38^1 2 2993

TABLE VI. Franck–Condon and Herzberg–Teller modes close to the origin and more intense combination bands for the ¹B₁ state.

Position (cm^{-1})	Excitation	FC intensity	Position (cm ⁻¹)	Excitation	FC intensity
0	0	68 260	1120	21^1	4 855
192	1^{2}	10 100	1122	7^1;1^6	444
346	$2^{\wedge}1$	75 880	1154	$4^{2};1^{2}$	463
383	$1^{\wedge}4$	3 4 9 0	1168	$5^{2};1^{2}$	426
538	$2^{1};1^{2}$	17 020	1172	6 [^] 2;1 [^] 2	281
548	7^{1}	15 550	1231	$2^{3};1^{2}$	1 243
575	1^{6}	1 337	1268	$2^{2};1^{6}$	892
693	2^{2}	30 740	1305	$2^{1};1^{10}$	333
709	$8^{\wedge}1$	1 586	1314	$7^{1};1^{8}$	174
728	<i>9</i> ^ <i>1</i>	10 230	1338	27^{1}	3 883
730	$2^{1};1^{4}$	5714	1345	$4^{2};1^{4}$	156
739	$7^{1};1^{2}$	3 507	1359	$5^{2};1^{4}$	149
767	1^{8}	537	1385	$2^{\wedge}4$	631
842	3^{2}	874	1419	$8^{\wedge}2$	133
884	$2^{2};1^{2}$	6973	1422	$2^{3};1^{4}$	427
901	$8^{1};1^{2}$	361	1448	$28^{\wedge}1$	38 450
920	$16^{\wedge}1$	123	1457	<i>9</i> ^ <i>2</i>	220
920	9 [^] 1;1 [^] 2	1 472	1459	$2^{2};1^{8}$	352
921	$2^{1};1^{6}$	2 1 2 8	1496	$2^{1};1^{12}$	136
931	$7^{1};1^{4}$	1 185	1577	$2^{4};1^{2}$	141
958	1^{10}	221	1614	$2^{3};1^{6}$	163
962	$4^{\wedge}2$	2 066	1643	7^3	158
976	5^{2}	1817	1651	$2^{2};1^{10}$	143
981	$6^{\wedge}2$	1 238	1651	$12^{^{^{^{^{^{^{^{^{^{^{^{^{^{^{^{}}}}}}}}$	508

TABLE VII. B3LYP CIS $2^{1}B_{2}$ state is root 6 of the TDDFT sequence, with f(r) = 0.1956, Franck–Condon profile. The 0–0 band is at 7.031 eV.

TABLE VIII. B3LYP CIS 2^1B_2 state is root 6 of the TDDFT sequence, with f(r) = 0.1956, Herzberg–Teller profile. The symmetries of the participating vibrations are indicated. The 0–0 band is at 7.031 eV.

Position (cm^{-1})	Excitation	HT intensity	Position (cm^{-1})	Excitation	HT intensity
0	0	128	1510	$22^{1};2^{1}(b2 + a1)$	1474
346	$2^{1}(a1)$	208	1547	$22^{1};1^{4}(b2 + 4a2)$	116
421	3^1(b2)	437	1548	$13^{1};2^{2}(b2 + 2a1)$	180
442	$2^{1};1^{1}(a1 + a2)$	63	1654	$26^{1};2^{1}(b2 + a1)$	1964
693	$2^{2}(2a1)$	125	1690	$26^{1};1^{4}(b2 + 4a2)$	149
759	10^{1} (b2)	1093	1719	$18^{1};2^{2}(a2 + 2a1)$	120
767	$3^{1};2^{1}(a1 + b2)$	415	1746	$19^{1};2^{2}(b2 + 2a2)$	1516
823	$11^{1}(a2)$	59	1798	$10^{1};2^{3}(b2 + 3a1)$	133
856	13^1(b2)	262	2129	$20^{1};19^{1}(a1 + b2)$	1501
951	$10^{1};1^{2}(b1 + 2a2)$	251	2476	$20^{1};19^{1};2^{1}(a1 + b2 + a1)$	1541
1026	18^1(a2)	307	2501	$28^{1};19^{1}(a1 + b2)$	2163
1053	19 [^] 1(b2)	2487	2553	$28^{1};10^{1};2^{1}(a1 + b1 + a1)$	1532
1076	$20^{1}(a1)$	70	2612	$28^{1};22^{1}(a1 + b2)$	1315
1105	$10^{1};2^{1}(b1 + a1)$	1735	2730	$26^{1};20^{1};2^{1}(b2 + a1 + a1)$	1175
1113	$3^{1};2^{2}(b2 + 2a1)$	264	2755	$28^{1};26^{1}(a1 + b2)$	1693
1164	$22^{1}(b2)$	1516	2847	$28^{1};19^{1};2^{1}(a1 + b2 + a1)$	2117
1202	23^1(a2)	188	2896	28 [^] 2(2a1)	356
1202	$13^{1};2^{1}(b2 + a1)$	438	2958	$28^{1};22^{1};2^{1}(a1 + b2 + a1)$	1322
1245	$19^{1};1^{2}(b2 + 2a2)$	565	3086	32^1(b2)	752
1307	$26^{1}(b2)$	1957	3101	$28^{1};26^{1};2^{1}(a1 + b2 + a1)$	1756
1356	$22^{1};1^{2}(b2 + 2a2)$	343	3146	36 [^] 1(b2)	929
1373	$18^{1};2^{1}(a2 + a1)$	317	3193	$28^{1};19^{1};2^{2}(a1 + b2 + 2a1)$	1384
1399	$19^{1};2^{1}(b2 + a1)$	2355	3278	$32^{1};1^{2}(b2 + 2a2)$	172
1436	$19^{1};1^{4}(b2 + 4a1)$	192	3433	$32^{1};2^{1}(b2 + a1)$	1131
1448	28^1(a1)	354	3577	$28^{1};20^{1};19^{1}(a1 + a1 + b2)$	1221
1452	$10^{1};2^{2}(b2 + 2a1)$	728	3924	$28^{1};20^{1};19^{1};2^{1}(a1 + a1 + b2 + a1)$	1295
1499	$26^{1};1^{2}(b2 + 2a2)$	441	4343	28 ³ (3a1)	160

There are again significant Herzberg–Teller contributions, especially in the binary combinations. The HT bands occur in binary and ternary combination bands. They have a simple effect of moving the center of gravity of the combined set to slightly higher energy.

H. NBD absorption in the 7.1–8.1 eV range: The $2^{1}B_{2}$ calculated state NV₄

The $2^{1}B_{2}$ calculated state, NV₄, has an even more complex wave-function, in contrast to that expected from the early analyses, as follows: 0.526 ($5b_{1}3a_{2}^{*}$) + 0.133 ($5b_{1}4a_{2}^{*}$) + 0.232 ($5a_{1}5b_{2}^{*}$) + 0.303 ($7a_{1}5b_{2}^{*}$).

This state, as with the $1^{1}B_{2}$ state, shows the presence of an imaginary frequency under C_{2V} conditions; as previously, this is eliminated by projection with the largest overlap with a ground state frequency. The labeling for this process is shown in the supplementary material under SM11. The onset and a selection of the fundamentals together with the most intense higher frequency bands for the Franck–Condon and Herzberg–Teller modes are shown in Tables VII and VIII, respectively. This band, shown in Fig. 10, has a very complex envelope.

In the Franck–Condon vibrations, the 0–0 band is intense; the most prominent a_1 fundamentals in the Franck–Condon modes have sequence numbers 2, 7, 8, 9, 16, 21, and 28, with frequencies 346, 548, 709, 728, 920, and 1448 cm⁻¹.

The Herzberg–Teller vibrations for the 2^1B_2 state shown in Table VIII are much more complex, and more vibrational states are shown in Table VIII to represent this situation. Most of the states contain b_2 modes, but all other symmetries are represented. Nearly



FIG. 10. The NBD VUV spectrum from 7.7 to 8.4 eV with the Franck–Condon profile superimposed. Although many of the peaks have irregular intensity, they show spacing consistent with the 2^1B_2 calculated state. The full set of FC peaks shown is the summation of two separate sequences from two computations. Both used the same Half-Width at Half-Maximum (HWHM) of 10 cm⁻¹. This leads to apparent discontinuities in the curves in some places. As with other relatively high energy bands here, the observed spectrum and its fit are not expected to cover all the peaks observed. There are clearly a number of absorption bands in these regions of energy.

10 000 vibrations containing up to five components were obtained, far more than for any other state studied. The sequence numbers in Table VIII also contain the symmetries of these contributors, and the mode sequence can be derived from the data given. The most intense fundamentals have b_2 symmetry; the most notable of these have frequencies 421, 759, 856, and 1053 cm⁻¹, but other vibrational state symmetries also occur, such as 823, 1026 and 1202 (a₂) and 1448 cm⁻¹ (a₁). Generally, the binary and higher combinations occur at higher frequencies as expected, but their effect is to move the center of gravity of the combined FC + HT bands to higher energy. The overall band is clearly more than 1 eV wide.

IV. CONCLUSIONS

We have obtained the vacuum ultraviolet absorption spectra for norbornadiene. The spectrum shows a wide energy range where the vibrational structure is present. A direct comparison of the earlier UV absorption, REMPI spectrum, and our VUV absorption, shown in the supplementary material, SM3, demonstrates that our VUV absorption spectrum shows all the features observed in the REMPI spectra, together with the additional sub-structure. This is also present in the high-resolution photoelectron spectrum, but is not apparent in the REMPI spectra, and hence provides evidence of an underlying valence state, NV₂.

The deceptively simple analysis applied to the NBD UV absorption spectra by previous studies is replaced by a more complex analysis, where many more $\sigma + \pi$ interactions occur. The TDDFT method is very successful in obtaining the equilibrium structure for a number of single excitation singlet and triplet states. The triplet state manifold was compared with electron impact spectra *en passant*, and that was the lead into the detailed study of the singlet manifolds. While the equilibrium structures for a number of singlet states were obtained with C_{2V} symmetry, it was found that several large basis sets showed a frequent preference for lower symmetry singlet states. This was attributed to relatively diffuse p-functions on the H atoms since the omission of these led to the desired C_{2V} structures, with few saddle points occurring.

The Franck–Condon and Herzberg–Teller vibrational structures for several low-lying valence states were determined and superimposed on the absorption spectrum. By choosing "best fits," the calculated 0–0 band energies were corrected to estimates of the adiabatic excitation energies for each of the NV₁ to NV₄ traditional valence states. For most states investigated, the HT intensities were only circa 10% of the FC ones. The presence of the ¹A₂ state in the optical absorption spectrum is attributed to HT vibrations.

We have also obtained the VUV absorption spectrum of quadricyclane; it is fundamentally different with a small vibrational structure, except that attributed to Rydberg states. The absence of a chromophore in QC is presumably responsible for this. Detailed analysis of the QC VUV spectrum is deferred, awaiting a further experimental study.

SUPPLEMENTARY MATERIAL

See the supplementary material for additional information on each of the following: (1) the cleaned NBD and QC spectral onsets (Figs. SM1a and SM1b); (2) theoretical methods: additional material; 28 June 2024 15:04:34

(3) the comparison of the details from the present VUV absorption spectrum with previous UV and REMPI data; (4) the experimental photoelectron spectral onset of NBD with the theoretical Franck–Condon profile superimposed; (5) the mixing of p- and s-atomic orbitals (AOs) in NBD; (6) molecular structures for some singlet and triplet valence states of NBD; (7) excited state C–H bond structural changes; (8) electron impact variations; (9) the a₁ frequencies for the ${}^{3}A_{2}$ and ${}^{3}B_{2}$ states; (10) adiabatic excitation energies (AEEs) and oscillator strengths at the TDDFT level; (11) the harmonic frequencies for the excited states; (12) the 7.5–8.5 eV region: a potential alternative assignment; and (13) the extended list of singlet state energies using the TDDFT method.

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DATA AVAILABILITY

The data that support the findings of this study are available from the corresponding author upon reasonable request.

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