

Self-Consistent Implementation of Kohn–Sham Adiabatic Connection Models with Improved Treatment of the Strong-Interaction Limit

Szymon Śmiga,* Fabio Della Sala, Paola Gori-Giorgi, and Eduardo Fabiano



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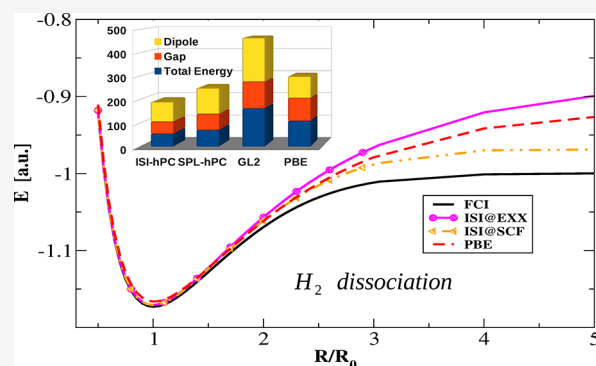
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ABSTRACT: Adiabatic connection models (ACMs), which interpolate between the limits of weak and strong interaction, are powerful tools to build accurate exchange–correlation functionals. If the exact weak-interaction expansion from the second-order perturbation theory is included, a self-consistent implementation of these functionals is challenging and still absent in the literature. In this work, we fill this gap by presenting a fully self-consistent-field (SCF) implementation of some popular ACM functionals. While using second-order perturbation theory at weak interactions, we have also introduced new generalized gradient approximations (GGAs), beyond the usual point-charge-plus-continuum model, for the first two leading terms at strong interactions, which are crucial to ensure robustness and reliability. We then assess the SCF–ACM functionals for molecular systems and for prototypical strong-correlation problems. We find that they perform well for both the total energy and the electronic density and that the impact of SCF orbitals is directly connected to the accuracy of the ACM functional form. For the H_2 dissociation, the SCF–ACM functionals yield significant improvements with respect to standard functionals also thanks to the use of the new GGAs for the strong-coupling functionals.



INTRODUCTION

Kohn–Sham (KS)¹ density functional theory (DFT) is the most used electronic structure computational approach for molecular and solid-state systems.^{2–4} Its accuracy depends on the choice of the approximation for the exchange–correlation (XC) functional^{5–7} which, at the highest-rung of the Jacob’s ladder,⁸ involves all the occupied and virtual KS orbitals as well as the eigenvalues. Then, the XC approximation is no more an explicit functional of the density and, to stay within the pure KS formalism, the optimized effective potential (OEP) method^{9,10} must be employed. Early OEP approaches included exact-exchange (EXX) and approximated the correlation using the second-order Görling–Levy perturbation theory (GL2).¹¹ However, this led to a large overestimation of correlation effects and to convergence problems.^{12–18}

Actually two different main approaches have been explored to solve this issue: going beyond the second-order approximation^{19–26} or using a semicanonical transformation.^{12,13,18} Another possible path is the adiabatic connection (AC) formalism^{27–29} which is a general, powerful tool for the development of XC functionals. For several decades, it has been used to justify the introduction of hybrid^{30–32} and double hybrid (DH) functionals^{33–35} and successively it has been directly employed to construct high-level XC functionals based on AC models (ACM) interpolating between known limits of

the AC integrand.^{36–43} Recently, it has also been employed in the context of the Hartree–Fock (HF) theory^{44,45} to develop corrections to the Møller–Plesset perturbation series.⁴⁶

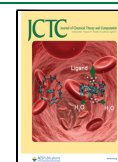
The XC functionals based on ACMs have the general form

$$E_{xc}^{ACM} = f^{ACM}(\mathbf{W}) = \int_0^1 W_\lambda^{ACM}(\mathbf{W}) d\lambda \quad (1)$$

where $\mathbf{W} = (W_0, W'_0, W_\infty, W'_\infty)$, with $W_0 = E_x$ being the exact exchange energy, $W'_0 = 2E_c^{GL2}$ being twice the GL2 correlation energy,¹¹ and W_∞ and W'_∞ being the indirect part of the minimum expectation value of the electron–electron repulsion for a given density and the potential energy of coupled zero-point oscillations around this minimum, respectively.^{39,47} The model W_λ^{ACM} is designed to mimic the exact but unknown W_λ , in particular by considering the known asymptotic expansions^{11,39,40,47}

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$$W_{\lambda \rightarrow 0} \sim W_0 + \lambda W'_0 + \dots \quad (2)$$

$$W_{\lambda \rightarrow \infty} \sim W_\infty + \frac{1}{\sqrt{\lambda}} W'_\infty + \dots \quad (3)$$

In recent years, several ACMs have been tested for various chemical applications showing promising results,^{48,49} especially in the description of non-covalent interactions.^{46,50} However, most of these recent studies have been performed within the HF–AC framework, that is, as post-HF calculations. Conversely, little attention has been devoted to DFT-based ACM functionals. The main reason for this is that in the HF case, the ACM is applied on top of the HF ground state,^{44,45} which is a simple and well defined reference; on the contrary, in the DFT framework, the ACM-based XC functional should be in principle applied inside the KS equations in a self-consistent-field (SCF) fashion. This requirement is not trivial because ACM-based functionals are in general not simple explicit functionals of the density but are instead complicated expressions depending on KS orbitals and orbital energies as well (through E_x and E_c^{GL2}). One notable exception are the MCY functionals⁵¹ which use semilocal approximations to set the interpolation points along the AC integrand, thus allowing for a relatively straightforward SCF implementation. In the most general case considered in this work, however, ACM functionals are fifth rung functionals and thus, in practice, also in the context of DFT, they are always applied in a post-SCF scheme using precomputed DFT densities and orbitals.^{48,52} In this way, the results depend significantly on the choice of the reference density and orbitals, making the whole method not fully reliable.⁴⁸ On the other hand, an exploratory study of the XC potential derived from ACM models has shown that this possesses promising features, indicating that SCF calculations with ACM-based functionals might be an interesting path to explore.⁵³

In this work, we tackle this issue by introducing an SCF implementation of the ACM potential and applying it to some test problems in order to verify its ability to describe different properties and systems. One important aim of this work is in fact to measure and assess the capabilities of some of the most popular ACM presently available in literature. To this purpose the use of a proper SCF procedure is crucial as the level of accuracy of such methods can be inspected independently of an arbitrary reference ground-state as in previous works. In fact, for any density functional, the energy error can be decomposed into a contribution due to the approximate nature of the functional (intrinsic error) and that due to the approximate density used in the calculation (relaxation error).^{54,55} When the functional is evaluated on an arbitrary (non-SCF) density, the relaxation error may become important and the whole performance can be influenced by the choice of the density. Indeed, recent studies have shown how this effect can be used to improve DFT results by choosing accurate non-SCF densities.^{55,56} Nevertheless, within this framework, it is difficult to really understand the accuracy of the functional form itself and therefore to plan new advances. On the other hand, the use of a proper SCF procedure provides a well-defined reference for assessing the intrinsic errors. This is an extremely important point to clarify in view of future ACM developments. Note that such a development of new and possibly more accurate ACMs will instead not be covered in this work but left to upcoming publications. The development work performed here will

instead focus on a second important goal aimed at solving some open problems with the ACM potential that hinder its straightforward SCF implementation. These problems originate mainly from the naive treatment used so far for the large- λ contributions W_∞ and W'_∞ which causes an unphysical behavior in the ACM potential. Hence, in this article, we develop new approximations for both W_∞ and W'_∞ that preserve the accuracy for energies and remove the limitations on the potential side. As a byproduct of this work, we obtain useful strong-correlation generalized gradient approximations that prove to be very robust for the description of the Harmonium atom and the H_2 dissociation.

In the following, we present the theory behind SCF implementation of ACM functionals and the construction of new W_∞ and W'_∞ approximations. Afterward, we present some interesting preliminary results obtained for model and real systems.

THEORY

To perform SCF ACM calculations we need to deal with the potential arising from the functional derivative of the energy of eq 1, that is⁵³

$$\begin{aligned} v_{xc,\sigma}^{\text{ACM}}(\mathbf{r}) &\equiv \frac{\delta E_{xc}^{\text{ACM}}}{\delta \rho_\sigma(\mathbf{r})} \\ &= D_{E_x}^{\text{ACM}} \frac{\delta E_x}{\delta \rho_\sigma(\mathbf{r})} + D_{E_c^{\text{GL2}}}^{\text{ACM}} \frac{\delta E_c^{\text{GL2}}}{\delta \rho_\sigma(\mathbf{r})} + D_{W_\infty}^{\text{ACM}} \frac{\delta W_\infty}{\delta \rho_\sigma(\mathbf{r})} \\ &\quad + D_{W'_\infty}^{\text{ACM}} \frac{\delta W'_\infty}{\delta \rho_\sigma(\mathbf{r})} \end{aligned} \quad (4)$$

where $D_j = \partial f^{\text{ACM}} / \partial j$ with $j = E_x, E_c^{\text{GL2}}, W_\infty, W'_\infty$. As discussed in ref 53, the potential in eq 4 requires a combination of OEP (for E_x and E_c^{GL2}) and generalized gradient approximation (GGA) approaches (for W_∞ and W'_∞). Thus, it resembles the OEP–SCF implementation of the DH functionals reported in refs 57 and 58. In more details, the $v_{x\sigma}(\mathbf{r}) = \frac{\delta E_x}{\delta \rho_\sigma(\mathbf{r})}$ and $v_{c\sigma}(\mathbf{r}) = \frac{\delta E_c^{\text{GL2}}}{\delta \rho_\sigma(\mathbf{r})}$ functional derivatives are obtained by solving the OEP equation which reads^{9,10,12,59–61}

$$\int X_\sigma(\mathbf{r}, \mathbf{r}') v_{A,\sigma}^{\text{OEP}}(\mathbf{r}') d\mathbf{r}' = \Lambda_{A,\sigma}(\mathbf{r}) \quad (5)$$

with $A = X, C$ denoting the exchange and correlation parts, respectively. The inhomogeneity on the right hand side of eq 5 is given by

$$\begin{aligned} \Lambda_{A,\sigma}(\mathbf{r}) &= \sum_p \left\{ \int \phi_{p\sigma}(\mathbf{r}) \sum_{q \neq p} \frac{\phi_{q\sigma}(\mathbf{r}) \phi_{q\sigma}(\mathbf{r}')}{\epsilon_{p\sigma} - \epsilon_{q\sigma}} \frac{\delta E_A}{\delta \phi_{p\sigma}(\mathbf{r}')} d\mathbf{r}' \right. \\ &\quad \left. + \frac{\delta E_A}{\delta \epsilon_{p\sigma}} |\phi_{p\sigma}(\mathbf{r})|^2 \right\} \end{aligned} \quad (6)$$

and the static KS linear response function is

$$X_\sigma(\mathbf{r}', \mathbf{r}) = 2 \sum_{ia} \frac{\phi_{i\sigma}(\mathbf{r}') \phi_{a\sigma}(\mathbf{r}') \phi_{a\sigma}(\mathbf{r}) \phi_{i\sigma}(\mathbf{r})}{\epsilon_{i\sigma} - \epsilon_{a\sigma}} \quad (7)$$

All quantities are evaluated using orbitals $\phi_{p\sigma}$ and eigenvalues $\epsilon_{p\sigma}$ in a given cycle of KS SCF procedure (further details can be found in refs 17, 57, 58, 62, and 63). We note,

however, that there is a significant difference between ACM and DH approaches: in the former, the coefficients $D_{E_x}^{\text{ACM}}$ and $D_{E_c}^{\text{ACM}}$ are not fixed empirical parameters as in DH, but are well-defined (non-linear) functions of E_x , E_c^{GL2} , W_∞ , W'_∞ .⁵³

Approximations for the Strong-Interaction Limit.

Another important issue to consider in the SCF implementation of the ACMs is related to the treatment of W_∞ and W'_∞ , which describe the $\lambda \rightarrow \infty$ limit of the AC integrand. It can be proven that both W_∞ and W'_∞ display a highly non-local density dependence.^{64–68} This is accurately described by the strictly correlated electron (SCE) formalism,^{39,47} which is however computationally very demanding and nontrivial to evaluate. Therefore, the $\lambda \rightarrow \infty$ limit is usually approximated by simple semilocal gradient expansions (GEA) derived within the point-charge-plus-continuum (PC) model³⁸

$$W_\infty^{\text{PC}}[\rho] = \int d^3\mathbf{r} A \rho^{4/3} (1 + \mu_w s^2) \quad (8)$$

$$W'_\infty^{\text{PC}}[\rho] = \int d^3\mathbf{r} C \rho^{3/2} (1 + \mu_w s^2) \quad (9)$$

where $s = |\nabla\rho|/[2(3\pi^2)^{1/3}\rho^{4/3}]$ is the reduced gradient of the density, $A = -9(4\pi/3)^{1/3}/10$, $C = 1/2(3\pi)^{1/2}$, $\mu_w = -3^{1/3}(2\pi)^{2/3}/35 \approx -0.1403$, and $\mu_w = -0.7222$ (slightly different estimates are possible for μ_w , see, e.g., refs 36 and 39). The GEAs of eqs 8 and 9 yield, at least for small atoms, energies that are quite close to the accurate SCE values. However, when s is large, for example, in the tail of an exponentially decaying density, they fail, giving functional derivatives that diverge.⁵³ This is a severe drawback that does not allow these approximations to be used directly in an SCF implementation.

To remedy this limitation we consider here a simple GGA approximation, named, harmonium PC (hPC) model, based on the Perdew–Burke–Ernzerhof (PBE) exchange enhancement factor⁶⁹ that recovers the GEAs of eqs 8 and 9 in the slowly varying regime, is well behaved everywhere, and reproduces as close as possible the SCE values for both W_∞ and W'_∞ . Thus, we have

$$W_\infty^{\text{hPC}} = \int d^3\mathbf{r} A \rho^{4/3} \left[\frac{1 + s^2 \mu_w \frac{\kappa_w + 1}{\kappa_w}}{1 + s^2 \mu_w / \kappa_w} \right] \quad (10)$$

$$W'_\infty^{\text{hPC}} = \int d^3\mathbf{r} C \rho^{3/2} \left[\frac{1 + s^2 \mu_w \frac{\kappa_w + 1}{\kappa_w}}{1 + s^2 \mu_w / \kappa_w} \right] \quad (11)$$

where $\kappa_w = -7.11$ and $\kappa_w = -99.11$ have been fixed such that W_∞^{hPC} and W'_∞^{hPC} recover exactly the corresponding SCE values for the harmonium atom at $\omega = 0.5$:⁷⁰ with this value, the degree of correlation resembles that of the He atom and a simple analytical density is obtained. We note that a previous attempt to develop GGAs for W_∞ and W'_∞ , the modified PC (mPC) model of ref 71, yields results that are quite far from both the PC and the SCE values, in particular W'_∞ does not even recover the PC model in the small s limit. In fact, the mPC GGAs have been derived for the quasi-two-dimensional density regime⁷¹ and their application in three-dimensional systems, for example for the total correlation of atoms, is highly based on an error cancellation between the quite inaccurate

values of W_∞ and W'_∞ .⁷¹ In particular, W_∞^{mPC} has been designed to compensate the inaccuracies of W_∞^{PC} for the ISI functional, but this error compensation cannot work for other ACMs (especially those, as SPL, using only W_∞).

To understand the performances of the different approximations for the strong-interaction functionals, we report in Figure 1 the differences between the values of W_∞ and W'_∞

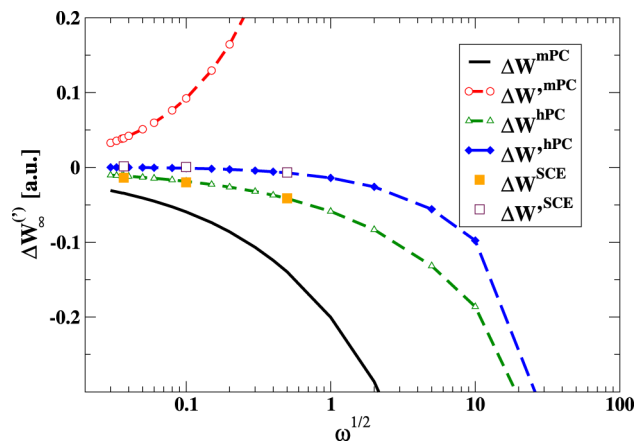


Figure 1. Differences between the values of W_∞ and W'_∞ computed with hPC and mPC formulas and the corresponding W_∞^{PC} and W'_∞^{PC} data ($\Delta W_\infty^{\text{method}} = W_\infty^{\text{method}} - W_\infty^{\text{PC}}$; $\Delta W'_\infty^{\text{method}} = W'_\infty^{\text{method}} - W'_\infty^{\text{PC}}$) for the harmonium atom at various values of the confinement strength ω . For reference, some available accurate SCE values are also reported.⁷⁰

computed with the two GGAs and the PC model, for the Hooke atom at different confinement strengths ω . The corresponding values for those instances of ω for which exact SCE reference data are available are also reported in Table 1.

Table 1. W_∞ and W'_∞ Energies (in Ha) for Three Values of ω for Which Hooke's Atom has Analytical Solutions⁷² and Exact SCE Reference Data Are Available^{70a}

ω	SCE	PC	hPC	mPC
		W_∞		
0.0365373	-0.170	-0.156	-0.167	-0.191
0.1	-0.304	-0.284	-0.303	-0.344
0.5	-0.743	-0.702	-0.743	-0.841
MARE		6.78%	0.70%	12.90%
		W'_∞		
0.0365373	0.022	0.021	0.021	0.060
0.1	0.054	0.054	0.053	0.146
0.5	0.208	0.215	0.208	0.562
MARE		2.64%	2.13%	171.10%

^aHooke's atom is usually considered to be in the strong correlation regime when the density displays a maximum away from the center of the harmonic trap, which happens⁷³ for $\omega \lesssim 0.0401$. The last line of each panel reports the MARE.

We see that, unlike mPC, the hPC model reproduces very well both the W_∞ and W'_∞ accurate SCE values,⁷⁰ being comparable to and even superior to the original PC model. This performance is not trivial because hPC was parameterized only on a single instance of the Hooke's atom ($\omega = 0.5$) but turns out to be very accurate for the whole range of confinement strengths. In particular, Figure 2 shows that in the small ω range (strong interaction limit of the Hooke's

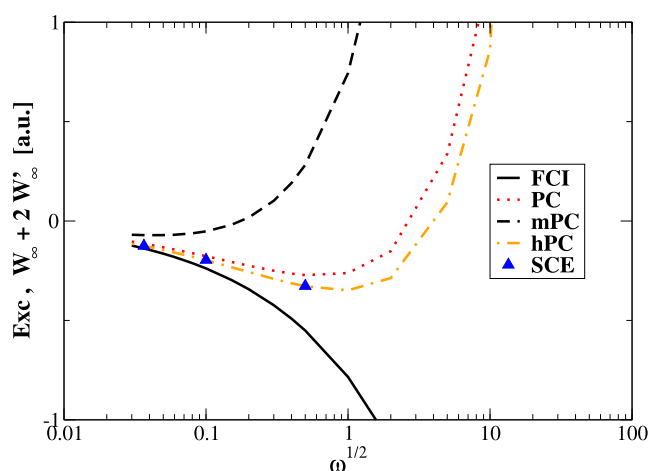


Figure 2. Comparison of the leading term of the XC energy ($E_{xc} = W_{\infty} + 2W'_{\infty}$) in the strong interacting regime of the Hooke's atom calculated using different models with FCI data.⁵²

atom) hPC yields the best estimation of the XC energy $E_{xc} = W_{\infty} + 2W'_{\infty}$, being slightly better than PC, while the mPC method fails completely.

Table 2. Values of W_{∞} and W'_{∞} for the He, Be, and Ne Atoms Obtained from Different Models and Using EXX Densities; We Use Atomic Units^a

	SCE	PC	hPC	mPC
		W_{∞}		
H	-0.3125	-0.3128	-0.3293	-0.4000
He	-1.500	-1.463	-1.492	-1.671
Be	-4.021	-3.943	-3.976	-4.380
Ne	-20.035	-20.018	-20.079	-21.022
MARE		1.15%	1.81%	13.31%
		W'_{∞}		
H	0	0.0426	0.0255	0.2918
He	0.621	0.729	0.646	1.728
Be	2.59	2.919	2.600	6.167
Ne	22	24.425	23.045	38.644
MARE		13.71%	3.05%	130.67%

^aThe results which agree best with SCE values^{39,47} are highlighted in bold. The last line of each panel reports the MARE [for W'_{∞} the H results are excluded]. The W'_{∞}^{SCE} reference data are reported with the same precision of as in ref 39.

An additional assessment is provided in Table 2 and Figure 3 where real atoms are considered both for SCE energies and SCE potentials. Also in this case the results of the hPC functional are in line with or better than the PC model, that was originally parametrized against the He atom, indicating once more the robustness of the hPC method. As anticipated, the mPC is instead quite far from the reference, especially for W'_{∞} .

COMPUTATION DETAILS

All calculations have been performed with a locally modified ACESII⁷⁴ software package. As in our previous studies,^{17,53,57,58,62,63,75,76} in order to solve OEP equations, we have employed the finite-basis set procedure of refs 77 and 78. In calculations, we employed the basis sets detailed below and

tight convergence criteria (SCF: 10^{-8}). In general, the convergence criteria were met within several cycles of the SCF procedure.

In order to solve algebraic OEP equations, the truncated singular-value decomposition (TSVD) of the response matrix was employed. The cutoff criteria in the TSVD procedure were set to 10^{-6} . For technical details on this type of calculations, we refer the reader to refs 17 and 63.

As reference data, we have considered the coupled-cluster single double and perturbative triple [CCSD(T)]⁷⁹ results obtained in the same basis set in order to make a comparison on the same footing and to reduce basis set related errors. In particular, we have considered a comparison with CCSD(T) relaxed densities, the corresponding KS potentials obtained via KS inversion,⁸⁰ and the total CCSD(T) energies. In the assessment, we have considered several properties, that is:

- **total energies:** the total energies have been calculated for the systems listed in Table 1 in ref 63 using an identical computational setup as in the same paper. A summary of the employed basis sets is also reported in the Supporting Information. We remark that, although total energies are not very important in practical chemical applications, they are important observables and are especially useful as indicators of the quality of the ACM interpolation.
- **Dipole moments:** for selected systems (H_2O , HF, HCl, H_2S , and CO), we have calculated the dipole moments using SCF densities for various methods. This is a direct test of the quality of self-consistent densities obtained within all approaches. The uncontracted aug-cc-pVTZ basis set of Dunning⁸¹ was used for all systems together with geometries taken from ref 82.
- **HOMO and HOMO–LUMO gap energies:** as in refs 63 and 83, we have computed the HOMO and HOMO–LUMO gaps, respectively, for the same set of systems as in the case of total energies. In the case of HOMO energies, the reference data have been taken from ref 83, whereas the HOMO–LUMO gap energies have been obtained from applying the KS inversion method⁸⁰ taking as a starting point the CCSD(T) relaxed density matrix as in ref 63.
- **correlation potentials and densities:** as in our previous studies,^{17,63,84,85} here we also investigate the quality of correlation potentials and densities^{17,86,87} looking at their spatial behavior. Both quantities are obtained from fully SCF calculations. The densities are analyzed in terms of correlation densities defined as $\Delta\rho_c = \rho^{\text{method}} - \rho^X$, where ρ^X is the density obtained from the exact exchange only ($X = \text{EXX}$)⁶⁰ or Hartree-Fock (HF) ($X = \text{HF}$) calculations for DFT and WFT methods, respectively. The Ne atom OEP calculations have been performed in a fully uncontracted ROOS-ATZD⁸⁸ basis set, whereas for the CO molecule, the uncontracted cc-pVTZ⁸⁹ basis sets were employed.
- **dissociation of H_2 :** fully self-consistent and post-SCF calculations, using OEP EXX orbitals, have been performed in the spin restricted formalism using the uncontracted aug-cc-pVTZ basis set. For comparison, PBE, MP2, GL2@EXX, and FCI data are also reported.
- **correlation energies of Hooke's atoms:** as previously,^{52,90,91} we have performed the calculation for various values of ω in the Hooke's atom model⁹² ranging

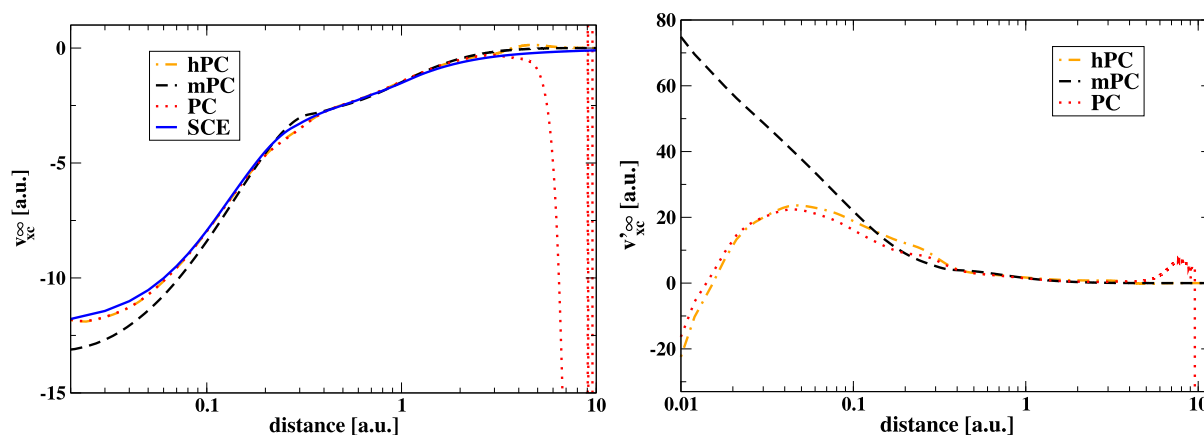


Figure 3. Comparison between (a) $v_{xc}^{\infty}(\mathbf{r}) = \delta W_{\infty} / \delta \rho(\mathbf{r})$ and (b) $v_{xc}^{\prime\infty}(\mathbf{r}) = \delta W_{\infty}^{\prime} / \delta \rho(\mathbf{r})$ potentials computed from different models for the Ne atom (using EXX densities).

Table 3. Total Energies (Ha) Calculated with Different Methods Self-Consistently (@SCF) or on top of EXX Orbitals (@EXX), for Several Functionals^a

system	@SCF			@EXX			
	ISI	SPL	GL2	ISI	SPL	GL2	CCSD(T)
He	-2.90089	-2.90043	-2.90780	-2.90191	-2.90148	-2.90925	-2.90253
Be	-14.67318	-14.67551	not. conv.	-14.67102	-14.67278	-14.69013	-14.66234
Ne	-128.93274	-128.94313	-128.98863	-128.92733	-128.93628	-128.97770	-128.89996
Mg	-199.86915	-199.86937	-199.88275	-199.86560	-199.86569	-199.87826	-199.82815
Ar	-527.51661	-527.53309	-527.58461	-527.51478	-527.53095	-527.58181	-527.45748
H ₂	-1.17039	-1.16972	-1.18107	-1.17019	-1.16953	-1.18060	-1.17273
He ₂	-5.80177	-5.80086	-5.81560	-5.80167	-5.80075	-5.81539	-5.80506
N ₂	-109.58263	-109.61715	-109.75090	-109.56105	-109.58609	-109.68725	-109.47628
Ne ₂	-257.86564	-257.88644	-257.97751	-257.85475	-257.87266	-257.95552	-257.80003
HF	-100.43787	-100.45019	-100.50368	-100.43148	-100.44188	-100.48965	-100.39579
CO	-113.35397	-113.38496	-113.51191	-113.32766	-113.34760	-113.43484	-113.25738
H ₂ O	-76.42686	-76.44091	-76.50285	-76.42076	-76.43270	-76.48790	-76.38692
HCl	-460.58531	-460.58876	-460.61411	-460.58227	-460.58550	-460.61020	-460.50933
Cl ₂	-919.93674	-919.94378	-919.99349	-919.92413	-919.93022	-919.97763	-919.77032
NH ₃	-56.55283	-56.56435	-56.62446	-56.54859	-56.55876	-56.61412	-56.52332
C ₂ H ₆	-79.80517	-79.82045	-79.92279	-79.79876	-79.81239	-79.90830	-79.76414
ME	-50.00	-61.08 ^b	-120.85	-43.14	-52.09	-99.17	
MAE	50.91	62.25 ^b	120.85	43.96	53.16	99.17	
MARE	0.055%	0.071% ^b	0.162%	0.048%	0.062%	0.150%	

^aCCSD(T) results are given as a reference. The last rows report the mean error (ME, in mHa), MAE (in mHa), and the MARE (in percent). For OEP-GL2, all the averages exclude the Be atom that for this functional has not converged. Not. conv.—not converged. ^bWithout Be.

between 0.03 (strong interaction) and 1000 (weak interaction) using a even-tempered Gaussian basis set from ref 93. For comparison, the ACM correlation energies have been calculated at both @EXX and @SCF reference orbitals.

RESULTS

We have performed a series of SCF ACM calculations to investigate the performance of these methods in the KS framework. In particular, we have considered the interaction-strength-interpolation (ISI)³⁶ and Seidl-Perdew-Levy (SPL)⁴⁰ ACMs. Unless explicitly stated, the hPC model has been used to describe the strong-interaction limit in all calculations. Moreover the bare GL2 (for SCF calculations OEP-GL2¹²) approach is also reported. The ISI model for W_{λ} has in general a larger deviation from linearity than SPL (which does not depend on W_{∞}^{\prime} too), whereas GL2 corresponds to the

linear approximation $W_{\lambda} = 2E_{\text{GL2}} \lambda$. Thus, the comparison of ISI with SPL and GL2 gives information on the importance of the shape of the ACM interpolation form.

In Table 3, we show the total energies computed with the various methods for a test set of 16 closed-shell atoms and small molecules, namely, He, Be, Ne, Mg, Ar, HF, CO, H₂O, H₂, He₂, Cl₂, N₂, Ne₂, HCl, NH₃, and C₂H₆.

We see that ISI@SCF and SPL@SCF perform quite well, giving errors roughly half that of OEP-GL2. For comparison, we acknowledge that the PBE functional⁶⁹ yields a mean absolute relative error (MARE) of 0.11%, which is twice as large as that of ISI@SCF.

Nevertheless, we have to acknowledge that the performance has further margins of improvement. For example the MAEs of MP2 and OEP2-sc (not reported) for the same test are 20 and 17 mHa, respectively. We can trace back most of this difference to the fact that the use of KS eigenvalues, as in ISI, SPL, and OEP-GL2, requires a quite large AC curvature (i.e., second

derivative with respect to λ) to yield accurate results, whereas this is not the case for MP2 and OEP2-sc that employ HF-quality eigenvalues. Then, KS-based methods need much more accurate ACMs to compete with HF-based ones. This is also confirmed observing that in Table 3, ISI is generally better than SPL, as the former is a more advanced ACM than the latter.

A second, related observation is that the ISI and SPL results suffer from a small relaxation error that worsens slightly the performance (with respect to using EXX orbitals). This effect might be related to the fact that the considered ACMs were developed in the context of post-SCF calculations and, as a result, may include some inherent error cancellation which is lost when they are evaluated using a (more accurate) SCF density. To better understand this trend, we define the quantity

$$\Delta[E] = |E@SCF - E^{\text{ref}}| - |E@EXX - E^{\text{ref}}| \quad (12)$$

which considers the absolute error difference [with respect the reference, i.e., CCSD(T)] going from EXX orbitals to SCF orbitals (a negative value means that SCF orbitals give better accuracy than EXX orbitals). The values of $\Delta[E]$ for ISI, SPL, and OEP-GL2/GL2 are 7.0, 9.1, and 16.9 mHa, respectively. Despite the $\Delta[E]$ values all being positive (i.e., calculations using EXX orbitals are more accurate) they decrease going from GL2 to SPL and then from SPL to ISI, showing again that increasing the complexity/accuracy of the ACM can yield better SCF potentials and relaxed total energies.

Interestingly, an opposite effect of the density relaxation is found in the harmonium atom, as shown in Figure 4, where if

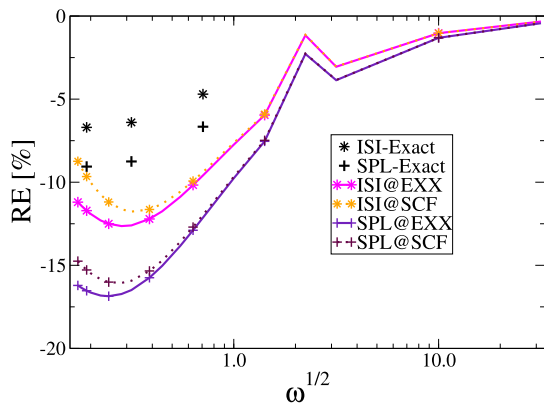


Figure 4. Relative error on correlation energies of harmonium atoms for various values of ω computed at @SCF and @EXX orbitals for ISI and SPL functionals using the hPC model for the strong-interaction functionals. The errors have been computed with respect to FCI data obtained in the same basis set.⁹³ The exact ISI and SPL values are taken from ref 70 and are obtained by inserting exact densities into the ISI and SPL functionals, including the exact treatment (SCE) of the strong-interaction limit.

we look at small values of the confinement strength, where the relaxation becomes more important, the SCF results are better with respect to the ones obtained using EXX orbitals and density (for both ISI and SPL). This depends on the fact that at these regimes, the true density is very different from the EXX one, and thus, the SCF procedure produces a significant improvement on the density. This also traces back to the use of hPC which yields accurate strong-correlation potentials; we note in fact that the accuracy of both ACMs with the hPC

model is very high (compare, e.g., with Figure 3 of ref 52). Conversely using the mPC model only ISI results are rather accurate because of error compensation effects between the W_{∞}^{mPC} and the W'_{∞}^{mPC} terms, while SPL ones, where only W_{∞}^{mPC} is used, are rather poor (see Figure S2 in the Supporting Information). This is an important indication of the importance of using proper strong-correlation approximations, delivering both good energies and potentials.

In Table 4, we report the dipole moments of some selected systems, from the SCF density. The results for CO are reported separately because they are qualitatively different and deserve a distinct analysis.

For H₂O, HF, HCl, and H₂S, a comparison with the CCSD(T) data shows that ISI is quite effective in predicting the dipole moments being slightly better than SPL and twice as good as GL2 [for comparison PBE gives in this case a mean absolute error (MAE) of 0.092 Debye with respect CCSD(T)]. Anyway, as already observed for the total energies, there are important margins of improvement as testified by the OEP2-sc performance that is definitely better than the ISI one. As already discussed, we can trace back the limitations of ISI and SPL not only in part to relaxation effects but also on the fact that, working in a pure SCF KS framework, it is very hard for the ACM to provide a proper curvature of the AC integrand curve as to get accurate KS orbital energies; consequently, the orbital-dependent energies are also negatively affected. For the case of CO, these effects are even more evident. In this case, in fact, OEPx predicts a qualitatively wrong dipole moment but GL2 largely over-corrects it, indicating that the linear behavior of the AC integrand needs to be significantly improved. Both ISI and SPL can partially achieve this task, halving the error with respect to GL2, but still they yield quite overestimated dipole moments.

As a next step, we consider in Table 5 the highest occupied molecular orbital (HOMO)–lowest unoccupied molecular orbital (LUMO) gaps obtained from different methods. As it could be expected both ACMs correct the general overestimation of gaps given by the OEPx but in doing so, they overestimate the correlation effects yielding gaps that are too small in most cases. Thus, we obtain MAEs of 0.68 and 0.52 eV for SPL and ISI, respectively, to be compared with the OEP2-sc MAE of 0.21 eV. We note anyway that the ISI and SPL results are clearly better than conventional semilocal functionals (PBE gives a MAE of 0.97 eV). Moreover, we note that by improving the quality of the ACM (GE2 \rightarrow SPL \rightarrow ISI) the description of the HOMO–LUMO gap is also significantly improved. Similar considerations apply as well for the HOMO energies (see Table 6). At the ISI level, the HOMO is shifted to higher energy with the almost the same MARE as OEPx (which is shifted to lower energy). Again, the ISI approach is better than SPL and much better than GL2 (as well as PBE with a MARE of 38.3%).

Then, we consider the correlation potentials for two typical systems, the Ne atom and the CO molecule. In the top panels of Figure 5, we see that the ACMs provide a quite good description of the correlation potential for the two systems, improving significantly over GL2. Nevertheless, with respect to reference data there are still some limitations, for example, a moderate overestimation of the correlation potential in valence regions. This characteristic corresponds to an overestimation of shell oscillations in the SCF density, as indicated in the bottom panels of Figure 5, where we report the correlation

Table 4. Dipole Moments (in Debye) for Some Selected Systems Calculated Using Self-Consistent Densities^a

method	H ₂ O	HF	HCl	H ₂ S	MAE		CO
					CCSD(T)	exp.	
OEPx	2.043	1.954	1.279	1.171	0.121	0.180	-0.265
GL2	1.616	1.531	1.061	1.004	0.187	0.145	1.703
SPL	1.758	1.654	1.085	1.024	0.110	0.080	0.940
ISI	1.809	1.699	1.093	1.030	0.083	0.060	0.692
OEP2-sc	1.885	1.786	1.185	1.094	0.018	0.073	0.355
CCSD(T)	1.904	1.809	1.170	1.079		0.065	0.153
Exp.	1.855	1.820	1.080	0.970			0.122

^aExperimental data are taken from ref 82. The MAE of H₂O, HF, HCl, and H₂S with respect to CCSD(T) and experimental results is also reported.

Table 5. HOMO–LUMO Energy Gap (eV) for Different Systems as Obtained from Several Methods^a

system	@SCF						KS[CCSD(T)]
	OEPx	GL2	OEP2-sc	SPL	ISI		
He	21.60	20.95	21.32	21.23	21.23	21.21	
Be	3.57	not conv.	3.63	3.40	3.47	3.61	
Ne	18.48	14.12	16.45	15.17	15.60	17.00	
Mg	3.18	3.40	3.33	3.38	3.38	3.36	
Ar	11.80	10.95	11.43	11.08	11.17	11.51	
H ₂	12.09	12.03	12.13	12.12	12.12	12.14	
He ₂	21.28	20.64	21.02	20.81	20.81	20.56	
N ₂	9.21	6.73	8.37	7.68	7.99	8.55	
Ne ₂	17.84	13.49	15.75	14.41	14.83	16.23	
HF	11.36	7.80	9.84	8.70	9.08	10.30	
CO	7.77	5.87	7.22	6.68	6.90	7.29	
H ₂ O	8.44	5.99	7.49	6.73	7.03	7.75	
HCl	7.82	7.10	7.52	7.11	7.14	7.55	
Cl ₂	3.90	2.65	3.35	2.74	2.78	3.29	
NH ₃	6.97	5.30	6.35	5.78	5.98	6.54	
C ₂ H ₆	9.21	8.24	8.85	8.51	8.62	8.95	
ME	+0.54	-1.13 ^b	-0.11	-0.64	-0.48		
MAE	0.52	1.15 ^b	0.21	0.68	0.52		
MARE	6.49%	12.16% ^b	1.86%	7.49%	5.71%		

^aThe last column reports the reference CCSD(T) data obtained from inverse method. The last lines report the MAE, and the MARE with respect to the CCSD(T) results. Not. conv.—not converged. ^bwithout Be.

density ρ_c , that is, the difference between the density obtained with a correlated method and its exchange-only version.

In the central panels of Figure 5, we report the values $\Delta[v_c(\mathbf{r})]$, which is defined, in analogy to eq 12 as

$$\Delta[v_c(\mathbf{r})] = |v_c@SCF(\mathbf{r}) - v_c^{\text{ref}}(\mathbf{r})| - |v_c@EXX(\mathbf{r}) - v_c^{\text{ref}}(\mathbf{r})| \quad (13)$$

These show, point-by-point whether or not the SCF procedure improves the correlation potential with respect to EXX orbitals. As we found for energies, the SCF correlation potentials are less accurate, but the error reduces with more accurate ACM functionals. This feature is also evident for the correlation density, see bottom panels. In this context, we should however also point out that the ACM-SCF density does not correspond to the exact linear response density.^{94–96}

As a final case, we consider in Figure 6 the potential energy surface for the dissociation of the H₂ molecule, in a restricted formalism,⁹⁷ which is one of the main DFT challenges,^{97,98} and was previously investigated in the ACM framework.^{43,99,100} While both MP2 and GL2@EXX diverge at large distances, ISI@SCF nicely reproduces the exact FCI curve, much better than ISI@EXX, see also ref 48. Thus, the SCF procedure turns out to be quite important showing that, despite some limitations discussed above, it is crucial to include important

correlation effects into the orbitals. For SPL (see Figure S1 in the Supporting Information), similar trends are found; the SPL@SCF curve for $R/R_0 > 2.5$ first increases and then decreases asymptotically, a behavior which is clearly incorrect and depends on some drawbacks of the SPL functional to describe the limit for large distances, which is more influenced by the strong correlation.

The limit for very large distances, well beyond $R/R_0 > 5$, is numerically tricky, but it can be computed exactly using the hydrogen atom with fractional spins, H(1/2,1/2), that is, with half spin up and half spin down.¹⁰¹ For this system, we have $E_{\text{GL2}} \rightarrow -\infty$ so that the ISI XC energy reduces to³⁶

$$E_{\text{xc}}^{\text{ISI}} \rightarrow W_{\infty} + 2W'_{\infty} \left(1 - \frac{1}{q} \ln(1+q) \right) \quad (14)$$

with $q = (E_x - W_{\infty})/W'_{\infty}$. The potential is thus a simple linear combination of the EXX potential and the GGA potential from W_{∞} and W'_{∞} . For the SPL approach, we have simply that $E_{\text{xc}}^{\text{SPL}} \rightarrow W_{\infty}$ and thus the potential is just $\delta W_{\infty}/\delta\rho(\mathbf{r})$.

The errors for different methods and orbitals are reported in Table 7.

Table 6. HOMO Orbital Energies (eV) for Different Systems as Obtained from Several Approaches^a

system	@SCF						CCSD(T)
	OEPx	GL2	OEP2-sc	SPL	ISI		
He	-24.98	-24.23	-24.55	-24.46	-24.39	-24.48	
Be	-8.41	not conv.	-8.89	-9.47	-9.32	-9.31	
Ne	-23.38	-17.66	-20.14	-18.98	-19.48	-21.47	
Mg	-6.88	-8.04	-7.33	-7.93	-7.91	-7.57	
Ar	-16.08	-14.94	-15.34	-15.11	-15.20	-15.63	
H ₂	-16.17	-16.34	-16.30	-16.25	-16.13	-16.41	
He ₂	-24.92	-24.14	-24.47	-24.38	-24.30	-24.48	
N ₂	-17.17	-11.32	-15.65	-13.09	-13.78	-15.51	
Ne ₂	-23.05	-17.45	-19.98	-18.80	-19.31	-21.34	
HF	-17.48	-12.16	-14.57	-13.52	-14.03	-15.96	
CO	-15.02	-10.64	-13.21	-12.18	-12.70	-13.94	
H ₂ O	-13.69	-9.01	-11.27	-10.39	-10.87	-12.50	
HCl	-12.92	-11.94	-12.28	-12.04	-12.08	-12.59	
Cl ₂	-12.06	-9.92	-10.85	-10.14	-10.22	-11.45	
NH ₃	-11.56	-8.37	-9.91	-9.34	-9.65	-10.78	
C ₂ H ₆	-13.21	-11.39	-12.20	-11.93	-12.07	-13.01	
ME	-0.65	+1.97 ^b	+0.59	+1.15	+0.93		
MAE	0.89	2.49 ^b	0.62	1.22	0.98		
MARE	6.12%	13.68% ^b	4.36%	8.28%	6.67%		

^aIn the last column, we report reference HOMO energies from ref 62. The last lines report the MAE, and the MARE calculated with respect to the CCSD(T) results. Not. conv.—not converged. ^bWithout be.

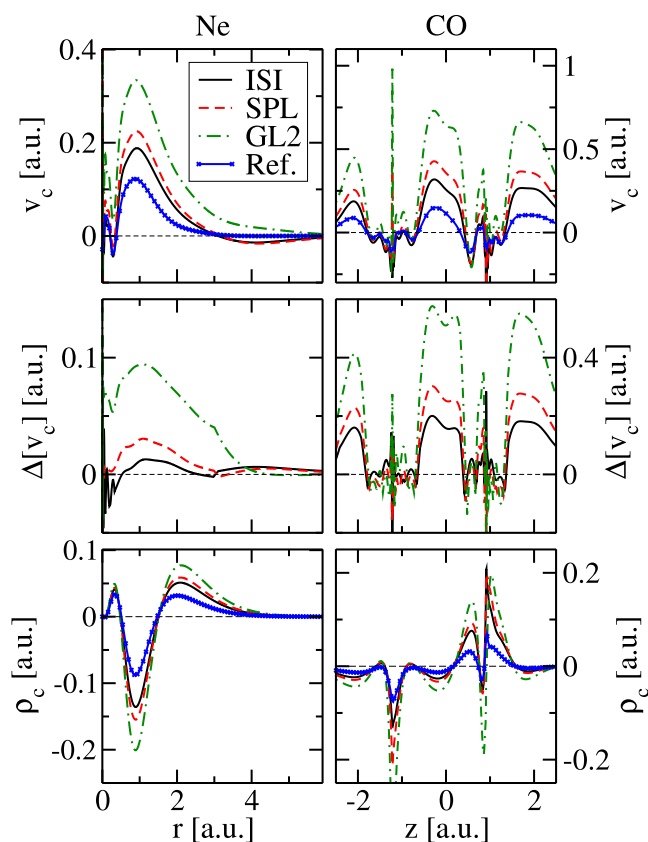


Figure 5. Correlation potentials (top panels), $\Delta[v_c]$ (middle), and correlation density (bottom) for the Neon atom (left) and CO molecule (right) obtained using several ACM-SCF methods. Reference means the CCSD(T) data using the method from ref 80.

At the exact density ($\rho(r) = \exp(-2r)/\pi$) SPL-PC gives an extremely accurate total energy but the same method fails for the SCF calculation. The SPL-mPC approach strongly

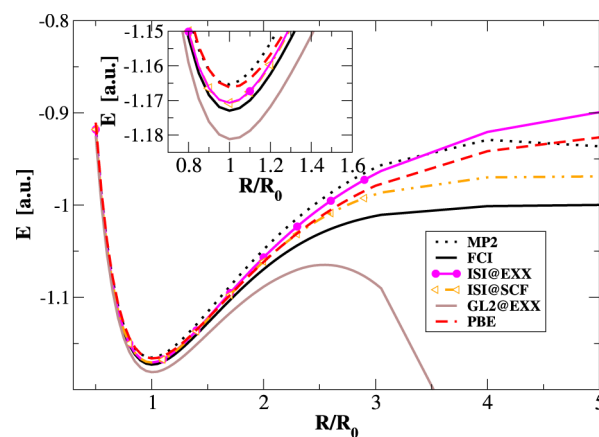


Figure 6. Total energy of the H₂ molecule as it is stretched calculated with the various methods. The inset presents the same data around the equilibrium distance.

underestimates the total energy, while the SPL-hPC gives a much lower error, both for the exact and the SCF densities. At the ISI level, all the energies are higher and the ISI-hPC@SCF is the most accurate approach. Note, however, that ISI-PC can be made exact with a proper choice of the parameters.³⁸ Note also that EXX fails for this system and PBE is also quite inaccurate.

When SCF effects are considered, PBE, EXX, ISI-mPC, and ISI-hPC yield a slight improvement with respect to the case when the exact density is used. Because the integrated density difference (IDD) is not zero in all cases, this is a clear signature that all methods display some error compensation effect. Moreover, some methods give important convergence issues: the simple PC model does not converge, as explained above; the mPC model converges but the errors are very large, about twice the PBE ones. Instead, the ISI-hPC is very good for both the considered densities, having the best accuracy among all functionals and performing even better than all the functionals

Table 7. Total Energy Error for H(1/2,1/2) in kcal/mol for Different Methods and Orbitals, Using a Geometric Series Basis-Set with 17 Uncontracted Gaussian Functions, 10⁴ as the Maximum Exponent, and 2.5 as the Geometric Progression Factor^a

	@EXACT	@SCF	IDD
PBE	54.7	51.5	0.103
EXX	196.1	178.6	0.260
SPL-PC	-0.4		
SPL-mPC	-109.8	-114.9	0.125
SPL-hPC	-21.0	-21.4	0.024
ISI-PC	27.4		
ISI-mPC	90.2	83.7	0.151
ISI-hPC	23.6	19.4	0.107

^aThe last column reports the IDD error, that is, $\int dr 4\pi r^2 |\rho(r) - \rho^{\text{exact}}(r)|$. Note that self-consistent PC calculations do not converge. The best two ACM results are reported in boldface.

considered in Table 5 of ref 97. Note that the good accuracy of the ISI-hPC with respect to ISI-mPC is not related to the previously mentioned error cancellation between an incorrect SCF density and an incorrect energy. In fact, the IDD error is significantly smaller going from ISI-hPC to ISI-mPC. Interestingly, the same arguments hold when comparing SPL-hPC to SPL-mPC, thus confirming the high quality of the hPC functional. Note that the almost vanishing IDD value for the SPL-hPC approach is a particular case, and all methods with $\text{IDD} \lesssim 0.1$ show a quite accurate density. The accuracy of the ISI-hPC@SCF approach for the H₂ dissociation limit is thus quite significant, considering that it uses full exact exchange and a combination of GL2 and a GGA functional without empirical parameters, in contrast to other approaches that use more complex constructions or extensive fitting on molecular data.^{98,102}

CONCLUSIONS

In this paper, we have shown that it is possible to use ACM-based XC functionals in a full SCF procedure. This solves a long-standing issue in DFT as all the previous calculations with ACM functionals had been done in a post-SCF fashion using GGA or EXX orbitals. This opens the way to new applications and even basic studies in this context, removing the need for a post-SCF procedure and all the related sources of inaccuracy. Of course, despite the ACM-SCF procedure presented here is well defined, conceptually clean and fully capable of producing important results, is it fair to state that the whole method is not yet optimized and straightforward to apply especially because it is strictly related to the OEP approach used for the treatment of the GL2 component, which requires itself some expertise to be handled. Nevertheless, several tricks and improvements can be used to make the OEP calculations simpler and more reliable,¹⁰³ thus various upgrades can be easily seen from the practical point of view for the SCF-ACM method. Anyway, these are left for future works, as in this paper we wanted to focus only on the core of problem without adding too many technical details.

Having been able to perform SCF ACM calculations on various systems, we could perform a thorough assessment of the functionals, finding important results. For strongly correlated systems, such as the harmonium atom and the hydrogen molecule at the dissociation limit, the ACM SCF calculations yield very accurate results taking advantage of the

incorporated strong-correlation limit and also thanks to the novel hPC functional for W_∞ and W'_∞ that proved to be very accurate for these cases. For molecular systems, we found that the overall accuracy using SCF orbitals depends on the quality of the underlying ACM, in line with the refs 24 and 25. In any case, the ISI-hPC yields already quite correct SCF potentials and total energies: nevertheless, its accuracy needs to be further verified for reactions and atomization energies.

Thus, we can finally conclude that, despite some limitations, the overall accuracy of the ISI functional (and partially also of the SPL one), when the full SCF solution is taken into account, is overall satisfactory, especially considering the following: (i) it does not employ any parameter obtained from molecular systems, and (ii) the approach is within a pure KS formalism with a local potential. These results and the availability of a working SCF procedure for general ACM formulas now open to the application and testing on other systems beyond the simple ones considered in this work. Moreover, it paves the path toward the development of more accurate ACM functional forms (see e.g. ref 46) as well as to further development of W_∞ and W'_∞ approximations, with improved accuracy for molecular systems.

INTERPOLATION FORMULAS

In the following, we report the ISI and SPL interpolation formulas.

ISI formula³⁸

$$W_\lambda^{\text{ISI}} = W_\infty + \frac{X}{\sqrt{1 + \lambda Y} + Z} \quad (15)$$

with

$$X = \frac{xy^2}{z^2}, \quad Y = \frac{x^2y^2}{z^4}, \quad Z = \frac{xy^2}{z^3} - 1 \quad (16)$$

$$x = -2W'_0, \quad y = W'_\infty, \quad z = W_0 - W_\infty \quad (17)$$

which yields for the XC energy

$$E_{\text{xc}}^{\text{ISI}} = W_\infty + \frac{2X}{Y} \left[\sqrt{1 + Y} - 1 - Z \ln \left(\frac{\sqrt{1 + Y} + Z}{1 + Z} \right) \right] \quad (18)$$

SPL formula⁴⁰

$$W_\lambda^{\text{SPL}} = W_\infty + \frac{W_0 - W_\infty}{\sqrt{1 + 2\lambda\chi}} \quad (19)$$

with

$$\chi = \frac{W'_0}{W_\infty - W_0} \quad (20)$$

The SPL XC functional reads

$$E_{\text{xc}}^{\text{SPL}} = (W_0 - W_\infty) \left[\frac{\sqrt{1 + 2\chi} - 1 - \chi}{\chi} \right] + W_0 \quad (21)$$

Note that this functional does not make use of the information from W'_∞ .

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.jctc.2c00352>.

Details on the basis set, dissociation curve of H_2 with the SPL functional, and further results for Hooke's atom (PDF)

AUTHOR INFORMATION

Corresponding Author

Szymon Śmiga – Institute of Physics, Faculty of Physics, Astronomy and Informatics, Nicolaus Copernicus University in Toruń, 87-100 Toruń, Poland; orcid.org/0000-0002-5941-5409; Email: szsmiga@fizyka.umk.pl

Authors

Fabio Della Sala – Institute for Microelectronics and Microsystems (CNR-IMM), Lecce, Via Monteroni 73100, Italy; Center for Biomolecular Nanotechnologies, Istituto Italiano di Tecnologia, Lecce 73010, Italy; orcid.org/0000-0003-0940-8830

Paola Gori-Giorgi – Department of Chemistry & Pharmaceutical Sciences and Amsterdam Institute of Molecular and Life Sciences (AIMMS), Faculty of Science, Vrije Universiteit, 1081HV Amsterdam, The Netherlands; orcid.org/0000-0002-5952-1172

Eduardo Fabiano – Institute for Microelectronics and Microsystems (CNR-IMM), Lecce, Via Monteroni 73100, Italy; Center for Biomolecular Nanotechnologies, Istituto Italiano di Tecnologia, Lecce 73010, Italy; orcid.org/0000-0002-3990-669X

Complete contact information is available at: <https://pubs.acs.org/10.1021/acs.jctc.2c00352>

Notes

The authors declare no competing financial interest.

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