

***m*-MTDATA on Au(111): Spectroscopic Evidence of Molecule–Substrate Interactions**

T. Zhang,^{1,*} C. Grazioli,² A. Guarnaccio,³ I. E. Brumboiu,⁴ V. Lanzilotto,⁵

F.O.L. Johansson,^{6,7,8} Klára Beranová,^{9,10} M. Coreno,³ M. de Simone,² B. Brena,⁶

and C. Puglia^{6,*}

¹ School of Integrated Circuits and Electronics, MIIT Key Laboratory for Low-Dimensional Quantum Structure and Devices, Beijing Institute of Technology, Beijing, China;

² IOM-CNR, Laboratorio TASC, Sincrotrone Trieste, 34149 Trieste, Italy;

³ ISM-CNR, Istituto di Struttura della Materia, 85050 Tito Scalo, Pz, Italy

⁴ Department of Chemistry, Pohang University of Science and Technology (POSTECH), 37673 Pohang, Republic of Korea;

⁵ Department of Chemistry, Sapienza Università di Roma, P.le A. Moro 5, 00185 Roma, Italy;

⁶ Department of Physics and Astronomy, Uppsala University, Box 516, SE-751 20 Uppsala, Sweden;

⁷ Division of Applied Physical Chemistry, Department of Chemistry, KTH Royal Institute of Technology, 10044 Stockholm, Sweden;

⁸ Institut des Nanosciences de Paris, UMR CNRS 7588, Sorbonne Université, F-75005 Paris, France.

⁹ Elettra-Sincrotrone Trieste S. C. p. A., Strada Statale 14, km 163.5, Basovizza, Trieste 34149, Italy;

¹⁰ FZU – Institute of Physics of the Czech Academy of Sciences, 18221 Prague, Czech Republic

Abstract

The starburst π -conjugated molecule based on triphenylamine (TPA) building blocks, 4,4',4''-tris(*N*-3-ethylphenyl-*N*-phenylamino)triphenylamine (C₅₇H₄₈N₄, *m*-MTDATA), is widely used in optoelectronic devices due to its electron-donating properties. The electronic structure of *m*-MTDATA adsorbed on an Au(111) surface was investigated by means of photoelectron spectroscopy (PES) and near edge X-ray absorption fine structure (NEXAFS) spectroscopy. The results were further compared to gas-phase measurements and DFT calculations. Our results clearly indicate a significant molecule–substrate interaction that induces considerable modifications on the electronic structure of the adsorbate

29 compared to the isolated molecule. The energy level alignment analysis shows that the HOMO–LUMO gap
30 is filled by new interface states.

31 Introduction

32 *m*-MTDATA (4,4',4''-tris(*N*-3-methylphenyl-*N*-phenylamino)triphenylamine, C₅₇H₄₈N₄) belongs to a group
33 of molecules called "starburst π -conjugated systems", for which the triphenylamine (TPA) is regarded as
34 their building block (Figure 1). (1–8) Starburst molecules are quite often found in organic optoelectronic
35 devices, like organic light emitting diodes (OLEDs), organic photovoltaics (OPVs), and more recently, solid
36 dye sensitized solar cells (DSSCs), proving to be a key factor of their high efficiency. (9–13) Our study of
37 gas-phase *m*-MTDATA (14) showed that the good electron-donating and charge-transfer properties of this
38 starburst molecule are largely related to its building block triphenylamine (TPA), particularly involving the
39 lone pair electrons of the N atoms. Moreover, the more complex molecular structure of *m*-MTDATA with
40 respect to TPA potentially promotes the formation of homogeneous amorphous and glassy molecular
41 films with higher thermal stability and better electron transport properties than films of TPA. (15–17)

42 Recent studies have shown that blends of electron-donating and electron-accepting molecules can
43 improve significantly the performance of OPVs and OLEDs. (9–11,18–24) The performance of such devices
44 is related both to the energy level matching between the donor and the acceptor components and to the
45 charge transfer/separation occurring at the interfaces between the organic/inorganic semiconductors or
46 between the donor and acceptor materials. For new developments and device optimizations,
47 fundamental studies of the electronic structure of components (donor, acceptor) and of the electronic
48 structure modifications occurring at the interfaces are therefore crucial for the understanding of the
49 charge separation at organic/inorganic and organic/organic heterojunctions. (25)

50 In this study, we characterize the adsorption of *m*-MTDATA on an Au(111) surface by core and valence
51 photoelectron spectroscopy (PES) and near-edge X-ray absorption fine structure (NEXAFS) spectroscopy.
52 The comparison between different molecular depositions on Au(111) reveals that the geometrical
53 arrangement of the molecules strongly depends on the coverage. Moreover, a strong modification of the
54 molecular electronic structure compared to the isolated *m*-MTDATA, especially for the
55 monolayer/interface and submonolayer case, is observed and is ascribed to the molecule/Au interaction.
56 This will have important implications on the performance of the devices in which such materials are
57 implemented.

58 Methods

59 Experimental Methods

60 The *m*-MTDATA thin film measurements were carried out at the Materials Science beamline of the Elettra
61 Synchrotron. (26) The PE spectra were recorded by the Specs Phoibos 150 hemispherical electron analyzer
62 mounted at the end station. During measurements, the base pressure of the analysis chamber was in the
63 10⁻¹⁰ mbar range. All the samples were prepared with a base pressure of high 10⁻¹⁰ mbar range. The clean
64 Au(111) substrate was cleaned by repeated Ar⁺ sputtering and annealing cycles until no contaminants
65 were observed by PE measurements.

66 The *m*-MTDATA (Sigma-Aldrich, purity 98%) were deposited onto the clean Au(111) via thermal
67 evaporation from a quartz crucible that was resistively heated to 190 °C by a tantalum wire. The thickness
68 of the *m*-MTDATA films was controlled by the evaporation time and estimated by the attenuation of the
69 PES Au 4f lines. We used multilayer samples (~3 molecular layers), interface samples (~1.3 molecular

70 layers), and low-coverage samples (~ 0.4 molecular layer) to shed light on the interfacial modifications to
71 the electronic structure of *m*-MTDATA. Due to the Volmer–Weber (island) type growth of the *m*-MTDATA
72 film (discussed later), it is difficult to achieve full monolayer coverage of *m*-MTDATA on an Au(111)
73 substrate. Thus, the ~ 1.3 molecular layer sample was chosen to represent the monolayer sample of *m*-
74 MTDATA/Au (i.e., the interface sample).

75 The C 1s and N 1s core level PE spectra were measured at normal emission (NE) with respect to the
76 electron analyzer using photon energies of 392 and 495 eV, respectively. At these photon energies, the
77 electron kinetic energy is close to the escape depth minimum and optimized to enhance surface
78 sensitivity. The overall resolutions were about 330 and 430 meV for C 1s and N 1s, respectively, estimated
79 from the width of the Fermi edge of the clean Au(111) crystal. Similarly, the overall resolution of the
80 valence spectra measured with photon energies of 40 and 100 eV was about 150 meV, where the
81 resolution using 100 eV photon energy is slightly worse than when using 40 eV.

82 The NEXAFS spectra at the C and N K-edges of the deposited molecules were recorded using partial Auger
83 yield. The measurements were performed at different scattering geometries: normal incidence (NI, 90°
84 between the incident light and the surface plane), normal emission (NE, 90° between the analyzer and
85 the surface plane or 60° between the incident light and the surface plane), and grazing incidence (GI, 10° ,
86 between the incident light and the surface plane). The photon energy scales of the NEXAFS were
87 calibrated by measuring the Au 4f PES lines by the first- and second-order light. The energy resolution for
88 the C and N K-edge NEXAFS spectra was estimated to be about 250 and 400 meV, respectively. The
89 measured spectra were normalized to the intensity of the photon flux measured simultaneously on a high-
90 transmission gold mesh. The C K-edge spectra were further normalized to the background spectra of the
91 clean Au(111) surface, in order to eliminate spectral features due to the carbon contamination on the
92 mesh and the beamline optics.

93 **Computational Methods**

94 The geometry of an isolated *m*-MTDATA molecule was optimized using the B3LYP (27) exchange and
95 correlation functional in combination with the 6-31G(d,p) (28) basis set. The optimization was carried out
96 in the Gaussian 16 (29) quantum chemistry software. An extensive characterization of *m*-MTDATA in the
97 gas phase has been carried out in ref (14), and in the present work we refer to the notation adopted
98 therein.

99 **Results and Discussion**

100 **Structure of Isolated *m*-MTDATA**

101 As shown in Figure 1, *m*-MTDATA contains 14 chemically nonequivalent C atoms and a total of 4 N atoms,
102 which includes one central (N_c) and three peripherals (N_p). It is evident that the core of *m*-MTDATA is a
103 TPA molecule (whose structure is shown in Figure 1c), which has a propeller-like form, with a torsion angle
104 of 41.7° of the phenyl rings. (19) The *m*-MTDATA molecule has a more complicated structure, where the
105 torsion angles of the phenyl rings connected to the central N (N_c) atom are approximately 41.7° , just like
106 in TPA. However, the torsion angles related to the peripheral N (N_p) atoms are different since the local
107 planes formed by the three C_{ipso} atoms with a N_p at the center, denoted [$N_p-(C_{\text{ipso}})_3$], are rotated with
108 respect to the central [$N_c-(C_{\text{ipso}})_3$] plane by about $2.2\text{--}2.4^\circ$ (see Figure 1b), even though all four N atoms
109 sit in the same molecular plane, i.e. [$(N)_4$ -plane]; see Figure 1a. Specifically, the average torsion angle

110 between each ring A and the $[N_P-(C)_3]$ planes is 44.1° . The torsion angle between the $[N_P-(C_{ipso})_3]$ planes
111 and phenyl rings B are approximately 39.9° . Finally, the torsion angle between the $[N_P-(C_{ipso})_3]$ planes and
112 the phenyl rings C are 41.2° . These local torsions mean that the phenyl rings B are twisted by
113 approximately 41.1° with respect to the molecular plane, i.e., $(N)_4$ -plane, defined by the four N atoms.
114 Similarly, the phenyl rings C are twisted by approximately 42.3° with respect to this molecular plane. We
115 stress here that the structural characteristics described above are for the isolated *m*-MTDATA molecule
116 and that the molecular geometry changes upon deposition on a substrate, as discussed below.

117 **C 1s PES**

118 The experimental and fitting of C 1s PE spectra of *m*-MTDATA in the gas phase and deposited on an
119 Au(111) single crystal at different thicknesses are shown in Figure 2. Three samples of different coverages
120 are analyzed, from low-coverage to interface to multilayer samples (see Experimental Methods). The
121 fitting was performed by using Gaussian curves, and the results are presented in Table 1. The chemical
122 shift between the main peak and the low-intensity peak is about 1 eV, similar to the gas-phase results. (14)

123 **1. Multilayer Sample**

124 We will start by analyzing the multilayer sample, where *m*-MTDATA keeps most of its molecular-like
125 character (Figure 2b), confirmed by the line profile of the corresponding C 1s PE spectrum, which is very
126 similar to the gas-phase results as further discussed in the following sections. The spectrum shows a high-
127 intensity peak at 285.0 eV and a structure of lower intensity at about 285.5 eV on the higher being energy
128 side. The spectrum has been fitted using two peaks. The intensity ratio of peak B vs peak B' is 2.8 as in the
129 gas phase (A/A', Figure 2a), (14) in agreement with the stoichiometry of the isolated molecule. In detail,
130 peak B' comes from methyl ($-CH_3$) and C_{ipso} atoms (directly bonded to the N atoms, $-C-N-$) while B comes
131 from the other C atoms within the phenyl rings, i.e., $15/42 = 1/2.8$.

132 As expected, for all the coverages the peaks of the C 1s PE spectrum (14) are broader than in the gas
133 phase. Besides the common solid-state effect, (30) reasons for this can be the different geometries
134 (disorder) that the adsorbed molecules can adopt in films on surfaces (as also confirmed later by C K-edge
135 NEXAFS), which would result in even more chemically inequivalent atoms contributing at different BEs.

136 **2. Interface and Low-Coverage Samples**

137 For the C 1s PE spectra of the thinner films, i.e., interface and low coverage shown in Figure 2c,d,
138 respectively, the peaks shift to lower BE, partly due to the more effective core-hole screening by the
139 Au(111) surface, (31) even if we cannot exclude any initial-state effects. The C 1s spectrum of the low-
140 coverage *m*-MTDATA/Au(111) sample is characterized by two peaks. One more intensive at 283.9 eV and
141 the other one at 284.9 eV. This spectrum has been fitted by using two peaks, D and D'. For the low-
142 coverage sample, the intensity ratio of peak D vs peak D' is only 1.25, deviating significantly from the
143 stoichiometric value (2.8). This indicates a significant modification of the molecular structure with respect
144 to the free molecule in Figure 1. The differences observed in the PE spectra can have many different
145 origins. For example, there can be shifts due to direct interactions of the C atoms with the surface, the
146 adsorption geometry may be very distorted with respect to the gas phase, or there may be diffraction
147 effects which can alter the intensity ratios of the peaks.

148 For the interface sample, the spectral features are less resolved, but we still can distinguish a more
149 intensive peak at 284.0 eV and the weaker one at about 284.9 eV. The fit reveals that the ratio between

150 the main and the shoulder peaks is different with respect to both the low-coverage and multilayer
151 samples. We get a good fit of the interface C 1s spectrum only if we include the first and second layer
152 components. After the decomposition, the intensity ratio of C1/C1' of the first layer was found to be like
153 that of the low-coverage sample, i.e., with a similar BE (shifted only +0.02 eV) and with a similar intensity
154 ratio (1.2). The second layer components are shifted by +0.41 eV to higher BE (with respect to the first
155 layer components) with an intensity ratio C2'/C2 of 2.6, more like the multilayer case. The fit seems to
156 suggest that the second layer has already a molecular-like character. Although the spectrum of the
157 interface sample clearly requires a second layer contribution, we note that the average thickness
158 estimated from the attenuation method (32) is much less than a full coverage (only ~ 1.7 Å). In view of
159 this, it is reasonable to consider the growth of the *m*-MTDATA layer to be a Volmer–Weber (island) type
160 from early stage of deposition instead of Stranski–Kastanov or Franck–van der Merwe growth, i.e.,
161 without the completion of full monolayers.

162 **N 1s PES**

163 The N 1s photoelectron spectra of *m*-MTDADA adsorbed on Au(111) at two coverages are shown in Figure
164 3. The N 1s BE is shifted from 399.30 eV (interface sample) to 399.90 eV (multilayer sample), likely due to
165 the different core-hole screening for the two coverages as already observed for the C 1s spectra of the
166 same samples. The N 1s line of the multilayer sample is slightly broader than the line of the interface
167 sample (1.25 vs 0.99 eV), possibly related to the increased disorder in the multilayer film, as also later
168 confirmed by N K-edge NEXAFS. The N 1s PE spectrum of the multilayer sample shows a broad low-
169 intensity feature at about 7 eV from the main line, similar to the N 1s shakeup of TPA measured in the gas
170 phase. (33) It is difficult to discern if such a feature is also present in the spectrum of the interface sample,
171 due to the noisy signal.

172 **C K-Edge NEXAFS**

173 In Figure 4 we show the angle-dependent C K-edge NEXAFS spectra of the different coverages of *m*-
174 MTDADA adsorbed on the Au(111) surface in comparison to the gas-phase results. (14) When the incident
175 angle of the linearly polarized synchrotron light with respect to the surface is varied, i.e., from grazing
176 incidence (GI, 10°) to normal incidence (NI, 90°), the cross-section of X-ray absorption of different orbitals,
177 namely, for the π^* (out-of-plane) or the σ^* (in-plane) orbitals, varies, allowing us to investigate the
178 orientation of the adsorbed molecules.

179 Both interface and multilayer samples do not show significant variation between π^* and σ^* orbital
180 intensities at the different experimental setups. According to free molecule calculations, (14) there is an
181 average twist angle of about 41–42° for all phenyl rings with respect to the horizontal molecular plane
182 formed by the 4 N atoms, giving the molecule a very complex geometric structure (Figure 1). It is then
183 difficult to draw any conclusions about the molecular orientation for the interface coverage.

184 The multilayer sample does not show significant variation between π^* and σ^* orbital intensities at the
185 different experimental setups, while the interface sample does show dichroism. For the interface sample,
186 the σ^* orbitals did not change much from NI to GI, but the π^* seems to respond more to the variation of
187 the incident angle of the beam. We observe an enhancement of the π^* resonances in GI, indicating a
188 preferential ordering of some of the phenyl rings, i.e., more parallel to the surface. Indeed, from our
189 previous study related to the isolated molecule, the first π^* resonance (peak A, at 285.0 eV) is related to
190 the contributions from all carbon atoms belonging to the phenyl rings with the exception of the carbons

191 directly bonded to nitrogen. (14) The second π^* resonance, at higher photon energy (A' , at 286.2 eV), is
192 instead attributed to the C_{ipso} atoms (covalently bonded to the N atoms) and to the methyl ($-CH_3$) C atoms.
193 Additionally, the adsorption of the building block molecule TPA on Au(111) shows a similar behavior of
194 the C K-edge NEXAFS at monolayer coverage. (34)

195 The results of the multilayer sample are quite similar to the C K-edge NEXAFS of *m*-MTDADA in the gas
196 phase, indicating that the phenyl rings of *m*-MTDATA in subsequent layers lose the preferential
197 orientation characteristic of the interface sample and are rotated at different angles similarly to the free
198 molecule structure, giving the layers a more “disordered” character.

199 **N K-Edge NEXAFS**

200 The N K-edge NEXAFS spectra of *m*-MTDATA adsorbed on Au(111) shown in Figure 5 were measured at
201 NI, GI, and NE and compared to the corresponding gas-phase spectrum (14). By contrast to the C K-edge
202 NEXAFS, we observe a strong polarization dependence of the π^* and σ^* resonances. The main resonance
203 of the adsorbed films (interface and multilayer) is found at 402.42 eV.

204 For the interface sample, shown in Figure 5c, the π^* resonances are enhanced at GI incidence, while the
205 σ^* are at NI. Then, as expected, at NE, the NEXAFS features have an intensity between what is observed
206 in the NI and GI spectra. This significant angle dependence of the NEXAFS spectra of the interface *m*-
207 MTDADA sample shows that all the planes defined by the $N_C-(C_{\text{ipso}})_3$ and $N_P-(C_{\text{ipso}})_3$ bonds are almost
208 parallel to the Au(111) surface at this coverage. This means that, although the free *m*-MTDATA is a quite
209 large molecule with phenyl rings and methyl groups twisted and forming a complex molecular structure,
210 when it is adsorbed on the Au(111) surface, the four planes defined by the $N_C-(C_{\text{ipso}})_3$ and $N_P-(C_{\text{ipso}})_3$ bonds
211 are forced, by the molecule–surface interaction, to be parallel to the Au(111) surface. It is then normal to
212 expect that this molecular rearrangement (unlike the gas-phase structure) has a significant impact on the
213 C 1s and N 1s PES results, as previously discussed.

214 In Figure 5b, a weaker angle dependence is observed for the multilayer *m*-MTDATA film, showing that the
215 molecular layers become more disordered and that the $N_C-(C_{\text{ipso}})_3$ and $N_P-(C_{\text{ipso}})_3$ planes are no longer
216 parallel to the surface. Furthermore, the asymmetry tail observed at the lower photon energy side of the
217 π^* resonance of the N K-edge NEXAFS spectrum (399.5–402.3 eV) of the multilayer sample, can most likely
218 be attributed to the same transitions contributing to the pre-edge of the gas phase (to LUMO+1 and
219 LUMO+2 having mostly in-plane contribution from the N $2p_{xy}$ component). (14)

220 For the interface *m*-MTDATA coverage on Au(111) the N K-edge NEXAFS clearly shows an extra broad pre-
221 edge low-intensity feature, between 398 eV and the main resonance (402.42 eV). The new pre-edge
222 feature has an origin different from that observed for the isolated *m*-MTDATA (Figure 5a), considering the
223 following: (i) the new pre-edge covers a wider photon energy range (~ 4 eV) than the pre-edge feature in
224 isolated *m*-MTDATA (~ 1 eV, Figure 5a); (ii) the angle dependence of the adsorbate NEXAFS spectra of the
225 interface sample reveals that this pre-edge peak can be ascribed to the out-of-plane π orbitals since its
226 intensity grows significantly at GI (especially evident in 399–400 eV in Figure 5c), whereas the pre-edge
227 peak for free *m*-MTDADA was ascribed to transitions involving mostly in-plane (N $2p_{xy}$) orbitals; (14) (iii)
228 this new interface pre-edge feature is almost absent in the multilayer spectrum (Figure 5b), which instead
229 shows only a similar pre-edge feature as for the gas-phase case, at around 401.8 eV.

230 In summary, the pre-edge feature for the interface sample can be attributed to the molecule–surface
231 interaction. In a previous study about 1,4-benzenediamine (BDA), a similar pre-edge feature was observed
232 and ascribed to N–Au interactions. (35) Our study presents another solid experimental proof for the origin
233 of such pre-edge states due to N–Au interaction. In fact, our study on triphenylamine (TPA) adsorbed on
234 Au(111) suggests a more complex mechanism as the origin of the pre-edge feature, ascribed to the
235 creation of new hybrid states originating from the interaction between the molecule and the gold
236 surface. (34) However, we cannot exclude that core-hole effects during the NEXAFS absorption process
237 could also contribute to the formation of new unoccupied states or to the increase of the strength of the
238 commonly known weak *m*-MTDATA/Au interaction.

239 Given the electron-donating properties of *m*-MTDATA, and in light of our recent findings regarding the
240 TPA/Au(111) system, we could speculate that the new out-of-plane component observed in the GI N K-
241 edge spectrum for the interface sample (Figure 5c) derives from new available valence states originating
242 from the molecule–surface interaction. These new states are available for new possible electron
243 transitions observed as the pre-edge intensity feature in the N K edge spectrum, as already observed and
244 discussed in our study about TPA/Au(111). Also, the C 1s and N 1s PE BE energy positions on the edge of
245 adsorption resonances (marked as bars in the Figures 4 and 5) confirm a quite significant interaction
246 between the molecule and the surface. (36)

247 Valence Band PES

248 To enhance the surface sensitivity, the valence band PES measurements were taken at the NI geometry
249 (i.e., grazing emission). The valence spectra taken with photon energies of 40 and 100 eV of the
250 interface *m*-MTDATA/Au(111) sample are depicted in Figure 6c,d, respectively. In the figure, we also show
251 the VB of clean Au(111) (measured in the same conditions) and the valence PES results of *m*-MTDATA in
252 the gas phase. This allows us to distinguish the molecular valence features from those of the substrate.
253 As expected, the contributions from the substrate dominate the spectrum in the binding energy region
254 between 2 and 8 eV of the VB spectrum taken at 100 eV of the interface sample. On the other hand, the
255 comparison between the isolated and interface samples helps in recognizing the molecular features
256 observed at binding energies >8 eV and at binding energies >6 eV in the spectrum taken with a photon
257 energy of 40 eV. The characteristic three-peak *m*-MTDATA outermost valence feature (14) is quite well
258 resolved, with the HOMO–2 peak a little hidden in the strong Au substrate signal. The *m*-MTDATA HOMO,
259 HOMO–1, and (trace of) HOMO–2 are, respectively, at 0.9, 1.5, and 2.1 eV, keeping the same energy
260 separation of 0.6 eV as in the gas phase.

261 Similarly, the VB PES results of the multilayer sample of *m*-MTDATA/Au(111) are shown in Figure 6a,b,
262 measured at 40 and 100 eV, respectively. Although the spectrum taken at 100 eV has a large Au(111)
263 substrate contribution in the energy region between 2 and 8 eV, the *m*-MTDATA valence photoemission
264 features are clearly visible, as observed from the similarity to the PES results of *m*-MTDATA measured in
265 the gas phase. The characteristic *m*-MTDATA outermost features are well observed. The HOMO, HOMO–
266 1, and (trace of) HOMO–2 are at 1.08, 1.68, and 2.28 eV, respectively. Compared to the interface sample,
267 these features are shifted (+0.18 eV) to higher binding energy, likely related to a less effective screening
268 effect from the Au(111) substrate for thicker molecular coverages. (25)

269 The lost resolution for the three expected features in the multilayer spectrum taken with 100 eV photon
270 energy (Figure 6b) can be related to different combined effects. The more bulk sensitive measurements

271 at such a photon energy (with respect to 40 eV, shown in Figure 6a) affects the spectroscopic results that
272 are the contributions from different molecular layers and from the Au(111) substrate. As also indicated
273 by the N K-edge NEXAFS results, the multilayer sample is characterized by a not ordered molecular
274 arrangement, which would cause a broadening of the spectroscopic lines. Moreover, the experimental
275 resolution using 100 eV photon energy is slightly worse than when using 40 eV. In principle, the beam
276 damage could also broaden the peaks, but we carefully checked that the characterized films were not
277 damaged.

278 **Energy Level Alignment**

279 The results of the valence level photoemission for the occupied states and of NEXAFS for the unoccupied
280 states can be aligned, according to the method introduced by Schnadt et al., (37) on a common energy
281 scale, giving an overview of the occupied and unoccupied density of states in the presence of the core-
282 hole. As shown in Figure 7, it is clearly observed that at the interface coverage, the unoccupied N states,
283 i.e., the pre-edge feature of the N K-edge NEXAFS, appear to extend and overlap with the Au Fermi level,
284 confirming the possibility of a charge redistribution channel between the Au surface and the *m*-MTDATA
285 molecule. The hybridization between the molecular orbitals and the metal electronic states gives rise to
286 these new interface states observed as a pre-edge feature in N K-edge NEXAFS. These new states are
287 filling the HOMO–LUMO gap leading to two important consequences: on one side, this can result in an
288 improved charge transfer efficiency through the substrate; on the other side, these states are filling the
289 molecular energy gap so important for many energy conversion applications.

290 In summary, through the energy alignment of the molecular levels with the bands of the Au substrate, we
291 can expect an effective charge transfer toward the substrate but losing, at interface coverage, the
292 semiconductor energy gap.

293 **Conclusions**

294 When *m*-MTDATA is deposited on Au(111), both the molecular and electronic structures undergo
295 significant modifications. The C 1s PE spectra of *m*-MTDATA for the low-coverage and interface coverage
296 show a significant change of the intensity ratio of the spectral lines with respect to gas-phase results,
297 suggesting a Volmer–Weber (island) type growth for *m*-MTDATA on Au(111) from the early stages of
298 deposition. For the interface sample, NEXAFS results indicate that all the four planes defined by the N_C–
299 (C_{ipso})₃ and N_P–(C_{ipso})₃ bonds are almost parallel to the surface upon adsorption on Au(111). Moreover, a
300 new pre-edge feature clearly observed in the N K-edge NEXAFS of the interface sample is ascribed to new
301 states formed by the interaction between the molecules and the gold substrate, appearing in the original
302 molecular energy gap. These results have important implications for single molecule devices and 2D layer
303 optoelectronics, which are highly dependent on the molecule–substrate or molecule–electrodes
304 interface. Although the formation of new interface states observed in our study can indicate the possibility
305 of a more effective charge transfer, these states also fill the molecular energy gap, affecting the optical
306 absorption properties of the molecule.

307 **Acknowledgments**

308 We thank the Carl Trygger Foundation for financial support and for making available the VG-Scienta SES-
309 200 photoelectron analyser at the Gas Phase beamline, Elettra, Italy. T.Z. is thankful for the financial

310 support from the National Natural Science Foundation of China (Nos. 61901038, 61971035, 61725107,
311 92163206), the Beijing Natural Science Foundation (Nos. Z190006, 4192054), and the National Key
312 Research and Development Program of China (2020YFA0308800, 2019YFA0308000) and the support from
313 the Beijing Institute of Technology Research Fund Program for Young Scholars and the Vice-Chancellor of
314 Uppsala University for financial support through the U4 collaboration. B.B. acknowledges the Swedish
315 Research Council for research grant (VR 2014-3776). F.O.L.J acknowledges support from the Swedish
316 Research Council (grants VR 2014-6463 and 2020-06409). The authors acknowledge the EU CERIC-ERIC
317 Consortium for the access to experimental facilities and financial support. We thank the staff at Materials
318 Science beamline, at Elettra, for all the help provided during the beamtimes. We thank G. Bortoletto and
319 C. Pedersini of the User Support Lab at Elettra. We also thank Prof. N. Mårtensson and Prof. S. Svensson
320 for helpful discussions.

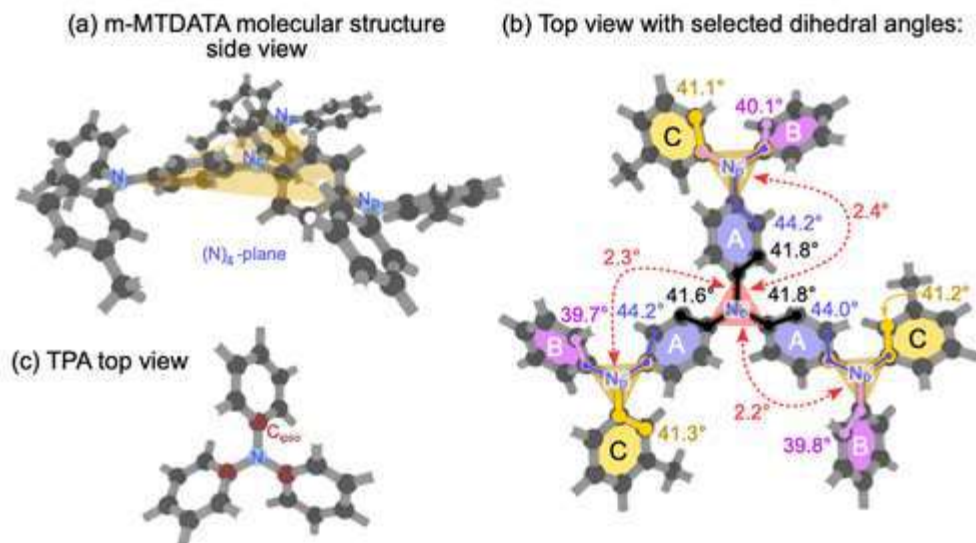
321 **References**

- 322 [1] Nishimura, K.; Kobata, T.; Inada, H.; Shiota, Y. Arylaldehyde and Arylketone Hydrazones as a New
323 Class of Amorphous Molecular Materials. *J. Mater. Chem.* 1991, 1 (5), 897, DOI:
324 10.1039/jm9910100897
- 325 [2] Ishikawa, W.; Inada, H.; Nakano, H.; Shiota, Y. Methyl-Substituted Derivatives of 1,3,5-
326 Tris(Diphenylamino)Benzene as a Novel Class of Amorphous Molecular Materials. *Chem. Lett.* 1991,
327 20 (10), 1731– 1734, DOI: 10.1246/cl.1991.1731
- 328 [3] Higuchi, A.; Inada, H.; Kobata, T.; Shiota, Y. Amorphous Molecular Materials: Synthesis and
329 Properties of a Novel Starburst Molecule, 4,4',4'' -Tri(N-Phenothiazinyl)Triphenylamine. *Adv. Mater.*
330 1991, 3 (11), 549– 550, DOI: 10.1002/adma.19910031105
- 331 [4] Ishikawa, W.; Inada, H.; Nakano, H.; Shiota, Y. Starburst Molecules for Amorphous Molecular
332 Materials: Synthesis and Morphology of 1,3,5-Tris(Diphenylamino)Benzene and Its Methyl-
333 Substituted Derivatives. *Mol. Cryst. Liq. Cryst. Sci. Technol. Sect. A. Mol. Cryst. Liq. Cryst.* 1992, 211
334 (1), 431– 438, DOI: 10.1080/10587259208025844
- 335 [5] Higuchi, A.; Ohnishi, K.; Nomura, S.; Inada, H.; Shiota, Y. Tri(Biphenyl-4-Yl)Amine and Tri(p-
336 Terphenyl-4-Yl)Amine as a Novel Class of Molecules for Amorphous Molecular Materials. *J. Mater.*
337 *Chem.* 1992, 2 (10), 1109, DOI: 10.1039/jm9920201109
- 338 [6] Inada, H.; Shiota, Y. 1,3,5-Tris[4-(Diphenylamino)Phenyl]Benzene and Its Methyl-Substituted
339 Derivatives as a Novel Class of Amorphous Molecular Materials. *J. Mater. Chem.* 1993, 3 (3), 319,
340 DOI: 10.1039/jm9930300319
- 341 [7] Ishikawa, W.; Noguchi, K.; Kuwabarau, Y.; Shiota, Y. Novel Amorphous Molecular Materials: The
342 Starburst Molecule 1,3,5-Tris[N-(4-Diphenyl- Aminophenyl)Phenylamino]Benzene. *Adv. Mater.* 1993,
343 5 (7–8), 559– 561, DOI: 10.1002/adma.19930050709
- 344 [8] Naito, K.; Miura, A. Molecular Design for Nonpolymeric Organic Dye Glasses with Thermal Stability:
345 Relations between Thermodynamic Parameters and Amorphous Properties. *J. Phys. Chem.* 1993, 97
346 (23), 6240– 6248, DOI: 10.1021/j100125a025

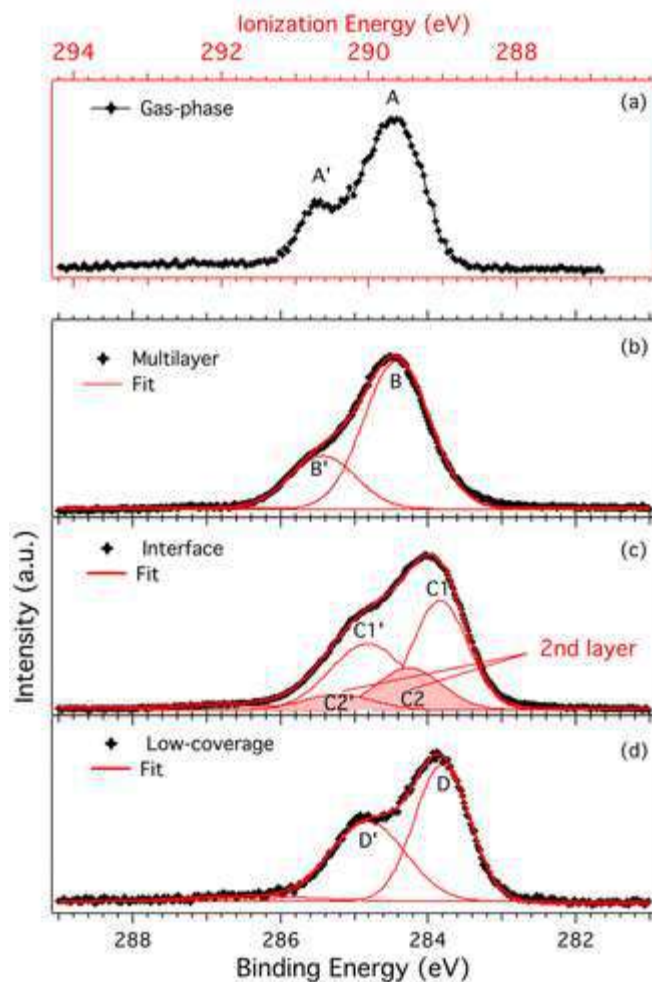
- 347 [9] Deotare, P. B.; Chang, W.; Hontz, E.; Congreve, D. N.; Shi, L.; Reuswig, P. D.; Modtland, B.; Bahlke,
348 M. E.; Lee, C. K.; Willard, A. P.; Bulovic, V.; Van Voorhis, T.; Baldo, M. A. Nanoscale Transport of
349 Charge-Transfer States in Organic Donor-Acceptor Blends. *Nat. Mater.* 2015, 14 (11), 1130– 1134,
350 DOI: 10.1038/nmat4424
- 351 [10] Goushi, K.; Adachi, C. Efficient Organic Light-Emitting Diodes through up-Conversion from Triplet to
352 Singlet Excited States of Exciplexes. *Appl. Phys. Lett.* 2012, 101 (2), 023306, DOI: 10.1063/1.4737006
- 353 [11] Wang, S.; Wang, X.; Yao, B.; Zhang, B.; Ding, J.; Xie, Z.; Wang, L. Solution-Processed Phosphorescent
354 Organic Light-Emitting Diodes with Ultralow Driving Voltage and Very High Power Efficiency. *Sci. Rep.*
355 2015, 5 (1), 12487, DOI: 10.1038/srep12487
- 356 [12] Johansson, E. M. J.; Karlsson, P. G.; Hedlund, M.; Ryan, D.; Siegbahn, H.; Rensmo, H. Photovoltaic and
357 Interfacial Properties of Heterojunctions Containing Dye-Sensitized Dense TiO₂ and Tri-Arylamine
358 Derivatives. *Chem. Mater.* 2007, 19 (8), 2071– 2078, DOI: 10.1021/cm062498v
- 359 [13] Johansson, E. M. J.; Odelius, M.; Karlsson, P. G.; Siegbahn, H.; Sandell, A.; Rensmo, H. Interface
360 Electronic States and Molecular Structure of a Triarylamine Based Hole Conductor on Rutile
361 TiO₂(110). *J. Chem. Phys.* 2008, 128 (18), 184709, DOI: 10.1063/1.2913245
- 362 [14] Zhang, T.; Brumboiu, I. E.; Lanzilotto, V.; Grazioli, C.; Guarnaccio, A.; Johansson, F. O. L.; Coreno, M.;
363 de Simone, M.; Santagata, A.; Brena, B.; Puglia, C. Electronic Structure Modifications Induced by
364 Increased Molecular Complexity: From Triphenylamine to m-MTDATA. *Phys. Chem. Chem. Phys.*
365 2019, 21 (32), 17959– 17970, DOI: 10.1039/C9CP02423A
- 366 [15] Shirota, Y.; Kobata, T.; Noma, N. Starburst Molecules for Amorphous Molecular Materials. 4,4',4''-
367 Tris(N,N-Diphenylamino)Triphenylamine and 4,4',4''-Tris[N-(3-Methylphenyl)-N-
368 Phenylamino]Triphenylamine. *Chem. Lett.* 1989, 18, 1145– 1148, DOI: 10.1246/cl.1989.1145
- 369 [16] Agarwala, P.; Kabra, D. A Review on Triphenylamine (TPA) Based Organic Hole Transport Materials
370 (HTMs) for Dye Sensitized Solar Cells (DSSCs) and Perovskite Solar Cells (PSCs): Evolution and
371 Molecular Engineering. *J. Mater. Chem. A* 2017, 5 (4), 1348– 1373, DOI: 10.1039/C6TA08449D
- 372 [17] Inada, H.; Ohnishi, K.; Nomura, S.; Higuchi, A.; Nakano, H.; Shirota, Y. Photo- and Electro-Active
373 Amorphous Molecular Materials: Morphology, Structures, and Hole Transport Properties of
374 Tri(Biphenyl-4-yl)Amine. *J. Mater. Chem.* 1994, 4 (2), 171, DOI: 10.1039/jm9940400171
- 375 [18] Kabe, R.; Adachi, C. Organic Long Persistent Luminescence. *Nature* 2017, 550 (7676), 384– 387, DOI:
376 10.1038/nature24010
- 377 [19] Wang, J.; Liu, K.; Ma, L.; Zhan, X. Triarylamine: Versatile Platform for Organic, Dye-Sensitized, and
378 Perovskite Solar Cells. *Chem. Rev.* 2016, 116 (23), 14675– 14725, DOI:
379 10.1021/acs.chemrev.6b00432
- 380 [20] Jailaubekov, A. E.; Willard, A. P.; Tritsch, J. R.; Chan, W. L.; Sai, N.; Gearba, R.; Kaake, L. G.; Williams,
381 K. J.; Leung, K.; Rossky, P. J.; Zhu, X. Y. Hot Charge-Transfer Excitons Set the Time Limit for Charge
382 Separation at Donor/Acceptor Interfaces in Organic Photovoltaics. *Nat. Mater.* 2013, 12 (1), 66– 73,
383 DOI: 10.1038/nmat3500

- 384 [21] Vandewal, K.; Albrecht, S.; Hoke, E. T.; Graham, K. R.; Widmer, J.; Douglas, J. D.; Schubert, M.;
385 Mateker, W. R.; Bloking, J. T.; Burkhard, G. F.; Sellinger, A.; Fréchet, J. M. J.; Amassian, A.; Riede, M.
386 K.; McGehee, M. D.; Neher, D.; Salleo, A. Efficient Charge Generation by Relaxed Charge-Transfer
387 States at Organic Interfaces. *Nat. Mater.* 2014, 13 (1), 63– 68, DOI: 10.1038/nmat3807
- 388 [22] Gelinas, S.; Rao, A.; Kumar, A.; Smith, S. L.; Chin, A. W.; Clark, J.; van der Poll, T. S.; Bazan, G. C.; Friend,
389 R. H. Ultrafast Long-Range Charge Separation in Organic Semiconductor Photovoltaic Diodes. *Science*
390 (80-) 2014, 343 (6170), 512– 516, DOI: 10.1126/science.1246249
- 391 [23] Cha, H.; Wu, J. Understanding What Determines the Organic Solar Cell Stability. *Joule* 2021, 5 (6),
392 1322– 1325, DOI: 10.1016/j.joule.2021.05.020
- 393 [24] Müller, K.; Schmidt, N.; Link, S.; Riedel, R.; Bock, J.; Malone, W.; Lasri, K.; Kara, A.; Starke, U.; Kivala,
394 M.; Stöhr, M. Triphenylene-Derived Electron Acceptors and Donors on Ag(111): Formation of
395 Intermolecular Charge-Transfer Complexes with Common Unoccupied Molecular States. *Small* 2019,
396 15 (33), 1901741, DOI: 10.1002/sml.201901741
- 397 [25] Zhang, T.; Brumboiu, I. E.; Lanzilotto, V.; Lüder, J.; Grazioli, C.; Giangrisostomi, E.; Ovsyannikov, R.;
398 Sassa, Y.; Bidermane, I.; Stupar, M.; de Simone, M.; Coreno, M.; Ressel, B.; Pedio, M.; Rudolf, P.;
399 Brena, B.; Puglia, C. Conclusively Addressing the CoPc Electronic Structure: A Joint Gas-Phase and
400 Solid-State Photoemission and Absorption Spectroscopy Study. *J. Phys. Chem. C* 2017, 121 (47),
401 26372– 26378, DOI: 10.1021/acs.jpcc.7b08524
- 402 [26] Tsud, N.; Acres, R. G.; Iakhnenko, M.; Mazur, D.; Prince, K. C.; Matolín, V. Bonding of Histidine to
403 Cerium Oxide. *J. Phys. Chem. B* 2013, 117 (31), 9182– 9193, DOI: 10.1021/jp404385h
- 404 [27] Becke, A. D. Density-functional Thermochemistry. III. The Role of Exact Exchange. *J. Chem. Phys.*
405 1993, 98 (7), 5648– 5652, DOI: 10.1063/1.464913
- 406 [28] Rassolov, V. A.; Pople, J. A.; Ratner, M. A.; Windus, T. L. 6-31G* Basis Set for Atoms K through Zn. *J.*
407 *Chem. Phys.* 1998, 109 (4), 1223– 1229, DOI: 10.1063/1.476673
- 408 [29] Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani,
409 G.; Barone, V.; Petersson, G. A.; Nakatsuji, H.; Li, X.; Caricato, M.; Marenich, A. V.; Bloino, J.; Janesko,
410 B. G.; Gomperts, R.; Mennucci, B.; Hratchian, H. P.; Ortiz, J. V.; Izmaylov, A. F.; Sonnenberg, J. L.;
411 Williams-Young, D.; Ding, F.; Lipparini, F.; Egidi, F.; Goings, J.; Peng, B.; Petrone, A.; Henderson, T.;
412 Ranasinghe, D.; Zakrzewski, V. G.; Gao, J.; Rega, N.; Zheng, G.; Liang, W.; Hada, M.; Ehara, M.; Toyota,
413 K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.;
414 Throssell, K.; Montgomery, J. A., Jr.; Peralta, J. E.; Ogliaro, F.; Bearpark, M. J.; Heyd, J. J.; Brothers, E.
415 N.; Kudin, K. N.; Staroverov, V. N.; Keith, T. A.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell,
416 A. P.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Millam, J. M.; Klene, M.; Adamo, C.; Cammi, R.;
417 Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Farkas, O.; Foresman, J. B.; Fox, D. J. *Gaussian 16*,
418 Revision B.01; Gaussian Inc.: Wallingford, CT, 2016.
- 419 [30] Tillborg, H.; Nilsson, A.; Mårtensson, N. Shake-up and Shake-off Structures in Core Level
420 Photoemission Spectra from Adsorbates. *J. Electron Spectrosc. Relat. Phenom.* 1993, 62 (1–2), 73–
421 93, DOI: 10.1016/0368-2048(93)80007-9

- 422 [31] Totani, R.; Grazioli, C.; Zhang, T.; Bidermane, I.; Lüder, J.; de Simone, M.; Coreno, M.; Brena, B.; Lozzi,
423 L.; Puglia, C. Electronic Structure Investigation of Biphenylene Films. *J. Chem. Phys.* 2017, 146 (5),
424 054705, DOI: 10.1063/1.4975104
- 425 [32] Graber, T.; Forster, F.; Schöll, A.; Reinert, F. Experimental Determination of the Attenuation Length
426 of Electrons in Organic Molecular Solids: The Example of PTCDA. *Surf. Sci.* 2011, 605 (9–10), 878–
427 882, DOI: 10.1016/j.susc.2011.01.033
- 428 [33] Zhang, T.; Brumboiu, I. E.; Grazioli, C.; Guarnaccio, A.; Coreno, M.; de Simone, M.; Santagata, A.;
429 Rensmo, H.; Brena, B.; Lanzilotto, V.; Puglia, C. Lone-Pair Delocalization Effects within Electron Donor
430 Molecules: The Case of Triphenylamine and Its Thiophene-Analog. *J. Phys. Chem. C* 2018, 122 (31),
431 17706–17717, DOI: 10.1021/acs.jpcc.8b06475
- 432 [34] Zhang, T.; Svensson, P. H. W.; Brumboiu, I. E.; Lanzilotto, V.; Grazioli, C.; Guarnaccio, A.; Johansson,
433 F. O. L.; Beranová, K.; Coreno, M.; de Simone, M.; Floreano, L.; Cossaro, A.; Brena, B.; Puglia, C.
434 Clarifying the Adsorption of Triphenylamine on Au(111): Filling the HOMO–LUMO Gap. *J. Phys. Chem.*
435 *C* 2022, 126, 1635, DOI: 10.1021/acs.jpcc.1c08877
- 436 [35] Balducci, G.; Romeo, M.; Stener, M.; Fronzoni, G.; Cvetko, D.; Cossaro, A.; Dell’Angela, M.; Kladnik,
437 G.; Venkataraman, L.; Morgante, A. Computational Study of Amino Mediated Molecular Interaction
438 Evidenced in N 1s NEXAFS: 1,4-Diaminobenzene on Au (111) -SI. *J. Phys. Chem. C* 2015, 119 (4), 1988–
439 1995, DOI: 10.1021/jp512146t
- 440 [36] Nilsson, A.; Björneholm, O.; Zdansky, E. O. F.; Tillborg, H.; Mårtensson, N.; Andersen, J. N.; Nyholm,
441 R. Photoabsorption and the Unoccupied Partial Density of States of Chemisorbed Molecules. *Chem.*
442 *Phys. Lett.* 1992, 197 (1–2), 12–16, DOI: 10.1016/0009-2614(92)86013-8
- 443 [37] Schnadt, J.; O’Shea, J. N.; Patthey, L.; Krempaský, J.; Mårtensson, N.; Brühwiler, P. A. Alignment of
444 Valence Photoemission, x-Ray Absorption, and Substrate Density of States for an Adsorbate on a
445 Semiconductor Surface. *Phys. Rev. B* 2003, 67 (23), 235420, DOI: 10.1103/PhysRevB.67.235420

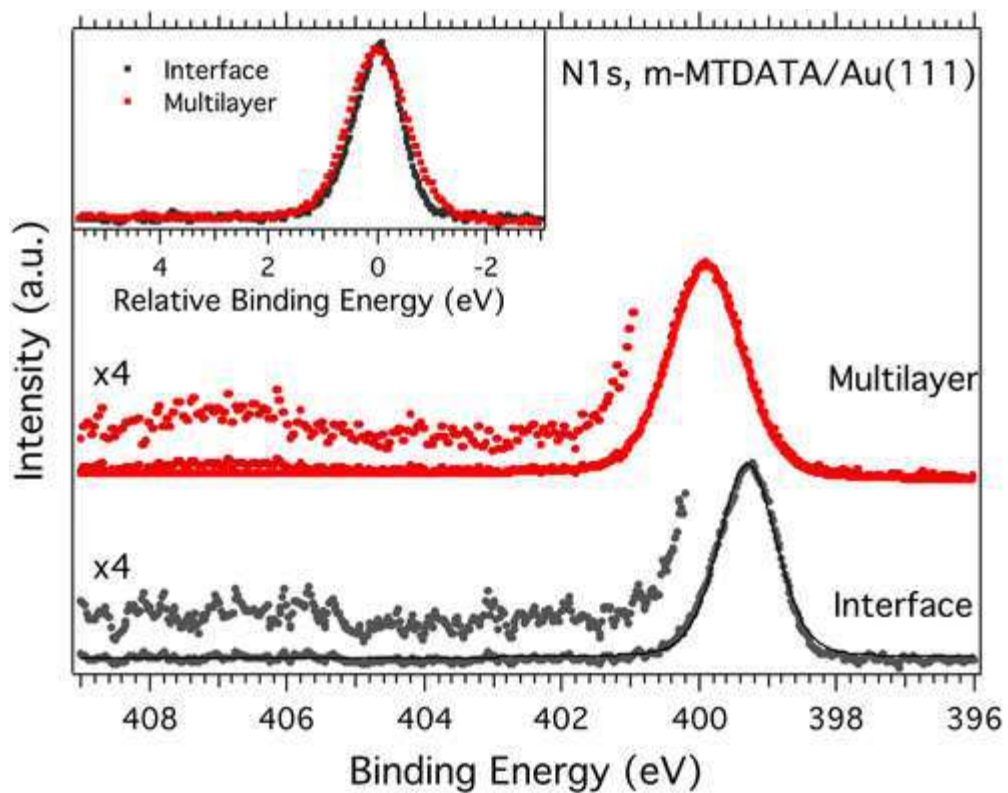


447 **Figure 1.** Structure of the *m*-MTDADA (a, b) molecule in comparison to its core molecule TPA (c). (a) Side
 448 view of *m*-MTDADA. The 4 N atoms (N_c + 3 N_p) are in the same plane, which defines the molecular N_c-
 449 plane of *m*-MTDADA. (b) There are three chemically different phenyl rings (A, B, and C), resulting from
 450 the 14 chemically distinct carbon atoms. The torsion angles of the phenyl rings are indicated.

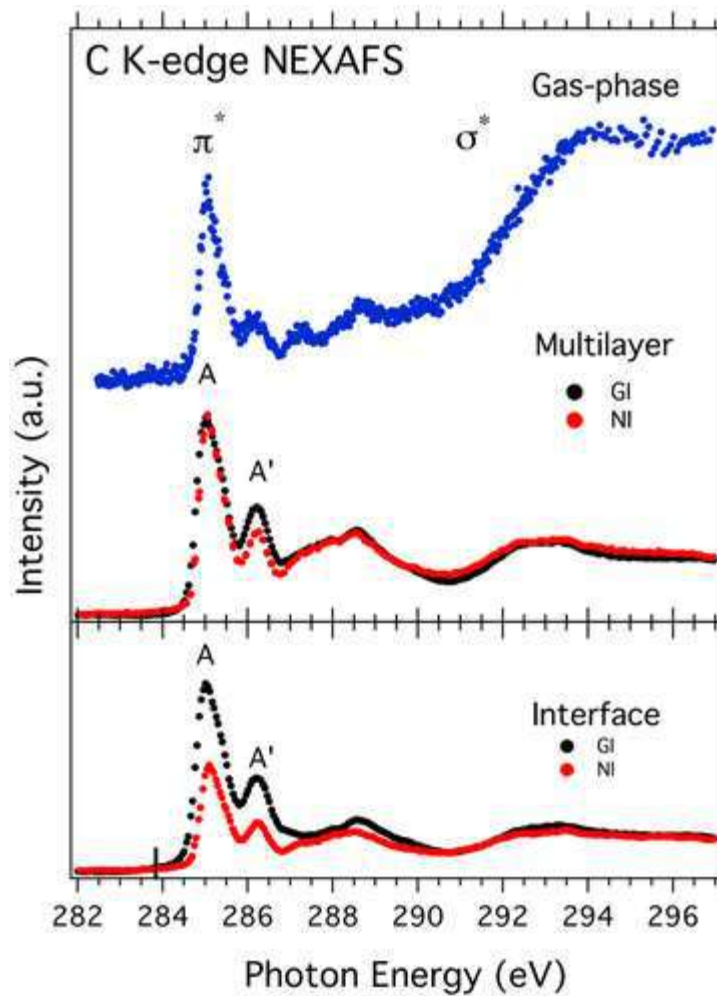


452 **Figure 2.** C 1s PE spectra of *m*-MTDATA, showing (a) gas-phase data from ref (14) versus the ionization
 453 energy scale (refer to the vacuum level, red axis) and (b)–(d) experimental results and fitting of the C 1s
 454 PE spectra (where a Shirley background has been removed) of *m*-MTDATA adsorbed on Au(111) at the
 455 indicated different coverage versus a binding energy scale (refer to the Fermi edge, black axis).

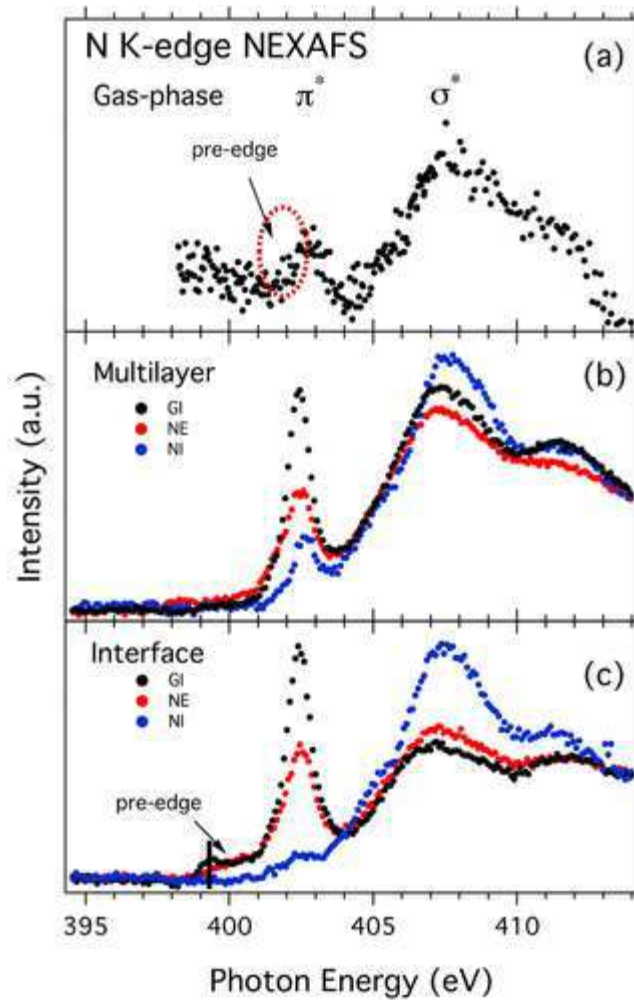
456



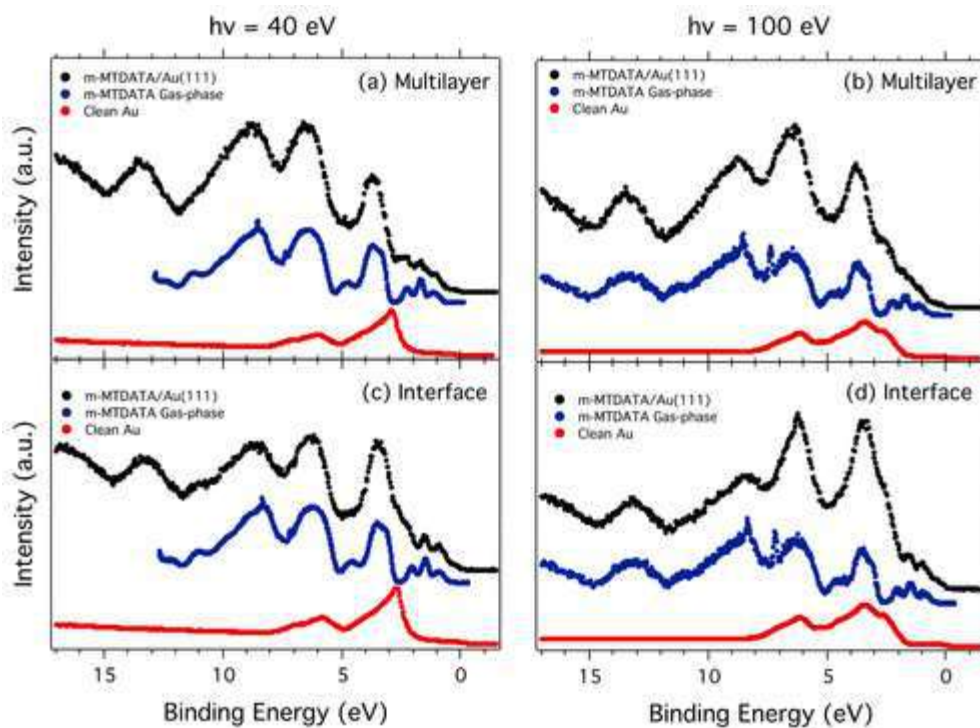
457 **Figure 3.** Comparison between N 1s PE spectra of interface and multilayer coverages of *m*-MTDATA
458 adsorbed on Au(111). In the inset, the comparison of the profiles of the two lines is shown on a relative
459 BE scale.



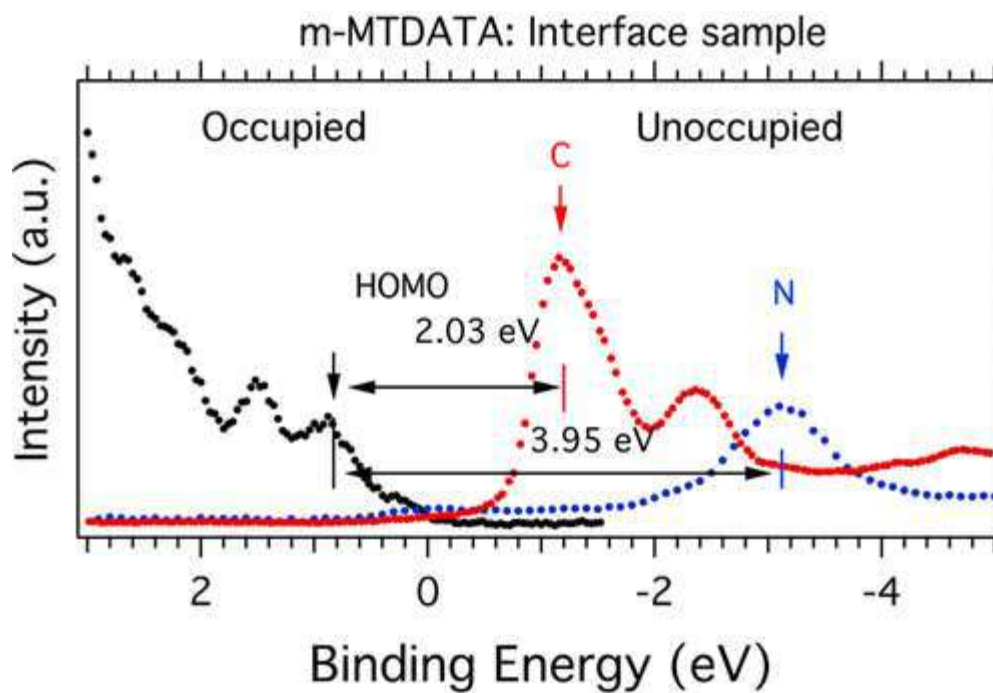
461 **Figure 4.** Comparison of the C K-edge NEXAFS of *m*-MTDATA/Au(111) at the interface and multilayer
 462 coverages, taken at different experimental setups and compared with the gas-phase results from
 463 ref (14). The vertical black bars indicate the binding energy of the corresponding non-ipso phenyl carbon
 464 C 1s PES.



466 **Figure 5.** N K-edge NEXAFS spectra of *m*-MTDATA/Au(111) films of interface and multilayer samples,
467 compared to the gas-phase result from ref (14). The vertical black bars indicate the binding energy of
468 the corresponding N 1s PES line.



470 **Figure 6.** Comparison of valence PE spectra of *m*-MTDATA/Au(111) (blue line with markers) of different
 471 thicknesses (multilayer and interface as indicated in the figure) taken at NI, with photon energies of 40
 472 eV (a, c) and 100 eV (b, d). In each panel, the clean Au(111) substrate (black dashed lines) and gas-phase
 473 (pink dashed lines, from ref (14)) spectra measured with the same photon energy are also shown. The
 474 gas-phase spectrum has been shifted -5.25 eV (-5.45 eV) to align with the multilayer (interface) results.



476 **Figure 7.** Energy level alignment of the occupied (valence PE results measured at 40 eV) and unoccupied
477 (in the presence of a core hole) (C and N K-edge NEXAFS results) states for *m*-MTDATA/Au at interface
478 sample.

Table 1. Fitting Results of C 1s PE Spectra of *m*-MTDATA/Au(111)

	BE (eV)	FWHM (eV)	chemical shift from main peak (eV)	area ratio/main
Multilayer				
main peak (B)	284.43	1.05		
low intensity peak (B')	285.43	1.13	0.99	2.8
Interface				
main peak (C1)	283.85	0.88		
low intensity peak (C1')	284.82	1.20	0.98	1.22 (first)
main peak (second layer, C2)	284.25	1.05	0.41	
low intensity peak (second layer, C2')	285.23	1.10		2.58 (second)
Low Coverage				
main peak (D)	283.82	0.80		
low intensity peak (D')	284.80	1.20	0.98	1.25