Supporting Information Living diatom microalgae for desiccation-resistant electrodes in biophotovoltaic devices

Cesar Vicente-Garcia,[†] Danilo Vona,[‡] Francesco Milano, \Box Gabriella Buscemi,[†] Matteo Grattieri,[†] Roberta Ragni,^{†,*} Gianluca M. Farinola,^{†,*}

† Dipartimento di Chimica, Università degli Studi di Bari "Aldo Moro", I-70126 Bari, Italy.

‡ Dipartimento di Scienze del Suolo, della Pianta e degli Alimenti, Università degli Studi di Bari"Aldo Moro", I-70126 Bari, Italy.

□ Istituto di Scienze delle Produzioni Alimentari, Consiglio Nazionale delle Ricerche, I-73100 Lecce, Italy.

* Corresponding author: RR: roberta.ragni@uniba.it; GMF: gianlucamaria.farinola@uniba.it

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Figure S1. Absorption spectrum of pigments extracted from *Phaeodactylum. tricornutum* in ethanol (black line). Light emission spectra of the white and red LED sources used for CA experiments (red dashed and red dotted lines, respectively). Maximum peaks in all spectra have been set to 1.



Figure S2. Average percentage of cells fallen from the biophotoanodes after a 45 min CA experiment carried out for each microalgal species. Fallen cells were recovered from the BPV electrolytic solution after the CA experiment.



Figure S3. Microscopy images of the ITO biophotoanodes prepared with each microalgal species. Bright field images at 20x (top row); chloroplast fluorescence images at 20x (middle row) and at 100x (bottom row). Fluorescence images were recorded using an excitation passband filter (λ_{ex} = 568 ± 10 nm) and emission filter (λ_{em} > 600 nm).



Figure S4. (a) Bright field image (top) of green microalga *Dunaliella tertiolecta* (*Dt*), and chloroplast fluorescence image (bottom), both at 100x. (b) Average photocurrent density at the third light cycle (left), and total chlorophyll content (right) of *Dt*, normalized to 10^6 cells. Tables report the pairwise tests showing the statistical differences in the chlorophyll content and photocurrent output for the investigated species.



Figure S5. (a) Mean fluorescein emission ($\lambda_{ex} = 488 \text{ nm}$, $\lambda_{em} = 511 \text{ nm}$) in the FDA viability assay of *Tw* and *Dt* cells from the bioanodes performed at day 0 and at day 15 after four CA experiments. (b) Pictures showing a slight detachment of *Tw* and *Dt* cells from the biophotoanodes at day 15 and after 4 CA with respect to day 0. (c) Bright field and chloroplast fluorescence images (both at 50x) of *Tw* and *Dt* coated electrodes at days 0 and after 15 respectively.

Calculation of the Internal Quantum Efficiency (IQE)

According to Equation 1, *IQE* is defined as the moles of electrons *e* per second pumped into the circuit (N_e) divided by the moles of absorbed photons (γ) per second (N_{ap}) .

$$IQE = N_e / N_{ap}$$
 Equation 1

Where N_e and N_{ap} can be calculated by Equations 2 and 3 respectively, as follows:

$$N_e = J/F$$
 Equation 2

$$N_{ap} = N_p \times (1 - T) = N_p \times A$$
 Equation 3

Where *J* is the maximum current density produced, *F* is the Faraday constant (96500 C/mol), T and *A* are the biophotoanode film transmittance and absorbance respectively and N_p is the number of photons emitted by the light source reaching the biophotoanode surface per second, that is determined by the **Equation 4**:

$$N_p = E \times \lambda / (h \times c)$$
 Equation 4

where *E* is the irradiance of the monochromatic light source, λ is the emission wavelength, *h* is the Planck's constant (6.626×10⁻³⁴ J×s) and *c* is the light speed (2.99×10⁸ m/s). Considering that the light source has E = 30 mW/cm² at $\lambda = 660$ nm, the numerical value of N_p results:

 $N_p = (30 \times 10^{-3} \text{ J/(cm^2 \times s)} \times 660 \times 10^{-9} \text{ m}) / (6.626 \times 10^{-34} \text{ J} \times \text{s} \times 2.99 \times 10^8 \text{ m/s}) = 9.99 \times 10^{16} \text{ } \gamma/(\text{cm}^2 \times \text{s})$

The average absorbance of Tw and Dt biophotoanodes are A = 0.0253 and 0.0529, respectively.

Therefore, we can calculate N_{ap} by using **Equation 3** obtaining:

$$(N_{ap})_{Tw} = 2.52 \times 10^{-15} \ \gamma/(\text{cm}^2 \times \text{s}); \ (N_{ap})_{Dt} = 5.28 \times 10^{-15} \ \gamma/(\text{cm}^2 \times \text{s})$$

The average current density values for Tw and Dt are $J = 0.296 \times 10^{-6}$ and 0.226×10^{-6} A/cm² respectively. These values are used to calculate N_e by using **Equation 2**, and finally *IQE* from **Equation 1**, resulting in the following values:

 $(IQE)_{Tw} = 0.073 \pm 0.015$ %; $(IQE)_{Dt} = 0.026 \pm 0.007$ %