



1 **Variability and Trends in Physical and Biogeochemical**  
2 **Parameters of the Mediterranean Sea during a Cruise with**  
3 **RV MARIA S. MERIAN in March 2018**

4  
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26

27 **Abstract**

28 The last decades have seen dramatic changes in the hydrography and biogeochemistry of the  
29 Mediterranean Sea. The complex bathymetry, highly variable spatial and temporal scales of  
30 atmospheric forcing and internal processes contribute to generate complex and unsteady  
31 circulation patterns and significant variability in biogeochemical systems. Part of this



1 variability can be influenced by anthropogenic contributions. Consequently, it is necessary to  
 2 document details and to understand trends in place to better relate the observed processes and  
 3 to possibly predict the consequences of these changes. In this context we report on data from  
 4 an oceanographic cruise in the Mediterranean Sea on the German research vessel MARIA S.  
 5 MERIAN (MSM72) in March 2018. The main objective of the cruise was to contribute to the  
 6 understanding of long-term changes and trends in physical and biogeochemical parameters,  
 7 such as the anthropogenic carbon uptake and to further assess the hydrographical situation after  
 8 the major climatological shifts in the eastern and western part of the basin, known as the Eastern  
 9 and Western Mediterranean Transients. During the cruise, multidisciplinary measurements  
 10 were conducted on a predominantly zonal section throughout the Mediterranean Sea,  
 11 contributing to the global GO-SHIP repeating hydrography program and adhering to the GO-  
 12 SHIP requirements.

13

14 **Data coverage and parameter measured**

15 Repository-Reference (table 1a and table 1b):

16

17 Table 1a. List of physical parameters from Maria S. Merian cruise MSM72 as seen in the

18 PANGAEA database. PI: Dagmar Hainbucher

Parameter Name	Short name	Unit	Method	Comments
<b>DATE/TIME</b>	Date/Time			Geocode
<b>LATITUDE</b>	Latitude			Geocode
<b>LONGITUDE</b>	Longitude			Geocode
<b>Pressure, water</b>	Press	dbar	CTD, SEA_BIRD SBE 911plus	
<b>Temperature, water</b>	Temp	°C	CTD, SEA_BIRD SBE 911plus	
<b>Salinity</b>	Sal		CTD, SEA_BIRD SBE 911plus	PSU
<b>Oxygen</b>	O2	µmol/l	CTD with attached oxygen sensor calibrated, corrected using Winkler titration	
<b>Pressure, water</b>	Press	dbar	UnderwayCTD (UCTD), Oceanscience	
<b>Temperature, water</b>	Temp	°C	UnderwayCTD (UCTD), Oceanscience	
<b>Salinity</b>	Sal		UnderwayCTD (UCTD), Oceanscience	PSU
<b>DEPTH, water</b>	Depth	m		



<b>Current velocity east-west</b>	UC	m/s	Shipboard Acoustic Doppler Current Profiling (SADCP)
<b>Current velocity north-south</b>	VC	m/s	Shipboard Acoustic Doppler Current Profiling (SADCP)
<b>DEPTH, water</b>	Depth	m	
<b>Current velocity east-west</b>	UC	m/s	lowered Acoustic Doppler Current Profiling (LADCP)
<b>Current velocity north-south</b>	VC	m/s	lowered Acoustic Doppler Current Profiling (LADCP)

1

2 Table 1b. List of biogeochemical parameters from Maria S. Merian cruise MSM72 as seen in  
 3 the CCHDO database. PI: Toste Tanhua

Variable	Unit
Dissolved Oxygen (O <sub>2</sub> )	μmol kg <sup>-1</sup>
Sulphurhexafluorid (SF <sub>6</sub> )	fmol kg <sup>-1</sup>
CCl <sub>2</sub> F <sub>2</sub> (CFC-12)	pmol kg <sup>-1</sup>
Nitrate (NO <sub>3</sub> <sup>-</sup> )	μmol kg <sup>-1</sup>
Nitrite (NO <sub>2</sub> <sup>-</sup> )	μmol kg <sup>-1</sup>
Phosphate (PO <sub>4</sub> <sup>2-</sup> )	μmol kg <sup>-1</sup>
Silicate (Si)	μmol kg <sup>-1</sup>
Dissolved Inorganic Carbon (DIC)	μmol kg <sup>-1</sup>
Total Alkalinity (TA)	μmol kg <sup>-1</sup>
pH	Total scale @ 25°C
Carbonate (CO <sub>3</sub> <sup>2-</sup> )	μmol kg <sup>-1</sup>
δ <sup>13</sup> C of DIC	Per mille
Total Dissolved Nitrogen (TDN)	μmol kg <sup>-1</sup>
Total Dissolve Phosphorus (TDP)	μmol kg <sup>-1</sup>
CHClF <sub>2</sub> (HCFC-22)	pmol kg <sup>-1</sup>



C <sub>2</sub> H <sub>3</sub> Cl <sub>2</sub> F (HCFC-141b)	pmol kg <sup>-1</sup>
C <sub>2</sub> H <sub>3</sub> ClF <sub>2</sub> (HCFC-142b)	pmol kg <sup>-1</sup>
CH <sub>2</sub> FCF <sub>3</sub> (HFC-134a)	pmol kg <sup>-1</sup>
C <sub>2</sub> HF <sub>5</sub> (HFC-125)	pmol kg <sup>-1</sup>
CHF <sub>3</sub> (HFC-23)	pmol kg <sup>-1</sup>

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2 <https://doi.pangaea.de/10.1594/PANGAEA.905902> (for CTD)

3 <https://doi.pangaea.de/10.1594/PANGAEA.913512> (for UCTD)

4 <https://doi.pangaea.de/10.1594/PANGAEA.913608> (for ADCP)

5 <https://doi.pangaea.de/10.1594/PANGAEA.913505> (for IADCP)

6 <https://doi.pangaea.de/10.1594/PANGAEA.905887> (for chemical data)

7 <https://doi.org/10.25921/z7en-hn85> (for pCO<sub>2</sub>)

8 A link to the summary page of the cruise MSM72 can be found in the PANGAEA data base  
9 under: <https://www.pangaea.de/?q=msm72&f.campaign%5B%5D=MSM72>

10 Coverage: 34°N-41°N, 6°W-28°E

11 Location Name: The Mediterranean Sea

12 Date/Time Start: 2. March 2018

13 Date/Time End: 3. April 2018

14

## 15 **1. Introduction**

16 Contrary to earlier ideas that the Mediterranean Sea is always in a steady state, we now know  
17 in the light of new research that the Mediterranean Sea is not but it is potentially sensitive to  
18 climatic changes (Malanotte-Rizzoli, 2014). Proof of this are the drastic changes that the eastern  
19 Mediterranean (EMed) has undergone in the past. The largest climatic event, named Eastern  
20 Mediterranean Transient (EMT), occurred in the EMed between the late 1980's and early  
21 1990's, where deep-water formation switched from the Adriatic to the Aegean Sea. This  
22 episode modified the thermohaline characteristics of the outflow through the Sicily Channel,  
23 changing the characteristics of the western Mediterranean (WMed) accordingly (Millot et al.,



1 2006, Schroeder et al., 2006). Thus, since 2005, the deep waters of the WMed have undergone  
2 significant physical changes, which are comparable to the EMT, both in terms of intensity and  
3 observed effects (Schroeder et al., 2008). This event is called the Western Mediterranean  
4 Transient (WMT). The existence of both transients contradicts the hypothesis of a steady state.  
5 On the other hand, it has also been proven that an EMT has never been observed before (Roether  
6 et al., 2013).

7 The characteristic of the Mediterranean Sea is also such that it has the potential to sequester  
8 large amounts of anthropogenic CO<sub>2</sub>, Cant, since the Mediterranean Sea has high alkalinity and  
9 temperature, which can be rapidly transported to deep by the overturning circulation (e.g.  
10 Schneider et al., 2010). The column inventories of Cant in the Mediterranean are among the  
11 highest found in the world oceans; the Mediterranean Sea thus stores a significant portion of  
12 the global anthropogenic emissions of Cant despite its relatively small volume.

13 Furthermore, marine dissolved organic carbon (DOC) represents the largest reservoir of  
14 reduced carbon ( $662 \cdot 10^{15}$  g C) on Earth (Hansell, 2009), it therefore plays a major role in the  
15 global carbon cycle. Its role in the functioning of marine ecosystems is equally crucial since  
16 DOC is released at all the levels of the food web, as a byproduct of many trophic interactions  
17 and/or metabolic processes and is the main source of energy for the heterotrophic prokaryotes  
18 (Carlson and Hansell, 2015). Although most of DOC is produced in-situ, external sources  
19 (atmosphere, rivers, sediments) may affect its concentration and distribution. Physical  
20 processes, such as deep-water formation, thermohaline circulation, vertical stratification and  
21 mesoscale activities have been reported to be the main drivers of DOC distribution in the  
22 Mediterranean Sea (Santinelli, 2015, Santinelli et al., 2015, Santinelli et al., 2013, Santinelli,  
23 2010).

24 The main scientific objective of the cruise reported here was to add knowledge to the different  
25 scales and magnitudes of variability and trends in circulation, hydrography, and  
26 biogeochemistry of the Mediterranean Sea. Key variables were measured in strategic regions  
27 in order to understand changes, the reason for occurrence, and the drivers.

28 The following science questions were addressed:

- 29 1. What are the long-term changes and/or trends in physics and biochemistry in the  
30 Mediterranean Sea, including all the sub-basins?
- 31 2. How is the hydrographic situation in the Mediterranean developing further after the EMT  
32 and WMT? Is there still a tendency of the system to return to the pre-EMT situation and is there



- 1 a similar trend in the WMed?
- 2 3. How are eddies distributed in the EMed and WMed during the cruise? Do they differ in
- 3 the sub basins? To what extent is heat and salt transferred into the vertical by eddies in the
- 4 WMed and EMed during the cruise period?
- 5 4. What is the uptake rate of the anthropogenic carbon in the Mediterranean and is this
- 6 changing over time?
- 7 5. What is the extent of the variability and trends in the inventory of biogeochemical
- 8 variables (including oxygen, nutrients and DOC)?
- 9 6. What are the baseline values of rarely measured Essential Ocean Variables (EOVs) such
- 10 as dissolved organic carbon and nitrous oxide?

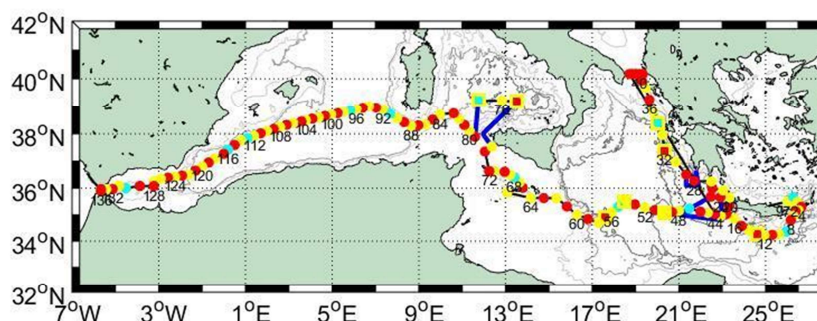
11

## 12 **2. Data Provenance**

13 The survey was carried out on the German RV Maria S. MERIAN from 2<sup>nd</sup> of March to 3<sup>rd</sup> of  
14 April 2018. The cruise started on Iraklion, Greece and ended in Cadiz, Spain. The main focus  
15 of the cruise was on an east-west transect across the Western and Eastern Mediterranean Sea  
16 (figure 1) starting east of Crete and ending near the Strait of Gibraltar, which is a repeating  
17 hydrographic line in GO-SHIP (MED1). Difficulties with diplomatic authorizations for Marine  
18 Scientific Research (MSR) in the disputed EEZ between Greek and Turkey made it impossible  
19 for us to carry out our measurements in this area, so that no data were obtained east of Kasos  
20 Strait.

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3 Figure 1: Station Map. Yellow dots: CTD without any chemical sampling, red dots: CTD with  
4 chemical sampling, cyan dots: CTD with chemical and additional sampling of isotopes, yellow  
5 squares: deployment of drifter and floats, blue lines: fine resolved uCTD and ADCP tracks.  
6 Black lines: Track with uCTD casts between CTD stations.

7

8 During the thirty-three days of the cruise we carried out measurements of hydrographic and  
9 biogeochemical variables along-track with the classical approach i.e. CTD, IADCP, uCTD  
10 instrumentation and bottle samples on highly resolved sections across the Mediterranean Sea.  
11 The high resolution of CTD stations, enhanced for the physical parameters by additional uCTD  
12 measurements, allowed us to resolve the eddy field on the sections, the analysis was also  
13 supported and complemented by satellite data.

14 Most sections and CTD-positions follow previous sampling strategies (cruise M84 and other  
15 along the GO-SHIP line MED-01, i.e. Tanhua et al., 2013) to allow long-term trend analyses.  
16 Along the different sections, CTD stations including sampling of chemical parameters were  
17 conducted approximately every 30 nm, CTD without sampling about every 15-20 nm and with  
18 even smaller spacing in the Straits. In addition, underway CTD measurements and ADCP  
19 measurements were performed between CTD stations.

20 The water sampling program included measurements of all level 1 variables as defined by GO-  
21 SHIP (i.e. oxygen, macronutrients, transient tracers and the carbonate system, <http://www.go-ship.org/DatReq.html>) and measurements of the biogeochemical EOVS <sup>13</sup>C, nitrous oxide  
22 (N<sub>2</sub>O) and dissolved organic carbon (DOC). These data were used to quantify trends and  
23



1 variability of ventilation and biogeochemical cycles, in particular uptake of anthropogenic  
2 carbon.

3 Sections were additionally conducted through the important passages: the Strait of Otranto,  
4 Kasos Strait, Antikythera Strait, Strait of Sicily and Strait of Gibraltar, in order to characterize  
5 the incoming and outgoing flows. CTD stations in the Eastern Ionian Sea were carried out to  
6 quantify the flow of the Levantine Surface Water (LSW) into the Adriatic Sea and to track the  
7 outflow of the Adriatic Deep Water (AdDW) into the Ionian.

8

### 9 **3. Methods**

#### 10 **3.1 CTD/rosette**

11 Altogether 136 CTD cast were performed from which 18 catalogued as isotopic (a full suite of  
12 observations), 65 as chemical (i.e. all GO-SHIP level 1 variables), and 59 as physical (i.e. only  
13 sampling for salinity). Due to the water amount needed, 2 casts were performed on most of the  
14 isotopic stations, the first cast was a full profile and the second a shallow one. During the  
15 physical stations water samples at 3 levels were taken for salinity analysis. The samples were  
16 then analyzed on board using a Guildline Autosol Salinometer. A total of 162 samples in 59  
17 stations were taken during the cruise with an offset with respect to standard water varying from  
18 0.0002 to 0.0030 depending on the laboratory temperature.

19 The primary CTD system (specifications see table 2) initially used on board was a Seabird  
20 SBE9plus + CTD s/n 0285 from the University of Hamburg connected to a SBE11 deck unit,  
21 configured with a 24- position SBE-32 pylon (from GEOMAR) with 10-liter Niskin bottles.  
22 Position of bottles #23 and #24 was occupied by the IADCP (specifications see table 3).  
23 Initially, the CTD was set up with two sensors for temperature and conductivity, an oxygen  
24 sensor, a fluorometer and an altimeter. To test the configuration and performance of the  
25 instrument a station was carried out on the Cretan Sea at the start of the cruise. Unfortunately,  
26 we had countless problems with instruments, sensors, cables and rosette during most of the  
27 campaign which forced us to change them very often with others available on board resulting  
28 in a continuous change of system configuration. Thus, all different configurations were  
29 carefully considered when post-processing the CTD data.

30 Temperature, salinity and pressure data were post-processed by applying Seabird software and  
31 MATLAB routines. At this stage, spikes were removed, 1 dbar averages calculated. A first  
32 attempt to assess the performance of the conductivity sensors installed on the CTD-Rosette was





1 done by comparing the salinity data with the bottle samples analyzed with the salinometer. The  
2 different hardware setups and configurations are taken carefully into account during post-  
3 processing. Overall accuracies are within the expected range of salinity (0.003).

4

5 Table 2: Used CTD instrument and sensors. Owner of instruments are either the University of  
6 Hamburg, Germany (IfM-HH), the National Institute of Oceanography and Geophysics (OGS),  
7 Italy or the property of the vessel MERIAN (MSM).

<i>Instrument/Sensor</i>	<i>Serial Number (owner)</i>
SBE 911plus / 917plus CTD	285 (IfM-HH) 806 (MSM) 807 (MSM)
Temperature 1: SBE-3-02/F	1717 (OGS) 5716 (MSM)
Conductivity 1: SBE-4-02/2	3442 (OGS) 4152 (MSM)
Temperature 2: SBE-3-02/F	1294 (IfM-HH) 5719 (MSM)
Conductivity 2: SBE-4-02/2	1106 (IfM-HH) 4159 (MSM)
Oxygen 1 SBE 43	3392 (OGS) 2417 (MSM) 0951 (MSM)
Oxygen 2 SBE 43	1761 (IfM-HH) 2418 (MSM) 0881 (MSM)
Fluorometer WETLAB SeaPoint	1755 (MSM) 1754 (MSM) SCF2874

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### 11 **3.2 Underway-CTD**

12 Underway CTDs measurements (uCTD, specifications see table 4) provide high-resolution  
13 profiles of temperature, conductivity and depth, which allow to characterize the upper ocean  
14 properties and to identify the position and characteristics of mesoscale structures. The  
15 advantage of this type of measurements is that it is not required to stop the vessel, but only to



1 maintain lower velocities (about 3 kn) during the deployments to reach greater depths. These  
2 measurements were made with an Ocean Science uCTD system.

3 The first uCTD deployment was done on March 5<sup>th</sup>, between CTD 015 and 016 stations, and  
4 we continued with this type of sampling between each CTD station to increase the sampling  
5 resolution. Unfortunately, several deployments were cancelled due to severe weather conditions  
6 and no uCTD cast was performed when the depth was shallower than 500m. Altogether 176  
7 casts were taken with depths ranging from 557 to 864 m.

8 Two probes were used during the cruise with a no time limit mode configuration (apart from  
9 the first cast configured to stop recording after 600 seconds, reaching 616 m depth) in order to  
10 get longer records. The probe tail spools were attached to the winch through a rope loop that  
11 was made new every day in the morning. Despite the probes can record several casts, data were  
12 downloaded right after each cast using a SBE software in order to avoid losing the data in case  
13 the probe was lost, and to free the memory. The probes were exchanged when the battery was  
14 running low (around 3.8V). In three occasions, no data were recorded because the magnet was  
15 taken off twice before deployment.

16 For calibration purposes, some additional casts were done right after the CTD cast in order to  
17 compare the data sets. The probes were also sent down with the starboard CTD in station 130.

18 Data files were processed using a set of Matlab routines. After extracting the downcast data, a  
19 first correction was done for removing inaccuracies in the descend rate based on the work of  
20 Ullmann and Hebert (2013). Additionally, the data were aligned to the comparable CTD data  
21 sets.

22

23 Table 3: Used UCTD sensors.

<i>Probe 1</i>	<i>Device Type</i>	<i>Serial Number (owner)</i>
0289	90745 uCTD /SBE49 FastCat CTD	702-0289 (IfM-HH)
0183	90745 uCTD /SBE 49 FastCat CTD	702-0183 (IfM-HH)

24

### 25 **3.3 IADCP Measurements**

26 The ocean current was studied by means of vertical profiles made with a IADCP-2 system  
27 (Workhorse RD Instruments type, table 3) which included two ADCPs operating at a frequency  
28 of 300 kHz, one looking upward and the other one looking downward. The system was placed  
29 in the rosette occupying the position of Niskin bottles 23 and 24. During the cruise, the IADCP  
30 batteries were changed twice: the first time on March 17<sup>th</sup> in Station 58 and the second time on



1 March 27<sup>th</sup> in Station 105. LADCP measurements were done at all CTD stations except for  
2 three (Station 73, 74, 80) with water depth less than 500 m. For these stations, the currents were  
3 observed by the ship mounted ADCP. At double (isotope) stations, LADCP profiles were only  
4 recorded from the deep cast. The gained data were processed with LDEO Matlab LADCP-  
5 processing system Version 10.15 (Turnherr, 2014). This software uses the raw LADCP data,  
6 processed CTD data and navigational data from the CTD. The resulting data are the u- and v-  
7 velocities at the depth. The bin size was set to 8m.

8

9 Table 4: Used LADCP.

Device Type	Serial Number (owner)
WHM300	Master s/n #22762 (IfM-HH)
WHM300	Slave s/n #22763 (IfM-HH)

10

### 11 Shipborne ADCP

12 During the whole campaign, underway current measurements were taken with two vessel-  
13 mounted VM-ADCPs Ocean Surveyor (ADCP) manufactured by RDI. The first, with work  
14 frequency of 75 kHz, covered approximately the top 500-700m of the water column. The  
15 number of bins was set to 100 with bin size of 8 m. The second, with work frequency of 38  
16 kHz, has a depth range of about 1600 m, set with the same bin number as the previous one and  
17 bin size of 16 m. Both instruments run in narrowband mode and were controlled by computers  
18 using the conventional RDI VMDAS software under a MS Windows system with a pinging set  
19 to fast as possible. No interferences with other used acoustical instruments were observed. The  
20 ADCP data was afterwards post-processed with the CODAS3 Software System  
21 ([https://currents.soest.hawaii.edu/docs/adcp\\_doc/](https://currents.soest.hawaii.edu/docs/adcp_doc/)), which allows extracting data, assigning  
22 coordinates, editing and correcting velocity data. Moreover, the data were corrected for errors  
23 in the value of sound velocity in water, and misalignment of the instrument with respect to the  
24 axis of the ship (about -2.8 degrees for 75 kHz ADCP and about -0.15 degrees for 38 kHz  
25 ADCP).

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### 1 **3.5 Underway CO<sub>2</sub> and O<sub>2</sub> Measurements**

2 Underway (UW) measurements of partial pressure of CO<sub>2</sub>, and dissolved oxygen  
3 concentrations in seawater were carried out by means of a Contros HydroC pCO<sub>2</sub> analyzer for  
4 pCO<sub>2</sub> and an Aanderaa optode for oxygen.

5 The instruments were placed in a cooling box in the hangar. Seawater was drawn from the  
6 ship's centrifugal pump for clean seawater that was continuously flowing through the cooling  
7 box with the inlet close to the instruments. Water was pumped through a SeaBird 5 salinity and  
8 temperature sensor and on to the HydroC instrument (Gerke et al., 2020).

9 The system operated reliably throughout the cruise, except when data acquisition was  
10 interrupted for the pCO<sub>2</sub> instrument for 2 days directly after the ship's centrifugal pump was  
11 switched off. This led to a gap 5-days period without data between March 5<sup>th</sup> and 10<sup>th</sup>. During  
12 the cruise 13 samples were taken from the cooling box for discrete measurements of pH and  
13 total alkalinity. The UW measurements started on March 2<sup>nd</sup> at 20:20 and stopped on April 1<sup>st</sup>,  
14 2018, at 14:00 (UTC).

15 The underway oxygen measurements were calibrated by comparing to the Winkler  
16 measurements taken for surface samples at the CTD stations

17

### 18 **3.6 Dissolved Oxygen**

19 Dissolved oxygen in seawater was measured at every station and depth along the cruise and  
20 reported in μmol/kg. Oxygen was measured following the automatic Winkler potentiometric  
21 method modified after Langdon (2010). Titrations were done within the sampling calibrated  
22 flasks using an Automatic Titrator Mettler Toledo T50 with a platinum combined electrode.

23 Reagents blank and Thiosulphate standardization were done daily by means of Potassium Iodate  
24 Standard 1.667 millimolar by OSIL, UK. About 1400 samples were analyzed on board. The  
25 precision of dissolved oxygen measurements was determined on five replicates, at the  
26 beginning and at the end of the cruise (table 5).

27 In addition, during the cruise 46 duplicates were analysed. The results are given in table 6.

28

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31



1 Table 5: Precision of dissolved oxygen. (STD = standard deviation, CV = Coefficient of  
 2 Variation)

Parameter	<i>Beginning of the cruise</i>			<i>End of the cruise</i>		
	Mean μM	STD μM	CV%	Mean μM	STD μM	CV%
DISSOLVED OXYGEN	196.07	0.13	0.07	198.84	0.14	0.07

3

4 Table 6: Results of duplicates. <sup>(1)</sup>AD=|duplicate #1 – duplicate #2|; <sup>(2)</sup>RPD%=Absolute  
 5 Difference \*100/mean (dupl. #1, #2).

Parameter	Range μM	mean Absolute Difference <sup>(1)</sup> μM	mean Relative Percentage Difference <sup>(2)</sup>
DISSOLVED OXYGEN	179-240	0.18	0.09

6

7

8 **3.7 Nutrients (nitrite, nitrate, phosphate, and silicate), Total Dissolved Nitrogen**  
 9 **(TDN) and Total Dissolved Phosphorus (TDP).**

10

11 ***Nutrients***

12 Analyses were performed at 40 °C on a four-channel, Quattro SEAL Analytical Continuous  
 13 Flow Analyzer s/n 8014549; [https://www.seal-analytical.com/Products/SegmentedFlow](https://www.seal-analytical.com/Products/SegmentedFlowAnalyzers/QuAAtro39AutoAnalyzer/tabid/814/language/en-US/Default.aspx)  
 14 [Analyzers/QuAAtro39AutoAnalyzer/tabid/814/language/en-US/Default.aspx](https://www.seal-analytical.com/Products/SegmentedFlowAnalyzers/QuAAtro39AutoAnalyzer/tabid/814/language/en-US/Default.aspx), according to  
 15 Hansen and Koroleff (1999). Nitrite was determined through the formation of a reddish-purple  
 16 azo dye, and measured at 520 nm (SEAL Method No. Q-030-04 Rev. 2). Nitrate was reduced  
 17 to nitrite in a copperized cadmium reduction coil and then determined as described for nitrite  
 18 (SEAL Method No. Q-035-04 Rev. 4). The determination of phosphate was based on the  
 19 reduced blue phospho-molybdenum complex, then measured at 880 nm (SEAL Method No. Q-  
 20 031-04 Rev. 1). Silicate was determined by means of acidic reduction of silicomolybdate to  
 21 molybdenum blue, then measured at 820 nm (SEAL Method No. Q-038-04 Rev. 0).



- 1 About 1400 nutrient samples were analyzed on board. The onboard precision of nutrient
- 2 measurements was determined on five replicates, at the beginning and at the end of the cruise.
- 3 The results are shown in table 7.
- 4 In addition, during the cruise 140 duplicates were analysed. The results are shown in table 8.
- 5 An internal quality check was daily performed by means of analyses of QUASIMEME samples
- 6 which provided results within the already certified ranges.
- 7 Table 7: On board precision of nutrient measurements

Parameter	Beginning of the cruise			End of the cruise		
	Mean	STD	CV%	Mean	STD	CV%
	$\mu\text{M}$	$\mu\text{M}$		$\mu\text{M}$	$\mu\text{M}$	
NITRITE (1)	0.01	0.01	100	0.03	0.01	56.5
NITRITE + NITRATE	4.94	0.01	0.2	9.01	0.02	0.2
PHOSPHATE	0.18	0.01	5.5	0.41	0.01	3.1
SILICATE	8.34	0.03	0.3	9.55	0.04	0.5

8  
9  
10  
11  
12  
13  
14



- 1 Table 8: Analysis of duplicates. (1)AD=|duplicate #1 – duplicate #2|; (2) RPD%=Absolute
- 2 Difference \*100/mean (dupl. #1, #2); (3) Nitrite statistics was given just for completeness,
- 3 since the concentration levels recorded were too low, often below the detection limit.

Parameter	Range μM	mean Absolute Difference (1) μM	mean Relative Percentage Difference (2)
NITRITE (3)	0-0.19	0.01	48.77
NITRITE+NITRATE	0.33-9.86	0.02	0.42
PHOSPHATE	0-0.47	0.01	5.13
SILICATE	0.93-11.00	0.04	0.72

4  
5

### 6 ***TDN and TDP***

7 About 550 samples for Total Dissolved Nitrogen and Total Dissolved Phosphorus (TDN and  
 8 TDP) on land-based laboratory analyses were collected and frozen at -20°C after filtration on  
 9 pre-combusted GF/F filter. The dissolved organic components, Dissolved Organic Nitrogen  
 10 (DON) and Dissolved Organic Phosphorus (DOP) were subsequently calculated by subtracting  
 11 their mineral constituents (NO<sub>3</sub>+NO<sub>2</sub>) and PO<sub>4</sub>, respectively.

12

### 13 **3.8 Discrete CO<sub>2</sub> System Measurements**

14 Discrete CO<sub>2</sub> variables were measured on board, being Dissolved Inorganic Carbon (DIC), pH,  
 15 Total Alkalinity (TA) and carbonate ion (CO<sub>3</sub><sup>2-</sup>) at selected stations and depths (table 9). In  
 16 addition, discrete samples for DIC, pH and TA were analyzed specifically from surface Niskin  
 17 bottles to be compared with the continuous water supply feeding the underway partial pressure  
 18 of CO<sub>2</sub> (pCO<sub>2</sub>) system in determined stations. For further details see Hainbucher et al. (2018).

19



1 Table 9: Total number of CO<sub>2</sub> system samples analyzed during the MSM72 cruise. Total  
2 number of fired bottles 1723.

	DIC	pH	TA	CO <sub>3</sub> <sup>2-</sup>	Surface
Sample s	47 9	116 0	94 9	391	22

3

#### 4 **DIC**

5 Samples for DIC were collected following transient tracers and dissolved oxygen, in 500 ml  
6 borosilicate bottles following standard procedures. No poison was added. Samples were left at  
7 room temperature in the dark until analysis, maximum 48 hours after collection. DIC samples  
8 were analyzed with a MARIANDA VINDTA 3D system coupled with a UIC 5011 coulometer.  
9 This analysis overall consists of extracting seawater CO<sub>2</sub> from a known volume of sample by  
10 adding phosphoric acid, followed by coulometric detection (Johnson et al., 1993). No  
11 calibration unit was available for the system. A new coulometric cell was prepared for every  
12 batch of analysis and the accuracy of the DIC measurements was assessed by using Certified  
13 Reference Material (CRM #158 & #170 provided by Prof. Dickson, UCSD). The calibration  
14 factor obtained from the CRM was used for adjusting the final DIC of each sample measured  
15 in the corresponding batch of analysis. In addition, substandard seawater (stabilized seawater  
16 from the Cretan Sea 700m salinity minimum, stored in the dark in a 30 L container) was  
17 analysed at the beginning and end of the batch analysis as a secondary quality control. The  
18 precision of the DIC measurements was checked by: 1) double analysis from the same sample  
19 and 2) replicate analysis from 4 to 5 samples collected from the same Niskin bottle. The  
20 precision is estimated to be 1 μmol kg<sup>-1</sup> and the accuracy 2 μmol kg<sup>-1</sup>.

21

#### 22 **pH**

23 Seawater spectrophotometric pH was measured following Clayton and Byrne (1993) at almost  
24 all depths in the chemical and isotope stations during the MSM72 cruise (Table 1). This method  
25 consists on adding a volume of indicator solution to the seawater sample, so that measuring the  
26 absorbance of the sample at different wavelengths and obtaining the ratio between two of the  
27 wavelengths absorbance is proportional to the sample pH. The indicator was a 2 mM solution  
28 of unpurified m-cresol purple (Sigma Aldrich®) prepared in seawater and maintained at dark,  
29 with no air contact (Absorbance Ratio 1.30). Samples were taken following standard procedures  
30 immediately after DIC and directly into cylindrical 10 cm path length optical glass cells. The





1 cells were thermostated at  $25 \pm 0.2^\circ\text{C}$  during one hour before analysis. Absorbance  
2 measurements were obtained in the thermostated chamber of a double beam UV 2600 Shimadzu  
3 spectrophotometer. The equipment was checked before the cruise for the absorbance and  
4 wavelength accuracy using holmium standards. pH values on the total scale were calculated  
5 and referred at  $25^\circ\text{C}$  by using the formula by Clayton and Byrne (1993). The injection of the  
6 indicator in the sample slightly changes the sample pH. Following standard operating  
7 procedures, double additions of the indicator were performed over a pH gradient in order to  
8 obtain the corresponding correction (Hainbucher et al., 2018). The pH accuracy was controlled  
9 measuring TRIS buffer solution samples (batch #72, provided by Prof. Dickson, UCSD). TRIS  
10 samples were stabilized at three different temperatures covering the pH range found during the  
11 MSM72 cruise. Differences between measured and theoretical TRIS pH varied between 0.009  
12 to 0.005. The pH precision was checked by replicate analysis from cells collected at the same  
13 Niskin from surface and deep waters. The precision is estimated to be 0.0004 pH units and the  
14 accuracy 0.005 pH units. During the cruise some samples were also analyzed with purified m-  
15 cresol purple provided by Prof. Byrne (USC).

16

## 17 **TA**

18 TA was analysed following a double end point potentiometric technique by Pérez and Fraga  
19 (1987) further improved by Pérez et al. (2000). This technique is faster than the whole curve  
20 titration, with comparable results (Mintrop et al., 2000). TA was measured at most stations and  
21 depths (Table 1). Seawater samples for TA were collected after pH samples in 600 ml  
22 borosilicate bottles following standard procedures. Samples were left at room temperature in  
23 the dark until analysis, maximum 48 hours after collection. TA was measured by titration with  
24 0.1 N hydrochloric acid dispensed with an automatic potentiometric titrator, Titrand  
25 Metrohm®, provided with a combination glass electrode coupled with a temperature probe. The  
26 electrode was standardized using a 4.41 pH ftalate buffer made in  $\text{CO}_2$  free seawater. The TA  
27 accuracy was assessed with  $\text{CO}_2$  CRM (batch #170, provided by Prof. Dickson, UCSD) In  
28 addition to the CRM calibration, a drift control was conducted by analyzing substandard  
29 seawater (big volume of seawater stored in the dark, as for DIC) at the beginning and at the end  
30 of the analysis session. Each sample was measured twice and the mean value is reported, with  
31 the mean standard deviation of all duplicate differences being  $0.6 \mu\text{mol kg}^{-1}$ . In addition, typical  
32 reproducibility analysis were performed from samples collected from the same niskin bottle at  
33 different stations along the cruise. The TA precision is estimated to be  $1 \mu\text{mol kg}^{-1}$  and the  
34 accuracy  $2 \mu\text{mol kg}^{-1}$ .



1 **CO<sub>3</sub><sup>2-</sup>**

2 The CO<sub>3</sub><sup>2-</sup> ion concentration was determined spectrophotometrically following Byrne and Yao  
3 (2008) incorporating the recent improvements by Patsavas et al. (2015), at selected stations and  
4 depths (Table 1) Samples for CO<sub>3</sub><sup>2-</sup> were collected after TA following the same procedure as  
5 for pH but within cylindrical optical quartz 10 cm pathlength cuvettes. The cells were stabilized  
6 at 25°C for one hour before the analysis, maximum 24 hours after collection. A solution of  
7 0.022 M of Pb(ClO<sub>4</sub>)<sub>2</sub> was added to the seawater sample and the PbCO<sub>3</sub> complex formed  
8 afterwards was detected spectrophotometrically in the UV spectra. Absorbance measurements  
9 were obtained in the thermostated chamber of a double beam UV 2600 Shimadzu  
10 spectrophotometer. The equipment was checked before the cruise for the absorbance and  
11 wavelength accuracy width using holmium standards. The CO<sub>3</sub><sup>2-</sup> in μmol kg<sup>-1</sup> is the  
12 concentration of ion carbonate at 25°C calculated using the formula by Patsavas et al.  
13 (2015). The CO<sub>3</sub><sup>2-</sup> precision was checked by replicate analysis from cells collected at the same  
14 niskin from surface and deep waters. It is estimated to be 1 μmol kg<sup>-1</sup>.

15

16 **3.9 Measurements of CFC-12 and SF<sub>6</sub>**

17 During the cruise, one gas chromatograph purge-and-trap (GC/PT) system was used for the  
18 measurements of the transient tracers CFC-12 and SF<sub>6</sub>. The system is modified versions of the  
19 set-up normally used for the analysis of CFCs (Bullister and Weiss, 1988). All samples were  
20 collected in 250 mL ground glass syringes, of which an aliquot about 200 mL was injected to  
21 the purge-and-trap system, normally within 5 hours from sampling.

22 The traps consisted of 100 cm 1/16" tubing packed with 70cm Heysep D kept at temperatures  
23 between -70 and -75°C during trapping. The traps were desorbed by heating to 120°C and  
24 passed onto the pre-column. The pre-column consisted of 20 cm Porasil C followed by 20 cm  
25 Molsieve 5A in a 1/8" stainless steel column. The main column was a 1/8" packed column  
26 consisting of 180 cm Carbograph 1AC (60-80 mesh) and a 50 cm Molsieve 5A post-column.  
27 Both columns were kept isothermal at 60°C. Detection was performed on an Electron Capture  
28 Detector (ECD).

29 Standardization was performed by injecting small volumes of gaseous standard containing  
30 CFC-12 and SF<sub>6</sub>. This working standard was prepared by the company Dueste-Steiniger (DS1.).  
31 The CFC-12 and SF<sub>6</sub> concentrations in the working-standard has been calibrated vs. a reference  
32 standard obtained from R.F Weiss group at SIO, and the CFC-12 data are reported on the SIO98  
33 scale. Calibration curves were measured roughly once a week in order to characterize the non-



1 linearity of the system, depending on workload and system performance. Point calibrations  
2 were always performed between stations to determine the short-term drift in the detector.  
3 Replicate measurements were taken except for near coastal stations due to high workload. To  
4 assess the reproducibility of the set-up, 50 replicates samples were run, and resulted in a  
5 reproducibility of 1.0 % or 0.01 pmol kg<sup>-1</sup> for CFC-12 and 2.3% or 0.03 fmol kg<sup>-1</sup> for SF<sub>6</sub>. In  
6 total we successfully measured 1084 samples on 68 stations for transient tracers.

7 In addition to the on-board analysis, on three stations (#52, #84, and #106) 1500 ml glass  
8 ampoules were flame sealed for later analysis in the lab in Kiel for the detection of novel  
9 halogenated tracers such as HFC134a and HCFC22 (Li and Tanhua, 2019).

10

### 11 **3.10 Dissolved Organic Carbon (DOC)**

12 Seawater samples for DOC were collected from the CTD-Rosette into 250 ml Polycarbonate  
13 Nalgene bottles. Samples were filtered through a 0.2 μm Nylon filter under high-purity air  
14 pressure. Filtered samples were collected in 60 ml Nalgene bottles, acidified and stored at 4°C  
15 and in the dark.

16 DOC measurements were carried out with a Shimadzu Total Organic Carbon analyzer (TOC-  
17 Vcsn), by high temperature catalytic oxidation. Samples were acidified with HCl 2N and  
18 sparged for 3 minutes with CO<sub>2</sub>-free pure air, in order to remove inorganic carbon. From 3 to 5  
19 replicate injections were performed until the analytical precision was lower than 1% (± 1 μM).  
20 A five-point calibration curve was done by injecting standard solutions of potassium hydrogen  
21 phthalate in the expected concentration range of the samples. At the beginning and end of each  
22 analytical day the system blank was measured using low carbon water (LCW) and the reliability  
23 of measurements was controlled by comparison of data with a DOC reference (CRM) seawater  
24 sample kindly provided by Prof. D.A. Hansell of the University of Miami  
25 (<http://yyy.rsmas.miami.edu/groups/biogeochem/CRM.html>).

26 In total 650 samples were collected in 38 stations. Samples were collected at the following  
27 depths: 10, 25, 50, 75, 100, 150, 200, 300, 400, 500, 750, 1000 and every 250 m until the  
28 bottom.

29

### 30 **3.11 Chromophoric dissolved organic matter (CDOM)**

31 Chromophoric dissolved organic matter (CDOM) is the fraction of DOM that absorbs light at  
32 visible and ultraviolet (UV) wavelengths. It plays a key role in the marine ecosystem by



1 regulating light penetration into the water column (Nelson and Siegel, 2013) and preventing  
2 cellular DNA damage (Herndl et al., 1993; Häder and Sinha, 2005). A fraction of CDOM re-  
3 emit part of the absorbed light and is called fluorescent DOM (FDOM). The study of the  
4 absorption properties of CDOM, together with the analysis of the excitation-emission matrixes  
5 (EEMs) through the parallel factorial analysis (PARAFAC) can give qualitative information on  
6 the different groups of chromophores (protein-like, humic-like and PAH-like) present in the  
7 DOM pool, their changes due to photodegradation and/or microbial transformation, the main  
8 sources of CDOM and an indirect estimation of its molecular weight and aromaticity degree  
9 (Stedmon and Nelson, 2015, Retelletti et al., 2015, Gonelli et al., 2016, Margolin et al., 2018).  
10 The CDOM data collected during the MSM72 cruise will represent an unique opportunity to:  
11 (i) Compare CDOM optical properties in the different water masses of the Mediterranean Sea  
12 with those collected in the Geotraces cruise (Spring-summer 2013) and to relate them to the  
13 different trophic conditions of the basin; (ii) Study the relationship between DOC and CDOM  
14 in the surface, intermediate and deep waters.

15

### 16 **3.12 Sampling for Measurements of Stable Carbon Isotopes on Dissolved** 17 **Inorganic Carbon (DIC)**

18 Samples for the determination of stable carbon isotopes ( $\delta^{13}\text{C}$ ) of Dissolved Inorganic Carbon  
19 (DIC) were taken on 11 stations (the “isotope stations”, normally performed as a double cast)  
20 in the various basins along the cruise-track. In total 214 samples were taken in 100 ml dark  
21 glass bottles immediately poisoned with 100  $\mu\text{L}$  saturated mercury chloride. The samples were  
22 measured off-line during fall of 2018 at the Centre for Isotope Research (CIO), Energy and  
23 Sustainability Research Institute Groningen (ESRIG), University of Groningen.

24

### 25 **3.13 $\text{NO}_3^-$ isotopes ( $\delta^{15}\text{N}$ & $\delta^{18}\text{O}$ )**

26 Samples for nitrogen (N) and oxygen (O) isotopes in nitrate ( $\text{NO}_3^-$ ) and nitrate+nitrite ( $\text{NO}_3^-$   
27  $+\text{NO}_2^-$ ) analysis were collected at 44 stations evenly distributed along the transect. In total, 790  
28 samples have been collected. High-resolution  $\text{NO}_3^-$   $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  measurements represent a  
29 powerful tool to unravel the sources and sinks of reactive (i.e., fixed) N at the scale of the  
30 Mediterranean Sea. Complemented with coral-bound  $\delta^{15}\text{N}$  records covering the last centuries,  
31 these measurements may also shed light on the contribution of industrially fixed N to the  
32 reactive N budget, by revealing the large-scale systematics required to interpret the records back  
33 in time.



1 Unfiltered samples for N and O isotopic composition of  $\text{NO}_3^-$  were collected in 60 mL plastic  
2 bottles and stored frozen ( $-20^\circ\text{C}$ ) until analysis.  $\text{NO}_3^- + \text{NO}_2^-$   $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  will be measured  
3 (2019-2020) at the Max Planck Institute using the denitrifier method (Sigman et al., 2001;  
4 Casciotti et al., 2002). Briefly, 3-20 nmol of  $\text{NO}_3^- + \text{NO}_2^-$  is quantitatively converted to  $\text{N}_2\text{O}$  gas  
5 by denitrifying bacteria (*Pseudomonas aureofaciens*) that lack an active  $\text{N}_2\text{O}$  reductase. The  
6  $\text{N}_2\text{O}$  is then analysed by gas chromatography-isotope ratio mass spectrometer (GC-IRMS;  
7 MAT253, Thermo) with on-line cryo-trapping (Weigand et al., 2016). Measurements are  
8 referenced to air  $\text{N}_2$  for  $\delta^{15}\text{N}$  and VSMOW for  $\delta^{18}\text{O}$  using the nitrate reference materials IAEA-  
9  $\text{NO}_3$  and USGS-34. For  $\text{NO}_3^-$   $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  analysis,  $\text{NO}_2^-$  is removed with the sulfamic acid  
10 method prior to the isotopic analysis (Granger and Sigman, 2009). The reproducibility is  
11 generally better than 0.1‰ for  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$ , respectively.

12

### 13 **3.14 LISST – DEEP**

14 The LISST-Deep instrument obtains in-situ measurements of particle size distribution, optical  
15 transmission, and the optical volume scattering function (VSF) at depths down to 3,000 meters.  
16 It is manufactured by Sequoia Inc., and owned by the Hellenic Centre for Marine Research  
17 (HCMR) – Greece.

18 Using a red 670nm diode laser and a custom silicon detector, small-angle scattering from  
19 suspended particles is sensed at 32 specific log-spaced angle ranges. This primary measurement  
20 is post-processed to obtain sediment size distribution, volume concentration, optical  
21 transmission, and volume scattering function. The LISST-Deep s/n 4004 is categorized as a  
22 type B instrument, which means that the range of particles it measures ranges from 1.25  $\mu\text{m}$  to  
23 250  $\mu\text{m}$ . The LISST-Deep must be powered externally at all times. This is typically achieved  
24 by connecting it to a rosette, getting power from the main CTD unit.

25 Parameters measured during the cruise were:

- 26 • Particle size distribution from 1.25-250 $\mu\text{m}$  or 2.5-500 $\mu\text{m}$
- 27 • Depth (3000 m max depth @ 0.8 m resolution)
- 28 • Optical transmission @ 0.1 % resolution
- 29 • Beam attenuation Coefficient @ 0.1  $\text{m}^{-1}$  resolution
- 30 • Volume concentration @ 0.1  $\mu\text{l/l}$  resolution
- 31 • Volume scattering function (VSF)



1 The measurement of these parameters provided important information on the number, size and  
2 quality (phytoplankton, sediment, etc.) of the suspended matter in the water column. Further  
3 information for the determination of water masses was provided by the estimation of the  
4 intrinsic optical properties. Finally, for the first ~ 100m we estimated the color of the sea and  
5 compared this estimation with satellite images, providing valuable information for the  
6 calibration of satellite algorithms.

7 For the cruise MSM72 the sampling of these optical estimates is in itself an important  
8 achievement because, for the first time LISST – DEEP was used to record data in a transect  
9 over the full length of the Mediterranean Sea. Furthermore, the estimation of these parameters  
10 combined with POC - PON estimation, and other physical and chemical parameters, improve  
11 the study of the dynamics of the Mediterranean Sea.

12 In general, the use of LISST – DEEP during the cruise follows the standard methods which are  
13 provided by Sequoia Inc, but with one important difference. For the estimation of the above  
14 parameters the use of a background file is required for normalization purposes. This file is  
15 normally produced in laboratory conditions with mili – q 2 filtered water. However, experience  
16 until now has proved that especially in the eastern Mediterranean Sea (which is characterized  
17 as ultra-oligotrophic) the use of this background file leads us to an overestimation of the  
18 parameters and especially of the beam attenuation coefficient. Therefore, during this cruise we  
19 used a sampled in situ background file chosen as the minimum of the sum of the digital counts  
20 in the 32 rings and where the LaserPower to LaserReference ( $L_p/L_r$ ) ratio is maximum.

21 The main problem which we faced was the frequent change of the CTD main unit and the  
22 different cables that we had to use for the instrument connection to the CTD. Fortunately, with  
23 the most valuable help of the cruise technician we managed to deploy the LISST – DEEP as  
24 much as possible. Additionally, the maximum depth limitation of the instrument (3000m)  
25 enforced us to remove it in deep casts achieving a total of 54 stations.

26

#### 27 **4. Discussion and Conclusion**

28 Discussion and conclusion will focus in this publication on the quality of the data of MSM72  
29 cruise. We will concentrate here on the basic physical and biogeochemical parameters, as  
30 selected examples, to show the relevance of the sampled data and so as to be able to answer the  
31 questions on the scale and variability of the circulation and biogeochemical cycle in the  
32 Mediterranean Sea (see Introduction).

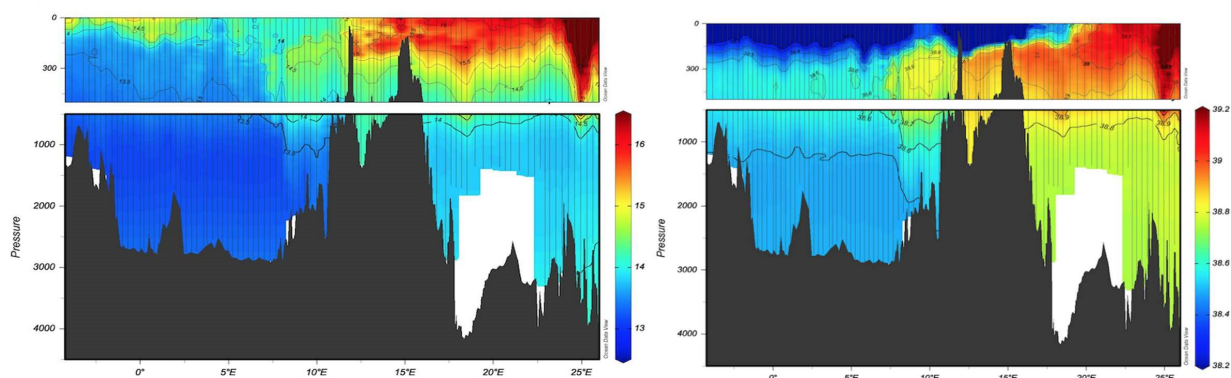


1

## 2 4.1 Physical parameters

3 The west east section (figure 2) is a typical example for the distribution of temperature and  
4 salinity in the Mediterranean Sea showing the different heat and salt content between the  
5 western and eastern basin. A clear intrusion of the salty Levantine Intermediate Water (LIW)  
6 from east to west in the first 500m is depicted while the low salinity Atlantic Water (AW)  
7 protrudes eastwards creating a front at about 20-22°E.

8



9

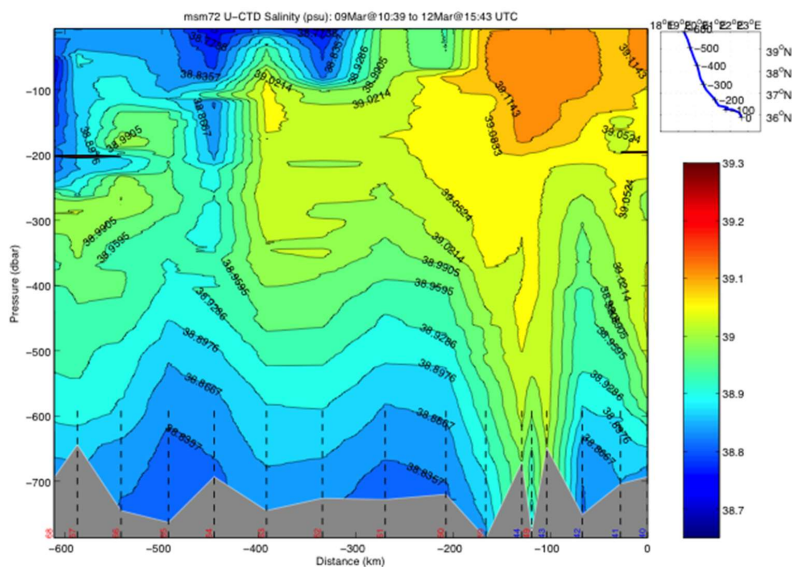
10 Figure 2: West-east temperature (left) and salinity (right) sections through the Mediterranean  
11 Sea.

12

13 The underway CTD data are a valuable addition to the classical CTD data. They enhance the  
14 resolution of data in the horizontal scale and give insight in eddy activity. Although the data do  
15 not reach to the bottom, the vertical resolution with about 1000 m is useful to characterize scales  
16 relevant for the LIW transport.

17 The uCTD salinity distribution of figure 3, located along the easternmost part of the northward  
18 transit in the Ionian Sea, shows that the Pelops gyre is well resolved.

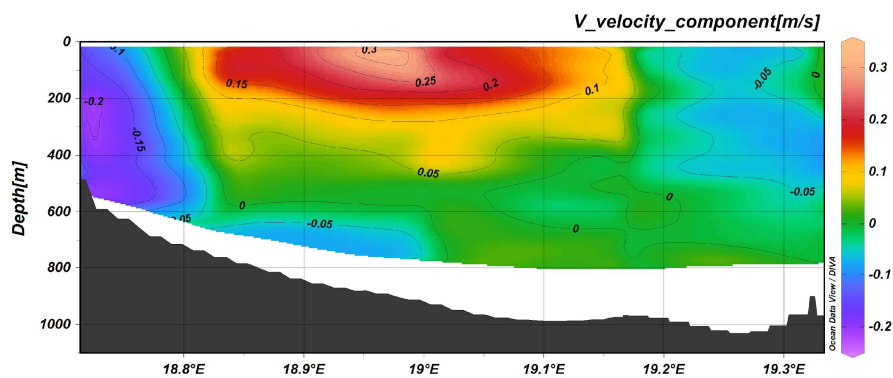




1

2 Figure 3: uCTD salinity transect. Location is shown in the upper right panel.

3 Considering the route of the ship during the cruise, it was possible to identify different ADCP  
4 transects that correspond to areas with the most important water mass dynamics. In particular  
5 the most important sections were: gyre activity in the area west of Crete and south of  
6 Peloponnese, the west Cretan, Otranto (figure 4) and Sicily Straits, the east boundary of the  
7 Ionian Sea and the west-east Mediterranean transect. The north-south current component (figure  
8 4) in the Strait of Otranto clearly shows the outflow of the Adriatic Deep Water (AdDW) along  
9 the western part while in the upper and intermediate layer of the central part the inflow of the  
10 Levantine Intermediate Water (LIW) proceeds.



11

12 Figure 4: Transect across the Strait of Otranto from ADCP 38, positive numbers correspond to  
13 northward currents.



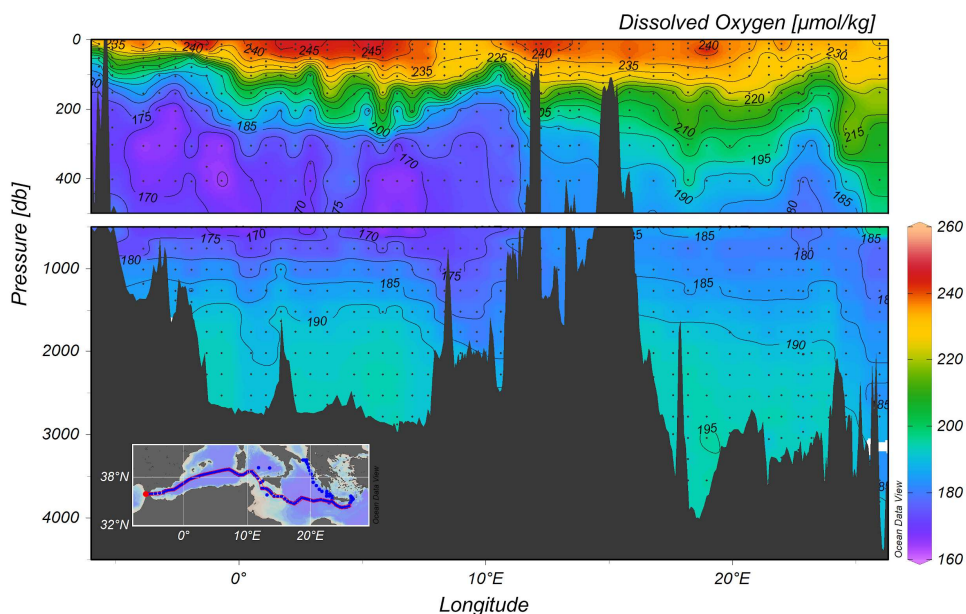


1

2

### 3 **4.2 Biogeochemical parameters**

4 The vertical distribution of dissolved oxygen along a section from the Cretan Sea to Gibraltar,  
5 including part of the Cretan Passage and the southern Ionian is shown in figure 5. This section  
6 shows the Oxygen Minimum Layer (<180  $\mu\text{mol/kg}$ ) which occupies the layer 500-1500m.  
7 Increased oxygen towards the bottom indicate the ventilation of deep water in the  
8 Mediterranean. The western part of the Ionian Sea appears to be better oxygenated than the  
9 eastern part due to the spreading of newly ventilated dense water from the Adriatic Sea via the  
10 Otranto Strait – a feature that is observed in the transient tracer section as well.

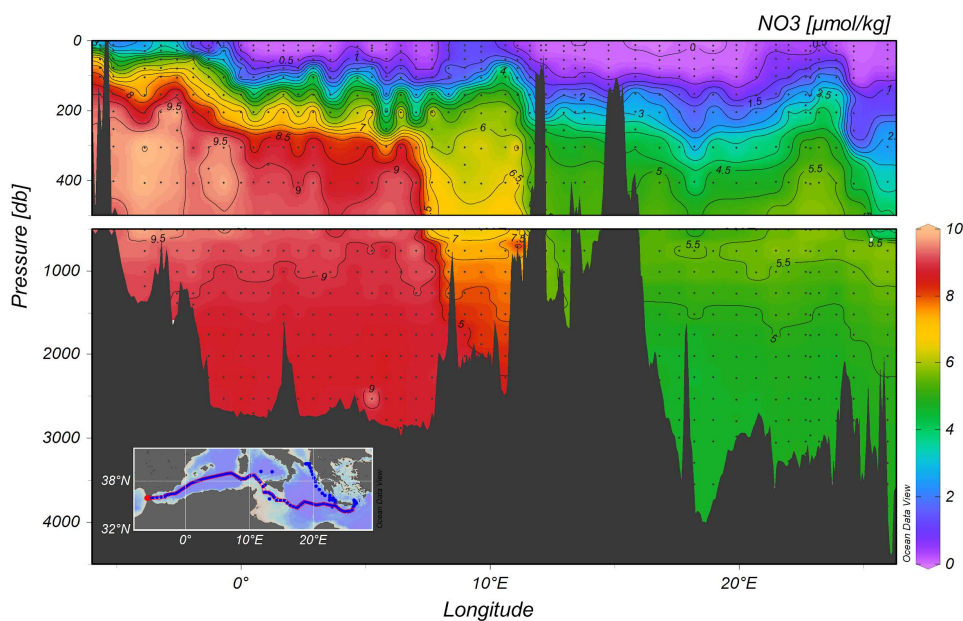


11

12 Figure 5: Distribution of dissolved oxygen along the trans-Mediterranean section.

13

14 Figure 6 illustrates the distribution of nitrate along the quasi-zonal section. Interesting features  
15 include: the maximum nutrient layer in the range of depth of 500-1500 m which is co-located  
16 to the minimum of transient tracers; the deepest layer shows an homogeneous distribution of  
17 nutrients and the nutrient impoverished upper layer is, not yet completely depleted of nutrients,  
18 likely do to subject to mesoscale dynamics (as, for example, south of Crete).

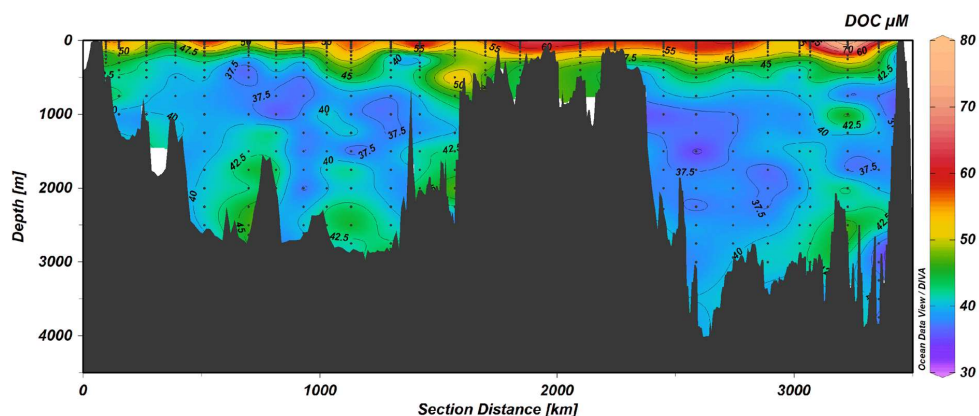


1

2 Figure 6: Distribution of nitrate along the trans-Mediterranean section.

3 The DOC data collected during the MSM72 cruise represents an unique opportunity to (i)  
4 investigate the long-term variation in DOC distribution in intermediate and deep waters on a  
5 basin scale; (ii) quantify the role of DOC in C export and sequestration in the Mediterranean  
6 Sea; (iv) estimate DOC mineralization rates; (v) asses the functioning of microbial loop in the  
7 different areas of the Mediterranean Sea.

8 DOC concentrations range between 34 and 80  $\mu\text{M}$  (figure 7). The highest values ( $> 50 \mu\text{M}$ )  
9 were observed in the upper 200 m, with a marked increase moving eastward. This feature has  
10 already been observed in the Med Sea (Santinelli, 2015; Santinelli et al., 2012) and can be  
11 explained by different processes such as nutrient limitation of heterotrophic prokaryotes  
12 growth, the occurrence of recalcitrant DOC that is not available for consumption. The high  
13 stratification, occurring in the easternmost stations, makes DOC accumulation more visible. A  
14 different functioning of the microbial loop has been reported for the western and eastern  
15 Mediterranean Sea and these data support that DOC dynamics in the surface layer of the two  
16 sub-basins is different. The lowest concentrations ( $< 40 \mu\text{M}$ ) are between 1000 and 2000 m,  
17 in the bottom waters a slight increase in DOC can be observed. This feature, already reported for  
18 the Mediterranean Sea, can be explained by the export of the DOC accumulated in the surface  
19 layer by deep water formation (Santinelli, 2015 and references herein).



1

2 Figure 7: DOC vertical distribution along the trans-Mediterranean section

3

#### 4 **5. Data access**

5 Data are published at the information system PANGAEA and CCHDO;

6

#### 7 **6. Acknowledgements**

8 We thank Captain Björn Maaß, his officers and the crew of R/V MARIA S. MERIAN for the  
9 support of our scientific programme, for their unending competent and friendly help.

10 The financial support for the cruise was provided by the project of the “Deutsche  
11 Forschungsgemeinschaft” U4600DFG040204. We gratefully acknowledge their support.

12 The CO<sub>2</sub> team was funded by an internal IEO grant MEDSHIP18, A.E.R. Hassoun was funded  
13 by a POGO grant. The OGS team was funded by an internal grant.

14

#### 15 **7. References**

16 Bullister, J. L., and Weiss, R. F.: Determination of CCl<sub>3</sub>F and CCl<sub>2</sub>F<sub>2</sub> in seawater and air,  
17 Deep-Sea Res., 35, 839-853. 1988.

18 Byrne, R. H., and Yao, W.: Procedures for measurement of carbonate ion concentrations in  
19 seawater by direct spectrophotometric observations of Pb(II) complexation. Mar. Chem., 112  
20 (1–2), 128–135, 2008.

21 Carlson, C.A. and Hansell, D.A.: Biogeochemistry of Marine Dissolved Organic Matter 2<sup>nd</sup> Ed.,  
22 66-126, 2015.



- 1 Casciotti, K.L., D.M. Sigman, M. Galanter Hastings, J.K. Böhlke, and A. Hilkert: Measurement  
2 of the oxygen isotopic composition of nitrate in seawater and freshwater using the denitrifier  
3 method. *Anal. Chem.* 74: 4905-4912, 2002.
- 4 Gerke, L.: pCO<sub>2</sub> in the Mediterranean Sea during the cruise MSM72. Master thesis, Department  
5 of Chemistry, Christian-Albrechts-University Kiel, 2020
- 6 Clayton, T.D. and Byrne, R.H.: Spectrophotometric seawater pH measurements: total hydrogen  
7 ion concentration scale concentration scale calibration of m-cresol purple and at-sea results.  
8 *Deep-sea Res. I*, 40, 10, 2115-2129, 1993.
- 9 Gonnelli, M., Galletti, Y., Marchetti, E., Mercadante, L., Brogi, S.R., Ribotti, A., Sorgente, R.,  
10 Vestri, S., Santinelli, C.: Dissolved organic matter dynamics in surface waters affected by oil  
11 spill pollution: Results from the serious game exercise, *Deep Sea Res. Part II Top. Stud.*  
12 *Oceanogr.*, 133 88–99, 2016.
- 13 Granger, J., and D.M. Sigman. 2009. Removal of nitrite with sulfamic acid for nitrate N and O  
14 isotope analysis with the denitrifier method. *Rapid Commun. Mass Spectrom.* 23: 3753-3762,  
15 doi:10.1002/rcm.4307.
- 16 Häder, D.-P. and Sinha, R. P.: Solar ultraviolet radiation-induced DNA damage in aquatic  
17 organisms: potential environmental impact. <https://doi.org/10.1016/j.mrfmmm.2004.11.017>,  
18 2005.
- 19 Hainbucher, D., Álvarez, M., Astray, B., Bachi, G., Cardin, V., Celentano, P., Chaikakis, S.,  
20 Chaves Montero, M. M., Civitarese, G., El Rahman Hassoun, H., Fajar, N. M., Fripiat, F.,  
21 Gerkel., Gogou, A., Guallart, F., Gülk, B., Lange, N., Rochner, A., Santinelli, C., Schroeder,  
22 K., Steinhoff t., Tanhua, T., Urbini, L., Velaoras, D., Wolf, F. and Welsch, A.: Variability and  
23 Trends in Physical and Biogeochemical Parameters of the Mediterranean Sea.  
24 <http://epic.awi.de/47567/2/msm71-74-expeditionsheft.pdf>, 2018.
- 25 Hansell, D.A., Carlson, C.A., Pepeta, D.I. and Schlitzer, R.: Dissolved Organic Matter in the  
26 Ocean: A controversy stimulates new insights. *Oceanog.* 22, 4, 202-211, 2009.
- 27 Hansen, H. P. & Koroleff e.: Determination of nutrients. In Grasshoff K., K. Kremling & M.  
28 Ehrhardt (eds), *Methods of seawater analysis*. Wiley VCH, Weinheim: 159–228, 1999.
- 29 Herndl, G.J., Müller-Niklas, G. and Frick, J.: Major role of ultraviolet-B in controlling bacterio  
30 plankton growth in the surface layer of the ocean. *Nature* 361:717-719, 1993.



- 1 Johnson, K.M., Wills, K.D., Butler, D.B., Johnson, W.K., Wong, C.S., (1993). Coulometric  
2 total carbon dioxide analysis for marine studies: maximizing the performance of an automated  
3 gas extraction system and coulometric detector. *Marine Chemistry*, 44, 167-187.
- 4 Langdon, C.: Determination of dissolved oxygen in seawater by winkler titration using the  
5 amperometric technique. IOCCP Report N°14, ICPO publication series N 134, 2010.
- 6 Li, P., Tanhua, T.: Medusa-Aqua system: simultaneous measurement and evaluation of novel  
7 potential halogenated transient tracers HCFCs, HFCs and PFCs in the ocean. *Ocean Sci.*  
8 *Discuss.* 2019, 1-34, 2019
- 9 P. Malanotte-Rizzoli, V. Artale, G. L. Borzelli-Eusebi, S. Brenner, G. Civitarese, A. Crise,  
10 J. Font, M. Gacic, N. Kress, S. Marullo, E. Ozsoy, M. Ribera d'Alcalà, W. Roether,  
11 K. Schroeder, S. Sofianos, T. Tanhua, A. Theocharis, M. Alvarez, Y. Ashkenazy,  
12 A. Bergamasco, V. Cardin, S. Carniel, F. D'Ortenzio, E. Garcia-Ladona, J. M. Garcia-  
13 Lafuente, A. Gogou, M. Gregoire, D. Hainbucher, H. Kontoyannis, V. Kovacevic,  
14 E. Krasakapoulou, G. Krokos, A. Incarbona, M. G. Mazzocchi, M. Orlic, A. Pascual, P.-  
15 M. Poulain, A. Rubino, J. Siokou-Frangou, E. Souvermezoglou, M. Sprovieri, I. Taupier-  
16 Letage, J. Tintoré, and G. Triantafyllou, 2013: Physical forcing and physical/biochemical  
17 variability of the Mediterranean Sea: a review of unresolved issues and directions for future  
18 research. *Ocean Science* Vol.10, Issue 3, 281-322, 2014
- 19 Margolin, A.R., Gonnelli, M., Hansel, D.A. and Santinelli, C.: Black Sea dissolved organic  
20 matter dynamics: Insights from optical analyses. *Limnol. Oceanogr.* 63, 3, 1425-1443, doi:  
21 10.1002/lno.10791, 2018.
- 22 Millot, C., Candela, J., Fuda, J.-L. and Tber, Y.; Large warming and salinification of the  
23 Mediterranean outflow due to changes in its composition. *Deep Sea Res. I: Oceanogr. Res.*  
24 *Pap.*, V. 53, Iss. 4, 656-666, 2006.
- 25 Mintrop, L., Pérez, F. F., González-Dávila, M., Körtzinger, A. and Santana-Casiano, J.M.:  
26 Alkalinity determination by potentiometry- intercalibration using three different methods.  
27 *Ciencias Marinas* , 26 , 1, 23-37., 2000.
- 28 Nelson, N.B. and Siegel, D.A.: The global distribution and dynamics of chromophoric dissolved  
29 organic matter. *Annu. Rev. Mar. Sci.*, 5, 447-476, 2013.
- 30 Patsavas, M.C., Byrne, R.B., Yang, B., Easley, R.A., Wanninkhof, R., Liu, X.: Procedures for  
31 direct spectrophotometric determination of carbonate ion concentrations: Measurements in US  
32 Gulf of Mexico and East Coast waters. *Mar. Chem.*, 168, 80-85, 2015.



- 1 Pérez, F.F. and Fraga F.: A precise and rapid analytical procedure for alkalinity determination.
- 2 *Marine Chemistry*, 21, 169-182, 1987.
- 3 Pérez, F.F., Ríos, A.F., Rellán, T. and Álvarez, M.: Improvements in a fast potentiometric
- 4 seawater alkalinity determination. *Ciencias Marinas*, 26, 463-478, 2000.
- 5 Retelletti Brogi, S. Gonelli, M., Vestri, S. and Santinelli, C.: Biophysical processes affecting
- 6 DOM dynamics at the Arno river mouth (Tyrrhenian Sea). *Biophysical Chemistry*, 197, 1-9,
- 7 2015.
- 8 Roether, W., Klein, B., and Hainbucher, D.: The Eastern Mediterranean Transient: Evidence
- 9 for Similar Events Previously? in: *The Mediterranean Sea: Temporal Variability and Spatial*
- 10 *Patterns*, edited by: Borzelli, G. L. E., AGU monographs, 2013
- 11 Santinelli, C., Follet C., Retelletti Brogi, S., Xu, L., Repeta, D.: Carbon isotope measurements
- 12 reveal unexpected cycling of dissolved organic matter in the deep Mediterranean Sea. *Mar*
- 13 *Chem.*, 177, 267-277, 2015.
- 14 Santinelli, C.: DOC in the Mediterranean Sea. In: *Biogeochemistry of Marine Dissolved*
- 15 *Organic Matter 2<sup>nd</sup> Ed.*, 579-608, 2015.
- 16 Santinelli, C., Hansell, D. A. and Ribera d'Alcala, M.: Influence of stratification on marine
- 17 dissolved organic carbon (DOC) dynamics: The Mediterranean Sea case. *Prog. Oceanogr.*, 119,
- 18 68-77, 2013.
- 19 Santinelli, C. Sempéré, R., Van Wambeke, F., Charrière B. and Seritti, A.: Organic carbon
- 20 dynamics in the Mediterranean Sea: An integrated study. *Global Biogeochemical Cycles*, 26,
- 21 4, GB4004, DOI: [10.1029/2011GB004151](https://doi.org/10.1029/2011GB004151), 2012.
- 22 Santinelli, C., Nannicini, L. and Seritti, A. :DOC dynamics in the meso and bathypelagic layers
- 23 of the Mediterranean Sea. *Deep-Sea Res.* **II**, 57, 1446-1459, 2010.
- 24 Schneider, A., Tanhua, T., Körtzinger, A., and Wallace, D. W. R.: High anthropogenic carbon
- 25 content in the eastern Mediterranean, *J. Geophys. Res.*, 115, C12050,
- 26 doi:10.1029/2010JC006171, 2010.
- 27 Schroeder, K., Gasparini, G. P., Tangherlini, M. and Astraldi, M.: Deep and intermediate water
- 28 in the western Mediterranean under the influence of the Eastern Mediterranean Transient.
- 29 *Geophys. Res. Lett.*, doi: 10.1029/2006GL027121, 2006.
- 30 Schroeder, K., Ribotti, A., Borghini, M., Sorgente, R., Perilli, A., and Gasparini, G. P.: An
- 31 extensive western Mediterranean deep water renewal between 2004 and 2006, *Geophys. Res.*



- 1 Letters, 35, 2008.
- 2 Stedmon, C.A. and Nelson, N.B.: The optical properties of DOM in the Ocean. In:  
3 Biogeochemistry of Marine Dissolved Organic Matter 2<sup>nd</sup> Ed., 481-508, 2015.
- 4 Sigman, D.M., K.L. Casciotti, M. Andreani, C. Barford, M. Galanter, and J.K. Böhlke: A  
5 bacterial method for the nitrogen isotopic analysis of nitrate in seawater and freshwater. Anal.  
6 Chem. 73: 4145-4153, 2001.
- 7 Tanhua, T., Hainbucher, D., Schroeder, K., Cardin, V., Álvarez, M., and Civitarese, G.: The  
8 Mediterranean Sea system: a review and an introduction to the special issue, Ocean Sci., 9, 789-  
9 803, doi:10.5194/os-9-789-2013.
- 10 Turnherr, A.M.: How To Process LADCP Data With the LDEO Software (Versions IX.7 –  
11 IX.10). The “go-ship” manual for LADCP data acquisition ftp:  
12 [//ftp.ldeo.columbia.edu/pub/LADCP/UserManuals](ftp://ftp.ldeo.columbia.edu/pub/LADCP/UserManuals), 2014.
- 13 Ullmann, D. S. and Hebert, D.: Processing of Underway CTD data. AMS  
14 <https://doi.org/10.1175/JTECH-D-13-00200.1>, 2014.
- 15 Weigand, M.A., J. Foriel, B. Barnett, S. Oleynik, and Sigman, D. M.: Updates to  
16 instrumentation and protocols for isotopic analysis of nitrate by the denitrifier method. Rapid  
17 Commun. Mass Spectrom. 30: 1365-1383, 2016.