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Supporting Information

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Enabling High-Performance NASICON-Based Solid-State Lithium Metal Batteries Towards Practical Conditions

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Figure S1: *a*) schematic representing the preparation of cathode and electrolyte and b) chemistry of polycarbonate decomposition.



Figure S2: AFM image of LAGP pellet surface.



Figure R3: SEM image of LAGP/ionic-liquid interlayer (scare bar 5 microns).







Figure S5: Cell performance of Li/LiTFSI-PEO interlayer/LAGP/LFP SSLMB at 80 °C.



Figures S6: cycling performance of the Li/ionic-liquid interlayer/LAGP/LFP cell at 60° C.



Figures S7: cycling performance of the Li/ionic-liquid interlayer/LAGP/LFP cell at 25° C.



Figure S8: Rietveld refinement of HEXRD patterns of LAGP/LFP mixture (a) before and (b) after heating.



Figure S9: Cross section SEM image of the Li/ionic-liquid interlayer/LAGP/LFP cellafter 50 cycles, top panelshows Li-interlayer interface, while bottom panel showsLFP-LAGPinterfaceaftercycling.





Figure S10: HEXRD and Rietveld refinement of LFP-LAGP-graphite pellet after 50 cycles.



Figure S11: Cycle performance of Li/ioniq-liquid interlayer/LAGP/LFP SSLMB with a weight ratio LFP:LAGP:Graphite of 55:35:10.